

Achieving broadband near-infrared emission with superior anti-thermal quenching by
optimizing excited-state population of Cr³⁺ in Gd₃ScGa₄O₁₂ garnet phosphors

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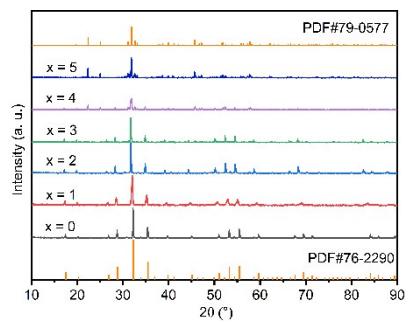


Fig. S1. XRD patterns of $\text{Gd}_3\text{Sc}_x\text{Ga}_{5-x}\text{O}_{12}:0.08\text{Cr}^{3+}$ ($x = 0\text{--}5$).

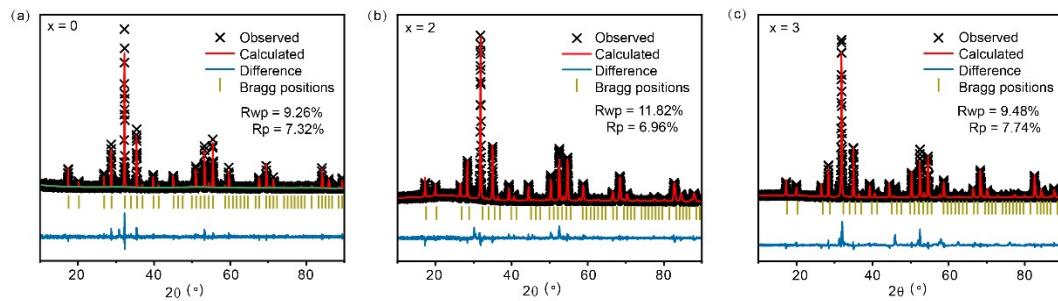


Fig. S2. (a–c) XRD Rietveld refinement of $\text{Gd}_3\text{Sc}_x\text{Ga}_{5-x}\text{O}_{12}$ ($x = 0, 2, 3$).

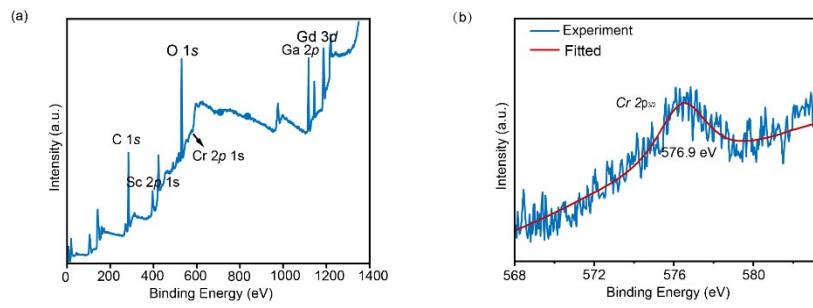


Fig. S3. (a) XPS full spectrum and (b) Cr 2p scan of $\text{Gd}_3\text{ScGa}_4\text{O}_{12}:0.08\text{Cr}^{3+}$.

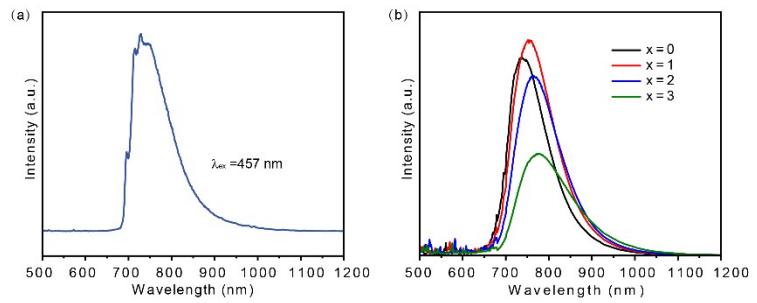


Fig. S4. (a) Emission spectrum of $\text{Gd}_3\text{ScGa}_4\text{O}_{12}:0.08\text{Cr}^{3+}$ at 77 K. (b) Relative emission intensity of $\text{Gd}_3\text{Sc}_x\text{Ga}_{5-x}\text{O}_{12}:0.08\text{Cr}^{3+}$ ($x = 0\text{--}3$).

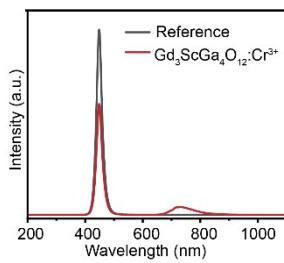


Fig. S5. IQE (82.75%), EQE (28.19%), and absorption coefficient (34.07%) of Gd₃ScGa₄O₁₂:0.08Cr³⁺.

