

Multidentate copper complex on magnetic biochar nanoparticles as a practical and recoverable nanocatalyst for the selective synthesis of tetrazole derivatives

Marwan Majeed Maseer¹, Tavan Kikhavani^{1,*}, Bahman Tahmasbi^{2,*}

¹ Department of Chemical Engineering, Faculty of Engineering, Ilam University, Ilam, Iran. E-mail:

t.kikhavandi@ilam.ac.ir

² Department of Chemistry, Faculty of Science, Ilam University, P.O. Box 69315516, Ilam, Iran. E-mail:

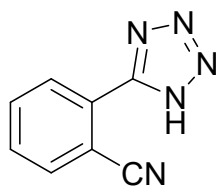
b.tahmasbi@ilam.ac.ir, bah.tahmasbi@gmail.com

Abstract:

In this work, biochar nanoparticles (B-NPs) were synthesized by pyrolysis of chicken manure. Then, the B-NPs were magnetized by Fe(0) nanoparticles. Then, the surface of biochar magnetic nanoparticles (FeB-MNPs) was modified by (3-chloropropyl)trimethoxysilane (3Cl-PTMS). Finally, a multidentate copper complex of 2,2'-(propane-1,3-diylbis(oxy))dianiline (P.bis(OA)) was immobilized on the surface of modified FeB-MNPs, which labelled as Cu-P.bis(OA)@FeB-MNPs. Cu-P.bis(OA)@FeB-MNPs was investigated as commercial, homoselective, practical, and recyclable nanocatalyst in the synthesis of 5-substituted-1H-tetrazole compounds through [3+2] cycloaddition of sodium azide (NaN₃) and organo-nitriles in PEG-400 as a green solvent. Cu-P.bis(OA)@FeB-MNPs was characterized by wavelength dispersive X-ray spectroscopy (WDX), scanning electron microscopy (SEM), thermogravimetric analysis (TGA), energy-dispersive X-ray spectroscopy (EDS), vibrating-sample magnetometer (VSM), atomic absorption spectroscopy (AAS) and N₂ adsorption-desorption (BET method) techniques. Cu-P.bis(OA)@FeB-MNPs was recovered and reused for several runs in the synthesis of tetrazoles.

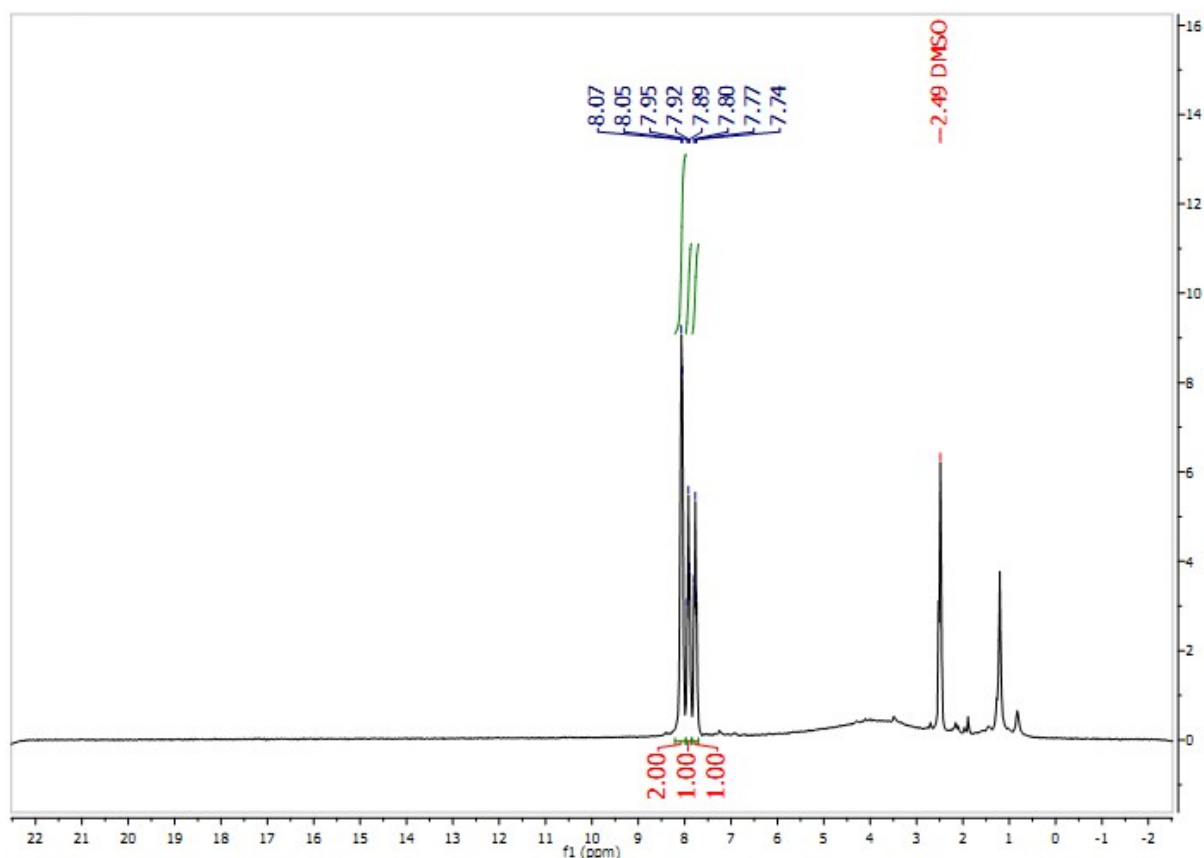
Keywords: Biochar nanoparticles, homoselective catalyst, magnetic nanoparticles, reusable nanocatalyst, 5-substituted-1H-tetrazoles, sodium azide, multidentate copper complex.

