Supporting Information

Pyrolysis of Naphthol Functionalized Polytriarylamine for Efficient Sodium-Ion Storage

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Table S1. Surface properties of PDNOHs.

PDNOH	S_{BET} [a] (m^2/g)	V_{total} [b] (cm^3/g)	Pore size [c] (nm)	S_{micro} [d] (m^2/g)	V_{micro} [e] (cm^3/g)
@650 °C	298.3	0.109	1.583	321.1	0.100
@800 °C	112.5	0.034	1.355	102.0	0.033

[a]BET surface area, [b]Total pore volume, [c]Average pore diameter, [d]Surface are of micropore calculated by t-Plot analysis, [e]Pore volume of micropore calculate by t-Plot analysis.

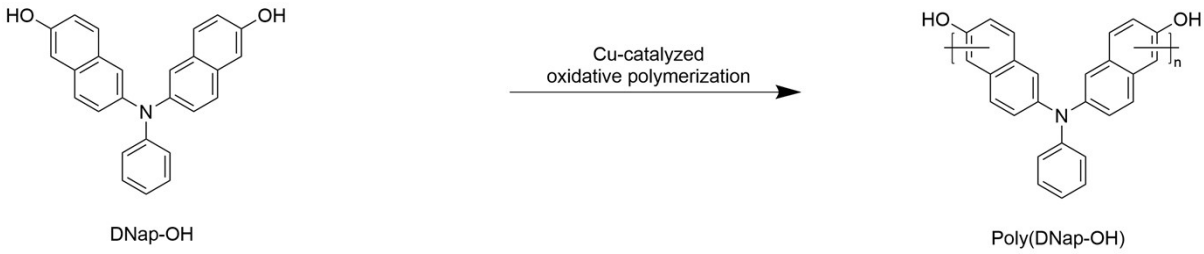

Figure S1. Polymerization scheme of Poly(DNap-OH).

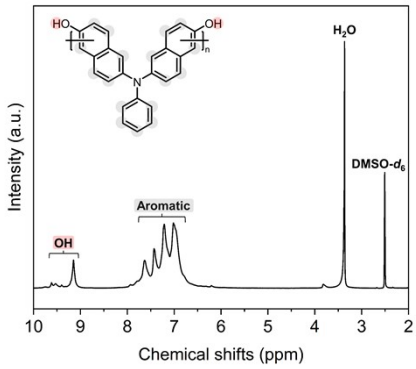

Figure S2. ^{1}H-NMR spectrum of Poly(DNap-OH) in DMSO-d_6.

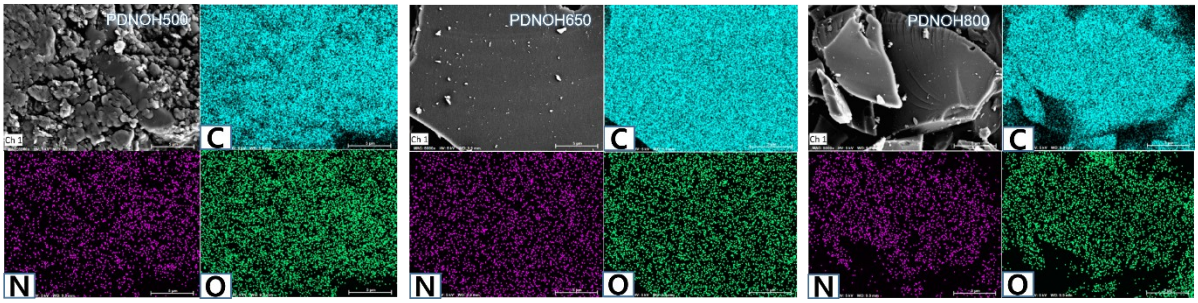

Figure S3. SEM-EDS mapping of PDNOHs.

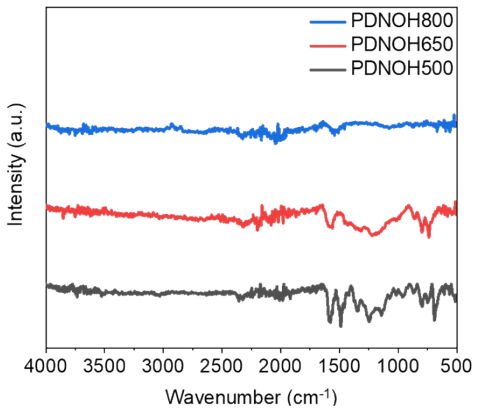

Figure S4. FT-IR spectra of PDNOHs.

