

# Construction and characterization of an inorganic-organic hybrid copper(I) iodide coordination polymer with the semiconducting luminescence

Hui Yang,<sup>abcd</sup> Xiaofei Kuang,<sup>\*abcd</sup> Ying-Hao Mi,<sup>ac</sup> Ming-Ming Wang,<sup>ac</sup> Yuqing Zhao,<sup>ad</sup> Fulin Lin<sup>ac</sup> and Can-Zhong Lu<sup>\*abcd</sup>

*a. State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian 350002, China.*

*E-mail: xfkuang@fjirsm.ac.cn, czlu@fjirsm.ac.cn*

*b. College of Chemistry and Materials Science, Fujian Normal University, Fuzhou 350007, China.*

*c. Xiamen Key Laboratory of Rare Earth Photoelectric Functional Materials, Xiamen Institute of Rare Earth Materials, Haixi Institutes, Chinese Academy of Sciences, Xiamen 361021, China.*

*d. Fujian College, University of Chinese Academy of Sciences, Fuzhou, Fujian 350002, China.*

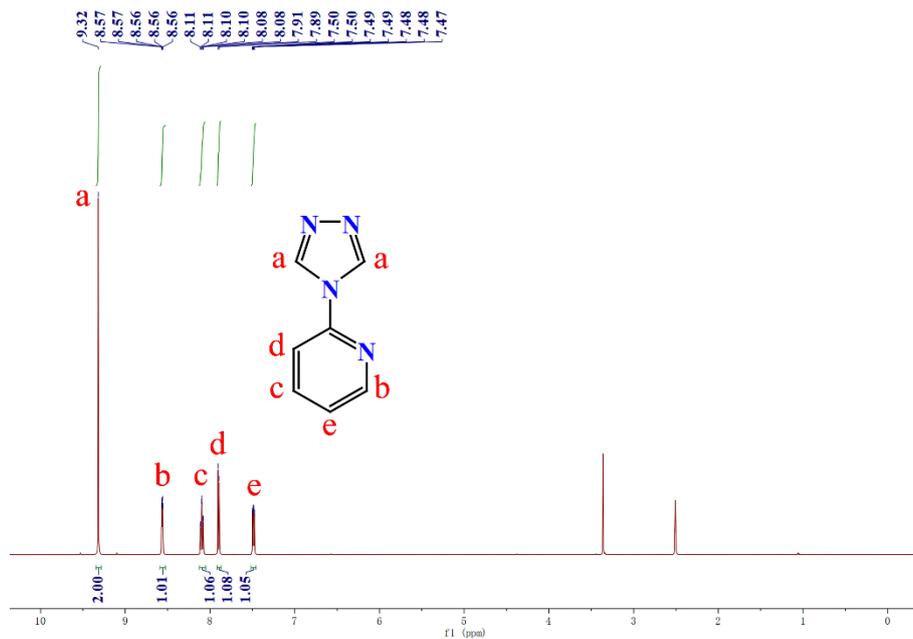


Fig. S1. The  $^1\text{H}$  NMR spectrum of Pytz in  $\text{DMSO-}d_6$ .

