

## Rationally constructing hollow N-doped carbon supported Ru catalyst for enhanced hydrogenation catalysis

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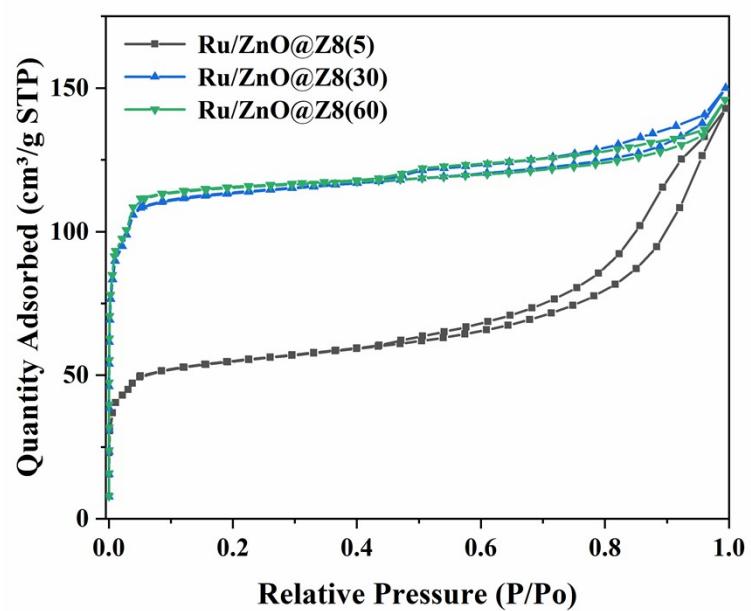


Figure S1 N<sub>2</sub> adsorption-desorption isotherms of Ru/ZnO@Z8(x) catalysts.

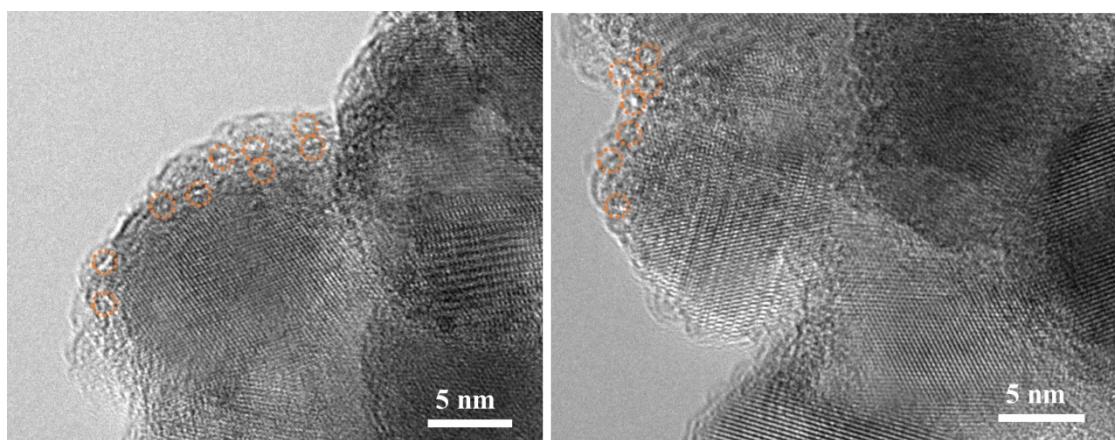


Figure S2 Representative HRTEM images of Ru/ZnO@Z8(5)-C catalyst. The circles indicate there are many white spots or even belts in the overlayer, suggesting the presence of porous structure.

