

Protein nanofibrils/polysaccharides aerogel beads as porous functional bio-adsorbent for water treatment

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List of abbreviation

βLG	β-lactoglobulin
CA	Citric Acid
CNFs	Cellulose nanofibers
CR	Congo red
Cs	Chitosan
CPCs	CNFs/Cs/PNFs aerogel beads
PNFs	Protein nanofibrils
PNFs-C	Cross-linked protein fibrils
ThT	thioflavin T

Assessment of protein nanofibrils formation

thioflavin T (ThT) fluorescence experiments were applied to explore the formation of the protein nanofibrils for purified (PNFs) and cross-linked protein fibrils (PNF-C) ^{1,2}. The sharp fluorescence intensity of ThT for all samples suggested that β -lactoglobulin (β LG) was actively involved in the fibril formation (Fig. S1A). In another test, equal amounts of purified PNFs and cross-linked PNFs solutions were drop-cast into a silicon mold. The solutions contained two different concentrations of citric acid (CA) (1 and 1.5 wt. %). The PNFs solution was evenly distributed on the substrate and then dried in an oven at 65°C for 10 h. The test result interestingly revealed (Fig. S1B-1) the brittleness of the purified PNFs film and its failure to form a uniform film, demonstrating the low mechanical property of PNFs ³. In contrast, the protein solutions incorporated with CA resulted in free-standing, homogeneous, flexible films (Fig. S1B-2). These valuable findings demonstrated the efficient integration of CA into the PNFs chains. However, increasing the CA concentration to 1.5 wt. % reduced the flexibility of the films (Fig. S1B-3). Consequently, PNF-C1 was selected as the highly effective cross-linked PNFs solution.

