

Supporting Information

Hydrangea-like $\delta\text{-MnO}_2$ anchored on GNS as a high-performance supercapacitor electrode material

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Experimental data

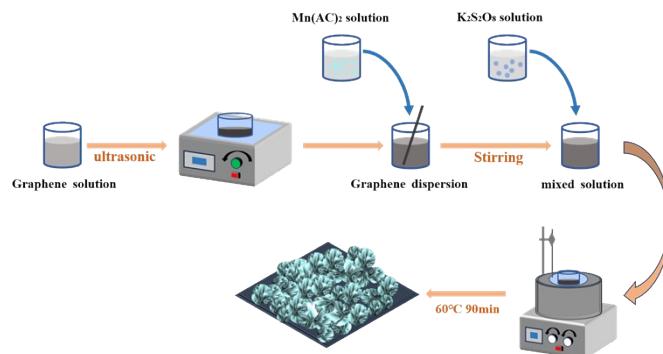


Figure 1. Schematic synthesis of $\delta\text{-MnO}_2/\text{GNS}$

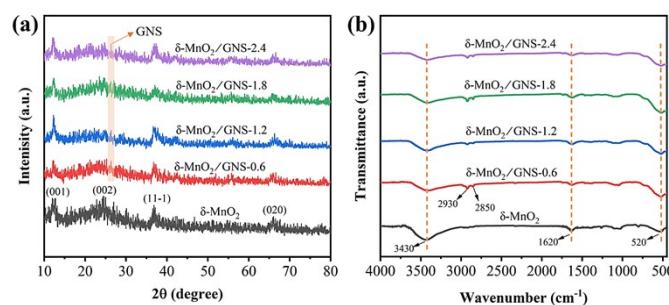


Figure 2. (a) XRD plots of $\delta\text{-MnO}_2$ and composites, (b) FTIR plots of $\delta\text{-MnO}_2$ and composites

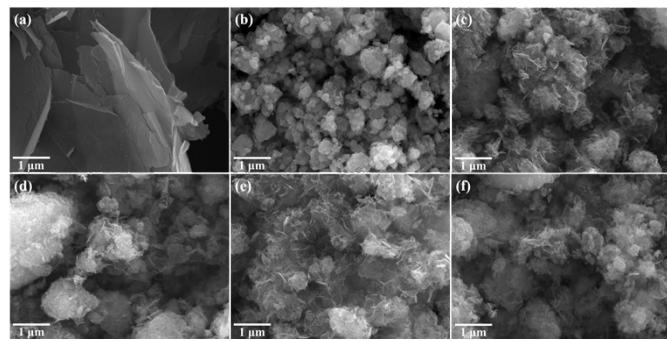


Figure 3. SEM image (a) GNS, (b) δ -MnO₂, (c) δ -MnO₂/GNS-0.6, (d) δ -MnO₂/GNS-1.2, (e) δ -MnO₂/GNS-1.8, (f) δ -MnO₂/GNS-2.4

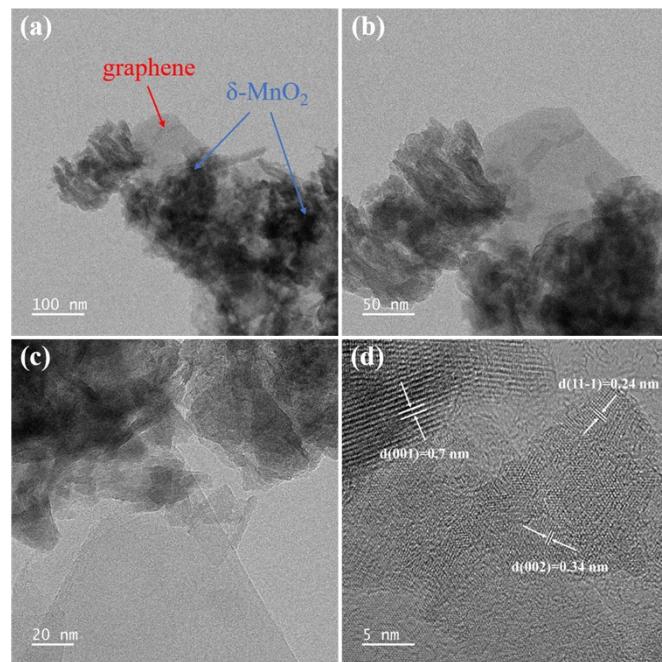


Figure 4. TEM image of δ -MnO₂/GNS-1.8. (a) 200 nm (b) 100 nm , (c) 20 nm, and (d) 5 nm