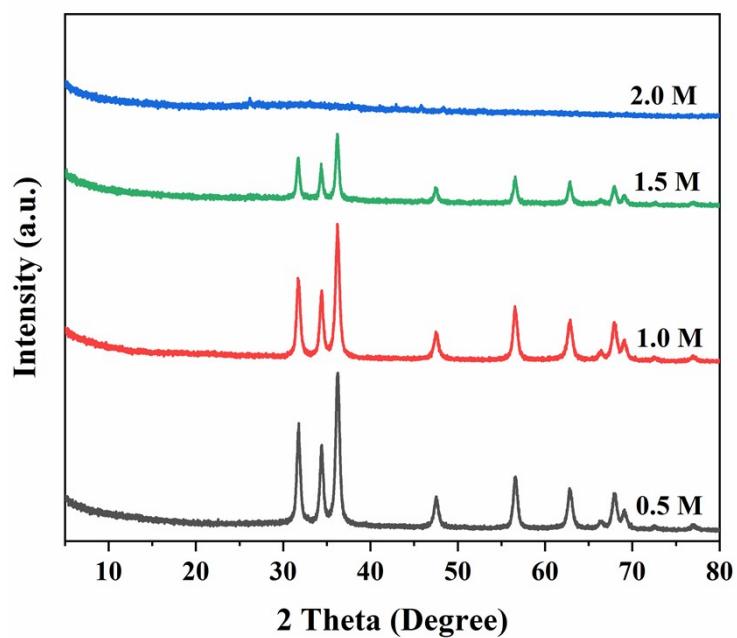


Figure S3 XRD patterns of Ru/ZnO@Z8(5)-C catalyst treated by NaOH solution with different concentration.

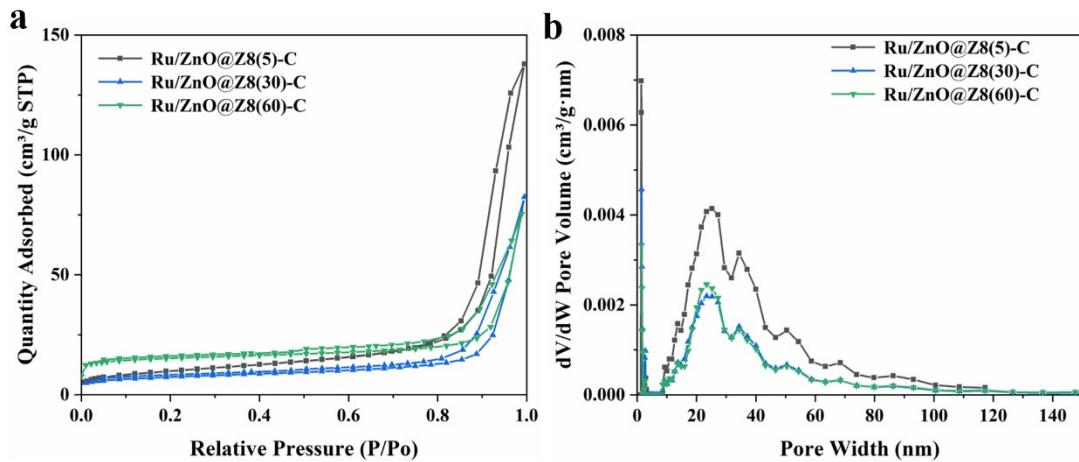


Figure S4  $\text{N}_2$  adsorption-desorption isotherms of  $\text{Ru/ZnO@Z8(x)-C}$  catalysts (a), the corresponding pore size distribution (b).

