Supporting Information for

Development of small-molecule fluorescent probes targeting neutrophils via N-formyl

peptide receptors

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1. Synthetic procedures, characterisation details and spectral data

General details. Unless otherwise stated, all reagents were purchased from chemical suppliers and used without further purification, room temperature corresponds to ambient temperature and yields refer to spectroscopically and chromatographically pure compounds. Reactions were monitored by thin layer chromatography (TLC) performed on commercially available glass plates pre-coated with Merck silica gel 60 F254 or Merck silica gel 60 RP-18 F254s. Visualisation was performed by the quenching of UV fluorescence (λ_{max} = 254/365 nm) and by staining with iodine or potassium permanganate. All flash chromatography was performed using slurry packed Merck 9325 Keiselgel 60 or Aldrich C18-reverse phase silica gel. Melting points were obtained using a Gallenkamp Melting Point apparatus and are uncorrected. NMR spectra were recorded using an internal deuterium lock at ambient probe temperatures on a Bruker Avance (400 MHz) instrument. Chemical shifts (δ) are quoted in ppm, to the nearest 0.01 ppm (for ¹H NMR spectra), or 0.1 ppm (for ¹³C NMR spectra), and are referenced to the residual nondeuterated solvent peak. Coupling constants (J) are reported in Hertz (Hz) to the nearest 0.1 Hz. Data are reported as follows: chemical shift, multiplicity (br= broad; s= singlet; d= doublet; t= triplet; m= multiplet, or as a combination of these, e.g. dd, dt, etc.), integration, assignment and coupling constant(s). Assignments were determined either on the basis of unambiguous chemical shift or coupling pattern, and analogy to fully interpreted spectra for related compounds. Liquid chromatography-mass spectrometry (LC-MS) experiments were performed on a Waters Aquity UPLC I-CLASS coupled with Waters LCT Premier (operating in ES⁺ or ES⁻ mode). High resolution masses (HRMS) for accurate mass determination were performed on the same equipment, and samples were referenced against leucine enkaphalin or sulfadimethoxine. For analytical HPLC, a Waters BEH Acquity C18 (50mm x 2.1mm) column was used and the mobile phase was composed of solvent A (99.9% Water, 0.1% Formic Acid) and solvent B (99.9% Acetonitrile, 0.1% Formic Acid) used in a linear gradient (time= 0 min, 95%A and 5%B; time= 3.2 min, 5%A and 95%B; time= 3.5 min, 95%A and 5%B; total run time 4 min; for compound **17**, time= 0 min, 95%A and 5%B; time= 5 min, 95%A and 5%B; time= 30 min, 5%A and 95%B; time= 35 min, 95%A and 5%B; total run time 35 min). The sample solutions were prepared at a concentration of 0.1 mg/1 mL. The injection volume was 10µL, the flow rate was 0.5mL/min, the column temperature was 40 °C and UV detection was carried out at 5 fixed wavelengths within the range 210-550 nm. The values of retention time (t_R) are given in

3

minutes. Electron spray ionisation (ESI) conditions were as follow: 2kV (ES⁺) and 2.5kV (ES⁻) capillary voltage; 30 V (ES⁺) and 150 V (ES⁺) sample cone voltage; 2.1kV MCP Voltage; $350 \degree$ C desolvation temperature; $120 \degree$ C source temperature; 10 L/h cone gas flow (N₂); 400 L/h desolvation gas flow (N₂). Mass values are reported within the error limits of ± 5 ppm mass units. UV–vis spectra of the probes were recorded using a Perkin-Elmer Lambda 25 UV-vis spectrophotometer, and fluorescence emission spectra were obtained on a Varian Cary Eclipse fluorescence spectrofluorometer, using 0.1-0.5 mg/mL sample solutions and 1.0 cm path-length quartz cuvettes (1.0-3.0 mL) at 25°C. Spectra were not corrected for light intensity or detector sensitivity. Data were recorded on-line and analysed by Excel software. 6-Methyl-4,5-dihydropyridazin-3(2*H*)-one **1**, 4-(3-methoxybenzyl)-6-methylpyridazin-3(2*H*)-one **5** and amino-BODIPY **15** were prepared by adopting previously reported procedures [37,44].

4-(Anthracen-9-yl-methyl)-6-methylpyridazin-3(2H)-one, 2.