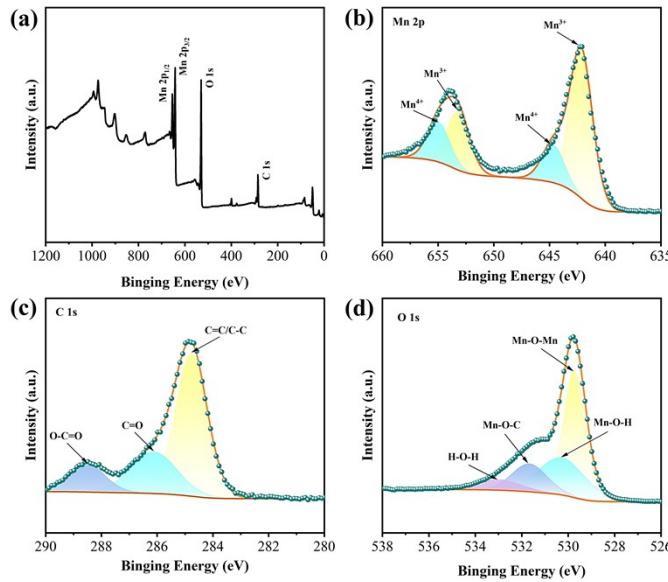


Figure 5. XPS spectra of δ -MnO₂/GNS-1.8 composite. (a) full spectrum, (b) Mn 2p spectrum, (c) Mn 3s spectrum, and (d) O 1s spectrum

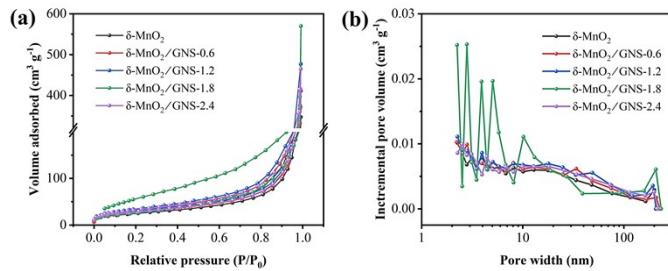


Figure 6. BET spectra of the synthesized materials. (a) Nitrogen adsorption/desorption isotherms , (b) pore size distribution

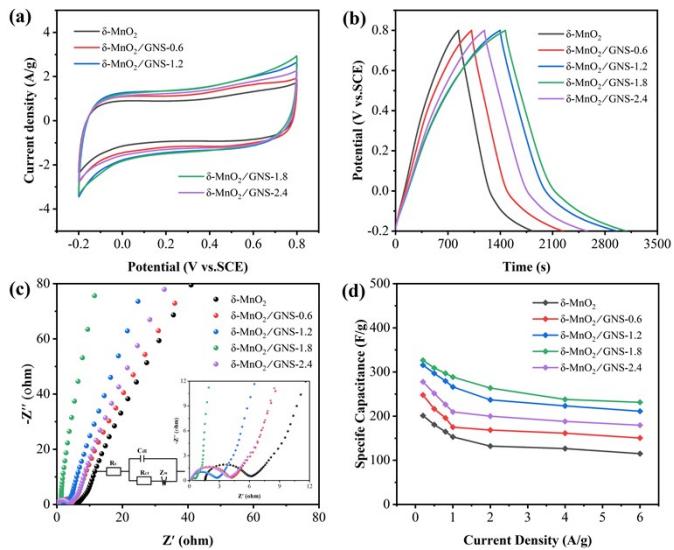


Figure 7. Electrochemical performance plots. (a) CV plot, (b) GCD plot, (c) EIS plot, (d) multiplicity performance plot

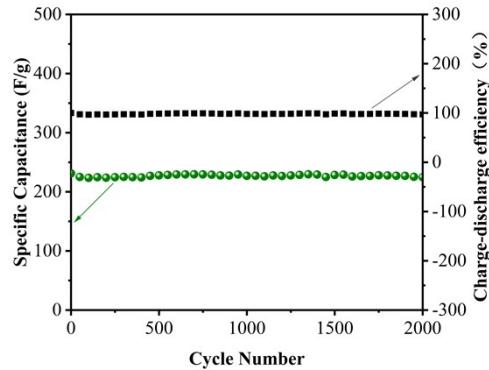


Figure 8. $\delta\text{-MnO}_2/\text{GNS-1.8}$ at 6 A/g 2000 cycles

Table 1. Experimental results for the same type of electrode material

| Electrode material | Potentiometric window | Specific capacitance | Ref |
|----------------------------------|-----------------------|----------------------|-----------|
| CNT@CNT@MnO ₂ | 0-1.0 V | 210 F/g 0.5 A/g | 11 |
| HRGO/MnO ₂ | -0.2-0.8 V | 270 F/g 0.5 A/g | 34 |
| 3DHG-MnO ₂ | -0.2-0.8 V | 192.2 F/g 0.5 A/g | 35 |
| MnO ₂ /rGO | -0.2-1.0 V | 255 F/g 0.5A/g | 37 |
| N-PCNs/MnO ₂ | -0.1-0.9 V | 295 F/g 0.5 A/g | 41 |
| $\delta\text{-MnO}_2/\text{GNS}$ | -0.2-0.8 V | 309 F/g 0.5 A/g | This work |
| $\delta\text{-MnO}_2/\text{GNS}$ | -0.2-0.8 V | 326.6 F/g 0.2 A/g | This Work |