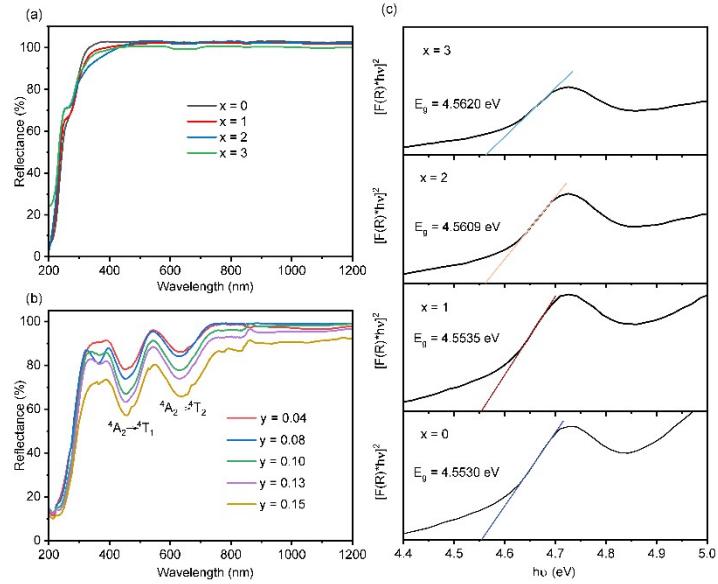


Fig. S6. (a) DRS of $\text{Gd}_3\text{Sc}_x\text{Ga}_{5-x}\text{O}_{12}$ ($x = 0\text{--}3$). (b) DRS of $\text{Gd}_3\text{ScGa}_4\text{O}_{12}:\text{yCr}^{3+}$ ($y = 0.04\text{--}0.15$). (c) The estimated optical band gaps of $\text{Gd}_3\text{Sc}_x\text{Ga}_{5-x}\text{O}_{12}$ ($x = 0\text{--}3$).

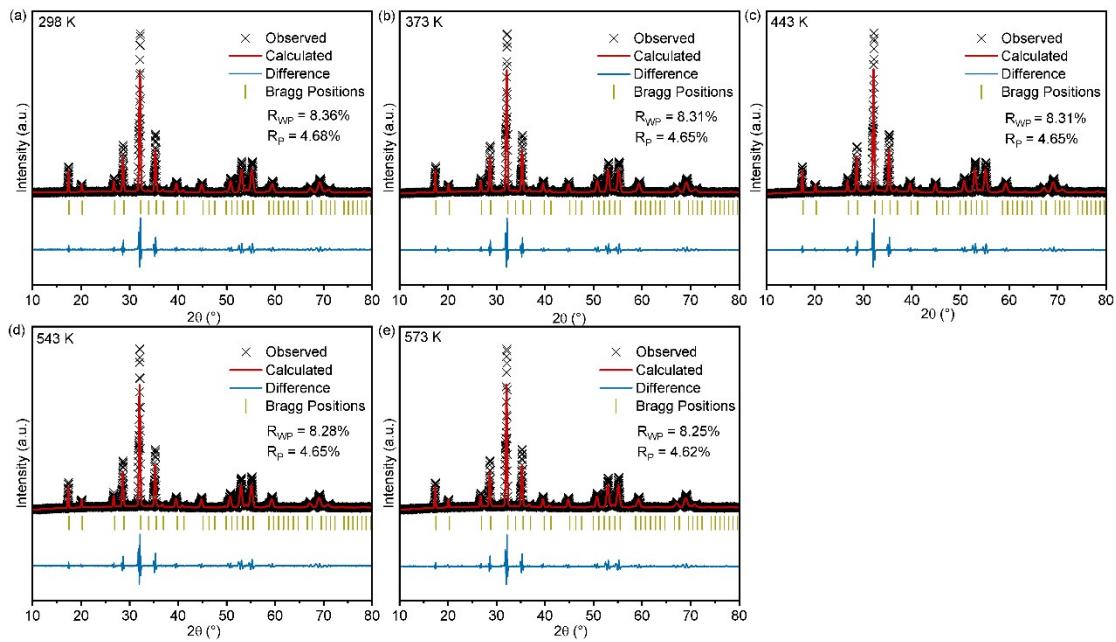


Fig. S7. XRD Rietveld refinement of (a) $\text{Gd}_3\text{ScGa}_4\text{O}_{12}:0.08\text{Cr}^{3+}$ at (a) 25 °C, (b) 100 °C, (c) 170 °C, (d) 270 °C, (e) 300 °C.

