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Supporting Information

Highly efficient selective oxidation of toluene to benzaldehyde over Cu-V oxides

supported on amorphous SiO₂

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Fig. S1 The corresponding energy dispersive X-ray spectrum (EDS) of 1Cu-4V/SiO₂

Catalyst	Cu loadings (wt%)	V loadings (wt%)
4Cu-1V/SiO ₂	9.85	0.31
2Cu-1V/SiO ₂	6.34	0.92
1 Cu- 1 V/SiO $_2$	4.01	1.89
1Cu-2V/SiO ₂	2.79	2.98
1Cu-4V/SiO ₂	2.56	5.57

Table S1 The loadings of Cu and V in the catalysts

Obtained by ICP-OES test



Fig. S3 Recycle experiment of $1Cu-4V/SiO_2$. Reaction conditions: toluene (2.5 mmol), PDS (5.0 mmol), H₂O/CH₃CN (5ml, volume ratio: 1:1), catalyst (46 mg), 60 °C, 6 h.

Experimental details for GC analysis

The reaction products are analyzed used the internal standard method with gas chromatography (Fuli 9790II, FID detector). Firstly, using ethylbenzene as the internal standard to prepare standard solutions of toluene and benzaldehyde in several concentrations, but the concentration of ethylbenzene remained unchanged. Then, the standard curves of toluene and benzaldehyde were obtained using the peak area ratio as the abscissa and the concentration ration as the ordinate.

The GC detecting conditions are as follows: using nitrogen as a carrier gas, the injection temperature is 280 °C and the detector temperature is 280 °C. The test procedures are maintaining one minute at 50 °C of the column temperature, and then heating up to 250 °C at a heating rate of 25 °C/min and subsequently maintain for one minute.



Fig. S4 The standard curve of toluene



Fig. S5 the standard curve of ethylbenzene



Fig. S6 The GC result for the oxidation of toluene (T = 2.774 min, toluene; T = 3.605 min, ethylbenzene; T = 4.461 min, benzaldehyde; T = 5.036 min, benzyl alcohol; T = 5.922 min, benzoic acid). The conversion of toluene and yield of benzaldehyde were determined with an internal standard method, the internal standard was ethylbenzene.



Fig. S7 GC-MS of benzaldehyde



Fig. S8 GC-MS of benzyl alcohol



Fig. S9 GC-MS of benzoic acid