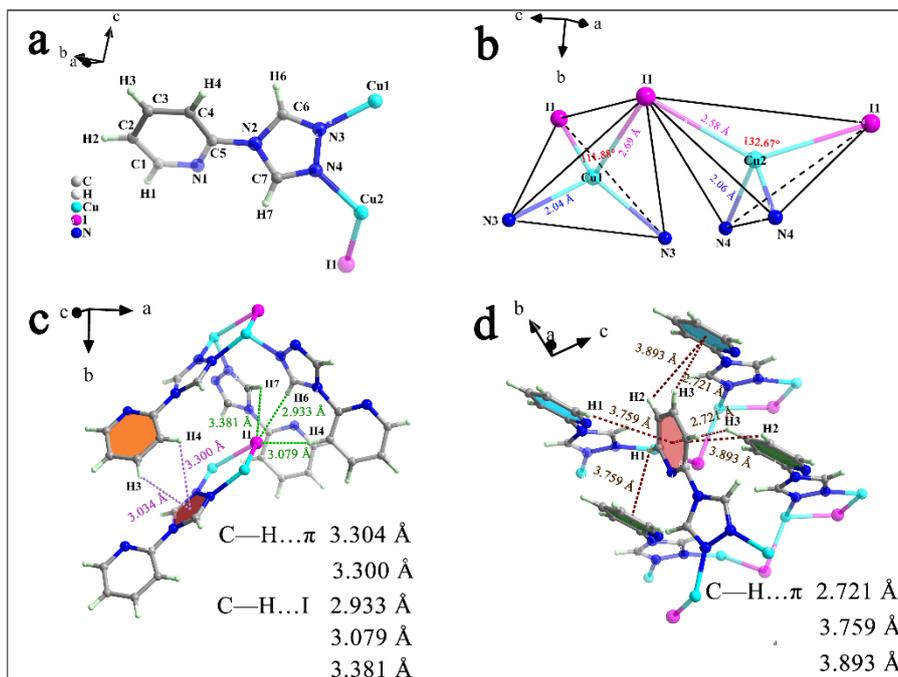


Fig. S2. (a) The asymmetric unit of  $\text{CuI-Pytz}$ . (b) The distorted tetrahedrons of  $\text{CuI-Pytz}$ . (c) and (d) The illustration of the  $\text{C-H}\dots\pi$  and  $\text{C-H}\dots\text{I}$  interactions stabilizing the supramolecular structure.

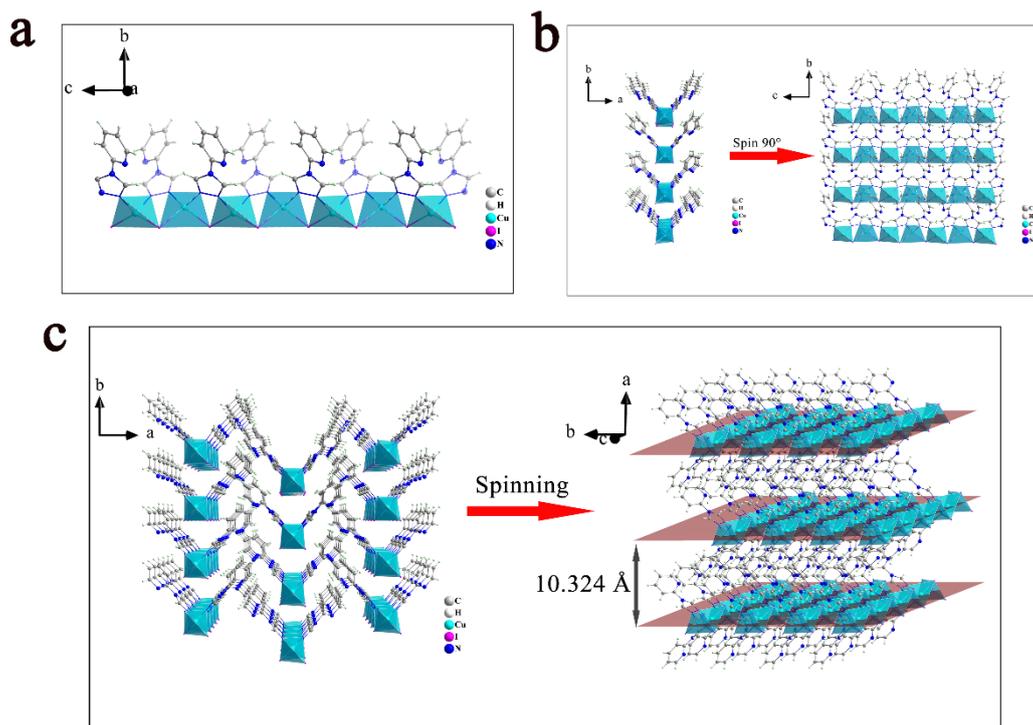


Fig. S3. (a) 1D chain of CuI-Pytz. (b) 2D layer of CuI-Pytz. (c) 3D structure of CuI-Pytz.