Compound **1** [37] (200.0 mg, 1.79 mmol, 1 eq.) and 9-anthracenecarboxaldehyde (405.0 mg, 1.96 mmol, 1.2 eq.) were sequentially added to a solution (7.0 mL) of KOH in absolute EtOH (5%, w/v). The reaction was refluxed under stirring for 2.5 h. After cooling, the mixture was concentrated *in vacuo*, diluted with ice-cold water (10.0 mL), and acidified with 6N HCl until pH= 3. The suspension was extracted with CH₂Cl₂ (3 x 25.0 mL), and the organic layer was dried over Na₂SO₄ and evaporated *in vacuo*. The resulting residue was purified by column flash chromatography using cyclohexane/ethyl acetate 3:1 as eluent, to obtain **2** as brownish solid. Yield = 71.1% (382.5 mg, 1.27 mmol); mp= 158-160 °C (EtOH). ¹H-NMR (CDCl₃) δ 10.50 (exch br s, 1H, NH), 8.50 (s, 1H, Ar), 8.07-8.09 (m, 2H, Ar), 8.01-8.03 (m, 2H, Ar), 7.50-7.52 (m, 4H, Ar), 6.13 (s, 1H, Ar), 4.89 (s, 2H, CH₂), 1.96 (s, 3H, CH₃). LC-MS (ESI): m/z calcd. for C₂₀H₁₆N₂O 301.1341 [M+H]⁺, found 301.1352.

2-[5-(Anthracen-9-yl-methyl)-3-methyl-6-oxopyridazin-1(6*H*)-yl]-*N*-(4bromophenyl)acetamide, 4.

 K_2CO_3 (352.0 mg, 2.55 mmol, 2 eq.) was added to a stirred solution of intermediate **2** (382.5 mg, 1.27 mmol, 1 eq.) in anhydrous acetonitrile (10.0 mL). After 10 min at 50 °C h, *N*-(4-bromophenyl)-2-chloroacetamide **3** [43] (472.1 mg, 1.90 mmol, 1.5 eq.) was added,

and the reaction was carried out at reflux for 3.5 h. The mixture was then cooled at room temperature to obtain **4** as a yellowish precipitate, which was filtered under vacuum, washed with acetonitrile (50.0 mL) and water (150.0 mL), and purified by crystallisation from EtOH. Yield = 69.1% (450.0 mg, 0.88 mmol); mp = 280-281 °C (EtOH). ¹H-NMR (CDCl₃) δ 9.05 (exch br s, 1H, NH), 8.51 (s, 1H, Ar), 8.06-8.10 (m, 2H, Ar), 7.97-8.01 (m, 2H, Ar), 7.44-7.53 (m, 8H, Ar), 6.17 (s, 1H, Ar), 5.03 (s, 2H, *CH*₂Ar), 4.91 (s, 2H, CH₂CO), 1.99 (s, 3H, CH₃). ¹³C-NMR (CDCl₃) δ 165.4 (C), 161.2 (C), 146.3 (C), 142.6 (C), 137.0 (C), 132.1 (2CH), 131.7 (2C), 131.1 (CH), 130.6 (2C), 129.5 (2CH), 128.4 (C), 127.6 (CH), 126.8 (2CH), 125.4 (2CH), 124.1 (2CH), 121.6 (2CH), 117.1 (C), 58.4 (CH₂), 27.7 (CH₂), 21.0 (CH₃). LC-MS (ESI): m/z calcd. for C₂₈H₂₂BrN₃O₂ 512.4 (MW), found 512.1/514.1 with a correct isotopic ratio 1:1 of ions species [M+H]⁺; *t*_R= 2.8. HRMS (ESI): m/z calcd. for C₂₈H₂₃BrN₃O₂ 512.0974 [M+H]⁺, found 512.0966. UV-vis (PBS buffer, 10 µM): λ /nm 357, 377, 398. Fluorescence (PBS buffer, PBS buffer): $\lambda_{max}(ex)$ 377 nm, $\lambda_{max}(em)$ 497 nm.

2-[(5-(Dimethylamino)naphthalen-1-yl)sulfonyl]-4-(3-methoxybenzyl)-6methylpyridazin-3(2*H*)-one, 7.