¹H NMR spectral data

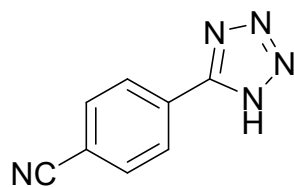


2-(1H-tetrazol-5-yl)benzonitrile

¹H NMR (250 MHz, DMSO): $\delta_{\text{H}} = 8.06$ (d, $J = 5.42$ Hz, 2H), 7.92 (t, $J = 7.05$ Hz, 1H), 7.76 (t, $J = 7.12$ Hz, 1H) ppm.

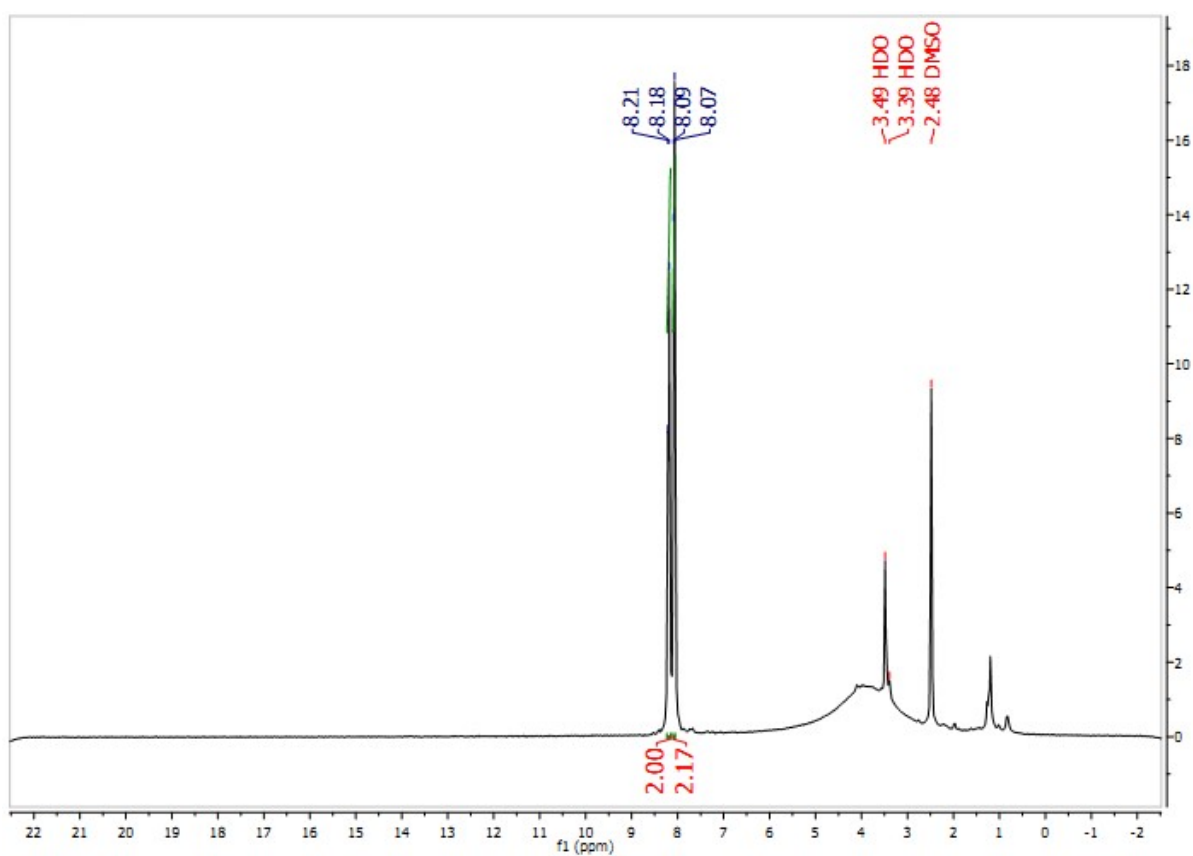


¹H NMR spectrum of 2-(1H-tetrazol-5-yl)benzonitrile from [3+2] cycloaddition of NaN_3 with phthalonitrile

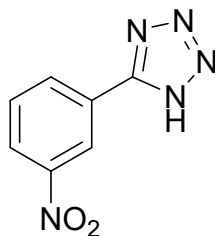


4-(1H-tetrazol-5-yl)benzotrile

^1H NMR (250 MHz, DMSO): $\delta_{\text{H}} = 8.20$ (d, $J = 8.22$ Hz, 2H), 8.08 (d, $J = 4.95$ Hz, 2H) ppm.

