

Supplementary Information

Improvement of chlorine evolution activity of $\text{SnO}_x@\text{IrO}_2\text{-Ta}_2\text{O}_5$ electrode and its application in electrolysis of extremely dilute chlorine-containing solution

Lin Liu, Ming Wang, Yiping Zhan, Zhiqian Lin, Shenglei Xiong, Hailin Ye, Yu Luo, Fenghe Fu,

Zhandong Ren and Yuchan Zhu**

School of Chemical and Environmental Engineering, Wuhan Polytechnic University, Wuhan,

430023, P. R. China.

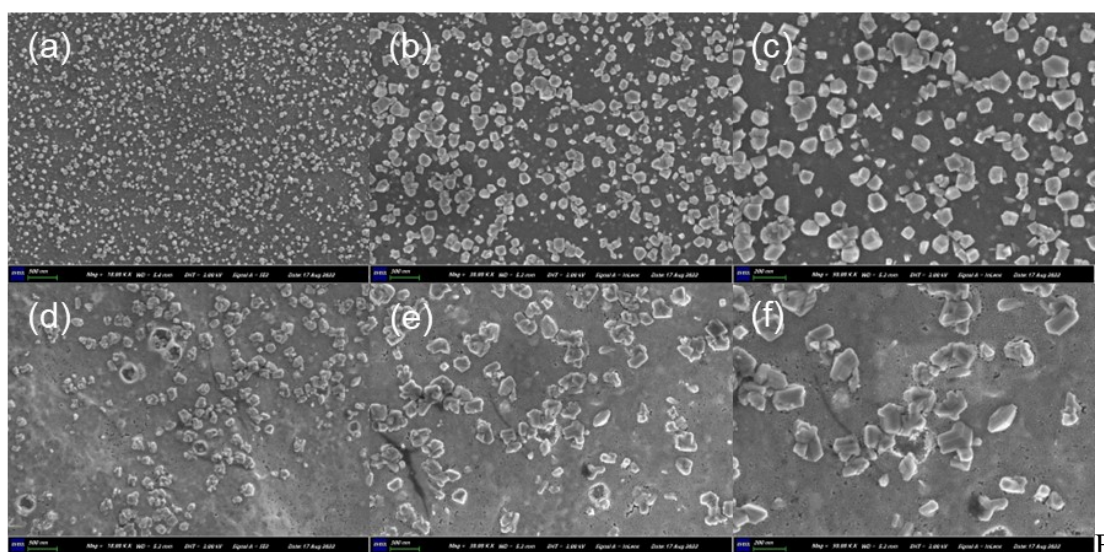


Figure S1 The SEM images of IrO₂-Ta₂O₅ (a-c) and SnO_x@IrO₂-Ta₂O₅-3 (d-f) electrodes.