Dansyl chloride 6 (281.0 mg, 1.04 mmol, 1 eg.) and Et₃N (0.44 mL, 3.13 mmol, 3 eg.) were added to a solution of intermediate 5 [37] (200.0 mg, 0.87 mmol) in anhydrous acetonitrile (3.0 mL). The reaction was stirred at room temperature for 16 h. After evaporation of the solvent, the residue was extracted with brine and CH₂Cl₂ (3 x 25.0 mL). The organic layer was dried over Na₂SO₄ and evaporated *in vacuo*. Lastly, the crude product was purified by column flash chromatography using cyclohexane/ethyl acetate 2:1 as eluent to obtain 7 as an orange solid. Yield = 11.5% (46.5 mg, 0.10 mmol); mp= 51-52 °C (EtOH). ¹H-NMR (CDCl₃) δ 8.64 (d, 1H, Ar, J= 8.8 Hz), 8.36-8.41 (m, 2H, Ar), 7.56-7.61 (m, 2H, Ar), 7.27 (t, 1H, Ar, J= 8.0 Hz, overlapped with CDCl₃ signal), 7.20 (d, 1H, Ar, J= 7.2 Hz), 6.98 (s, 1H, Ar), 6.85 (d, 1H, Ar, J= 8.8 Hz), 6.77-6.80 (m, 2H, Ar), 4.06 (s, 2H, CH₂), 3.81 (s, 3H, OCH₃), 2.89 (s, 6H, 2 x NCH₃), 2.51 (s, 3H, CCH₃). ¹³C-NMR (CDCl₃) δ 160.0 (C), 159.9 (C), 159.3 (C), 137.4 (2C), 134.5 (2C), 131.9 (CH), 131.3 (CH), 130.3 (CH), 130.0 (CH), 129.9 (2C), 128.7 (CH), 121.7 (2CH), 115.1 (2CH), 112.7 (2CH), 55.2 (CH₃), 45.5 (CH₃), 35.0 (CH₂), 21.5 (CH₃). LC-MS (ESI): m/z calcd. for C₂₅H₂₅N₃O₄S 463.5 (MW), found 464.2 $[M+H]^+$; t_R = 2.7. HRMS (ESI): m/z calcd. for C₂₅H₂₆N₃O₄S 464.1644 [M+H]⁺, found 464.1630. UV-vis (PBS buffer, 10 μM): λ/nm 299. Fluorescence (PBS buffer, 10 μM): $\lambda_{max}(ex)$ 299 nm, $\lambda_{max}(em)$ 511 nm.

3-(6-Oxo-1,4,5,6-tetrahydropyridazin-3-yl)propanoic acid, 9.[48]

To an ice-cold solution of 4-oxoheptanedioic acid **8** (550.0 mg, 3.14 mmol, 1 eq.) in EtOH (8.0 mL), hydrazine hydrate (60% in water, 0.19 mL, 3.77 mmol, 1.2 eq.) was added dropwise and the solution was stirred at 60 °C for 1.5 h. The solvent was evaporated, and the residue was extracted with brine (5.0 mL) and CH₂Cl₂ (3 x 15.0 mL). The combined organic layers were dried over Na₂SO₄ and evaporated under vacuum. Lastly, the crude product was purified by crystallisation from EtOH to yield **9** as an amorphous white solid. Yield = 67.7% (362.0 mg, 2.13 mmol); mp = 186-187 °C (EtOH). ¹H-NMR (DMSO-d₆) δ 12.08 (exch br s, 1H, OH), 10.47 (exch, 1H, NH), 2.45-2.48 (m, 4H, NCO(CH₂)₂), 2.42 (t, 2H, *CH*₂COOH, *J*= 8.4 Hz), 2.23-2.28 (m, 2H, *CH*₂COOH).

3-[5-(3-Methoxybenzyl)-6-oxo-1,6-dihydropyridazin-3-yl]propanoic acid, 10.

To a stirred solution of **9** (360.0 mg, 2.11 mmol, 1 eq.) in 10.0 mL of KOH/absolute EtOH (5% w/v), 3-methoxybenzaldehyde (0.33 mL, 2.74 mmol, 1.3 eq.) was added, and the reaction was refluxed for 5 h. After cooling, the mixture was evaporated under vacuum, diluted with ice-cold water (5.0 mL), acidified with 6N HCl (pH= 3), and extracted with CH₂Cl₂ (3 x 25.0 mL). Removal of the solvent resulted in the crude product, which was purified by column flash chromatography using CH₂Cl₂/MeOH (gradient 95:5→90:10) as eluent to obtain **10** as yellowish solid. Yield = 36.5% (222.0 mg, 0.77 mmol); mp = 145-146 °C (EtOH). ¹H NMR (CD₃OD) δ 7.26-7.21 (m, 1H, Ar), 7.08 (s, 1H, Ar), 6.86-6.80 (m, 3H, Ar), 3.83 (s, 2H, Ar*CH*₂), 3.79 (s, 3H, CH₃), 2.85 (t, 2H, N=CCH₂*CH*₂, J= 7.1 Hz), 2.66 (t, 2H, N=C*CH*₂CH₂, J= 7.1 Hz). ¹³C NMR (DMSO-d₆) δ 173.50 (C), 160.65 (C), 159.31 (C), 146.29 (C), 141.89 (C), 139.57 (C), 130.64 (CH), 129.43 (CH), 121.18 (CH), 114.80 (CH), 111.73 (CH), 54.93 (CH₃), 34.69 (CH₂), 31.14 (CH₂), 28.75 (CH₂). LC-MS (ESI): m/z calcd. for C₁₅H₁₆N₂O₄ 288.3 (MW), found 289.1 [M+H]⁺; *t*_R= 1.4. HRMS (ESI): m/z calcd. for C₁₅H₁₇N₂O₄ 289.1188 [M+H]⁺, found 289.1188.