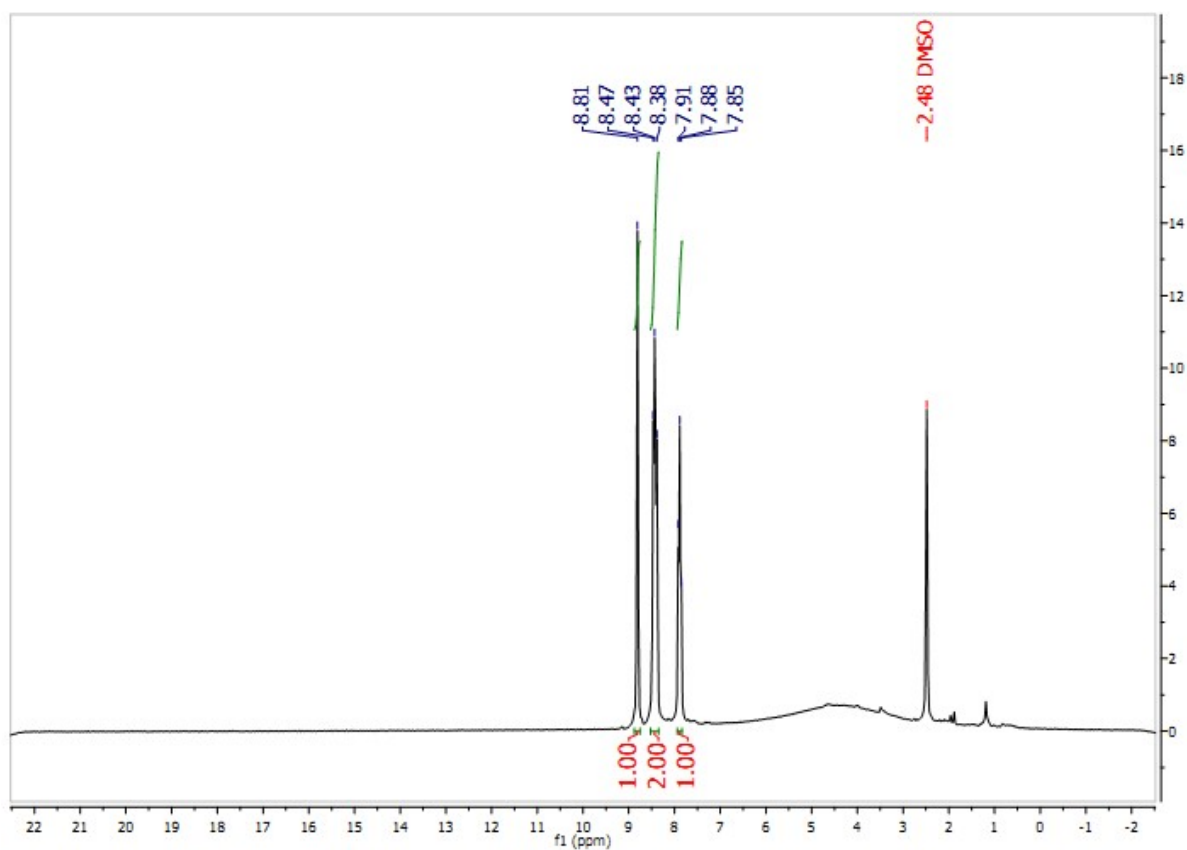


^1H NMR spectrum of 4-(1H-tetrazol-5-yl)benzotrile from [3+2] cycloaddition of NaN_3 with terephthalonitrile

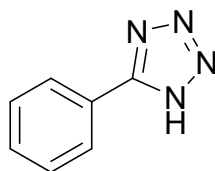


5-(3-nitrophenyl)-1H-tetrazole