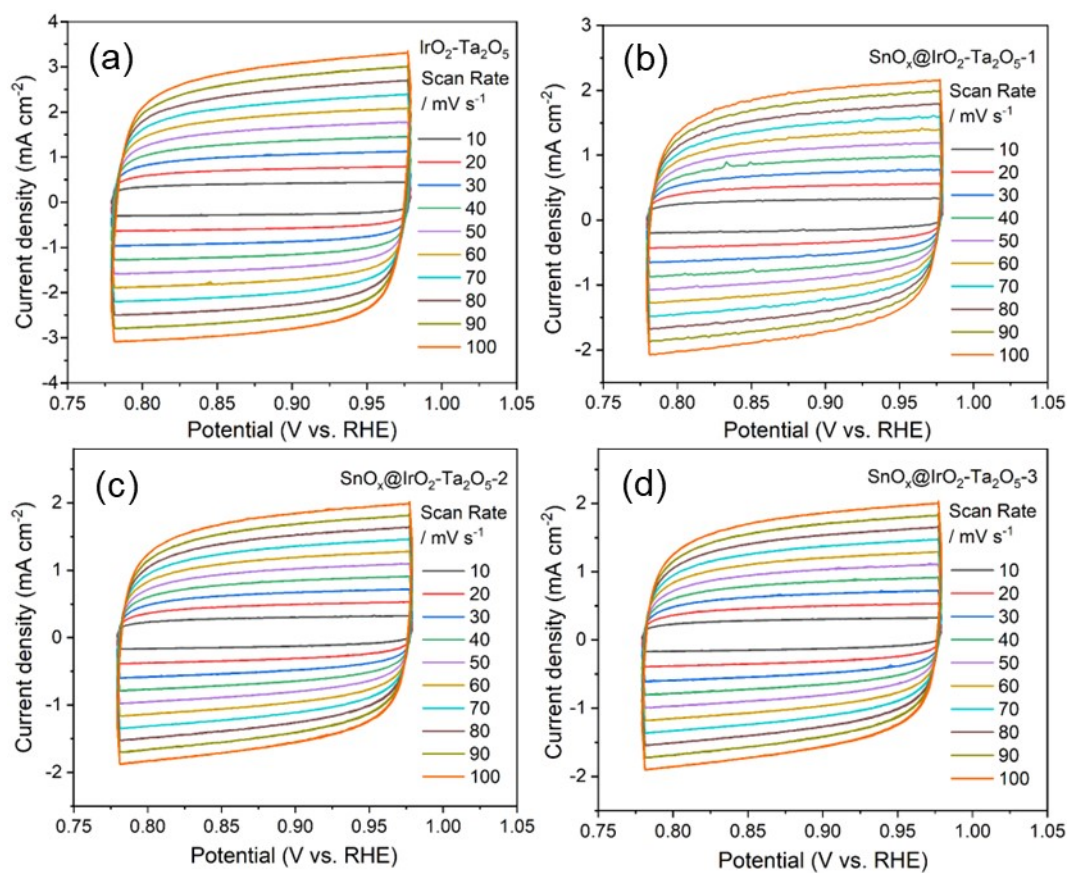


Figure S2 The cyclic voltammograms of IrO₂-Ta₂O₅ (a) and SnO_x@IrO₂-Ta₂O₅-1(2,3) (b-d) electrodes with different scan rate at potential of 0.78 ~ 0.98 V.