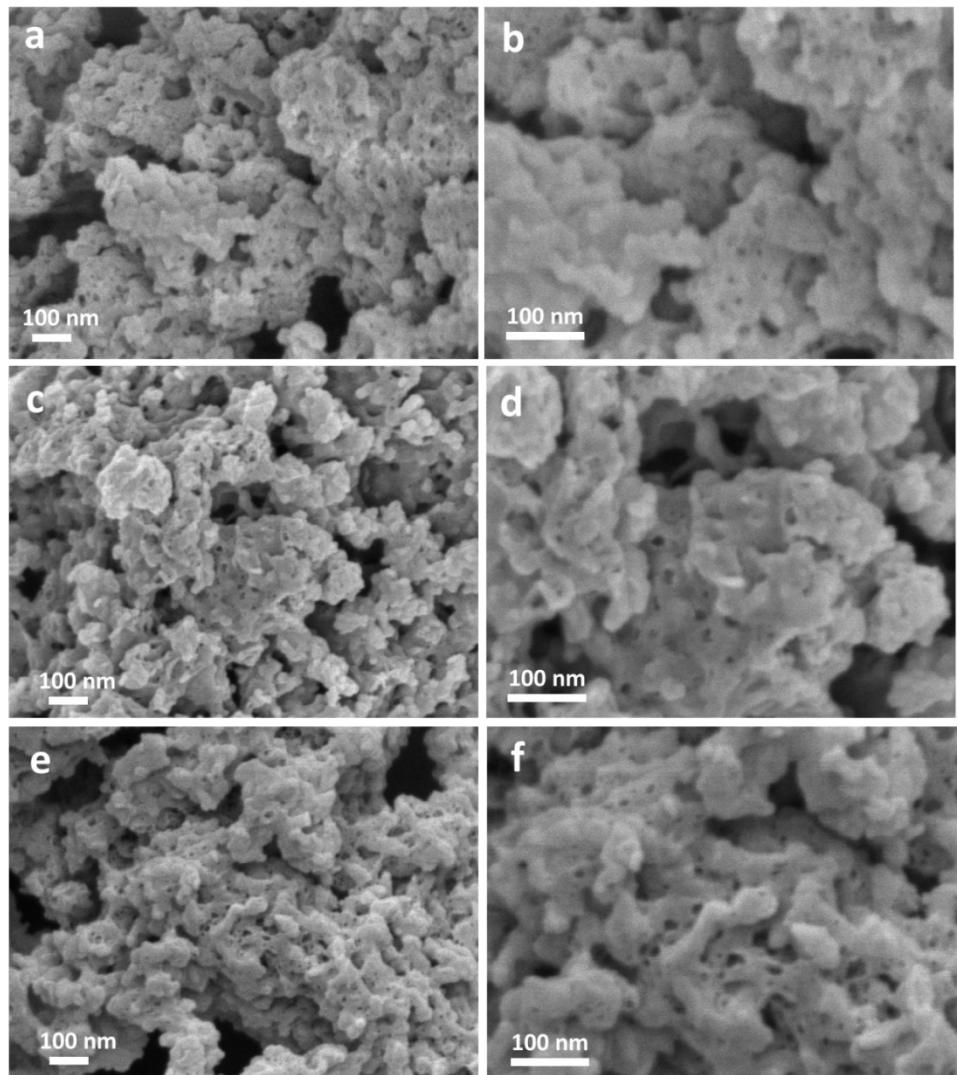


Figure S5 SEM images of Ru/H(5)-NC (a, b), Ru/H(30)-NC (c, d), Ru/H(60)-NC (e, f) catalysts.

