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Supplementary Information

Copper Sulfide-Incorporated Layered Porous Sulfur-Doped Graphitic

Carbon Nitride Nanosheets for an Efficient Catalytic

Reduction of 4-Nitrophenol

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Figure S1. Tauc's plot of $(\alpha hv)^2 vs$. Band gap energy of s-g-C₃N₄ (Black line) and CuS/ s-g-C₃N₄ (Red line) samples.



Figure S2. SEM micrographs of (A) s-g- C_3N_4 and (B) CuS/s-g- C_3N_4 samples.



Figure S3. HR-TEM image of the CuS/ S-g-C₃N₄ composite used for the determination of d-spacing CuS (A) and s-g-C₃N₄ (B).

SI No	Substrate	Abbreviations	Reduction rate (%)	Time (min)
1	OH NO ₂ CH ₃	4-methyl-2- nitrophenol	99.3	3
2	NH ₂ NO ₂	2-nitroamine 98.7		6
3	OH NO ₂	2-nitrophenol	99.3	2
4	NH ₂ NO ₂	3-nitroamine	99.5	3
5	NH ₂ NO ₂ NO ₂	2,4-Dinitroamine	89.3	3
6	NO ₂ NH ₂ NH ₂	4-nitrodiamine	98.6	2
7	NO ₂ NH ₂	4-nitroamine	98.8	6

Table S1. The catalytic reduction of various nitro scaffolds using CuS/s-g-C₃N₄.

Table S2. Comparison of the apparent rate constant of 4-NP reduction with variouscatalysts using the Langmuir-Hinshelwood equation.

Catalysts	4-NP	NaBH ₄	k	Time	References
			(min⁻¹)	(min)	
PANI/ZnO/MnO ₂	0.2 mM	0.1 M	219	10	1
CuO nanoparticles	0.36	30	0.022	15	2
GO/Au	0.1 mM	0.1 M	-	2100	3
TiO2/rGO NCs	10 mg/L of 4-NP	-	0.0216	32	4
	+ H ₂ O ₂ (1 Mm)				
PANI Nanofibers	0.2 mM	0.002 M	48.8	40	5
Ca ²⁺⁻ doped AgInS ₂	15 mg/L	-	-	120	6
Ag ₃ PO ₄ /g-C ₃ N ₄	30 mg/L	20mM	0.01277	5	7
ZnO/Ag ₂ O	1.0 mM	0.01 M	229.65	21	8
Bi ₂ O ₃	0.5 mM	0.2 M	91.42	50	9
CuS/s-g-C ₃ N ₄	20 mg L ^{−1}	0.1 M	2.357	3	Present work

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