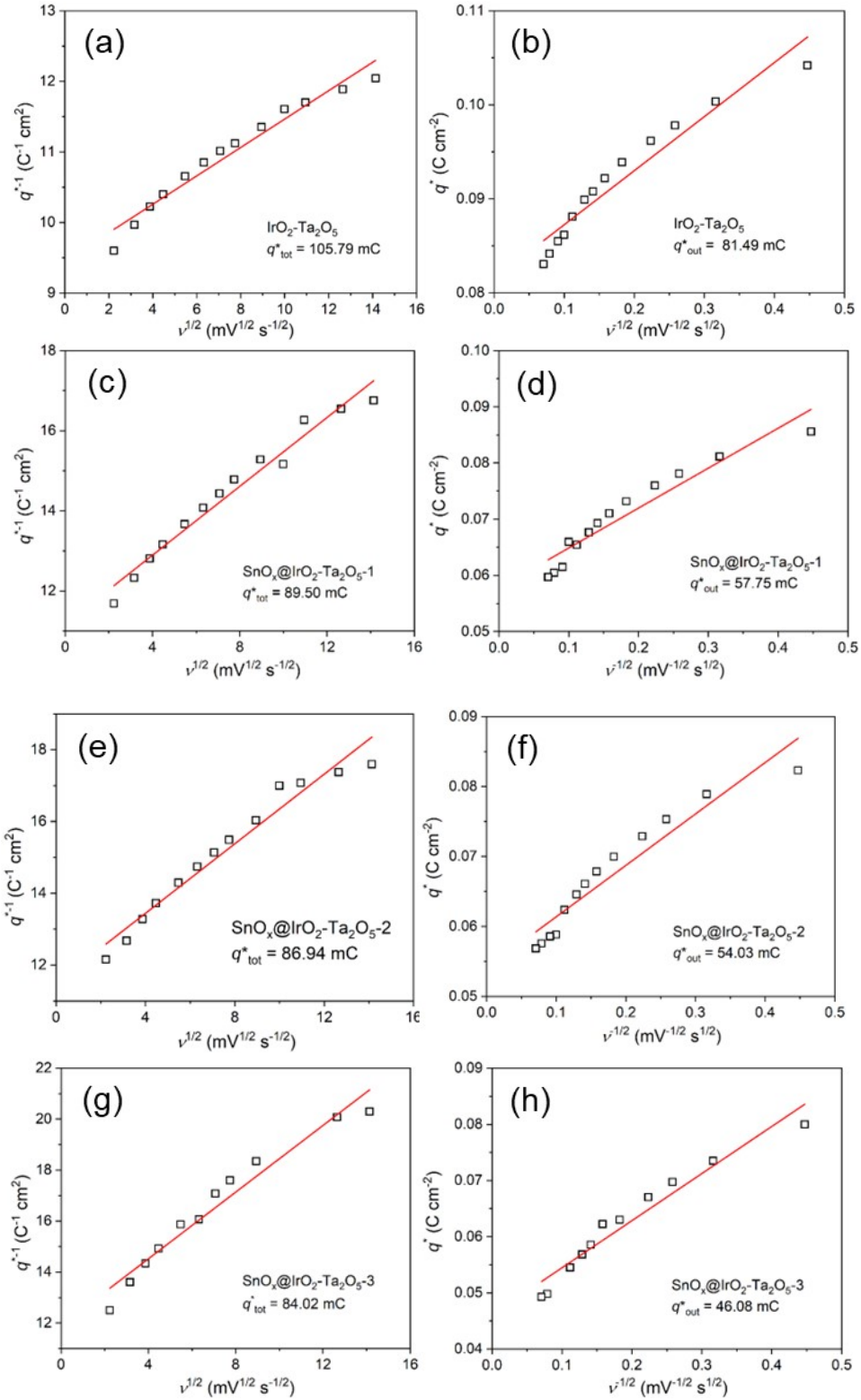


Figure S3 The total surface charge (q^{*}_{tot}) and outer surface charge (q^{*}_{out}) of IrO₂-Ta₂O₅ (a-b) and SnO_x@IrO₂-Ta₂O₅-1(2,3) (c-h) electrodes.

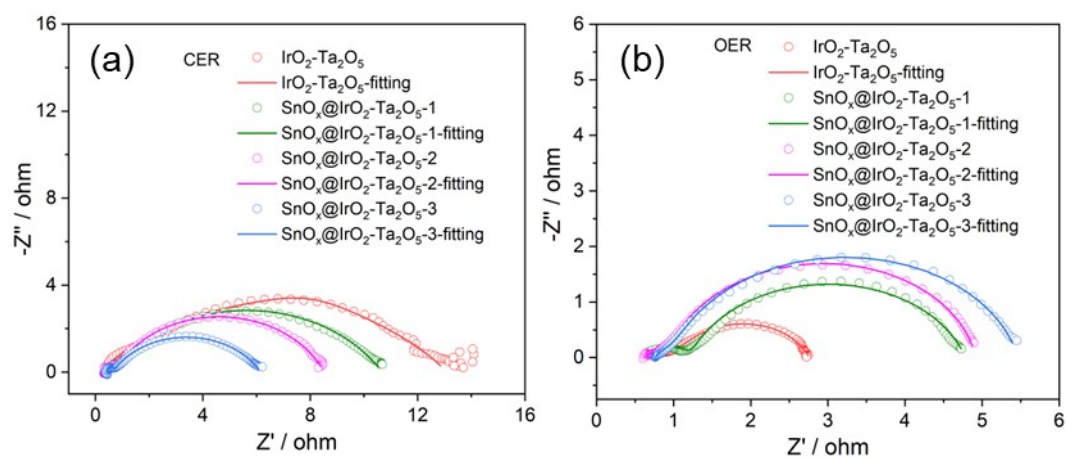


Figure S4 The electrochemical impedance spectroscopy (EIS) measurements of IrO₂-Ta₂O₅ and SnO_x@IrO₂-Ta₂O₅-1(2,3) electrodes at 1.48 V in the CER (a) and at 1.64 V OER (b).