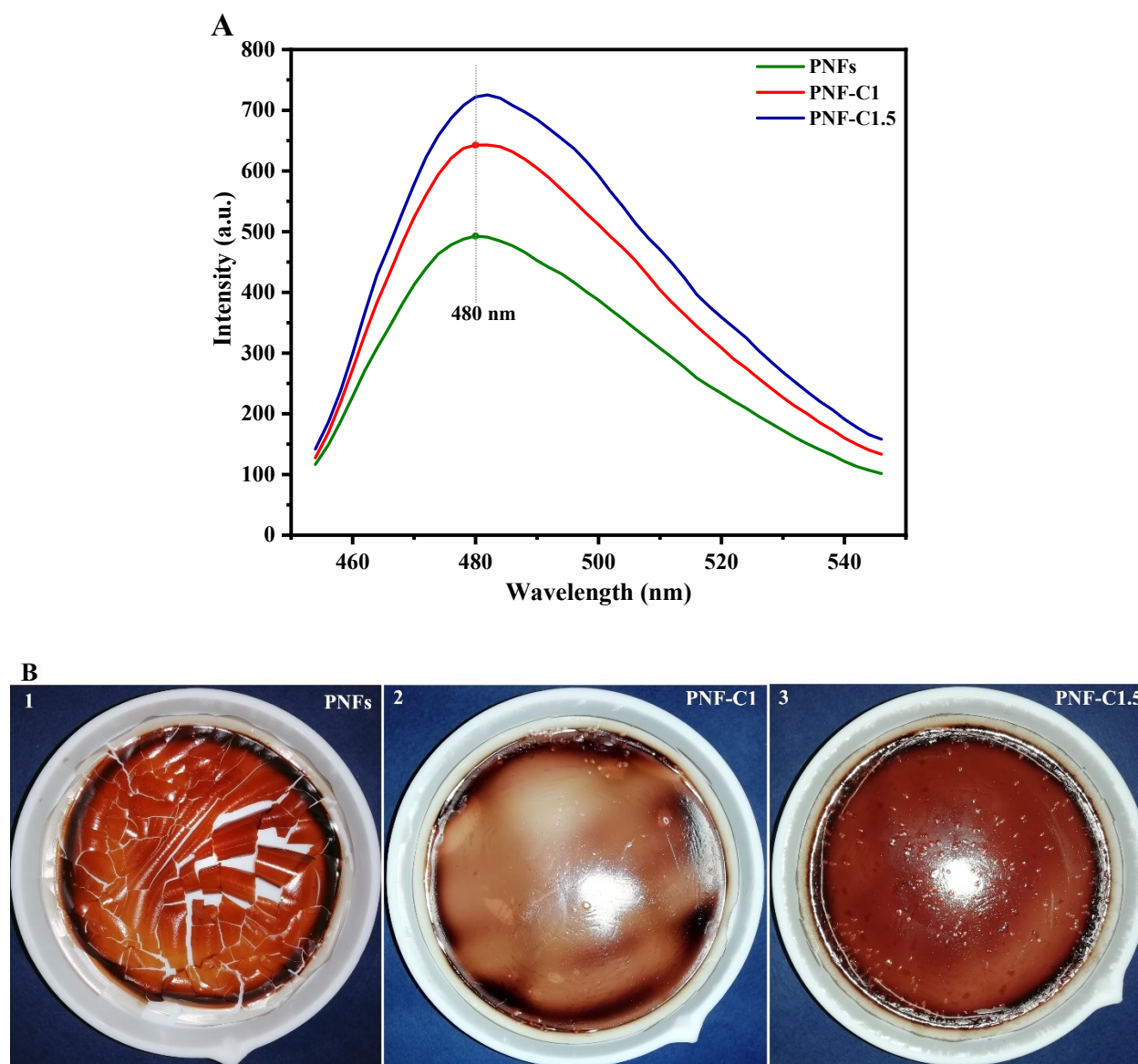


Fig. S1 (A) ThT fluorescence spectra of PNFs in the absence (PNFs, represented by the green line) and presence of CA with different concentrations (PNFs-C, represented by the red and blue lines). (B) Digital images of the PNFs films were captured before and after the addition of CA. (1) PNFs, (2) PNFs with 1 wt. % (PNFs-C1), and (3) 1.5 wt. % (PNFs-C1.5) of CA.

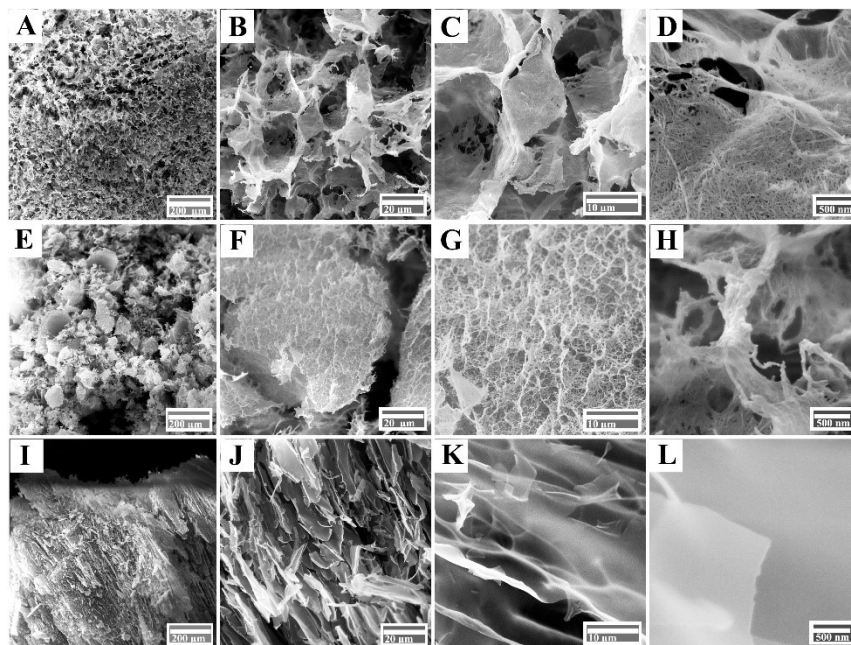


Fig. S2 Surface morphology of the freeze-dried biomaterials: FE-SEM images of (A) Cellulose nanofibers (CNFs) aerogel, (B-D) High-resolution images of CNFs aerogel illustrating the porous structure of the aerogel. E) Neat chitosan (Cs) aerogel, (F-H) High-resolution images of Cs aerogel demonstrating the integration of fibrils in the formation of the interconnected network. (I) PNFs aerogel, (J-L) High-resolution images of PNFs revealing compact layered sheets

