

Supporting Information

Cascade fractionation of poplar into xylose, glucan oligomers and less-condensed lignin via synergistic formic acid–LiBr molten salt hydrate pretreatment

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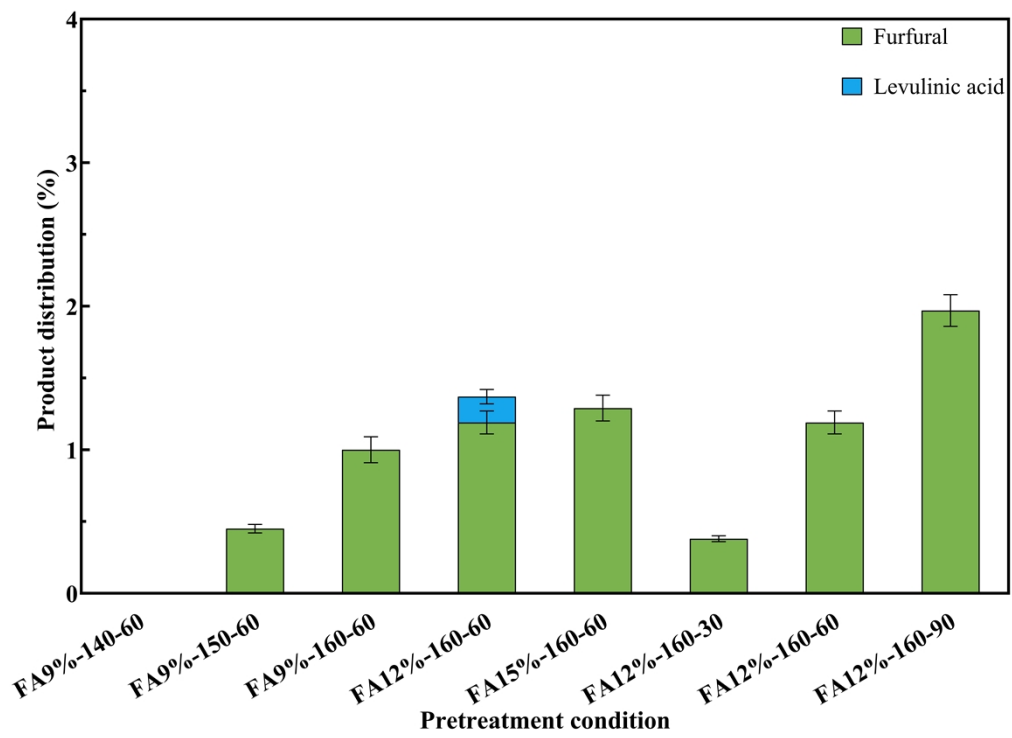


Fig. S1 The content of furfural and levulinic acid in the hydrolysate during FA pretreatment