Ethyl 3-{5-[(3-methoxyphenyl)methyl]-6-oxo-1,6-dihydropyri dazin-3-yl}propanoate,

11.

600 mg (2.08 mmol) of compound **10** was dissolved in 30 mL of anhydrous ethanol, 4 drops of concentrated sulfuric acid was added. The reaction was refluxed for 4h. Then the solution was concentrated under vacuum and diluted by dichloromethane. The organic phase was washed twice with water and dried over Na₂SO₄. The organic solvent was removed under vacuum and the solid residue was purified by flash column chromatography (eluent: hexane/ethyl acetate, 3:2) to obtain **11** as a pale-yellow solid. Yield = 76% (503 mg, 1.59 mmol); mp = 74-76 °C. ¹H NMR (CDCl₃) δ 7.29-7.25 (m,1H, Ar), δ 6.88-6.69 (m, 4H, Ar); δ 4.11 (q, 2H, OCH₂CH₃, J= 7.0), 3.87 (s, 2H, Ar-*CH*₂), 3.81 (s, 3H, OCH₃), 2.83 (t, 2H, N=CCH₂*CH*₂, J= 7.1), 2.64 (t, 2H, N=C*CH*₂CH₂, J= 7.1), 1.23 (t, 3H, OCH₂*CH*₃, J= 7.0 Hz). ¹³C NMR (CDCl₃) δ 172.6 (C), 161.7 (C), 160.1 (C), 147.4 (C), 143.7 (C), 138.5 (C), 130.9 (CH), 130.0 (CH), 121.9 (CH), 115.4 (CH), 112.5 (CH), 60.8 (CH₂), 55.4 (CH₃), 35.5 (CH₂), 31.7 (CH₂), 29.4 (CH₂), 14.3 (CH₃). HRMS(ESI): m/z calcd. for C₁₇H₂₀N₂O₄ 317.1501 [M+H]⁺, found 317.1505.

Ethyl 3-{1-[2-(4-bromophenylamino)-2-oxoethyl]-5-(3-methoxybenzyl)-6-oxo-1,6dihydropyridazin-3-yl}propanoate, 12.

K₂CO₃ (511 mg, 3.70 mmol, 3 eq.) was added to a stirred solution of compound **11** (390 mg, 1.23 mmol, 1.0 eq.) and N-(4-bromophenyl)-2-chloroacetamide **3** [43] (368 mg, 1.48 mmol, 1.2 eq.) in 10 mL of acetonitrile. The reaction was refluxed for 6 h. Then the solution was diluted with dichloromethane and extracted twice with water. The organic phase was dried, and the solvent was removed under vacuum. The solid residue was purified by flash column chromatography (hexane/dichloromethane/ethyl acetate, 8:9:3) to obtain **12** as pale-yellow solid. Yield=69% (447 mg, 0.85 mmol); mp = 53-54 °C. ¹H NMR (CDCl₃) δ 8.92 (s,1H, CONH), 7.39–7.35 (m, 4H, Ar), 7.28–7.24 (m, 1H, Ar), 6.84–6.77 (m, 4H, Ar), 4.91 (s, 2H, NCH₂CO), δ 4.09 (q, 2H, OCH₂CH₃, J= 7.1), 3.89 (s, 2H, Ar), 3.79 (s, 3H, OCH₃), 2.87 (t, 2H, N=CCH₂CH₂, J= 7.1 Hz), 2.68 (t, 2H, N=CCH₂CH₂, J= 7.1 Hz), 1.25 (t, 3H, OCH₂CH₃, J= 7.1 Hz). ¹³C NMR (CDCl₃) δ 172.6 (C), 165.3 (C), 161.2 (C), 160.1 (C), 147.6 (C), 143.5 (C), 138.3 (C), 136.9 (C), 131.9 (CH), 130.8 (2CH), 130.0 (CH), 121.8 (2CH), 121.6 (CH), 117.0 (C), 115.4 (CH), 112.4 (CH), 60.9 (CH₂), 58.4 (CH₃), 55.4 (CH₃), 36.2 (CH₂), 31.6 (CH₂), 29.5 (CH₂), 14.3 (CH₃). HRMS(ESI): m/z calcd. for C₂₅H₂₆BrN₃O₅ 528.1134, 530.1114 [M+H]⁺, found 528.1135, 530.1117.