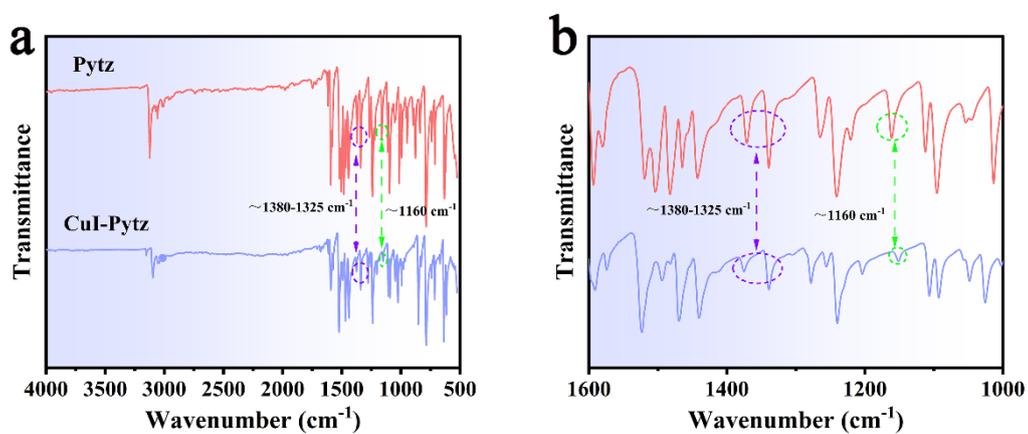


Fig. S4. (a) FT-IR spectra of Pytz and CuI-Pytz. (b) The fine structure of IR spectra.

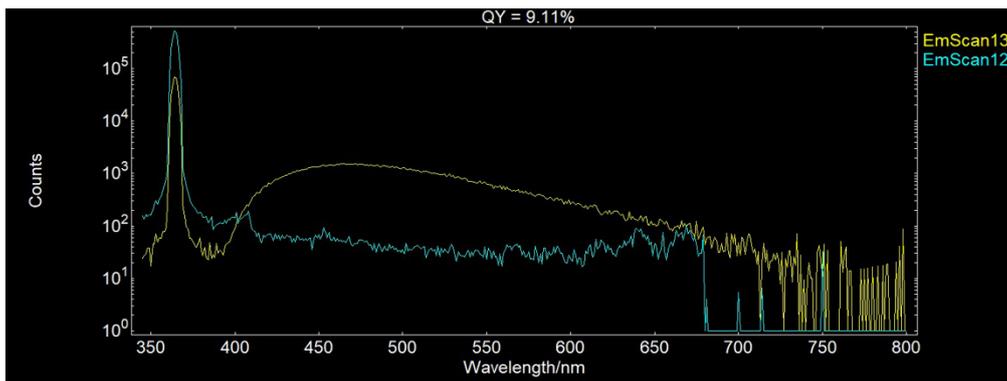


Fig. S5. The PL quantum yield ( $\Phi_{\text{PL}}$ ) of CuI-Pytz under 365 nm excitation light.