^1H NMR (250 MHz, DMSO): $\delta_{\text{H}} = 8.81$ (s, 1H), 8.42 (t, $J = 8.25$ Hz, 2H), 7.88 (t, $J = 7.62$ Hz, 1H) ppm.

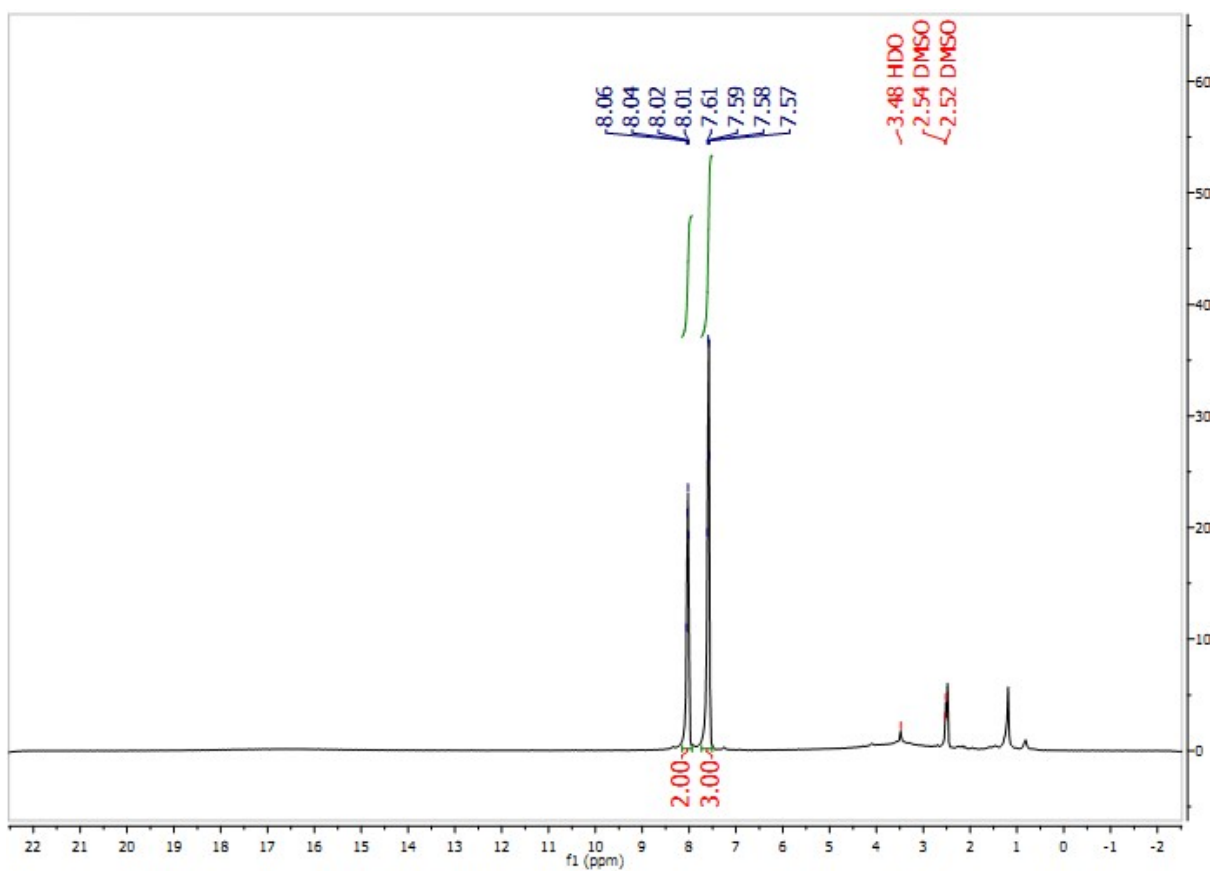


^1H NMR spectrum of 5-(3-nitrophenyl)-1H-tetrazole from [3+2] cycloaddition of NaN_3 with 3-nitrobenzonitrile