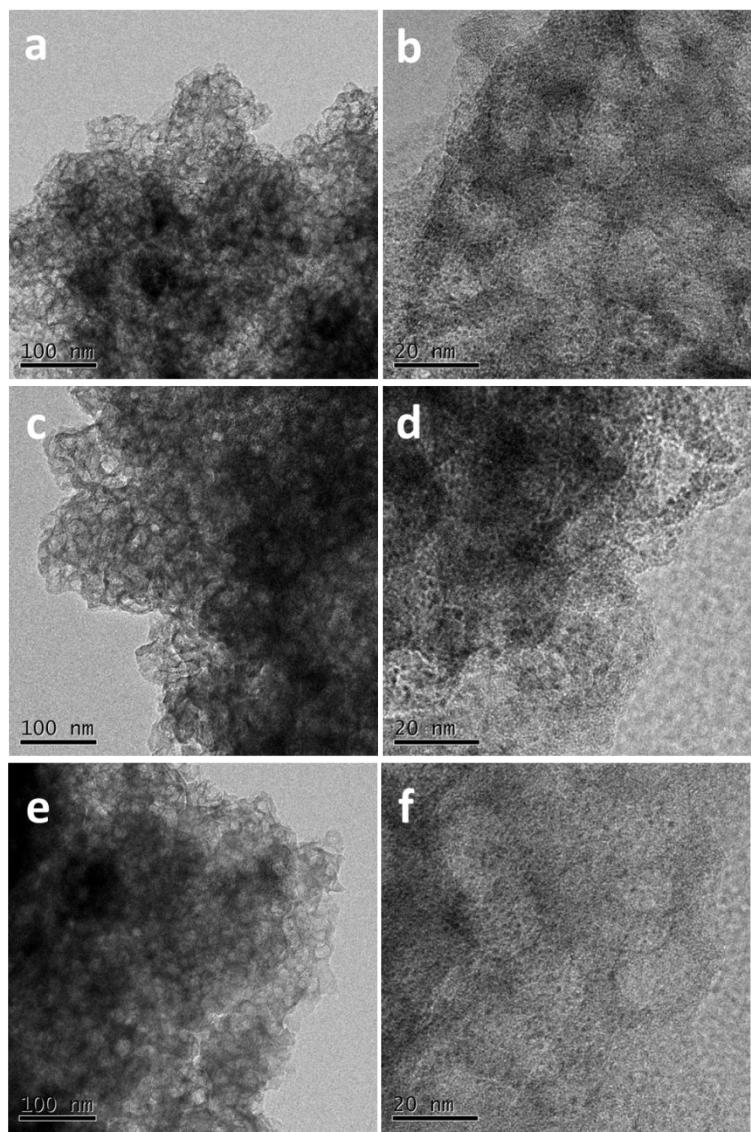


Figure S6 HRTEM images of Ru/H(5)-NC (a, b), Ru/H(30)-NC (c, d) and Ru/H(60)-NC (e, f) catalysts.

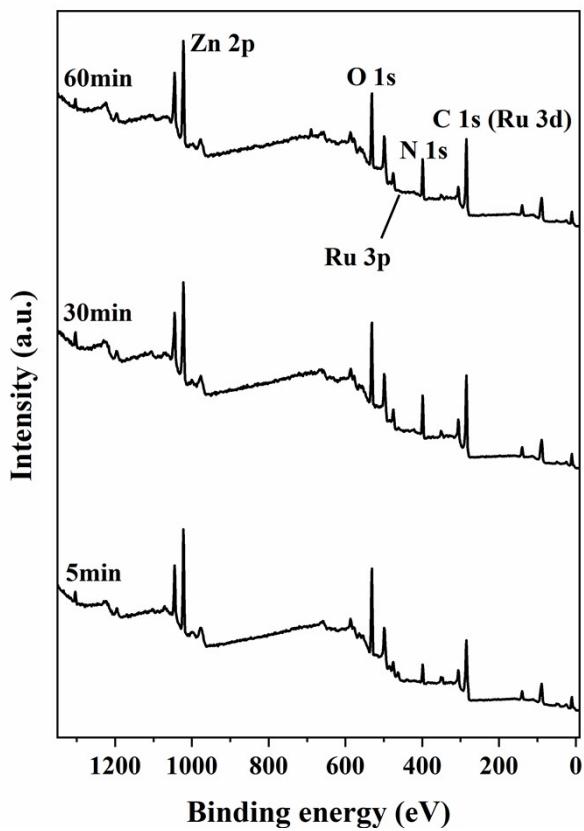


Figure S7 The full spectra of Ru/H(x)-NC catalysts by XPS analysis.

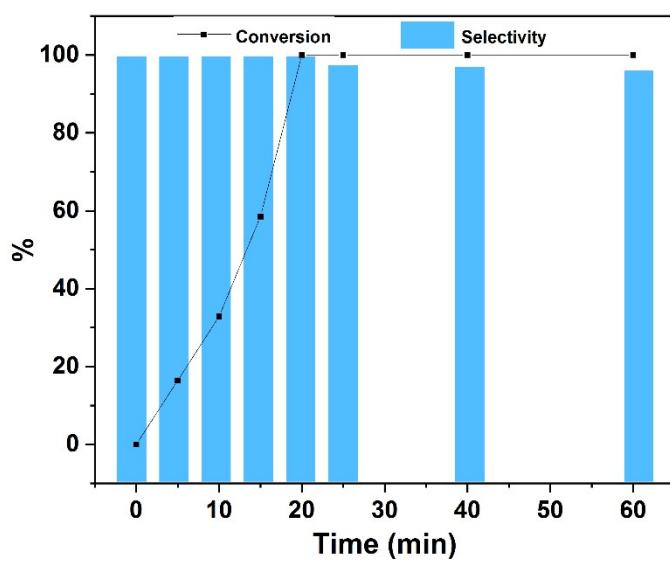


Figure S8 Conversion (Selectivity)-time diagram of Ru/H(5)-NC catalyst.

