

Supporting Information (SI)

Flexible-Design in Controllable Synthesis of Ru Catalyst toward Enzymatic and Electrochemical Hydrogen Peroxides Performance

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Xi Hu and Yujun Yang contributed equally in this work.

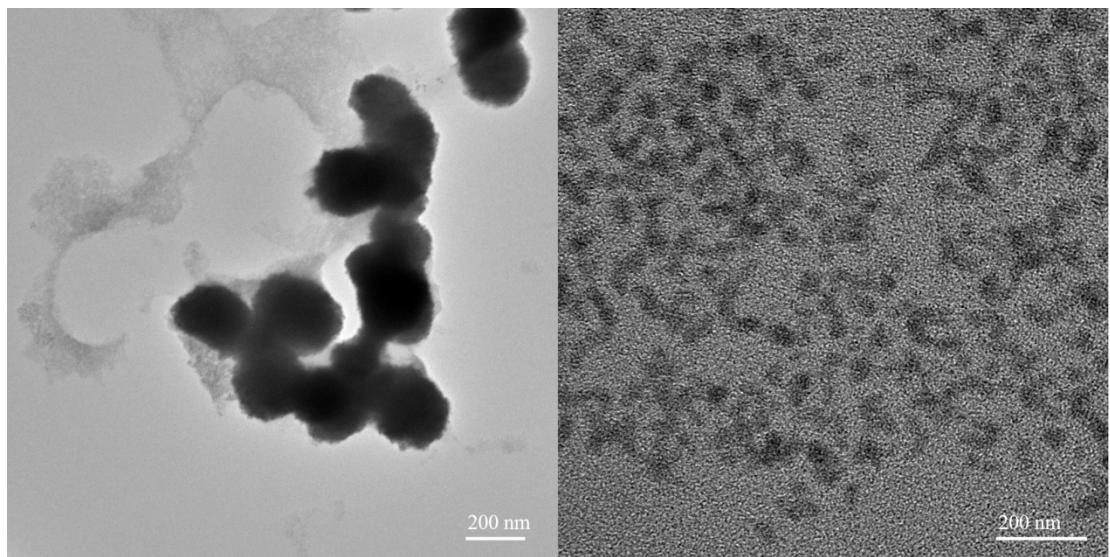


Figure S1. TEM images of as synthesized Ru NPs at room temperature for one hour.

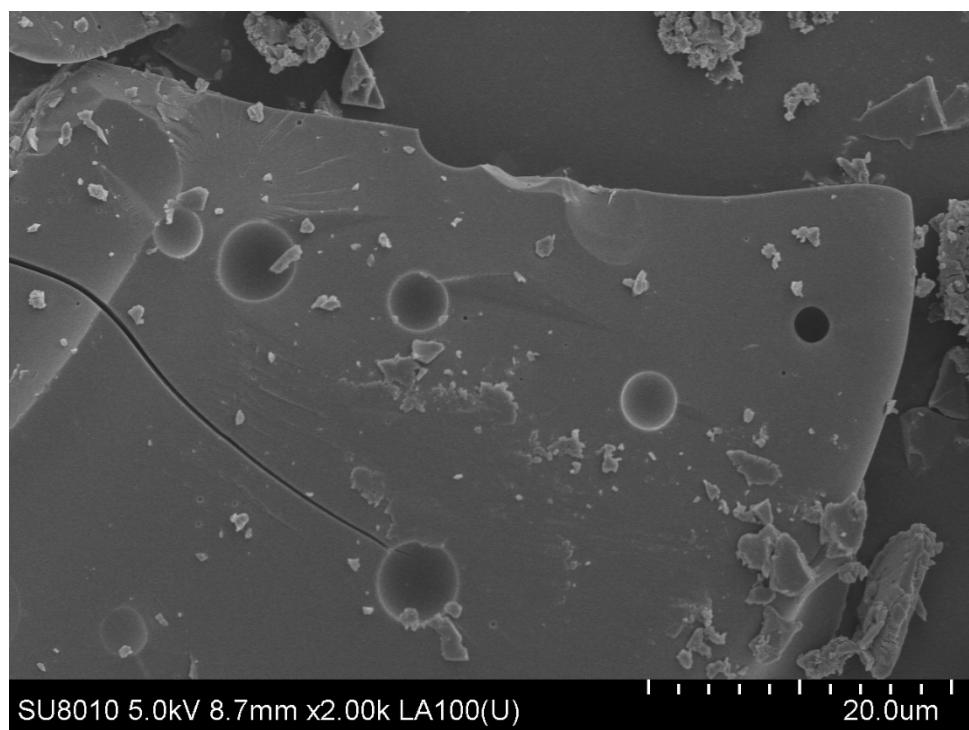


Figure S2. SEM image of as synthesized Ru NPs.