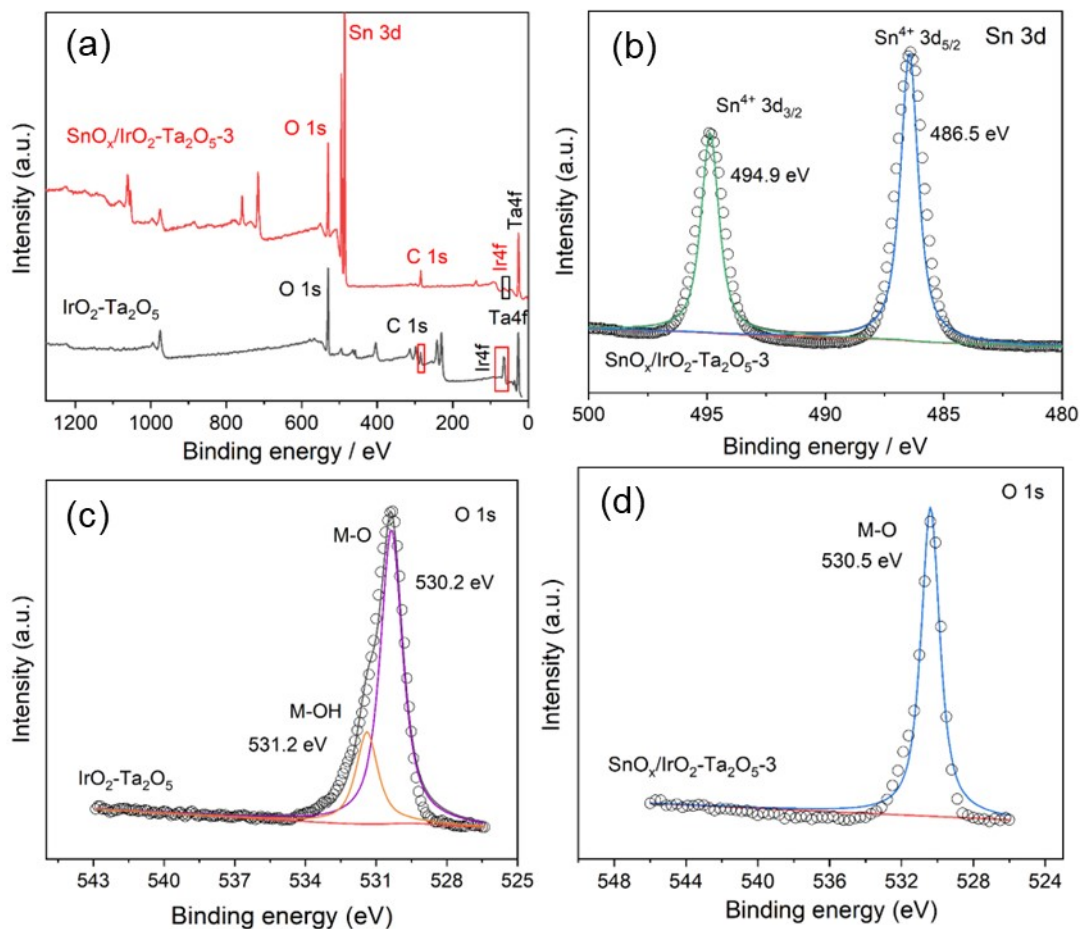


Figure S5 The XPS whole spectra of $\text{IrO}_2\text{-Ta}_2\text{O}_5$ (c) and $\text{SnO}_x@\text{IrO}_2\text{-Ta}_2\text{O}_5\text{-3}$ (a). The XPS core-level spectra of Sn 3d obtained from $\text{SnO}_x@\text{IrO}_2\text{-Ta}_2\text{O}_5\text{-3}$ (b). The XPS core-level spectra of O 1s obtained from $\text{IrO}_2\text{-Ta}_2\text{O}_5$ (c) and $\text{SnO}_x@\text{IrO}_2\text{-Ta}_2\text{O}_5\text{-3}$ (d).