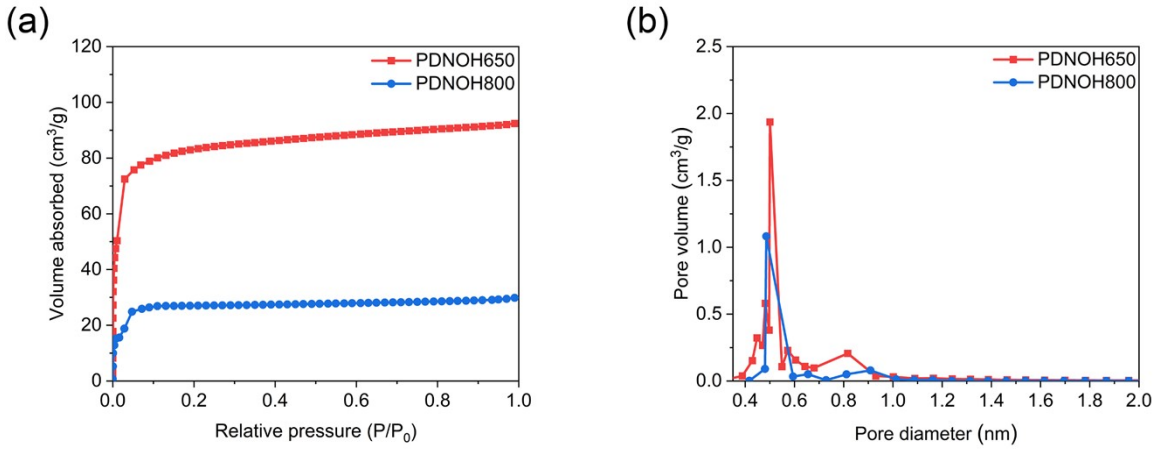

Figure S5. (a) Ar adsorption isotherms, (b) Pore-size distribution of PDNOHs.

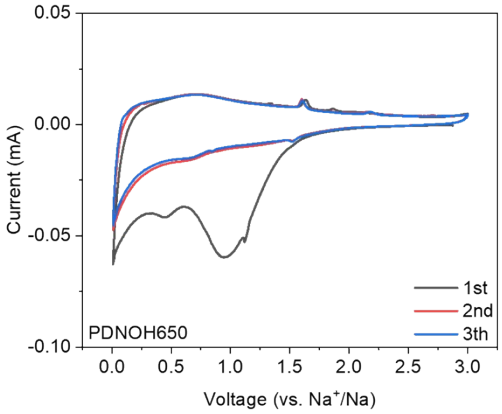

Figure S6. CV results of PDNOH-650.

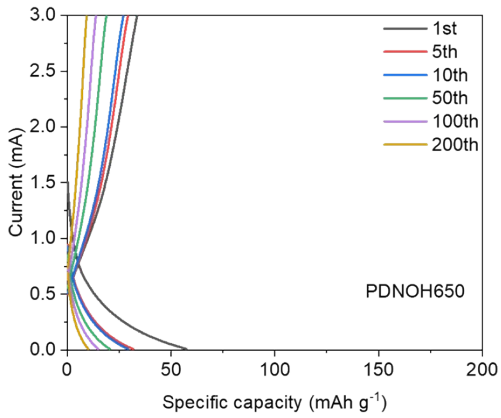

Figure S7. Charge/discharge curves of PDNOH-650.

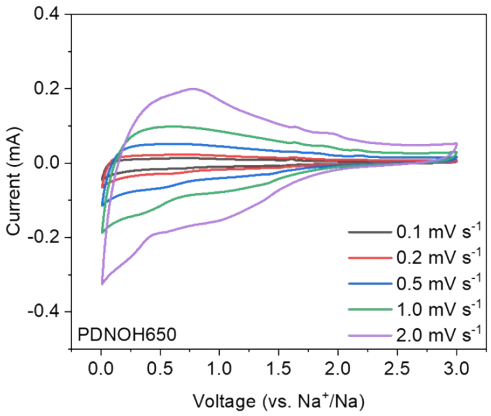

Figure S8. CV results of PDNOH-650 at various sweep rates.

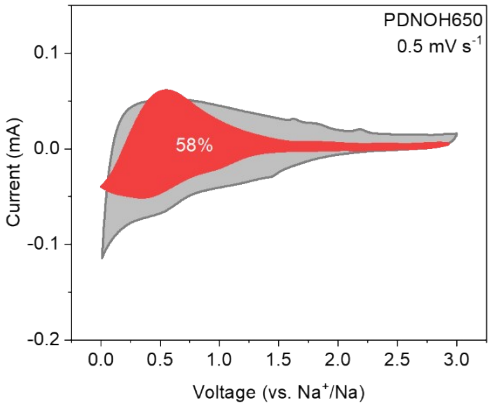

Figure S9. Contribution of surface-induced Na^+ storage in PDNOH-650 (0.5 mV s^{-1}).

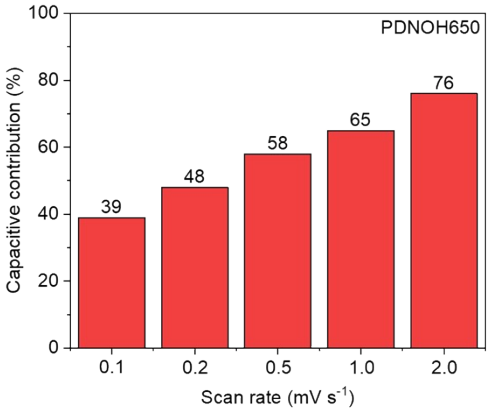

Figure S10. Contribution of surface-induced Na^+ storage at various sweep rates in PDNOH-650.

References

[1] T. Kim, T. Lee, Y. R. Yoon, W. S. Heo, S. Chae, J. W. Kim, B.-K. Kim, S. Y. Kim, J. Lee, J. H. Lee, Small, 2024, 2400333.