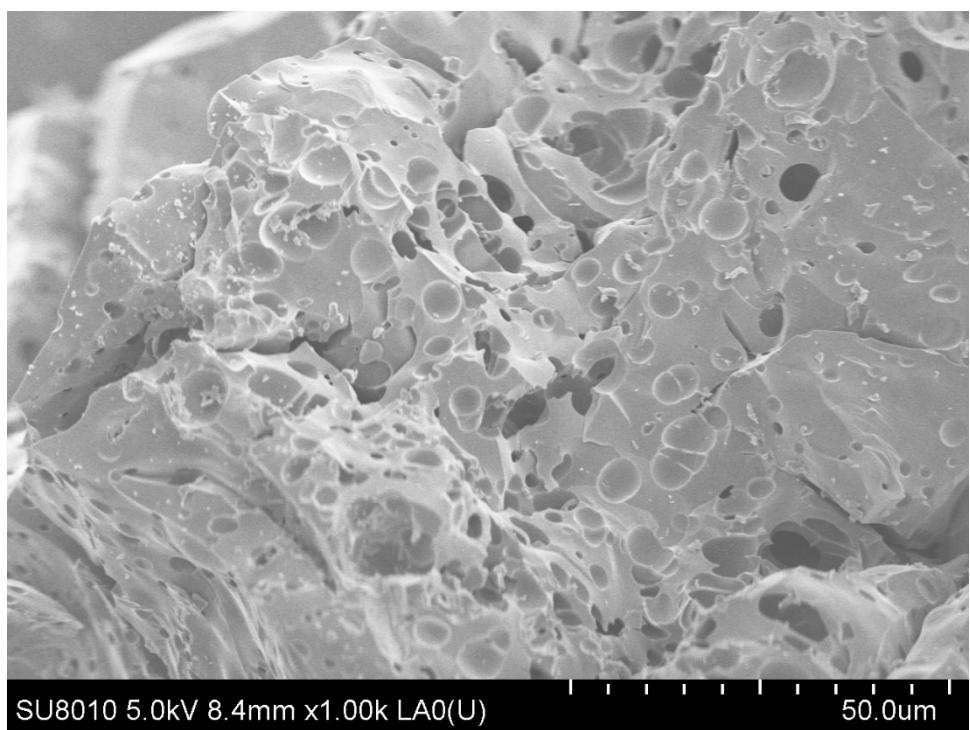


Figure S3. SEM image of annealed Ru NPs.