3-{1-[2-(4-Bromophenylamino)-2-oxoethyl]-5-(3-methoxybenzyl)-6-oxo-1,6dihydropyridazin-3-yl}propanoic acid, 13.

360 mg (0.68 mmol) of compound **12** was dissolved in 10 mL of tetrahydrofuran and 10 mL of 2 M NaOH solution was added. The reaction was kept at room temperature overnight. After completion of the reaction, the solution was neutralized by 1 N hydrochloric acid. The aqueous solution was then extracted with dichloromethane three times. The organic phase was collected and dried over Na₂SO₄. Subsequently, the solvent was removed under vacuum to obtain 13 as pale-yellow solid. Yield= 100% (340 mg, 0.68 mmol); mp = 133-135 °C (EtOH). ¹H NMR (CD₃OD) δ 7.55–7.47 (m, 4H, Ar), 7.26 (t, 1H, Ar, J=7.9 Hz), 7.11 (s, 1H, CH), 6.88–6.83 (m, 3H, Ar), 4.97 (s, 2H, NCH₂CO), 3.89 (s, 2H, Ar-CH₂), 3.81 (s, 3H, OCH₃), 2.90 (t, 2H, N=CCH₂CH₂, J=7.1), 2.69 (t, 2H, N=CCH₂CH₂, J=7.1 Hz). ¹³C NMR (CD₃OD) δ 176.3 (C), 167.4 (C), 162.4 (C), 161.5 (C), 149.0 (C), 144.0 (C), 140.2 (C), 138.8 (C), 132.9 (2CH), 132.3 (CH), 130.7 (CH), 122.8 (2CH), 122.6 (CH), 117.7 (C), 115.9 (CH), 113.4 (CH), 56.6 (CH₂), 55.6 (CH₃), 36.6 (CH₂), 32.4 (CH₂), 30.4 (CH₂). LC-MS (ESI): m/z calcd. for C₂₃H₂₂BrN₃O₅ 500.3 (MW), found 500.1/502.1 [M+H]⁺, 517.1/519.1 [M+NH₄]⁺, 522.1/524.1 [M+Na]⁺, 497.9/499.9 [M-H]⁻ all with a correct isotopic ratio 1:1 of ions species; $t_{\rm R}$ = 1.4. HRMS(ESI): m/z calcd. for C₂₃H₂₂BrN₃O₅ 500.0821, 502.0801 [M+H]⁺, found 500.0811, 502.0786.

5-({2-[3-(1-(2-[(4-bromophenyl)amino]-2-oxoethyl)-5-(3-methoxybenzyl)-6-oxo-1,6dihydropyridazin-3-yl)propanamido]ethyl}amino)naphthalene-1-sulfonic acid, 16. HOBt (7.6 mg, 0.06 mmol), Et₃N (0.011 mL, 0.08 mmol, 1.3 eq.) and 1-ethyl-3-(3dimethylaminopropyl)carbodiimide (EDC, 0.014 mL, 0.08 mmol, 1.3 eq.) were sequentially added to a stirred solution of **13** (40.0 mg, 0.08 mmol, 1.3 eq.) in 1.0 mL of anhydrous DMF. A solution of 5-[(2-aminoethyl)amino]naphthalene-1-sulfonic acid (EDANS) **14** (14.9 mg, 0.06 mmol, 1eq.) in anhydrous DMF (4 mL) was added, and the mixture was stirred under N₂ for 48 h at room temperature. After evaporation of the solvent, the residue was and extracted with brine and CH₂Cl₂ (3 x 15 mL). The organic layers were combined, dried over Na₂SO₄, and evaporated under vacuum to obtain a crude residue, which was purified by column flash chromatography using CH₂Cl₂/MeOH 90:10 as eluent to obtain **16** as a brownish solid. Yield = 92.4% (41.5 mg, 0.05 mmol); mp = 60-70 °C (dec). ¹H-NMR