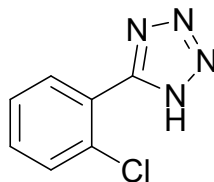


5-phenyl-1H-tetrazole

^1H NMR (250 MHz, DMSO): $\delta_{\text{H}} = 8.04$ (m, 2H), 7.59 (m, 3H) ppm.

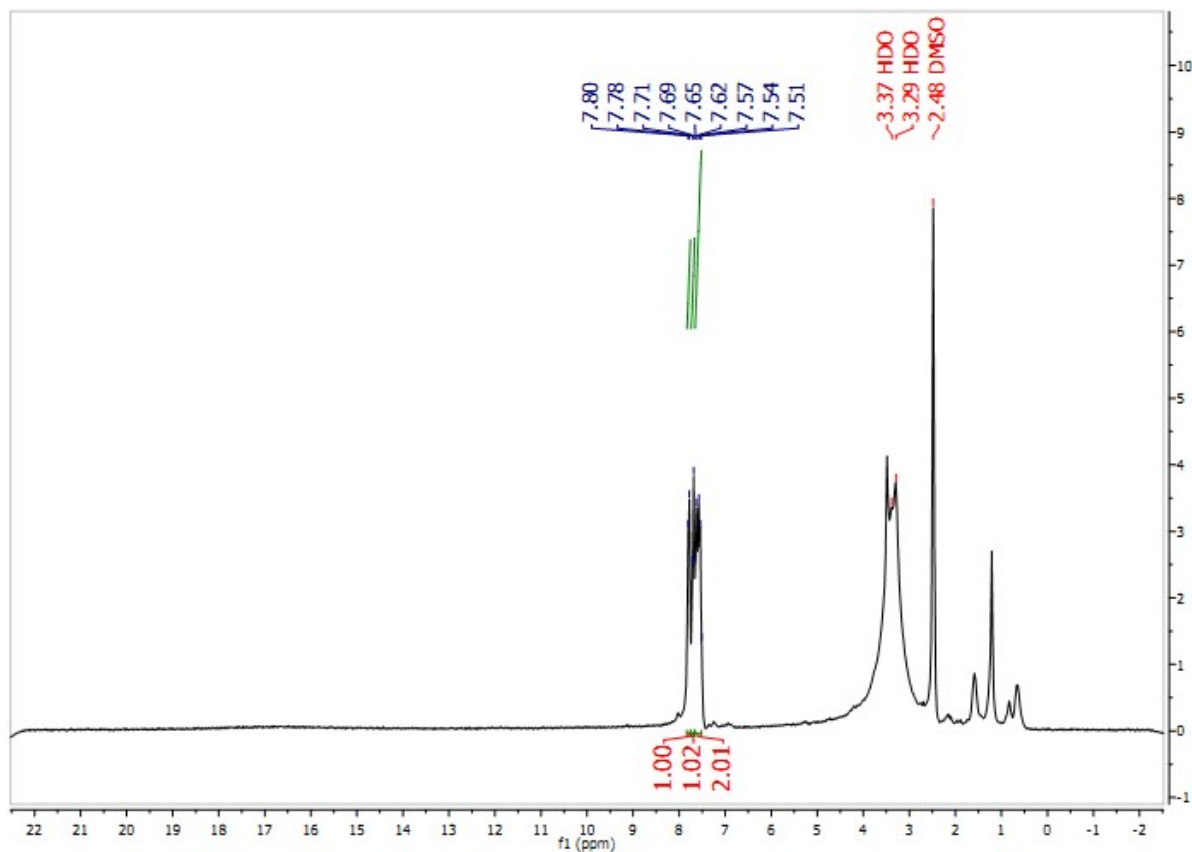


^1H NMR spectrum of 5-phenyl-1H-tetrazole from [3+2] cycloaddition of NaN_3 with 3-nitrobenzotrile

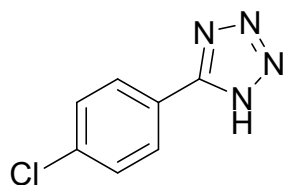


5-(2-chlorophenyl)-1H-tetrazole

^1H NMR (250 MHz, DMSO): $\delta_{\text{H}} = 7.79$ (d, $J = 5.90$ Hz, 1H), 7.70 (d, $J = 6.35$ Hz, 1H), 7.58 (m, 2H) ppm.