Table S1 Characteristics of Congo red (CR) ⁴

CA Index name	1-Naphthalenesulfonic acid, 3,3'-[[1,1'-biphenyl]-4,4'-diylbis(2,1-diazenediyl)]bis[4-amino-,sodium salt (1:2)]
CAS No.	573-58-0
Molecular formula	C ₃₂ H ₂₂ N ₆ Na ₂ O ₆ S ₂
Molecular weight	696.66 g mol ⁻¹
Chemical Structure	
Molecular surface area	557.6 cm ²
Density	0.995 g cm ⁻³ at 25 °C

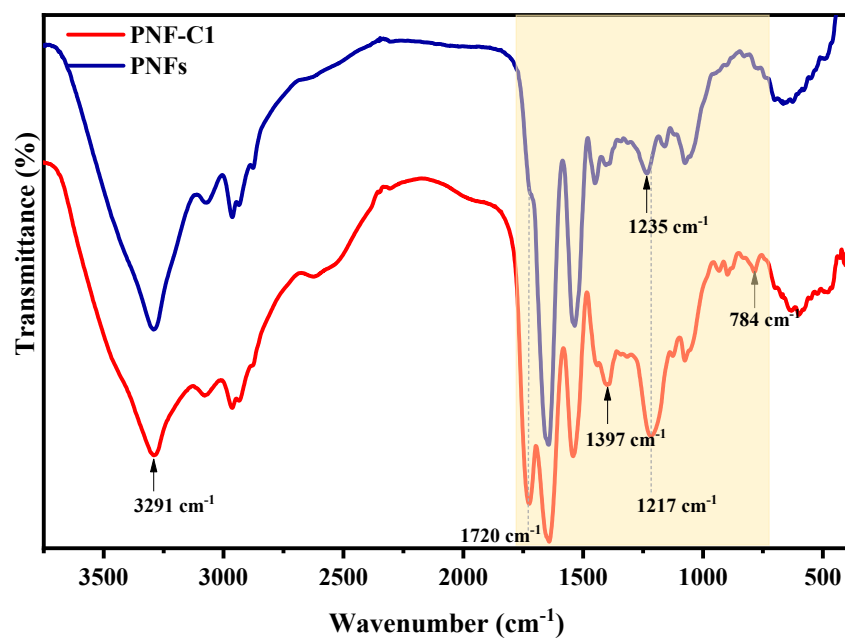


Fig. S3 FTIR spectra of PNFs before (PNFs, blue line) and after crosslinking with CA (PNF-C1, red line) shown in full range 3800-500 cm^{-1} .

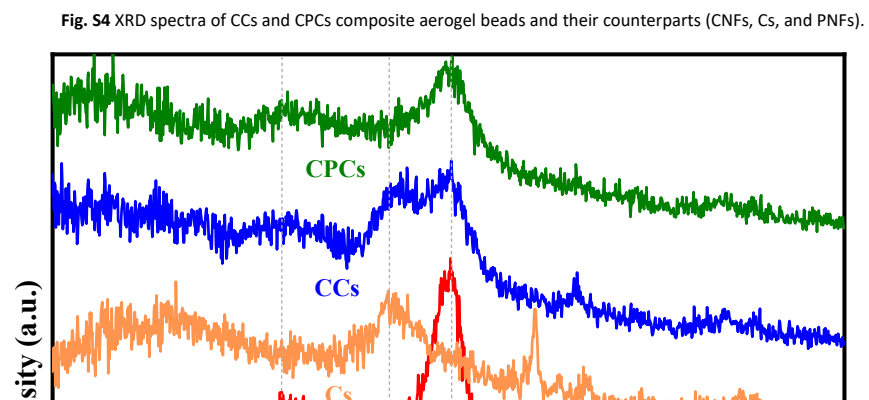


Fig. S4 XRD spectra of CCs and CPCs composite aerogel beads and their counterparts (CNFs, Cs, and PNFs).