(CD₃OD) δ 8.14-8.20 (m, 2H, Ar), 8.02 (d, 1H, Ar, J= 8.4 Hz), 7.50 (d, 2H, Ar, J= 8.8 Hz), 7.42 (d, 2H, Ar, J= 8.8 Hz), 7.29-7.38 (m, 2H, Ar), 7.23 (t, 1H, Ar, J= 8.0 Hz), 6.99 (s, 1H, Ar), 6.79-6.83 (m, 3H, Ar), 6.56 (s, 1H, Ar, J= 7.6 Hz), 4.59 (s, 2H, NCH₂CO), 3.76 (s, 3H, OCH₃); 3.77 (s, 2H, CH₂Ar), 3.52 (t, 2H, CH₂NH, J= 5.6 Hz), 3.28 (t, 2H, CH₂NH, J= 5.6 Hz), 2.89 (t, 2H, CH₂CH₂CONH, J= 6.8 Hz), 2.81 (exch br s, 1H, SO₃H), 2.58 (t, 2H, CH_2CH_2CONH , J= 6.8 Hz). ¹³C-NMR (CD₃OD, signals marked with * correspond to additional peaks due to the presence of a minor rotamer) δ 175.7(C), 167.4 (C), 162.1 (C), 161.4 (C), 161.3 (C*), 148.8 (C), 145.4 (C), 144.0 (C), 141.8 (C), 140.2 (C), 138.8 (C), 132.8 (2CH), 132.0 (CH), 131.4 (C), 130.7 (CH), 128.6 (CH), 128.54 (CH*), 126.7 (CH), 125.4 (C), 125.3 (CH), 123.6 (CH), 122.8 (2CH), 122.6 (CH), 120.4 (C), 118.6 (CH*), 117.7 (CH*), 116.4 (CH*), 116.3 (CH), 116.0 (CH), 113.5 (CH*), 113.3 (CH), 105.1 (CH*), 104.9 (CH), 56.4 (CH₂), 55.9 (CH3^{*}), 55.6 (CH₃), 45.5 (CH₂), 45.0 (CH₂^{*}), 40.3 (CH₂^{*}), 39.7 (CH₂), 36.5 (CH₂), 34.4 (CH₂), 31.0 (CH₂). LC-MS (ESI): m/z calcd. for C₃₅H₃₄BrN₅O₇S 748.6 (MW), found 748.2/750.2 with a correct isotopic ratio 1:1 of ions species [M+H]⁺ and 746.1/748.1 with a correct isotopic ratio 1:1 of ions species [M-H]⁻; t_{R} = 2.1. HRMS (ESI): m/z calcd. for C₃₅H₃₅BrN₅O₇S 748.1441 [M+H]⁺, found 748.1481, 750.1464. UV-Vis (PBS buffer, 10 μ M): λ /nm 250, 334. Fluorescence (PBS buffer, 10 μ M): $\lambda_{max}(ex)$ 334 nm, $\lambda_{max}(em)$ 505 nm.

3-{1-[2-(4-bromophenylamino)-2-oxoethyl]-5-(3-methoxybenzyl)-6-oxo-1,6dihydropyridazin-3-yl}-*N*-(5,5-difluoro-1,3,7,9,10-pentamethyl-5H-4l4,5l4dipyrrolo[1,2-c:2',1'-f][1,3,2]diazaborinin-2-yl)propanamide, 17.

Triethylamine (24.2 µL, 0.174 mmol, 3.5 eq.) was added to a solution of **13** (29 mg, 0.058 mmol, 1 eq.) in 10 mL of anhydrous tetrahydrofuran under N₂ atmosphere. The solution was kept at -5 °C for 30 min, followed by addition of 6.9 mg (0.064 mmol, 1.1 eq.) of ethyl chloroformate at 0°C. One hour later, the ice bath was removed, and 22 mg (0.0794 mmol, 1.4 eq.) of **15** were added. After 16 h at room temperature, the organic solvent was removed under vacuum, and the solid residue was purified by flash column chromatography (dichloromethane/methanol, 100:1) performed on aluminum oxide (neutral, 150 mesh, Merk) to obtain **17** as orange solid. Yield = 85%; mp = 136-137 °C. ¹H NMR (DMSO-d₆) δ 7.37 (s, 1H, CONH); 7.53~7.48 (m, 4H, Ar), 7.24~7.21 (m, 2H, Ar), 6.85-6.23 (m, 3H, Ar), 6.23 (s, 1H, H-pyrrole), 4.84 (s, 2H, NCH₂CO), 3.78 (s, 2H, Ar-CH₂), 3.71 (s, 3H, OCH₃), 2.88 (t, 2H, N=CCH₂CH₂, J= 7.2 Hz), 2.66 (t, 2H, N=CCH₂CH₂, J= 7.2