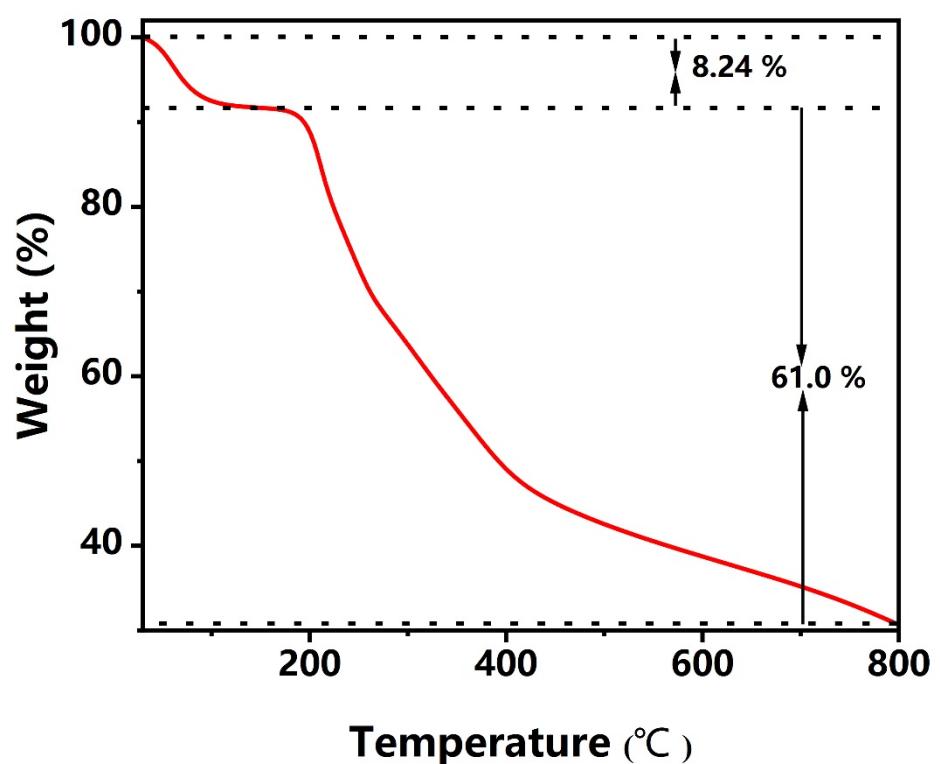


Figure S4. TGA curves of Ru NPs.

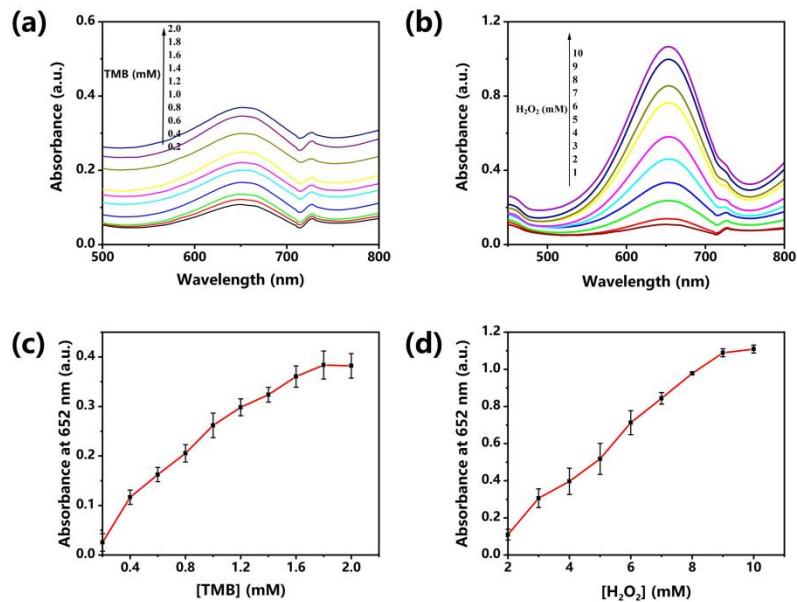


Figure S5. HRP-like activity of Ru NPs during oxidation of TMB. (a, c) UV-vis absorption spectra and (b, d) absorbance at 652 nm of Ru NPs ($200 \mu\text{g mL}^{-1}$) incubated at different concentrations of TMB and H₂O₂ for 10 mins.

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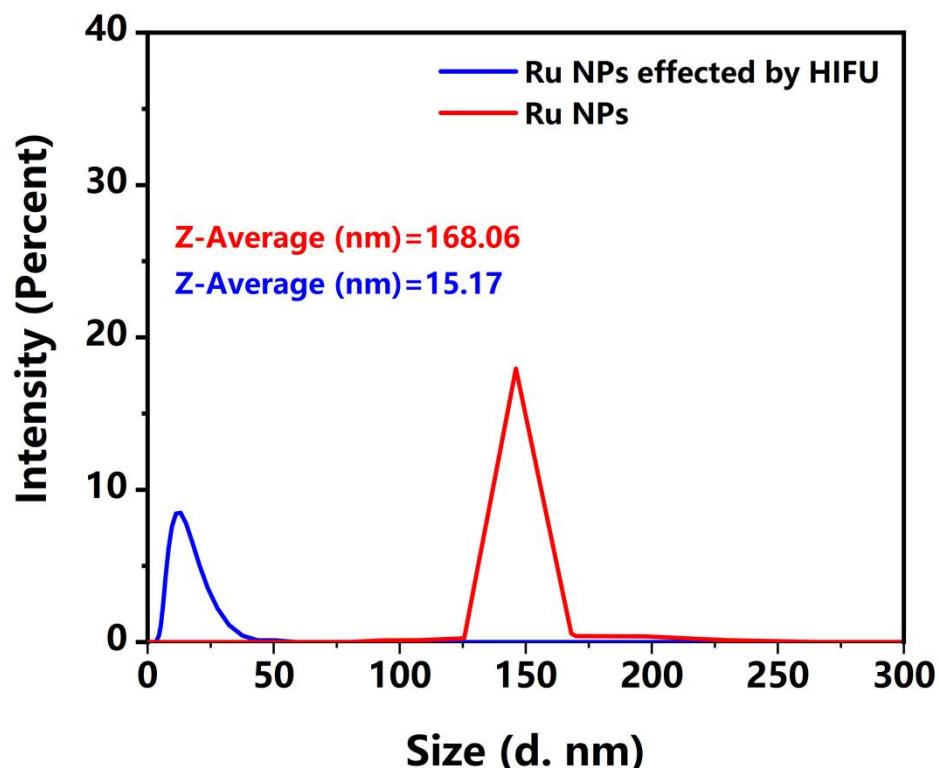