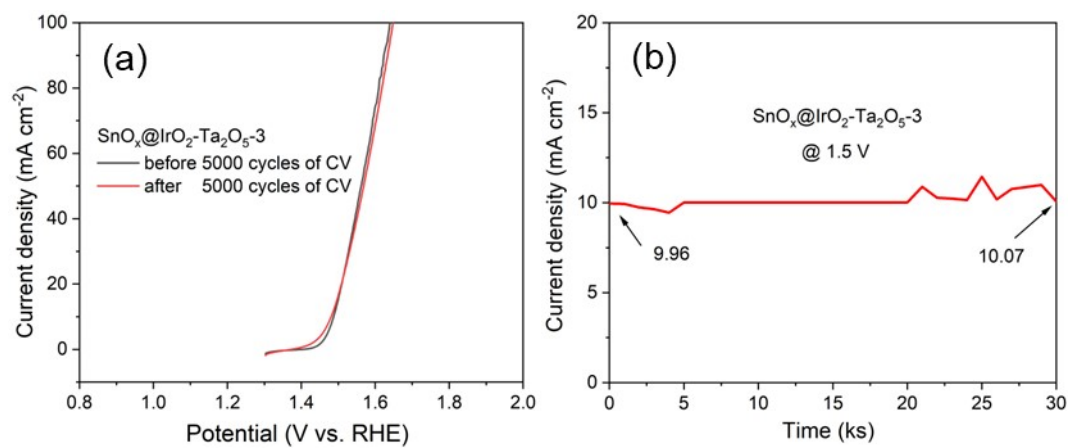


Figure S6 The accelerated lifetime tests of $\text{SnO}_x@ \text{IrO}_2\text{-Ta}_2\text{O}_5\text{-3}$ electrode after 5000 cycles of CVs (a) and the 30 ks of chronoamperometry testing (b).