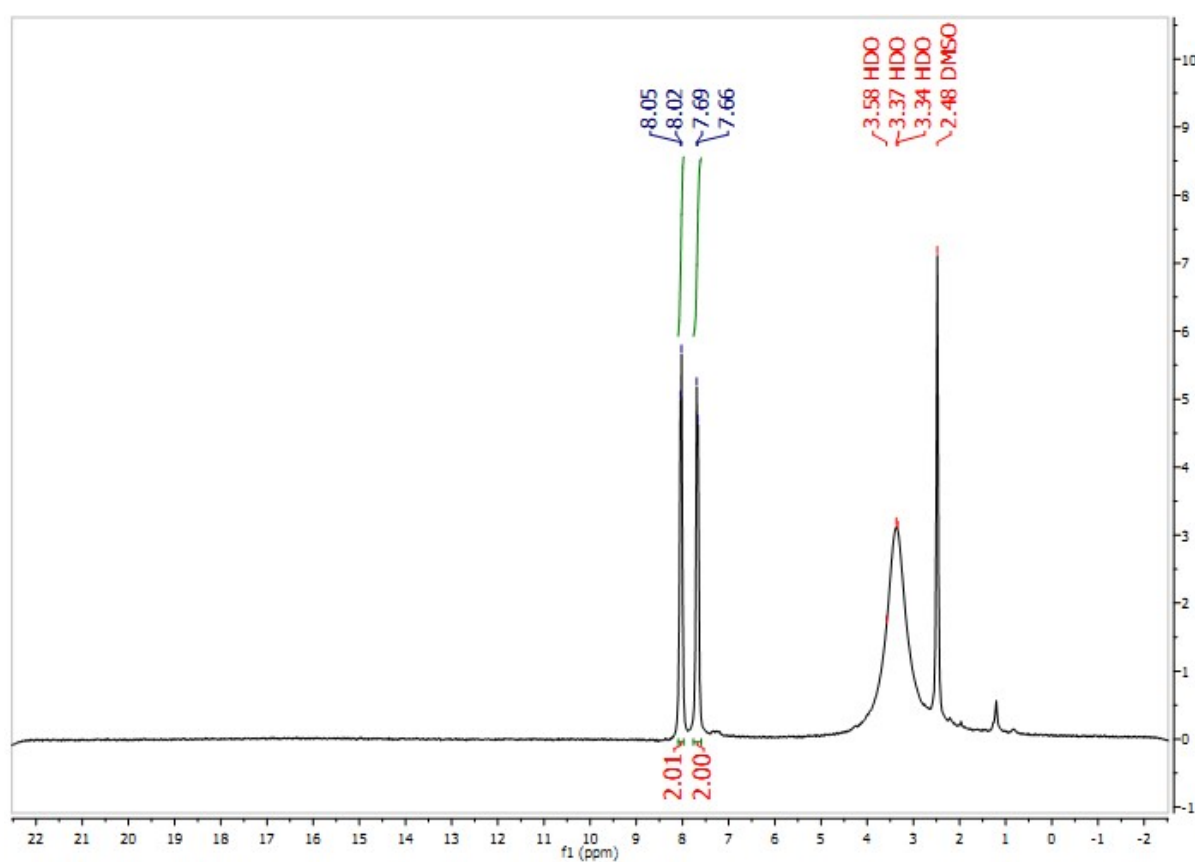


^1H NMR spectrum of 5-(2-chlorophenyl)-1H-tetrazole from [3+2] cycloaddition of NaN_3 with 2-chlorobenzonitrile



5-(4-chlorophenyl)-1H-tetrazole

^1H NMR (250 MHz, DMSO): $\delta_{\text{H}} = 8.03$ (d, $J = 6.65$ Hz, 2H), 7.67 (d, $J = 6.75$ Hz, 2H) ppm.



^1H NMR spectrum of 5-(4-chlorophenyl)-1H-tetrazole from [3+2] cycloaddition of NaN_3 with 4-chlorobenzonitrile