

Supplementary Martial

Synergetic Efficiency: In Situ Growth of Novel 2D/2D Chemically Bonded $\text{Bi}_2\text{O}_3/\text{Cs}_3\text{Bi}_2\text{Br}_9$ S-Scheme Heterostructure for Improved Photocatalytic Performance and Stability

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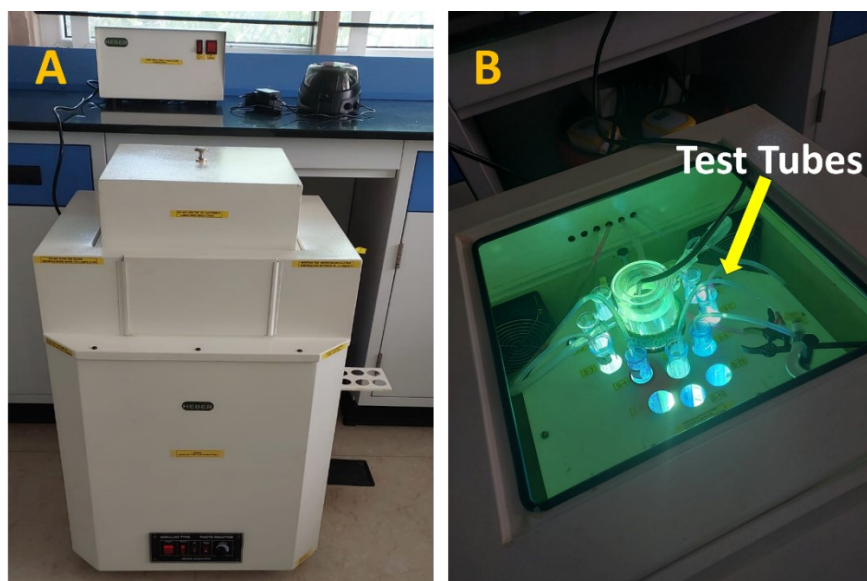


Fig. S1. A. Photocatalytic instrument, B. Top view of the instrument

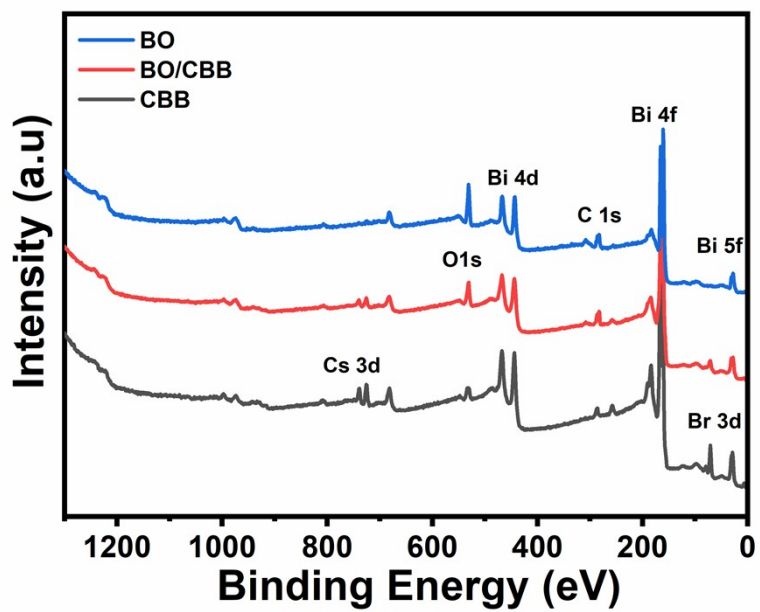


Fig. S2. Survey spectra of BO, CBB, and BO/CBB 28%

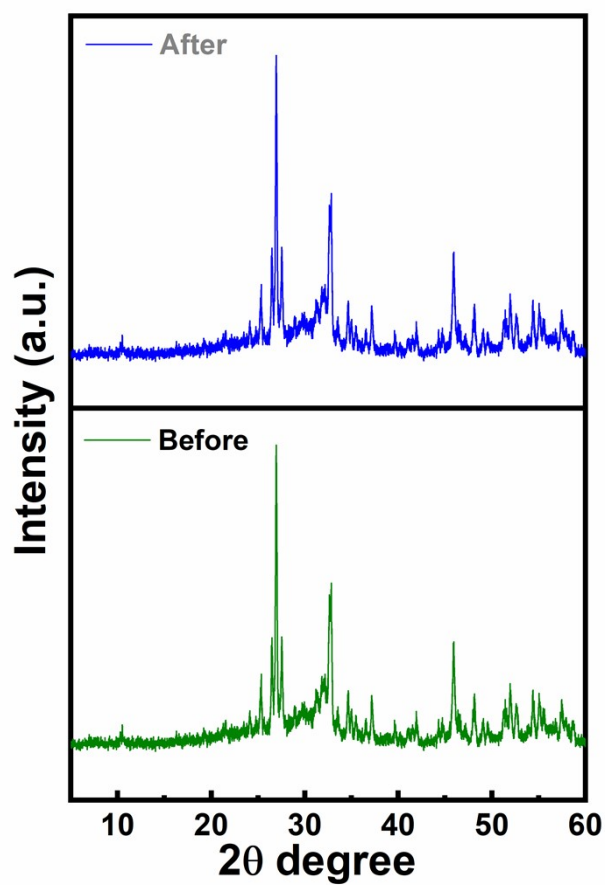


Fig. S3. XRD patterns of BO/CBB28% before and after four repeated cycles.