Hz), 2.58 (s, 3H, Ar-*CH*₃), 2.42 (s, 3H, Ar-*CH*₃), 2.41 (s, 3H, Ar-*CH*₃), 2.20 (s, 3H, Ar-*CH*₃), 2.14 (s, 3H, Ar-*CH*₃). ¹³C-NMR (101 MHz, DMSO) δ 170.70 (C), 165.28 (C), 159.62 (C), 159.34 (C), 152.98 (C), 149.20 (C), 146.43 (C), 142.63 (C), 141.69 (C), 141.56 (C), 139.35 (C), 138.11 (C), 134.92 (C), 131.65 (2×CH), 131.46 (C), 130.62 (CH), 129.45 (2×CH), 129.19 (C), 128.06 (C), 121.24 (CH), 121.16 (CH), 120.95 (CH), 115.00 (C), 114.83 (CH), 111.85 (CH), 55.17 (CH₂), 54.91 (CH₃), 35.15 (CH₂), 32.94 (CH₂), 29.59 (CH₂), 16.86 (CH₃), 16.27 (CH₃), 14.07 (CH₃), 13.54 (CH₃), 11.87 (CH₃). *t*_R = 16.43 min. HRMS(ESI): m/z calcd. for C₃₇H₃₈BBrF₂N₆O₄ 781.2097, 783.2076 [M+Na]⁺, found 781.2216, 783.2206. UV-vis (methanol, 1 μM): λ/nm 500. Fluorescence (methanol, 1 μM): λ/max(ex) 500 nm, λ_{max}(em) 536 nm.

4-(Anthracen-9-yl-methyl)-6-methylpyridazin-3(2H)-one, 2









2-[5-(Anthracen-9-yl-methyl)-3-methyl-6-oxopyridazin-1(6H)-yl]-N-(4-bromophenyl) acetamide, 4



Elemental Composition Report

Single Mass Analysis Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions 237 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass) Elements Used: C: 28-28 H: 0-200 N: 0-10 O: 0-10 Na: 0-1 Br: 1-1

A.CILIBRIZZI AC135 PPT ms17536a 235 (2.843) Cm (230:238)



Page 1









7.519e-1







2-[(5-(Dimethylamino)naphthalen-1-yl)sulfonyl]-4-(3-methoxybenzyl)-6-methylpyridazin-3(2H)-one, **7**



Elemental Composition Report

Page 1

Single Mass Analysis Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions 472 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass) Elements Used: C: 25-25 H: 0-200 N: 0-10 O: 0-10 Na: 0-1 S: 0-1 A.CILIBRIZZI AC119 2ND ms17099 227 (2.746) Cm (227:229)







3-(6-Oxo-1,4,5,6-tetrahydropyridazin-3-yl)propanoic acid, 9





Elemental Composition Report

Page 1

1: TOF MS ES+

Single Mass Analysis Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions 229 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass) Elements Used:

C: 15-15 H: 0-200 N: 0-10 O: 0-10 Na: 0-1 ACILIBRIZZI AC133C ms17684 107 (1.294) Cm (105:107)







3.3e+003



Ethyl 3-{5-[(3-methoxyphenyl)methyl]-6-oxo-1,6-dihydropyri dazin-3-yl}propanoate, 11









Ethyl 3-{1-[2-(4-bromophenylamino)-2-oxoethyl]-5-(4-methoxybenzyl)-6-oxo-1,6dihydropyridazin-3-yl}propanoate,**12**









3-{1-[2-(4-bromophenylamino)-2-oxoethyl]-5-(3-methoxybenzyl)-6-oxo-1,6-dihydro pyridazin-3-yl}propanoic acid, **13**















5,5-Difluoro-1,3,7,9,10-pentamethyl-5H- $4\lambda^4$,5 λ^4 -dipyrrolo[1,2-c:2',1'-f][1,3,2]diazaborinin-2-amine, **15**



3-{1-[2-(4-Bromophenylamino)-2-oxoethyl]-5-(3-methoxybenzyl)-6-oxo-1,6-dihydro pyridazin-3-yl}propanamidoethyl-2-(5-aminonaphthalene)-1-sulfonic acid, **16**



Elemental Composition Report

Page 1

Single Mass Analysis Tolerance = 8.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions 345 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass) Elements Used: C: 35-35 H: 0-200 N: 0-10 O: 0-15 Na: 0-1 S: 1-1 Br: 1-1