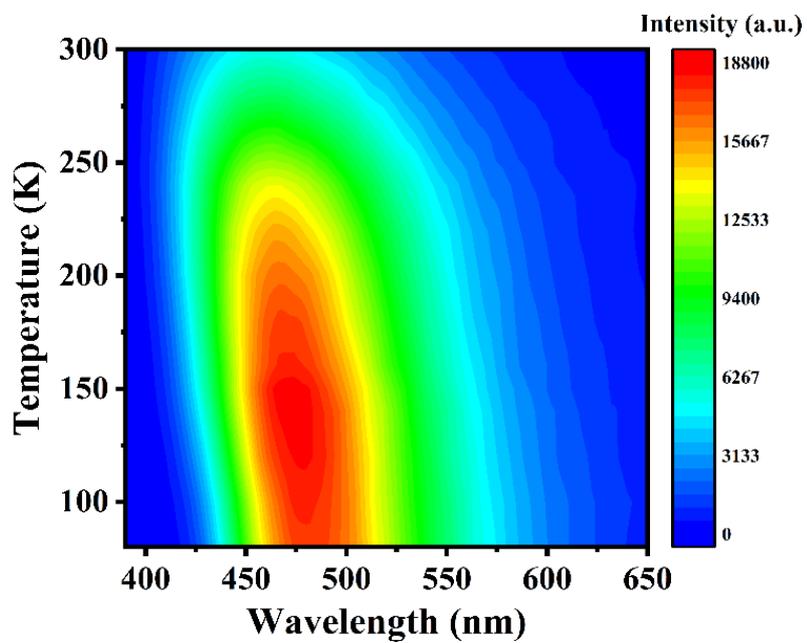


Fig. S6. Temperature and PL emission 2D contour mapping.

Table S1. Selected bond lengths (Å) and bond angles (°) for **CuI-Pytz**.

Cu1—I1 <sup>1</sup>	2.6879(4)	N3 <sup>3</sup> —Cu1—I1 <sup>1</sup>	117.04(8)
Cu1—I1 <sup>2</sup>	2.6879(4)	N3—Cu1—I1 <sup>1</sup>	99.51(8)
Cu1—N3	2.042(3)	N3—Cu1—I1 <sup>2</sup>	117.04(8)
Cu1—N3 <sup>3</sup>	2.041(3)	N3 <sup>3</sup> —Cu1—N3	112.81(16)
Cu2—I1	2.5809(3)	I1 <sup>1</sup> —Cu2—I1	132.67(3)
Cu2—I1 <sup>2</sup>	2.5809(3)	N4 <sup>1</sup> —Cu2—I1	100.09(8)
Cu2—N4	2.059(3)	N4 <sup>1</sup> —Cu2—I1 <sup>1</sup>	109.72(8)
Cu2—N4 <sup>2</sup>	2.059(3)	N4—Cu2—I1	109.72(8)
I1 <sup>1</sup> —Cu1—I1 <sup>2</sup>	111.88(2)	N4—Cu2—I1 <sup>1</sup>	100.09(8)
N3 <sup>3</sup> —Cu1—I1 <sup>2</sup>	99.51(8)	N4—Cu2—N4 <sup>1</sup>	100.66(16)

Symmetry codes: <sup>1</sup>+X, +Y, 1+Z; <sup>2</sup>1-X, +Y, 1-Z; <sup>3</sup>1-X, +Y, 2-Z (Bond lengths).

<sup>1</sup>1-X, +Y, 1-Z; <sup>2</sup>+X, +Y, 1+Z; <sup>3</sup>1-X, +Y, 2-Z; <sup>4</sup>+X, +Y, -1+Z (Bond angles).

Table S2. Temperature-dependent lifetime values of **CuI-Pytz**.

Temp (K)	A <sub>1</sub>	τ <sub>1</sub> (μs)	A <sub>2</sub>	τ <sub>2</sub> (μs)	τ <sub>ave</sub> (μs)
80	0.684 (50%)	24.500	0.684 (50%)	29.945	27.495
100	0.699 (50%)	21.439	0.699 (50%)	26.203	24.059
120	0.741 (50%)	17.902	0.741 (50%)	21.880	20.089
140	0.864 (50%)	14.059	0.864 (50%)	17.183	15.777
160	0.982 (50%)	11.054	0.982 (50%)	13.510	12.405
180	0.800 (50%)	9.880	0.800 (50%)	12.075	11.087
200	0.290 (30.8%)	3.033	0.651 (69.2%)	7.382	6.709
220	0.340 (36.4%)	1.972	0.594 (63.6%)	5.587	4.980
240	0.386 (40.8%)	1.603	0.561 (59.2%)	4.431	3.868
260	0.403 (43.2%)	1.058	0.529 (56.8%)	3.219	2.786
280	0.367 (39.2%)	0.623	0.569 (60.8%)	2.141	1.901
300	0.379 (41.2%)	0.381	0.541 (58.8%)	1.310	1.153