Adsorptive and photo-Fenton properties of bimetallic MIL-100(Fe,Sn) and MIL-100(Fe,Ir) MOFs toward removal of tetracycline from aqueous solutions

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Table S1. Dark adsorption and photo-Fenton degradation efficiency of the synthesized MOFs

Synthesized MOFs	Dark adsorption	Photo-Fenton degradation
	TC removal (%)	TC removal (%)
MIL-100(Fe)	13.2 ± 2.6	57.3 ± 6.3
MIL-100(Fe,Sn-1.8)	12.3 ± 3.0	27.9 ± 4.8
MIL-100(Fe,Sn-3.5)	16.3 ± 1.7	38.3 ± 7.7
MIL-100(Fe,Sn-7.1)	18.0 ± 2.0	58.0 ± 5.6
MIL-100(Fe,Sn-14.2)	16.6 ± 0.9	51.2 ± 9.6
MIL-100(Fe,Ir-26.7)	32.1 ± 3.1	55.9 ± 1.2
MIL-100(Fe,Ir-50.2)	43.9 ± 0.7	44.8 ± 1.0
MIL-100(Fe,Ir-62.3)	47.6 ± 2.6	43.2 ± 0.3
MIL-100(Fe,Ir-100.5)	52.2 ± 3.5	38.5 ± 2.7

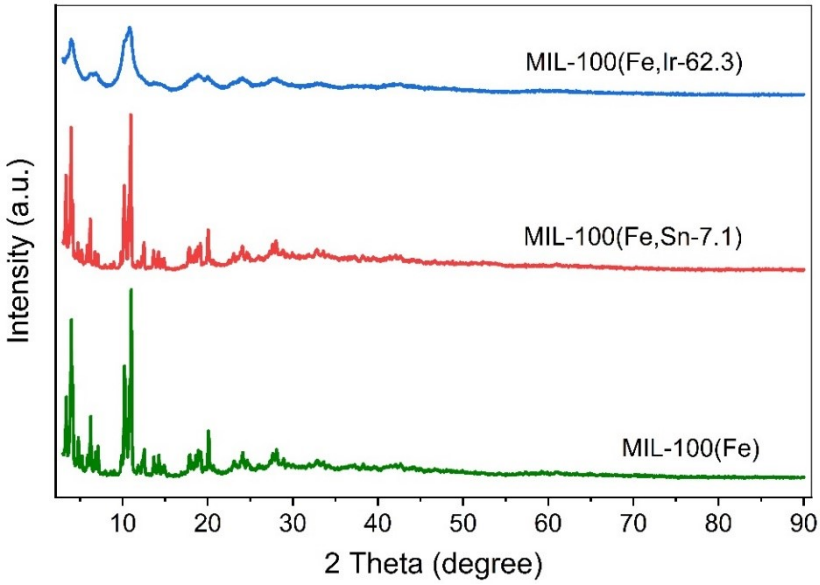

Fig. S1. XRD spectra of MIL-100(Fe), MIL-100(Fe,Sn-7.1), and MIL-100(Fe,Ir-62.3).

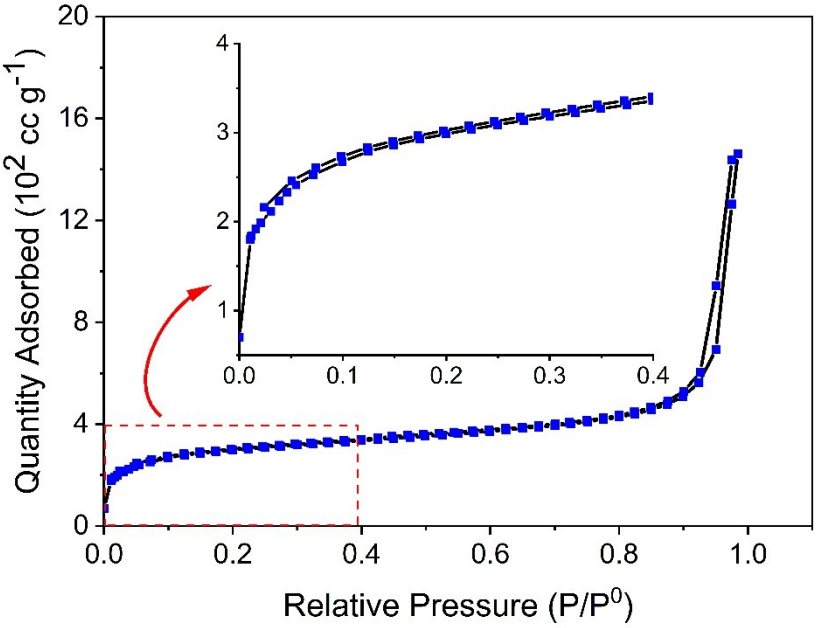

Fig. S2. Nitrogen adsorption-desorption isotherms of MIL-100(Fe,Ir-62.3).

Table S2. Doping metal content of the initial reactants and synthesized MOFs

MOFs	Dopant	$mol_{dopant}/(mol_{dopant} + mol_{Fe})$	
		Initial reactants	Synthesized MOFs
MIL-100(Fe,Sn-7.1)	Sn^{2+}	9.9%	3.6%
MIL-100(Fe,Ir-62.3)	Ir^{3+}	49%	1.6%