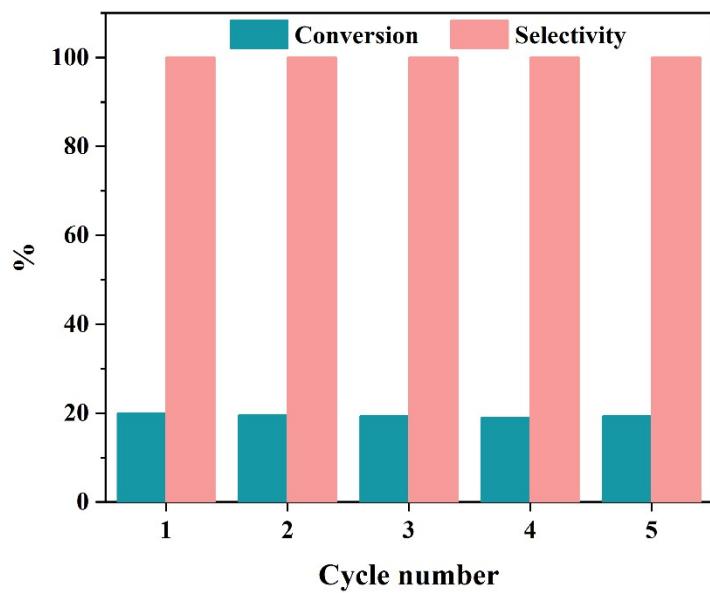


Figure S9 Stability test with Ru/H(5)-NC catalyst under low conversion.

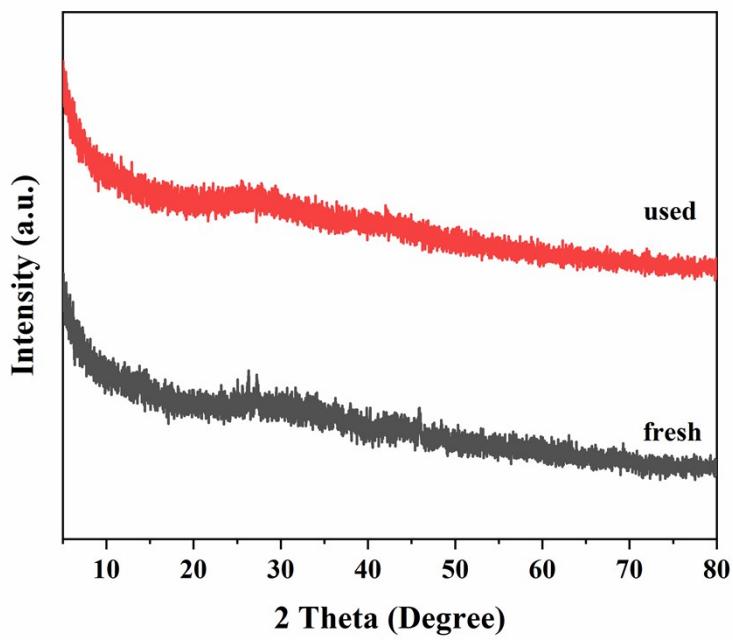


Figure S10 XRD patterns of Ru/H(5)-NC catalyst before and after hydrogenation reaction.