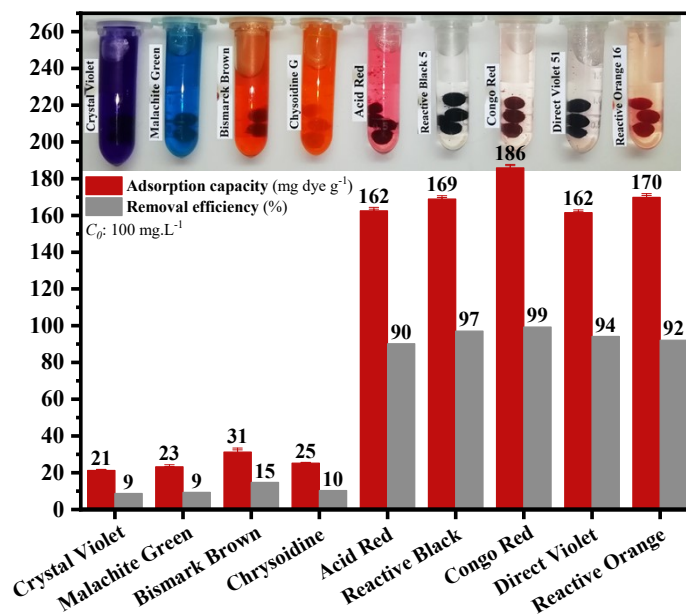
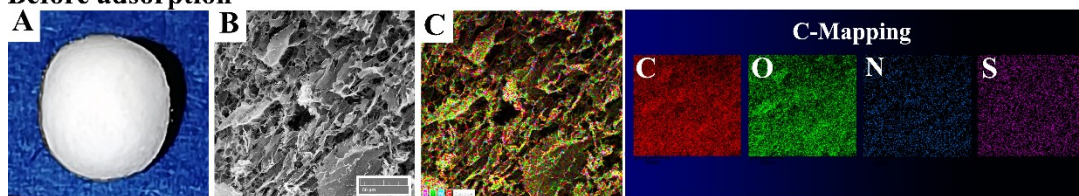


Fig. S5 The adsorption capacity and removal efficiency of CNFs/Cs/PNFs aerogel beads (CPCs) for removing various cationic and anionic dyes (The inset images representing remained dye solutions at adsorption equilibrium time).

Before adsorption



After adsorption

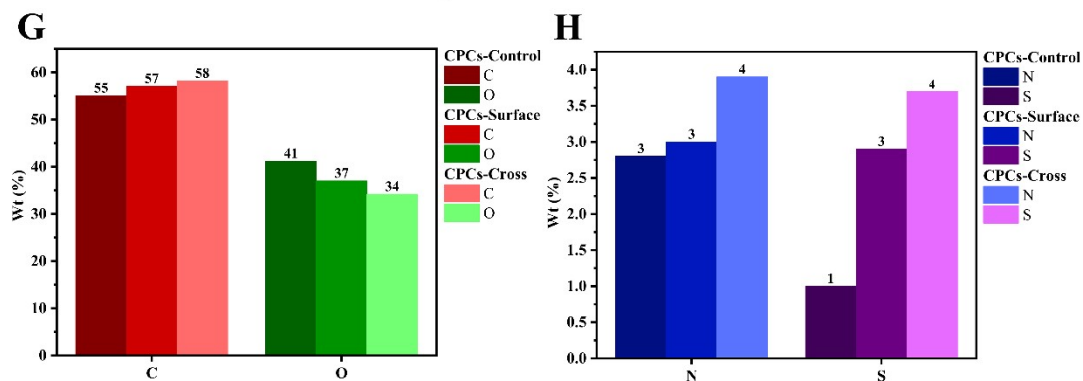
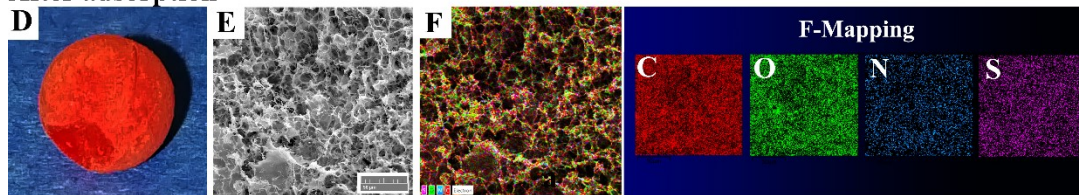


Fig. S6 Digital and FE-SEM images with corresponding EDX spectra and elemental mapping of C, O, N, and S for CPCs aerogel. (A-C) Before, and (D-F) After CR dye adsorption. Calculated values from the elemental maps of CPCs aerogel: (G) C and O, (H) N, and S.