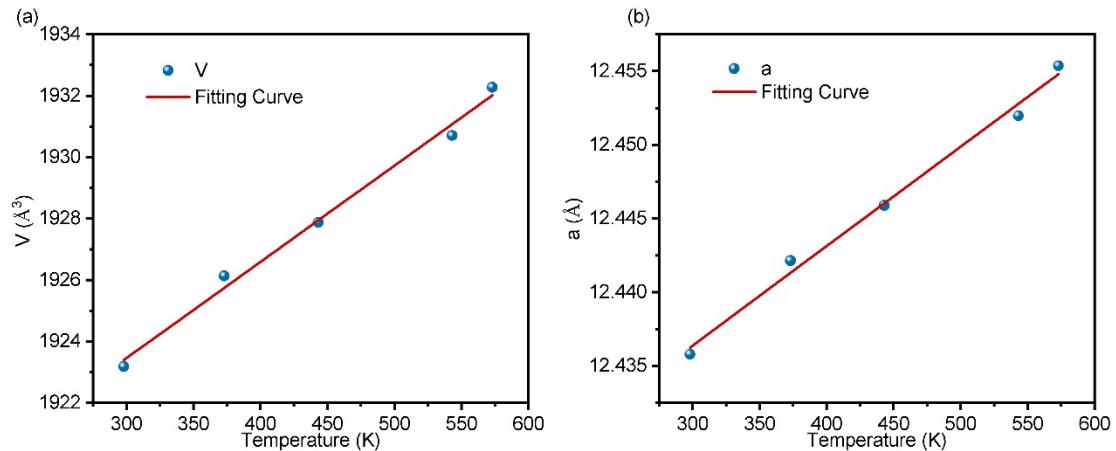


Fig. S8. Temperature-dependent (a) volume and (b) lattice parameters for $\text{Gd}_3\text{ScGa}_4\text{O}_{12}:0.08\text{Cr}^{3+}$.

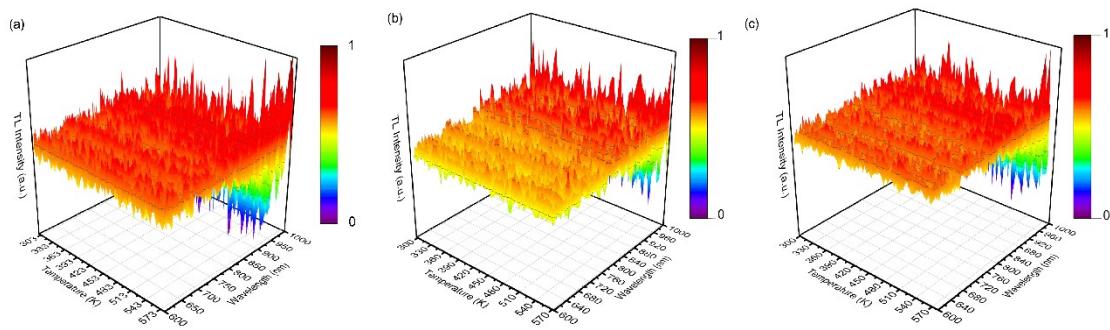


Fig. S9. Thermoluminescence curves of $\text{Gd}_3\text{ScGa}_4\text{O}_{12}:0.08\text{Cr}^{3+}$ measured without preheating (a) and with preheating at (b) 100 °C and (c) 150 °C under 460 nm excitation.

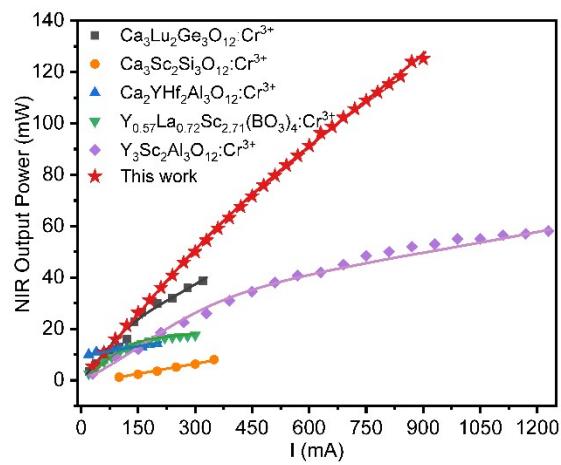


Fig. S10. A comparison of the NIR output power with other garnet NIR phosphors reported previously.^[1-5]

Table S1. Crystallographic data of $\text{Gd}_3\text{Sc}_x\text{Ga}_{5-x}\text{O}_{12}$ ($x = 0\text{--}3$).

