

Supplementary material

Table S1. Data collection and refinement details for **Cl2**

Chemical formula	H <sub>4</sub> Cl <sub>2</sub> EuO <sub>2</sub>
$M_r$	258.89
Crystal system, space group	Monoclinic, <i>C2/c</i>
Temperature (K)	100
$a, b, c$ (Å)	11.636 (3), 6.358 (2), 6.637 (2)
$\beta$ (°)	105.81 (3)
$V$ (Å <sup>3</sup> )	472.4 (2)
$Z$	4
Radiation type	Mo $K\alpha$
$\mu$ (mm <sup>-1</sup> )	14.24
Crystal size (mm)	0.21 × 0.12 × 0.06
Diffractometer	Xcalibur, Ruby
$T_{\min}, T_{\max}$	0.194, 0.541
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	3765, 653, 642
$R_{\text{int}}$	0.023
$(\sin \theta/\lambda)_{\text{max}}$ (Å <sup>-1</sup> )	0.701
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.013, 0.033, 1.23
No. of reflections	653
No. of parameters	32
H-atom treatment	All H-atom parameters refined
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e Å <sup>-3</sup> )	0.44, -1.07

Computer programs: *CrysAlis PRO*, Agilent Technologies, Version 1.171.35.19 (release 27-10-2011 CrysAlis171 .NET) (compiled Oct 27 2011, 15:02:11), *SHELXS97* (Sheldrick, 1990), *SHELXL97* (Sheldrick, 1997).

Table S2. Data collection and refinement details for **Cl6**

Chemical formula	Cl <sub>2</sub> EuH <sub>12</sub> O <sub>6</sub>
$M_r$	330.96
Crystal system, space group	Trigonal, <i>P321</i>
Temperature (K)	80
$a, c$ (Å)	7.923 (2), 4.089 (1)
$V$ (Å <sup>3</sup> )	222.29 (10)
$Z$	1.0
Radiation type	Mo $K\alpha$
$\mu$ (mm <sup>-1</sup> )	7.63
Crystal size (mm)	0.18 × 0.07 × 0.05
Diffractometer	Xcalibur, Ruby, Gemini ultra
$T_{\min}, T_{\max}$	0.437, 0.752

No. of measured, independent and observed [ $F > 6\sigma(F)$ ] reflections	27961, 2091, 2091
$R_{\text{int}}$	0.039
$(\sin \theta/\lambda)_{\text{max}}$ ( $\text{\AA}^{-1}$ )	1.186
$R[F > 6\sigma(F)]$ , $wR(F^2)$ , $S$	0.009, 0.0223, 0.75
No. of reflections	2091
No. of parameters	66
$\Delta\rho_{\text{max}}$ , $\Delta\rho_{\text{min}}$ ( $\text{e \AA}^{-3}$ )	0.72, -2.86
Absolute structure parameter	-0.003 (8)

Computer programs: *CrysAlis PRO* 1.171.42.50a (Rigaku OD, 2022), *SHELXS*, Volkov *et al.*, (2006).

Table S3. Data collection and refinement details for **Br1**

Chemical formula	$\text{Br}_2\text{H}_2\text{EuO}$
$M_r$	329.80
Crystal system, space group	Orthorhombic, <i>Pnma</i>
Temperature (K)	100
$a$ , $b$ , $c$ ( $\text{\AA}$ )	11.349 (1), 4.278 (1), 9.144 (1)
$V$ ( $\text{\AA}^3$ )	443.95 (12)
$Z$	4
Radiation type	Mo $K\alpha$
$\mu$ ( $\text{mm}^{-1}$ )	31.93
Crystal size (mm)	$0.13 \times 0.08 \times 0.05$
Diffractometer	Xcalibur, Ruby, Gemini ultra
$T_{\text{min}}$ , $T_{\text{max}}$	0.122, 0.344
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	3250, 667, 632
$R_{\text{int}}$	0.024
$(\sin \theta/\lambda)_{\text{max}}$ ( $\text{\AA}^{-1}$ )	0.698
$R[F^2 > 2\sigma(F^2)]$ , $wR(F^2)$ , $S$	0.012, 0.026, 1.13
No. of reflections	667
No. of parameters	32
$\Delta\rho_{\text{max}}$ , $\Delta\rho_{\text{min}}$ ( $\text{e \AA}^{-3}$ )	0.72, -0.57