Figure S6. DLS data of Ru NPs and Ru NPs effected by HIFU (10 MPa, 4000 W).

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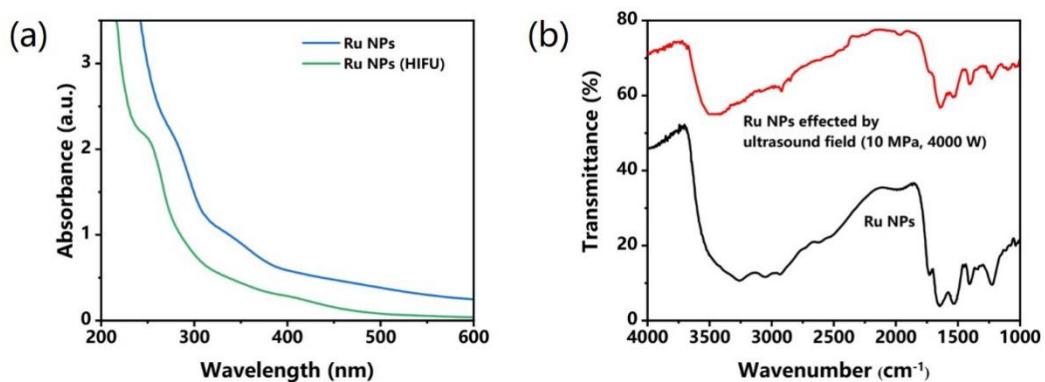


Figure S7. (a, b) UV-vis absorbance spectra and FTIR spectra of Ru NPs effected by ultrasound field (10 MPa, 4000 W).

Table S1. Comparison of the kinetic parameters of different catalysts.

Catalyst	Substrate	K _m (mM)	V _{max} (×10 ⁻⁷ M s ⁻¹)	Ref.
HRP	TMB	0.434	1.00	1
	H ₂ O ₂	3.70	0.871	
Ir NPs	TMB	0.33	12.7	2
	H ₂ O ₂	44.28	13.7	
Pt NPs	TMB	0.86	17.4	2
	H ₂ O ₂	7.92	53.5	
Pt nanoplates	TMB	0.17	10.1	3
	H ₂ O ₂	82.7	17.7	
RuO ₂ NPs	TMB	0.24	1.9	4
	H ₂ O ₂	212	2.05	
Ru NPs	TMB	0.234	4.95	5
	H ₂ O ₂	2.206	34.96	
Ru frames	TMB	0.0603	1.34	6
	H ₂ O ₂	318	0.741	
Co ₃ O ₄	TMB	0.36	2.49	7
	H ₂ O ₂	6.53	130	
Fe ₃ O ₄	TMB	0.098	0.344	1
	H ₂ O ₂	154	0.978	

Table S2. Comparison of the detection H₂O₂ of different catalysts.

Probe	Linear range(μM)	LOD (μM)	Ref.
Ru NPs	500-100000	1.45	This work
Ag/graphene oxide	100-20000	1.9	8
Pt/carbon nanotube	5-25000	1.5	9
FePPOPs-SO ₃ H	50-1800	26.7	10
Au/Co ₃ O ₄ -CeO _x	10-1000	5.29	11
PtPd-Fe ₃ O ₄ /C	0.02-67	0.005	12

References

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