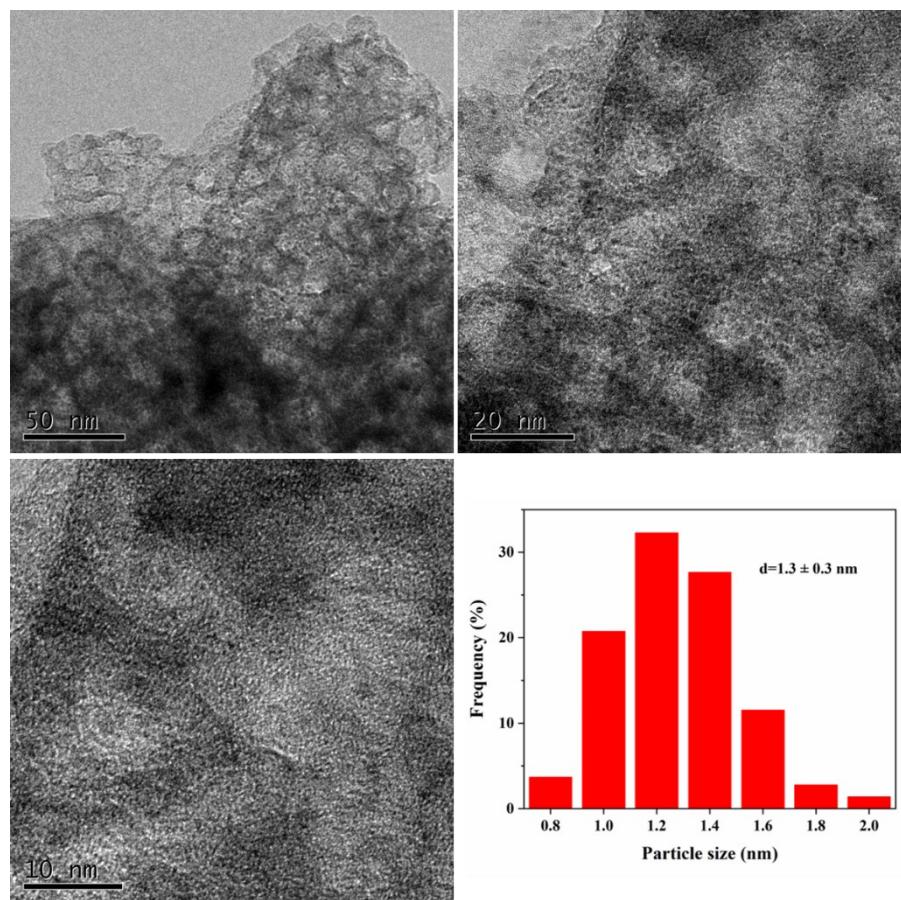


Figure S11 HRTEM images of Ru/H(5)-NC catalyst after cyclic experiments.

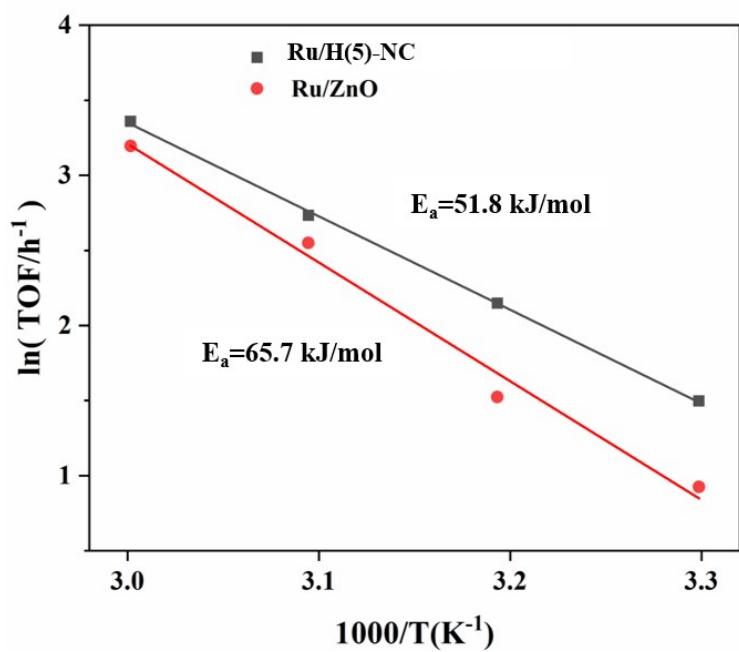


Figure S12 Activation energy tests for Ru/H(5)-NC and Ru/ZnO catalysts.