Table S1: Comparison of photocatalytic performance of various photocatalysts for degradation of dyes.

Photocatalyst	Irradiation	Pollutant	Time	Degradation	Ref.
CsSnBr ₃	Visible light	crystal violet dye	120 min	73.1%	[S1]
Cs ₃ Bi ₂ I ₉	Visible light	RhB	180 min	93%	[S2]
Cs ₃ Bi ₂ I ₉ -OA	Visible light	MB in water	60 min	62.1%	[S3]
Cs ₃ Bi ₂ I ₉ -OA	Visible light	MB in water	60 min	62.6%	[S3]
Cs ₃ Bi ₂ Br ₉ - OA/Ag ₂ S	Visible light	MB in water	60 min	40.13%	[S3]
Cs ₃ Bi ₂ Br ₉ -OA/TiO ₂	Visible light	MB in water	60 min	27.6%	[S3]
Cs ₃ Bi ₂ Br ₉	Visible light	MB in isopropanol	60 min	66.3%	[S3]
Cs ₃ Bi ₂ Br ₉ -OA NCs	Visible light	MB in isopropanol	60 min	58.8%	[S3]
Cu doped TiO ₂	Visible light	MB in water	300 min	81.22%	[S4]
Ni doped TiO ₂	Visible light	MB in water	300 min	71.18%	[S4]
Co doped TiO ₂	Visible light	MB in water	300 min	66.17%	[S4]
Fe doped TiO ₂	Visible light	MB in water	300 min	1.41%	[S4]
Mn doped TiO ₂	Visible light	MB in water	300 min	16.41%	[S4]
10%Bi ₂ O ₃ @TiO ₂	Visible light	MB in water	300 min	11.47%	[S4]
α - β Bi ₂ O ₃	sunlight	RhB in water	120 min	99.6%	[S5]
Bi ₂ O ₂ CO ₃	UV light	MB in water	300 min	64.19%	[S6]
Cs ₃ Bi ₂ Br ₉	Visible light	MO in water	90 min	92%	[S7]
Cs ₃ Bi ₂ Br ₉	Visible light	MB in water	90 min	80%	[S7]
Cs ₃ Bi ₂ Br ₉	Visible light	RhB in water	90 min	85%	[S7]
Bi ₂ O ₃	Visible light	MB in water	60 min	65%	This work
Bi ₂ O ₃ /Cs ₃ Bi ₂ Br ₉ 56%	Visible light	MB in water	60 min	89%	This work
Bi ₂ O ₃ /Cs ₃ Bi ₂ Br ₉ 28%	Visible light	MB in water	60 min	92%	This work

Effect of pH level: The impact of the pH level plays an important factor in achieving high-efficiency photocatalytic activity. The photocatalytic degradation of MB by the optimal BO/CBB 28% composite was carried out at different pH levels (3, 7, 9). Fig. S4 revealed that the studied sample did not show much difference in the degradation of MB with different pH levels. The photocatalytic activity in an acidic medium (pH = 3) achieved the best MB degradation efficiency among the three levels. Whereas the MB degradation in the alkaline medium (pH = 9) by the sample was lower than at pH = 7. However, this slight difference in the degradation of MB by the BO/CBB 28% composite might be attributed to the fact that the

changes in the charge surface of the photocatalyst at different pH media. This result is consistent with reported studies [S8, S9].

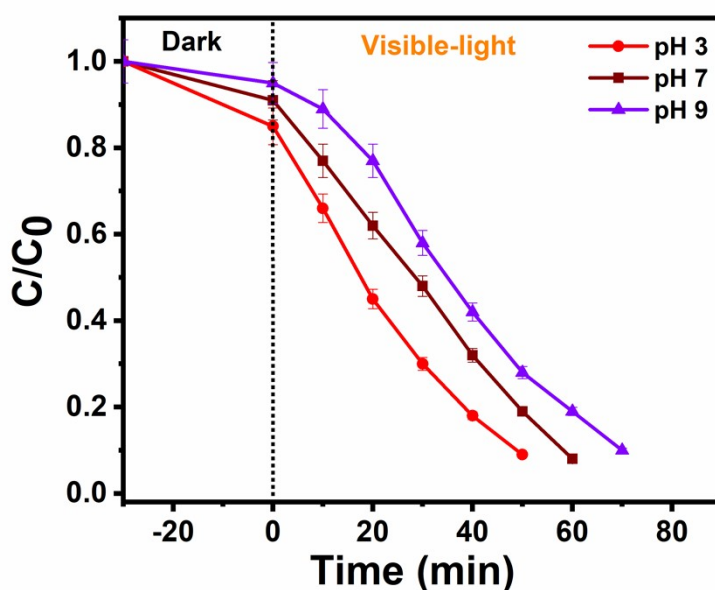


Fig. S4 pH level effect

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