A.CILIBRIZZI AC160 ms21329 175 (2.121) Cm (174:175)

1: TOF MS ES+















32

8.8e+001





3-{1-[2-(4-bromophenylamino)-2-oxoethyl]-5-(3-methoxybenzyl)-6-oxo-1,6dihydropyridazin-3-yl}-N-(5,5-difluoro-1,3,7,9,10-pentamethyl-5H-4l4,5l4-dipyrrolo[1,2c:2',1'-f][1,3,2]diazaborinin-2-yl)propanamide, **17**



Chemical Formula: C₃₇H₃₈BBrF₂N₆O₄ Exact Mass: 758.2199 Molecular Weight: 759.4598







2. Optical profiles of probes



Fig. S2.1. The absorbance (blue) and emission (orange) profile for compound **4**, **7**, **16** and **17**.



Fig. S2.2. The maximal absorption of compound **16** at different concentrations. The molar extinction coefficient was calculated to be 4600 M⁻¹•cm⁻¹ by linear regression.



Fig. S2.3. The linear plot of fluorescence versus absorbance for fluorescein (the standard for relevant quantum yield calculation) and compound **16**. The quantum yield (QY) of compound **16** was given by the equation: QY_{Compound 16}=QY_{Quinine sulphate}•Slope _{Compound 16}/Slope _{Quinine sulphate}.



Fig S2.4. The maximal absorption of compound 17 at different concentrations. The molar extinction coefficient was calculated to be 74400 M⁻¹•cm⁻¹ by linear regression.



Fig S2.5. The linear plot of fluorescence versus absorbance for fluorescein (the standard for relevant quantum yield calculation) and compound **17**. The quantum yield (QY) of compound **17** was given by the equation: QY_{Compound 17}=QY_{Fluoresein}•Slope _{Compound 17}/ Slope _{Fluoresein}.

3. Representative dose-response curves from Ca²⁺ experiments (for 16 and 17)



Fig. S3.1. Flow Cytometry analysis for the expression levels of FPR1 in FPR1transfected HL-60 cells *vs.* wild-type HL-60 cells. FITC-anti-FPR1 is the FPR1 antibody used in the test, from BioLegend, San Diego, CA, USA, cat# 391603.



Figure S3.2. Evaluation of agonist and antagonist effects of compound **16** in FPR1-HL60 and FPR2-HL60 Cells. FPR1-HL60 (upper panel) and FPR2-HL60 (lower panel) cells were treated with the indicated concentrations of compound **16** or 1% DMSO (negative control not shown), and [Ca²⁺]_i was measured to evaluate agonist activity (red lines and symbols). To evaluate antagonist activity, the cells were incubated with compound **16** or 1% DMSO (negative control not shown) for 10 min, followed by activation with 5 nM *f*MLF (upper panel, blue line and symbols) or WKYMVM (lower panel, blue line and symbols) and subsequent monitoring of [Ca²⁺]_i, as described. The data shown are presented as the mean ± SD from one experiment that is representative of three independent experiments with similar results.



Figure S3.3. Evaluation of agonist and antagonist effects of compound **17** in FPR1-HL60 and FPR2-HL60 Cells. FPR1-HL60 (upper panel) and FPR2-HL60 (lower panel) cells were treated with the indicated concentrations of compound **17** or 1% DMSO (negative control not shown), and $[Ca^{2+}]_i$ was measured to evaluate agonist activity (round symbols). To evaluate antagonist activity, the cells were incubated with compound **17** or 1% DMSO (negative control (negative control not shown) for 30 min, followed by activation with 5 nM *f*MLF (upper panel, square symbols) or WKYMVM (lower panel, square symbols) and subsequent monitoring of $[Ca^{2+}]_i$, as described. The data shown are presented as the mean ± SD from one experiment that is representative of three independent experiments with similar results.











Figure S3.6. Evaluation of agonist and antagonist effects of probes on human neutrophils. Human neutrophils were treated with the indicated concentrations of compounds **4** (Panel A), **7** (Panel B), or **16** (Panel C), or 1% DMSO (negative control not shown), and $[Ca^{2+}]_i$ was measured to evaluate agonist activity (red lines and symbols). To evaluate antagonist activity, neutrophils were incubated with compounds **4** (Panel A), **7** (Panel B), or **16** (Panel C), or 1% DMSO (negative control not shown) for 10 min, followed by activation with 5 nM *f*MLF and subsequent monitoring of $[Ca^{2+}]_i$, as described. The data shown are presented as the mean ± SD from one experiment that is representative of three independent experiments with similar results