Supporting Information

Platinum Nanoparticles Wrapped in Carbon-Dot-Films as Oxygen Reduction Reaction Catalysts Prepared by Solution Plasma Sputtering

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Fig. S1. (a) Relationship between gas temperature and electron temperature; (b) Current–voltage (I–V) curves in one cycle during discharge. The linear dashed line described by Temp(gas) = Temp(electron) show the equilibrium state.



Fig. S2. (a-1) Solution photo after solution plasma sputtering; (b-1) Solution photo after 100,000 rpm ultracentrifugation; (a-2) and (b-2) Absorbance of (a-1) and (b-1), respectively.



Fig. S3. TEM image of pH = 8. (a) original TEM image; (b) inverse Fourier Transform (FFT) after removing spatial frequencies from carbons on FFT, (c) the measurement results of the plane spacing.



Fig. S4. X-ray photoelectron spectra of C 1s for the obtained catalyst with (a) Vulcan support; (b) pH = 4.4; (c) pH = 7; (d) pH = 8; (e) pH = 11; (f) 20 wt% commercial Pt/C obtained by the Gaussian deconvolutions.



Fig. S5. Output performance of the fuel cell using commercial 46.7 wt.% Pt/C as an anode with the pH = 8 sample and commercial 46% Pt/C as the cathode. (a) I–V and power output curves under H_2 – O_2 atmosphere in membrane electrode assembly (MEA); (b) cyclic voltammetry curves under H_2 – N_2 atmosphere in MEA.



Fig. S6. Benchmark of the oxygen reduction reaction performance of different catalysts

from references. 1-21

Table S1. Ink composition for the membrane electrode assembly measurements.

	Weight (mg)	H₂O (mg)	IPA (mg)	Nafion (mg)	Membrane
					thickness (μm)
Commercial	75	600	770	200	
Pt/C (46.7 wt%)	75	600	770	200	150 and 50
pH = 8 (Pt% =	30	243	310.7	135	350
8.4%)					

Table S2. Loading amount on electrodes and membrane electrode assemblymeasurement performances.

	Anode	Loading	Cathode	Loading	ECSA (m²/g)
		amount (µg)		amount (µg)	
Commercial Pt/C (46.7 wt%)	150 μm		50 μm		Courset ha
	Commercial	165	Commercial	33.0	calculated
	Pt/C (46 wt.%)		Pt/C (46 wt%)		
	150 μm 46%				
pH=8 (Pt% = 8.4%)	Commercial	162	350 μm pH = 8 Pt/C	20.4	223.20
	Pt/C				

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