Computer programs: *CrysAlis PRO* 1.171.41.80a (Rigaku OD, 2020), *SHELXS*, *SHELXL2018/3* (Sheldrick, 2018).

Table S4. Data collection and refinement details for **Br6**

Chemical formula	$\text{Br}_2\text{H}_{12}\text{EuO}_6$
$M_r$	419.88
Crystal system, space group	Trigonal, <i>P321</i>
Temperature (K)	100

$a, c$ (Å)	8.200 (2), 4.138 (1)
$V$ (Å <sup>3</sup> )	240.96 (13)
$Z$	1
Radiation type	Mo $K\alpha$
$\mu$ (mm <sup>-1</sup> )	14.78
Crystal size (mm)	0.17 × 0.10 × 0.08
Diffractometer	Xcalibur, Ruby, Gemini ultra
$T_{\min}, T_{\max}$	0.194, 0.434
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	3999, 454, 453
$R_{\text{int}}$	0.040
$(\sin \theta/\lambda)_{\text{max}}$ (Å <sup>-1</sup> )	0.702
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.012, 0.029, 1.13
No. of reflections	454
No. of parameters	25
H-atom treatment	All H-atom parameters refined
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e Å <sup>-3</sup> )	0.46, -0.51
Absolute structure parameter	0.022 (18)
Computer programs: <i>CrysAlis PRO</i> 1.171.41.80a (Rigaku OD, 2020), <i>SHELXS</i> , <i>SHELXL2018/3</i> (Sheldrick, 2018).	

Table S5. Data collection and refinement details for **I6**

Chemical formula	EuH <sub>12</sub> I <sub>2</sub> O <sub>6</sub>
$M_r$	513.86
Crystal system, space group	Trigonal, $P321$
Temperature (K)	100
$a, c$ (Å)	8.565 (2), 4.241 (1)
$V$ (Å <sup>3</sup> )	269.43 (14)
$Z$	1
Radiation type	Mo $K\alpha$
$\mu$ (mm <sup>-1</sup> )	11.53
Crystal size (mm)	0.17 × 0.11 × 0.05
Diffractometer	Xcalibur, Ruby, Gemini ultra
$T_{\min}, T_{\max}$	0.251, 0.603
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	5433, 514, 510
$R_{\text{int}}$	0.031
$(\sin \theta/\lambda)_{\text{max}}$ (Å <sup>-1</sup> )	0.699
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.021, 0.051, 1.41
No. of reflections	514

No. of parameters	25
$\Delta\rho_{\max}, \Delta\rho_{\min}$ (e Å <sup>-3</sup> )	2.84, -3.69
Absolute structure parameter	0.027 (14)

Computer programs: *CrysAlis PRO* 1.171.42.50a (Rigaku OD, 2022), *SHELXS*, *SHELXL2018/3* (Sheldrick, 2018).

---Table S6. Hydrogen bonds and interactions.