Table S2 Comparison of maximum adsorption capacity of various polysaccharide and PNFs-based aerogels for CR.

Adsorbant	pH _{zpc}	Q _{max} (mg g ⁻¹)	Isotherm	Year	Ref.
Polysaccharide-based materials					
chitosan hydrobeads	6.4	93.71	Langmuir	2007	5
chitosan/montmorillonite	7	81.23	Langmuir	2007	6
chitosan/multiwalled carbon nanotubes	5	450.4	Langmuir	2010	7
cellulose/Fe ₃ O ₄ /activated carbon composites	natural pH	66.09	Langmuir	2011	8
cellulose/ chitosan hydrogel beads	7	40	Langmuir	2015	9
Tea waste	10	32.26	Langmuir	2015	10
Ni/Al mixed oxide microspheres	6	1229.59	SIP	2015	11
CaCO ₃ -loaded cellulose aerogel	natural pH	75.81	-	2015	12
Chitosan Beads	6	166.67	Langmuir	2016	13
diammonium tartrate modified chitosan	8	1597	SIP	2017	14
Modified chitosan hydrogel beads	5	1994	SIP	2017	15
Cellulose/chitosan aerogel	7	381.7	Langmuir	2018	16
graphene–chitosan composite hydrogel	7	384.62	Langmuir	2018	17
dialdehyde micro fibrillated cellulose/ chitosan composite film	5.5	152.5	Langmuir	2018	18
chitosan-Fe(OH) ₃ beads	8.39	445.32	Langmuir	2018	19
cellulose-chitosan foams	7.4	1170.2	Langmuir	2018	20
DAC-crosslinked cellulose-chitosan foam	7.4	1548.2	SIP	2019	21
Cationized rice husk cellulose	7.27	580.09	Langmuir	2019	22
pine bark	3.4	2.51	Freundlich	2019	23
graphene oxide/waste-newspaper cellulose aerogel	unadjusted pH	140.4	Langmuir	2020	24
Cellulose Nanofibril/Carbon Nanomaterial aerogel	natural pH	585.3	Langmuir	2020	25
chitosan/cellulose hydrogel	3	1480	Langmuir	2022	26
κ-carrageenan/polyacrylamide double network aerogel containing graphene oxide (GO)	7	20.718	Langmuir	2022	27
carboxyl cellulose nanofibers/montmorillonite/polyethyleneimine aerogel	unadjusted pH	3114	Langmuir	2022	28
waste bamboo paper and chitosan biohybrid aerogel	6.28	559.6	SIP	2022	29
magnetic hydroxyethyl cellulose/Fe ₃ O ₄	7	308.64	Langmuir	2023	30
abundant sodium alginate (SA)/gellan gum aerogel	natural pH	576.5	Langmuir	2023	31
Extracted cellulose/chitosan aerogel	Origin pH	255.1	Langmuir	2023	32
Chitosan–Quinoa Bran Aerogel	5	182.48	Langmuir	2023	33
Chitosan/UiO-67 hydrogel	7.2	1001.2	Langmuir	2023	34
micro/nano MIL-88A (Fe, Al, Fe-Al)/ chitosan composite sponge	7.1-8.3	607-1312	SIP	2023	35
Protein nanofibrils and Protein Isolates					

powdered eggshell	8.8	95.25	Langmuir	2013	³⁶
Industrial waste eggshell	7.09	49.5	Freundlich	2017	³⁷
regenerated silk materials	not mentioned	above 96 percent rejection	not mentioned	2017	³⁸
silk nanofibril (SNF)/ hydroxyapatite (HAP)		Flux rejection	not mentioned	2017	³⁹
modified eggshell	4.5	117.65	Langmuir	2019	⁴⁰
gluten materials	<6	211.1	Langmuir	2019	⁴¹
Protein Amyloid Fibrils aerogel	5.1	97.5	not mentioned	2022	⁴²
Cellulose nanofibers/Chitosan/Protein nanofibrils aerogel beads	7.4	1349.8	SIP	2023	This work