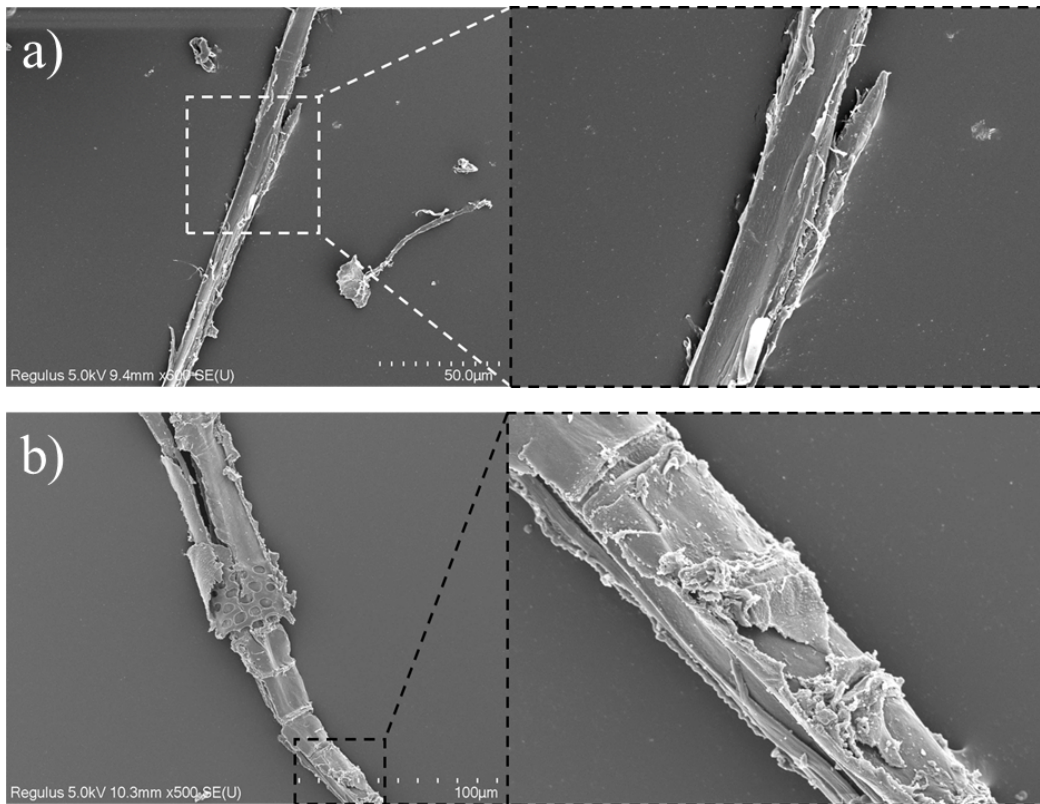


Fig. S2 Morphological characteristics of the control poplar (a) and FA-pretreated substrate (b)

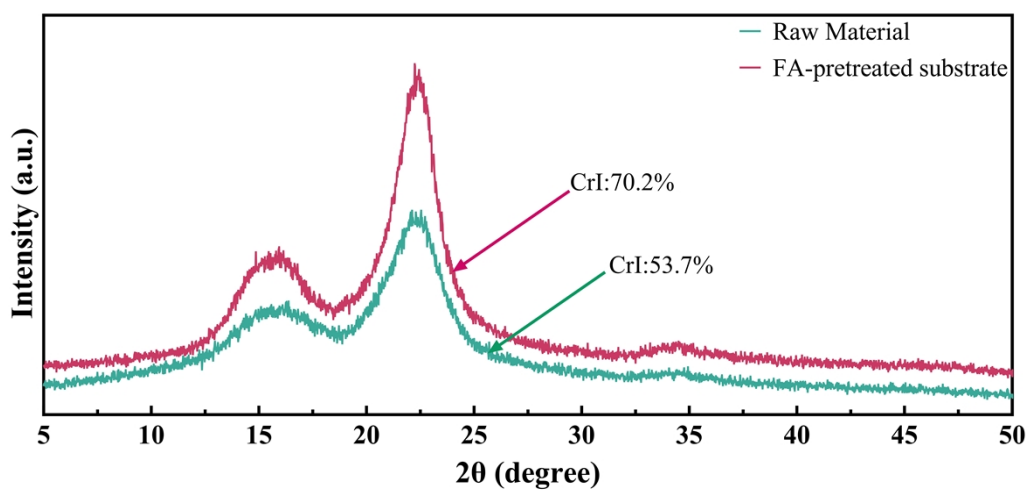


Fig. S3 The XRD pattern of raw material and FA-pretreated substrate

Table S1 Comparison of results of different treatment

	Ref ¹	Ref ²	This Work
Raw material	Corn stover	Cellulose	Poplar
Pretreatment	Auto-hydrolysis	/	12 wt% FA
Reaction conditions	190 °C, 20 min	/	160 °C, 1h
Hemicellulose removal rate	94.4%	/	91.24%
MSH	60 wt% LiBr	60 wt% LiBr	60 wt% LiBr
Reaction conditions	130 °C, 2h	130 °C, 5h	100°C, 0.5h, 30mM HCl
Glucan oligomers yield	59.9%	90.4%	42.68%
Glucose yield	21.4%		41.87%

Table S2 The purity and yield of lignin samples

Lignin samples (L _{a/b-c})	Yield of acid-insoluble lignin (%)	Yield of acid-soluble lignin (%)	Purity (%)
L _C /40 mM-110°C	81.35	6.18	87.53
L _{FA} /20 mM-90°C	40.77	3.32	44.09
L _{FA} /20 mM-100°C	73.22	3.57	76.79
L _{FA} /20 mM-110°C	88.80	3.32	92.12
L _{FA} /30 mM-90°C	62.84	3.66	66.50
L _{FA} /30 mM-100°C	91.73	3.84	95.57
L _{FA} /30 mM-110°C	91.86	3.44	95.30
L _{FA} /40 mM-90°C	69.09	3.58	72.67
L _{FA} /40 mM-100°C	92.68	3.70	96.38
L _{FA} /40 mM-110°C	94.04	3.34	97.38

