

Supporting information

| Mass | Oxide ions | Argide ions | Nitride ions | Nitrogen monoxide ions |
|-------------------|---|---|--|---|
| ¹⁸⁶ W | ¹⁷⁰ Er ¹⁶ O ⁺ (14.93%); ¹⁷⁰ Yb ¹⁶ O ⁺ (3.04%); | ¹⁴⁶ Nd ⁴⁰ Ar ⁺ (17.2%) | ¹⁷² Yb ¹⁴ N ⁺ (21.83%) | ¹⁵⁶ Gd ¹⁴ N ¹⁶ O ⁺ (20.47%) ¹⁵⁶ Dy ¹⁴ N ¹⁶ O ⁺ (0.06%); |
| | ¹⁶⁸ Er ¹⁶ O ⁺ (26.78%); ¹⁶⁸ Yb ¹⁶ O ⁺ (0.13%); | ¹⁴⁴ Nd ⁴⁰ Ar ⁺ (23.8%); ¹⁴⁴ Sm ⁴⁰ Ar ⁺ (3.07%) | ¹⁷⁰ Er ¹⁴ N ⁺ (14.93%); ¹⁷⁰ Yb ¹⁴ N ⁺ (3.04%) | ¹⁵⁴ Sm ¹⁴ N ¹⁶ O ⁺ (22.75%); ¹⁵⁴ Gd ¹⁴ N ¹⁶ O ⁺ (2.18%), |
| ²⁰³ Tl | ¹⁸⁷ Os ¹⁶ O ⁺ (13.24%); ¹⁸⁷ Re ¹⁶ O ⁺ (62.60%) | ¹⁶³ Dy ⁴⁰ Ar ⁺ (24.9%) | ¹⁸⁹ Os ¹⁴ N ⁺ (16.15%) | ¹⁷³ Yb ¹⁴ N ¹⁶ O ⁺ (16.13%) |
| | ¹⁸⁹ Os ¹⁶ O ⁺ (16.15%) | ¹⁶⁵ Ho ⁴⁰ Ar ⁺ (100%) | ¹⁹¹ Ir ¹⁴ N ⁺ (37.3%) | ¹⁷⁴ Yb ¹⁴ N ¹⁶ O ⁺ (31.83%); ¹⁷⁴ Hf ¹⁴ N ¹⁶ O ⁺ (0.16%) |

Table S 1. The polyatomic interference during Tl isotope analysis. The relative abundances of isotopes are from Rosman and Taylor (1998) and shown as a percentage.

1. K. J. R. Rosman and P. D. P. Taylor, *Pure and Applied Chemistry*, 1998, **70**, 217-235.