References:

- 1 K. Gade Malmos, L. M. Blancas-Mejia, B. Weber, J. Buchner, M. Ramirez-Alvarado, H. Naiki and D. Otzen, *Amyloid Int. J. Exp. Clin. Investig. Off. J. Int. Soc. Amyloidosis*, 2017, **24**, 1–16.
- 2 B. Ma, X. You and F. Lu, *Int. J. Biol. Macromol.*, 2014, **64**, 162–167.
- 3 M. Peydayesh, M. Bagnani and R. Mezzenga, *ACS Sustain. Chem. Eng.*, 2021, **9**, 11916–11926.
- 4 S. Chatterjee, S. Chatterjee, B. P. Chatterjee and A. K. Guha, *Colloids Surfaces A Physicochem. Eng. Asp.*, 2007, **299**, 146–152.
- 5 S. Chatterjee, S. Chatterjee, B. P. Chatterjee and A. K. Guha, *Colloids Surfaces A Physicochem. Eng. Asp.*, 2007, **299**, 146–152.
- 6 L. Wang and A. Wang, *J. Hazard. Mater.*, 2007, **147**, 979–985.
- 7 S. Chatterjee, M. W. Lee and S. H. Woo, *Bioresour. Technol.*, 2010, **101**, 1800–1806.
- 8 H. Y. Zhu, Y. Q. Fu, R. Jiang, J. H. Jiang, L. Xiao, G. M. Zeng, S. L. Zhao and Y. Wang, *Chem. Eng. J.*, 2011, **173**, 494–502.
- 9 M. Li, Z. Wang and B. Li, *Desalin. Water Treat.*, 2016, **57**, 16970–16980.
- 10 M. Foroughi-Dahr, H. Abolghasemi, M. Esmaili, A. Shojamoradi and H. Fatoorehchi, *Chem. Eng. Commun.*, 2015, **202**, 181–193.
- 11 W. Huang, X. Yu and D. Li, *RSC Adv.*, 2015, **5**, 84937–84946.