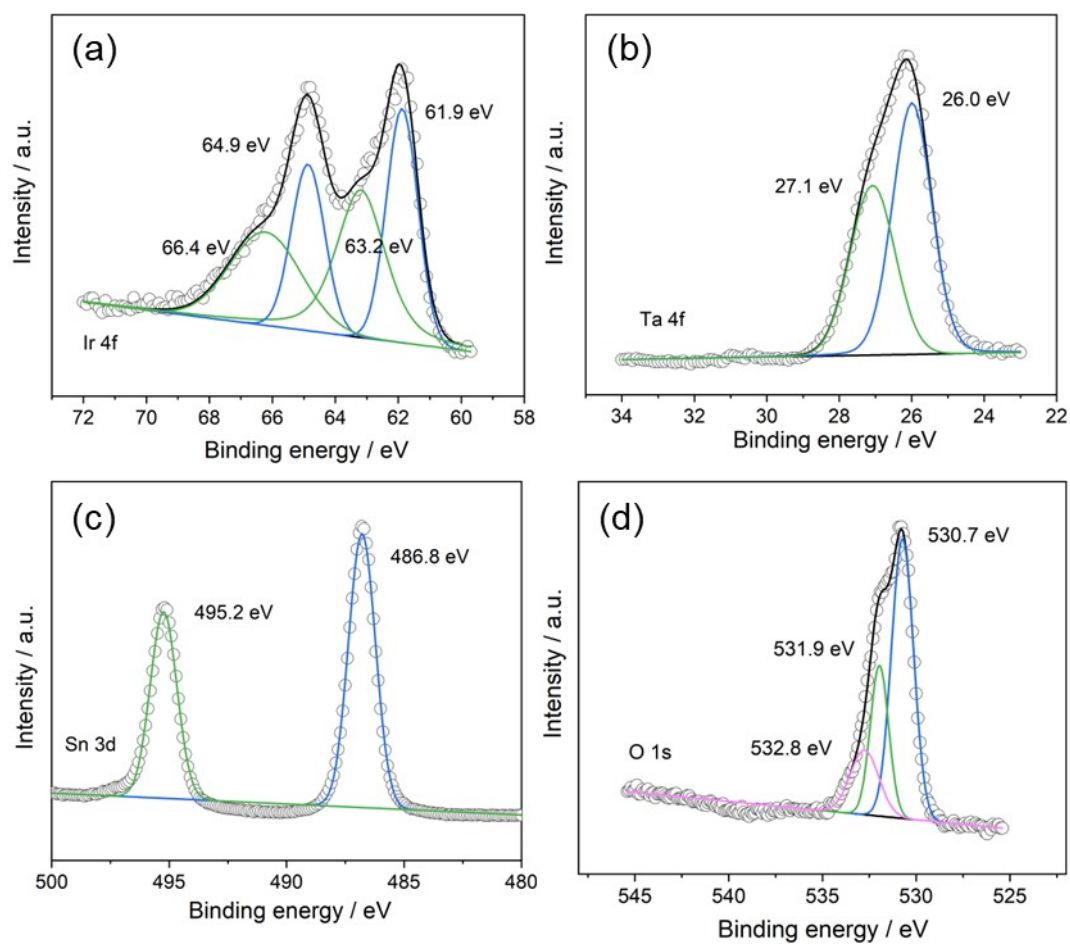


Figure S7 The XPS core-level spectra of Ir 4f (a), Ta 4f (b), Sn 3d (c) and O1s (d) obtained from $\text{SnO}_x@\text{IrO}_2\text{-Ta}_2\text{O}_5\text{-3}$ electrode after accelerated service life test.

Table S1 The loading of Ir, Ta and Sn of the IrO₂-Ta₂O₅ and SnO_x@IrO₂-Ta₂O₅ electrodes by XRF

Electrode	Ir:Ta:Sn (molar ratio)	The loading / $\mu\text{g cm}^{-2}$		
		Ir	Ta	Sn
IrO ₂ -Ta ₂ O ₅		169.40	138.52	
SnO _x @IrO ₂ -Ta ₂ O ₅ -1	4.2:5.0:0.8	109.40	138.36	24.01
SnO _x @IrO ₂ -Ta ₂ O ₅ -2	3.7:5.0:1.3	123.16	146.12	41.22
SnO _x @IrO ₂ -Ta ₂ O ₅ -3	3.3:5.0:1.7	127.31	152.31	54.17

Table S2 Fitting parameters of equivalent circuit of IrO₂-Ta₂O₅ and SnO_x@IrO₂-

Ta ₂ O ₅ -1(2,3) electrodes			
Reaction	Electrodes	$R_s/\Omega \text{ cm}^2$	$R_{ct}/\Omega \text{ cm}^2$
CER	IrO ₂ -Ta ₂ O ₅	0.370	11.0
	SnO _x @IrO ₂ -Ta ₂ O ₅ -1	0.413	10.1
	SnO _x @IrO ₂ -Ta ₂ O ₅ -2	0.367	7.81
	SnO _x @IrO ₂ -Ta ₂ O ₅ -3	0.425	5.54
OER	IrO ₂ -Ta ₂ O ₅	0.697	1.99
	SnO _x @IrO ₂ -Ta ₂ O ₅ -1	0.630	3.56
	SnO _x @IrO ₂ -Ta ₂ O ₅ -2	0.581	3.76
	SnO _x @IrO ₂ -Ta ₂ O ₅ -3	0.763	4.44