Table S1 The BET surface area and pore volume results for different catalysts.

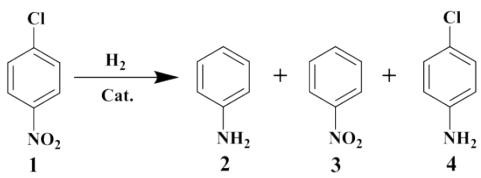
Catalyst	BET Surface Area(m <sup>2</sup> /g)	Pore Volume (cm <sup>3</sup> /g)
Ru/ZnO@Z8(5)	173	0.19
Ru/ZnO@Z8(30)	347	0.21
Ru/ZnO@Z8(60)	349	0.20
Ru/ZnO@Z8(5)-C	30	0.082
Ru/ZnO@Z8(30)-C	17	0.043
Ru/ZnO@Z8(60)-C	14	0.039
Ru/H(5)-NC	271	0.60
Ru/H(30)-NC	216	0.51
Ru/H(60)-NC	197	0.45

Table S2 TEM and XRD analyses of the cavity size in Ru/H(x)-NC catalyst

Catalyst	$d_1^a$ (nm)	$d_2^b$ (nm)
Ru/H(5)-NC	$17.4 \pm 2.0$	17.8
Ru/H(30)-NC	$16.1 \pm 2.0$	16.3
Ru/H(60)-NC	$15.4 \pm 1.9$	15.7

<sup>a</sup> The cavity size of the catalyst obtained by TEM characterization was calculated by measuring 200 hollow structure, <sup>b</sup> the particle size of the etched residual ZnO was obtained from XRD data by Scherrer's formula.

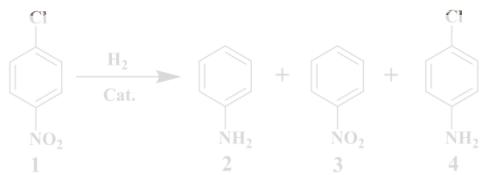
Table S3 The effect of solvent for the hydrogenation reaction with Ru/H(5)-NC catalyst.



Solvent	Conv. (%)	Sel. (%)		
		2	3	4
toluene	98.7	0	0	>99
ethanol	97.2	8.7	0	91.3
methanol	98.2	6.3	0	93.7
tetrahydrofuran	31.3	0	0	>99
isopropanol	40.6	0	0	>99

Reaction condition: 50 °C, 0.5 MPa H<sub>2</sub>, 0.5 mmol p-chloronitrobenzene, 0.25 mmol 1,3,5-trimethylbenzene (internal standard), 5 mL solvent, 19 min.

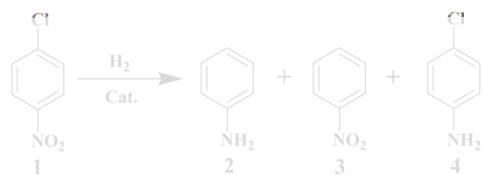
Table S4 The effect of temperature for the hydrogenation reaction with Ru/H(5)-NC catalyst.



T (°C)	Conv. (%)	Sel. (%)		
		2	3	4
30	72.8	0	0	>99
40	85.2	0	0	>99
50	98.2	0	0	>99
60 <sup>a</sup>	98.5	5.3	0	94.7

Reaction condition: 0.5 MPa H<sub>2</sub>, 0.5 mmol p-chloronitrobenzene, 0.25 mmol 1,3,5-trimethylbenzene (internal standard), 5 mL toluene, 19 min. <sup>a</sup> 15 min.

Table S5 The effect of pressure for the hydrogenation reaction with Ru/H(5)-NC catalyst.



P (MPa)	Conv. (%)	Sel. (%)		
		2	3	4
0.3	83.8	0	0	>99
0.5	98.2	0	0	>99
0.7	100	0	0	>99
1 <sup>a</sup>	100	0	0	>99

Reaction condition: 50 °C, 0.5 mmol p-chloronitrobenzene, 0.25 mmol 1,3,5-trimethylbenzene (internal standard), 5 mL toluene, 19 min. <sup>a</sup> 15 min.