L_{a/b-c}: a-Conventional or Formic Acid, b-HCl concentration, c-Temperature

Table S3 2D HSQC NMR data for the signal assignments in lignin

Lable	$\delta_C/\delta_H(\text{ppm})$	Assignment
A $_{\alpha}$	71.58/4.86	C $_{\alpha}$ -H $_{\alpha}$ in β -O-4 (A)
OMe	56.63/3.72	C-H in Methoxy
A $_{\gamma}$	59.46/3.23-3.64	C $_{\gamma}$ -H $_{\gamma}$ in β -O-4 (A)
I $_{\gamma}$	61.42/4.09	C $_{\gamma}$ -H $_{\gamma}$ in <i>p</i> -hydroxycinnamyl alcohol end-groups
C $_{\alpha}$	86.48-87.61/5.44-5.59	C $_{\alpha}$ -H $_{\alpha}$ in Phenylcoumarin substructure
C $_{\gamma}$	62.49/3.70	C $_{\gamma}$ -H $_{\gamma}$ in Phenylcoumarin substructure
B $_{\alpha}$	83.19-84.87/4.81-4.65	C $_{\alpha}$ -H $_{\alpha}$ in β - β resinol substructures (B)
B $_{\beta}$	53.25/3.03	C $_{\beta}$ -H $_{\beta}$ in β - β resinol substructures (B)
B $_{\gamma}$	70.18-71.10/3.82-4.19	C $_{\gamma}$ -H $_{\gamma}$ in β - β resinol substructures (B)
X $_2$	69.68/3.51	C $_2$ -H $_2$ in β -D-xylopyranoside (X)
X $_3$	72.27/3.21-3.43	C $_3$ -H $_3$ in β -D-xylopyranoside (X)
X $_4$	74.17/3.22-3.50	C $_4$ -H $_4$ in β -D-xylopyranoside (X)
A $_{\beta}$ (G/H)	83.47/4.26-4.45	C $_{\beta}$ -H $_{\beta}$ in β -O-4 linked to G/H (A)
A $_{\beta}$ (S)	85.72-87.11/3.96-4.08	C $_{\beta}$ -H $_{\beta}$ in β -O-4 linked to S (A)
BD $_{\alpha}$	75.48/4.89	C $_{\alpha}$ -H $_{\alpha}$ in benzodioxane substructures
HK $_{\alpha}$	44.34/3.65	C $_{\alpha}$ -H $_{\alpha}$ in Hibbert ketone
HK $_{\gamma}$	66.99/4.15	C $_{\gamma}$ -H $_{\gamma}$ in Hibbert ketone
S $_{2,6}$	103.42/6.70	C $_{2,6}$ -H $_{2,6}$ in syringyl units (S)
S' $_{2,6}$	106.41/7.30	C $_{2,6}$ -H $_{2,6}$ in oxidized S units (S')
S'' $_{2,6}$	106.68/6.49	C $_{2,6}$ -H $_{2,6}$ in condensed S units (S'')
G $_2$	110.47/6.96	C $_2$ -H $_2$ in guaiacyl units (G)
G $_5$	114.69/6.78	C $_5$ -H $_5$ in guaiacyl units (G)
G $_6$	118.74/6.78	C $_6$ -H $_6$ in guaiacyl units (G)
PB $_{2,6}$	131.17/7.50-7.87	C $_{2,6}$ -H $_{2,6}$ in benzoate
I $_{\alpha}$	128.18/6.22	C $_{\alpha}$ -H $_{\alpha}$ in acylated cinnamyl alcohol
I $_{\beta}$	128.39/6.45	C $_{\beta}$ -H $_{\beta}$ in acylated cinnamyl alcohol
I' $_{\gamma}$	125.95/6.77	C $_{\gamma}$ -H $_{\gamma}$ in acylated <i>p</i> -hydroxycinnamyl alcohol end-groups

References

1. Q. Liu, L. Zhou, X. Xie, D. Fan, X. Ouyang, W. Fan and X. Qiu, *Green Chem.*, 2022, **24**, 8812-8819.
2. Q. Liu, Q. Ma, S. Sabnis, W. Zheng, D. G. Vlachos, W. Fan, W. Li and L. Ma, *Green Chem.*, 2019, **21**, 5030-5038.