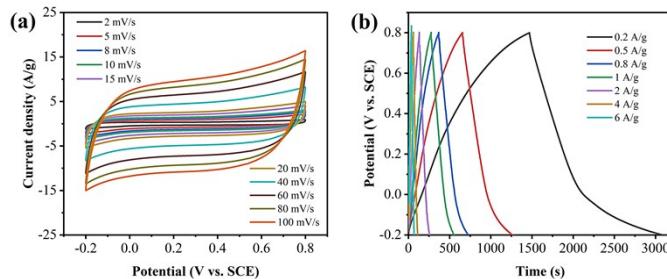


Figure 9. Electrochemical performance plots. (a) CV curves of $\delta\text{-MnO}_2/\text{GNS-1.8}$ with different sweep speeds, (b) GCD curves of $\delta\text{-MnO}_2/\text{GNS-1.8}$ with different current densities

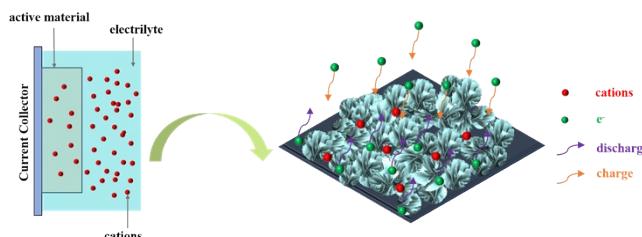


Figure 10. Electrochemical mechanism diagram of $\text{MnO}_2/\text{GNS-1.8}$

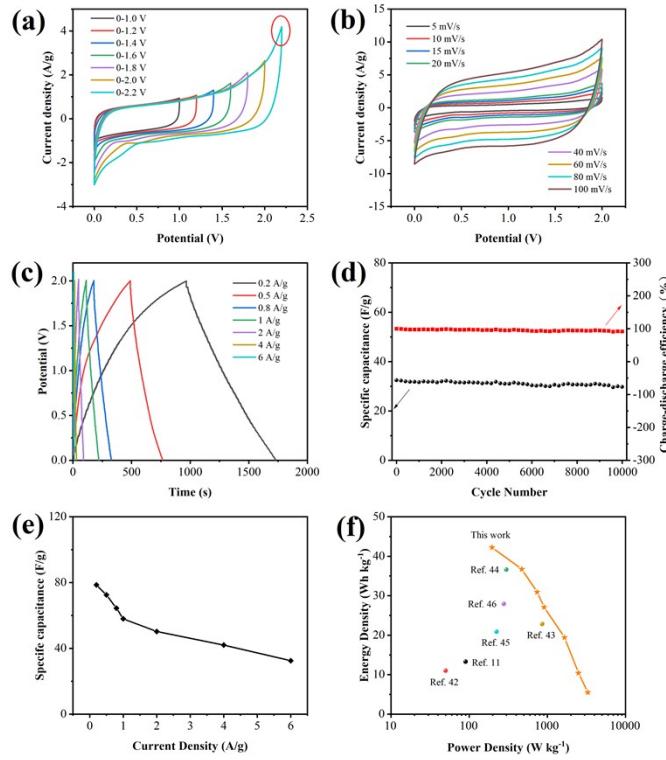


Figure 11. Performance diagram of $\delta\text{-MnO}_2/\text{GNS//AC}$ capacitor. (a) CV curves in different voltage windows at 10 mV/s scan rate, (b) CV curves at different scan rates, (c) GCD curves for different current densities, (d) 5000 consecutive charge/discharge cycles at 6 A/g current density, (e) Specific capacitance at different current densities, (f) Ragone diagram

Table 2. Table comparing the performance of $\delta\text{-MnO}_2/\text{GNS//AC}$ with other supercapacitor devices reported in the literature

| Asymmetric supercapacitors | Working potential (V) | Energy Density (Wh/kg) | Power density (W/kg) | Ref |
|--|-----------------------|------------------------|----------------------|-----------|
| CNT@CNT@MnO ₂ //CNT@NCT | 1.8 | 13.3 | 90 | 11 |
| MnO ₂ @Carbon black//AC | 1.6 | 11 | 50 | 42 |
| GF/CNT/MnO ₂ /GF/CNT/Ppy | 2.0 | 22.8 | 860 | 43 |
| 3D-MnO ₂ //AC | 2.0 | 36.6 | 300 | 44 |
| N-CNTs/MnO ₂ -2//N-CNTs | 1.8 | 20.9 | 224 | 45 |
| rGO/Mn ₃ O ₄ // rGO/Mn ₃ O ₄ | 0.8 | 27.92 | 277.8 | 46 |
| $\delta\text{-MnO}_2/\text{GNS//AC}$ | 2.0 | 42.2 | 196.8 | This work |