- 12 K. Y. Chong, C. H. Chia, S. Zakaria, M. S. Sajab, S. W. Chook and P. S. Khiew, *Cellulose*, 2015, **22**, 2683–2691.
- 13 N. P. Raval, P. U. Shah, D. G. Ladha, P. M. Wadhvani and N. K. Shah, *Desalin. Water Treat.*, 2016, **57**, 9247–9262.
- 14 A. Zahir, Z. Aslam, M. S. Kamal, W. Ahmad, A. Abbas and R. A. Shawabkeh, *J. Mol. Liq.*, 2017, **244**, 211–218.
- 15 C. Lin, S. Li, M. Chen and R. Jiang, *J. Dispers. Sci. Technol.*, 2017, **38**, 46–57.
- 16 Y. Wang, H. Wang, H. Peng, Z. Wang, J. Wu and Z. Liu, *Fibers Polym.*, 2018, **19**, 340–349.
- 17 S. Omid and A. Kakanejadifard, *RSC Adv.*, 2018, **8**, 12179–12189.
- 18 X. Zheng, X. Li, J. Li, L. Wang, W. Jin, Y. Pei and K. Tang, *Int. J. Biol. Macromol.*, 2018, **107**, 283–289.
- 19 XinxinYang, Y. Li, H. Gao, C. Wang, X. Zhang and H. Zhou, *Int. J. Biol. Macromol.*, 2018, **117**, 30–41.
- 20 U.-J. Kim, D. Kim, J. You, J. W. Choi, S. Kimura and M. Wada, *Cellulose*, 2018, **25**, 2615–2628.
- 21 U. J. Kim, S. Kimura and M. Wada, *Carbohydr. Polym.*, 2019, **214**, 294–302.
- 22 Z. Jiang and D. Hu, *J. Mol. Liq.*, 2019, **276**, 105–114.
- 23 K. Litefti, M. S. Freire, M. Stitou and J. González-Álvarez, *Sci. Rep.*, 2019, **9**, 1–11.
- 24 C. Feng, P. Ren, Z. Li, W. Tan, H. Zhang, Y. Jin and F. Ren, *New J. Chem.*, 2020, **44**, 2256–2267.
- 25 Z. Yu, C. Hu, A. B. Dichiara, W. Jiang and J. Gu, *Nanomaterials*, 2020, **10**, 1–20.
- 26 D. Li, J. Zhang, L. P. Li, G. Cai, W. Zuo, W. Zhan, P. Wang and Y. Tian, *J. Clean. Prod.*, 2022, **371**, 133650.
- 27 S. Tarashi, H. Nazockdast, S. Shafaghsorkh and G. Sodeifian, *Sep. Purif. Technol.*, 2022, **287**, 120587.
- 28 K. Fan, T. Zhang, S. Xiao, H. He, J. Yang and Z. Qin, *Int. J. Biol. Macromol.*, 2022, **211**, 1–14.
- 29 C. Qiu, Q. Tang, X. Zhang, M. C. Li, X. Zhang, J. Xie, S. Zhang, Z. Su, J. Qi, H. Xiao, Y. Chen, Y. Jiang, C. F. de Hoop and X. Huang, *J. Clean. Prod.*, 2022, **338**, 130550.
- 30 Y. Hui, R. Liu, L. Li, Q. Sun, Z. Xiao, A. Xu and S. Liu, *J. Porous Mater.*, 2023, **30**, 1735–1751.
- 31 Z. Qin, K. Dong, Y. Zhang, Y. Jiang, L. Mo and S. Xiao, *Bioresour. Technol.*, 2023, **370**, 128576.

- 32 Y. Liu, Y. Ke, Q. Shang, X. Yang, D. Wang and G. Liao, *Chem. Eng. J.*, 2023, **451**, 138934.
- 33 M. Tan, H. Lv, Q. Zhao, B. Wang, S. Zheng and K. Li, *Environ. Eng. Sci.*, 2023, **40**, 233–243.
- 34 Z. Jing, Y. Li, Y. Zhang, M. Wang, B. Chen, Y. Sun, K. Chen, Q. Du, X. Pi, Y. Wang, S. Zhao and Y. Jin, *ChemistrySelect*, 2023, **8**, 202204367.
- 35 Y. Jin, Y. Li, Q. Du, B. Chen, K. Chen, Y. Zhang, M. Wang, Y. Sun, S. Zhao, Z. Jing, J. Wang, X. Pi and Y. Q. Wang, *Microporous Mesoporous Mater.*, 2023, **348**, 112404.
- 36 M. A. Zufikar and H. Setiyanto, *Int. J. ChemTech Res.*, 2013, **5**, 1532–1540.
- 37 M. A. Abdel-Khalek, M. K. Abdel Rahman and A. A. Francis, *J. Environ. Chem. Eng.*, 2017, **5**, 319–327.
- 38 L. Lv, X. Han, L. Zong, M. Li, J. You, X. Wu and C. Li, *ACS Nano*, 2017, **11**, 8178–8184.
- 39 S. Ling, Z. Qin, W. Huang, S. Cao, D. L. Kaplan and M. J. Buehler, *Sci. Adv.*, 2017, **3**, 1–12.
- 40 S. Parvin, B. K. Biswas, M. A. Rahman, M. H. Rahman, M. S. Anik and M. R. Uddin, *Chemosphere*, 2019, **236**, 124326.
- 41 X. Zhang, Y. Li, M. Li, H. Zheng, Q. Du, H. Li, Y. Wang, D. Wang, C. Wang, K. Sui, H. Li and Y. Xia, *J. Colloid Interface Sci.*, 2019, **556**, 249–257.
- 42 X. Jia, M. Peydayesh, Q. Huang and R. Mezzenga, *Small*, 2022, **18**, 2105502.