Samples	$x = 0$	$x = 1$	$x = 2$	$x = 3$
Crystal system		Cubic		
Space group		Ia $\bar{3}$ d		
$a=b=c$ (Å)	12.3691	12.4523	12.5496	12.5747
$\alpha=\beta=\gamma$	90	90	90	90
Volume (Å ³)	1892.389	1930.835	1976.485	1988.344
Sc-O ₆ (Å)	1.999	2.012	2.076	2.087
R_p	0.0732	0.0731	0.0696	0.0774
R_{wp}	0.0926	0.0936	0.1182	0.0948
χ^2	1.35	1.299	1.827	4.937
Gd1	(0.125,0.000,0.250)	(0.125,0.000,0.250)	(0.125,0.000,0.250)	(0.125,0.000,0.250)
Ga1	(0.000,0.000,0.000)	(0.000,0.000,0.000)	(0.000,0.000,0.000)	(0.000,0.000,0.000)
Ga2	(0.375,0.000,0.250)	(0.375,0.000,0.250)	(0.375,0.000,0.250)	(0.375,0.000,0.250)
O1	(0.967,0.053,0.149)	(0.968,0.052,0.150)	(0.968,0.053,0.150)	(0.969,0.053,0.151)
Sc1	(0.000,0.000,0.000)	(0.000,0.000,0.000)	(0.000,0.000,0.000)	(0.000,0.000,0.000)

Table S2. The survey of ion-doped (excluding Cr³⁺-doped) inorganic phosphors with anti-TQ performance.

Host materials	Activators	Ex/Em (nm)	FWHM (nm)	IQE	I _T /I _{RT}	Ref.
Ba ₂ ZnGe ₂ O ₇	Bi ³⁺	346/500	129	—	138%@423K	[6]
Na ₃ Ca ₄ TeP ₃ O ₁₅	Tb ³⁺ , Eu ³⁺	376/545, 617	—	—	109%@423K	[7]
Lu ₂ SrAl ₄ SiO ₁₂	Eu ²⁺	342/420	—	—	106%@423K	[8]
Sc _{1.8} Zr _{0.1} Nb _{0.1} W _{2.85} O _{11.7}	Yb/Er ³⁺	522/980	—	—	114%@448K	[9]
Ca _{3.6} In _{3.6} (PO ₄) ₆	Dy ³⁺	349/575	—	30.3%	108.2%@423K	[10]
Ba ₃ EuAl ₂ O _{7.5}	Eu ³⁺	270/593	—	—	140%@443K	[11]
Ca ₂ InSbO ₆	Eu ³⁺	395/616	—	82%	114.5%@380K	[12]
LaSc ₃ (BO ₃) ₄	Eu ³⁺	395/618	—	88.3%	129.2%@548K	[13]
KAl ₁₁ O ₁₇	Mn ²⁺	450/510	~50	—	160%@548K	[14]
Cs ₂ MP ₂ O ₇ (M=Ba,Sr,Ca)	Eu ²⁺	360/548	—	98.9%	120%@448K	[15]
Mg ₃ Y ₂ Ge ₃ O ₁₂	Eu ³⁺ , Mn ⁴⁺	420/660	—	—	127%@500K	[16]
CsMoO ₂ F ₃	Mn ⁴⁺	450/633	—	54%	129.3%@423K	[17]
LaAlO ₃	Ca ²⁺ , Bi ³⁺ , Mn ⁴⁺	340/734	—	89.3%	103%@423K	[18]
In _{0.5} Sc _{1.5} (MoO ₄) ₃	Eu ³⁺	246/615	—	57.8%	108.2%@473K	[19]

Table S3. Crystallographic data of $\text{Gd}_3\text{ScGa}_4\text{O}_{12}:0.08\text{Cr}^{3+}$ at different temperatures.

Materials	$\text{Gd}_3\text{ScGa}_4\text{O}_{12}:0.08\text{Cr}^{3+}$				
Measuring temperature/°C	25	100	170	270	300
Crystal system	Cubic				
Space group	$\text{Ia}\bar{3}\text{d}$				
$a=b=c$ (Å)	12.4440	12.4498	12.4563	12.4652	12.4687
$\alpha=\beta=\gamma$	90°	90°	90°	90°	90°
Volume (Å ³)	1926.995	1929.669	1932.691	1936.864	1938.468
Average bond length of Cr-O (Å)	2.053	2.055	2.057	2.058	2.061
Volume of CrO ₆ octahedron (Å ³)	11.4548	11.4942	11.5276	11.5486	11.6141
R_p	0.0469	0.0466	0.0466	0.0466	0.0464
R_{wp}	0.0838	0.0832	0.0833	0.0829	0.0827
χ^2	2.08	2.86	2.92	2.79	2.67

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