H-bond	D-H (Å)	H...A (Å)	D...A (Å)	$\angle(\text{D-H}\cdots\text{A})$ (°)
<b>Cl2</b>				
O-H1...Cl <sup>i</sup>	0.87(5)	2.22(5)	3.080(3)	170(4)
O-H2...Cl <sup>ii</sup>	0.77(4)	2.54(4)	2.225(3)	149(4)
<b>Cl6</b>				
O1W-H1W1...Cl <sup>iii</sup>	0.99	2.21	3.1882(11)	172
O2W-H1W2...Cl <sup>iii</sup>	0.98	2.65	3.5813(6)	159
<b>Br1</b>				
OW-H1...Br2 <sup>iv</sup>	0.70(5)	2.84(5)	3.310(3)	128(5)
OW-H1...Br1 <sup>v</sup>	0.70(5)	2.92(4)	3.383(3)	126.14(2)
OW-H1...Br1 <sup>vi</sup>	0.70(5)	2.92(4)	3.383(3)	126.14(2)
OW-H2...Br2 <sup>vii</sup>	0.78(6)	2.77(4)	3.298(3)	126.33(2)
OW-H2...Br2 <sup>viii</sup>	0.78(6)	2.77(4)	3.298(3)	126.33(2)
<b>Br6</b>				
O1W-H1W1...Br <sup>ix</sup>	0.82(4)	2.55(4)	3.3469(18)	165(5)
O2W-H1W2...Br <sup>x</sup>	0.74(4)	2.63(4)	3.3450(12)	165(4)
<b>I6</b>				
O1W-H1W1...I <sup>ix</sup>	0.87(13)	2.72(14)	3.564(7)	162(14)
O2W-H1W2...I <sup>xi</sup>	0.97(13)	2.89(15)	3.566(5)	127(12)
O2W-H1W2...I <sup>x</sup>	0.97(13)	3.02(15)	3.811(5)	140(13)

Symmetry codes: <sup>i</sup> x, 1-y, -1/2+z; <sup>ii</sup> 1/2-x, 1/2+y, 3/2-z; <sup>iii</sup> 1+x-y, 1-y, -z; <sup>iv</sup> -1/2+x, 3/2-y, 3/2-z; <sup>v</sup> 1/2-x, 1-y, 1/2+z; <sup>vi</sup> 1/2-x, 2-y, 1/2+z; <sup>vii</sup> 1/2-x, 1-y, -1/2+z; <sup>viii</sup> 1/2-x, 2-y, -1/2+z; <sup>ix</sup> x-y, -y-z; <sup>x</sup> 1+x, y, z; <sup>xi</sup> 1+x, y, 1+z;

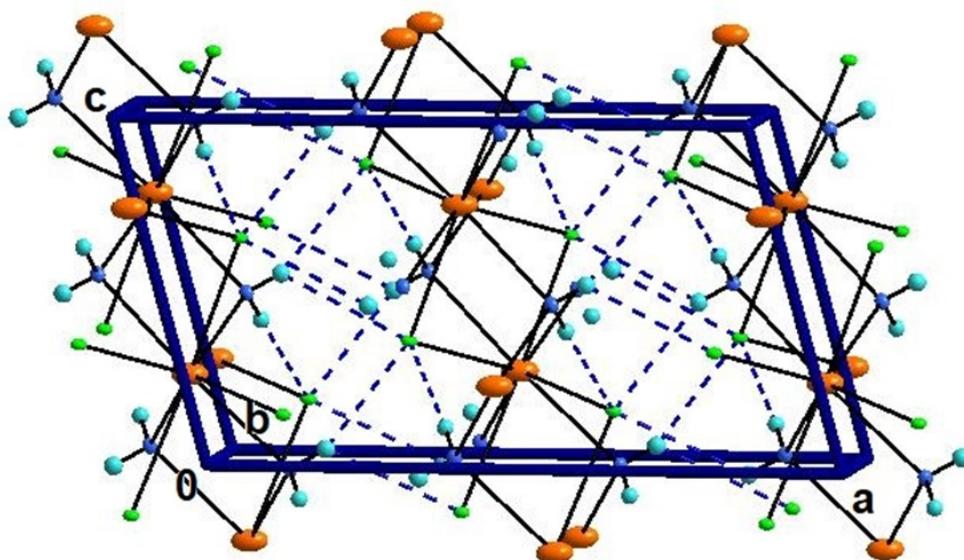


Fig. S1. Packing diagram of **Cl2**. The Eu ellipsoids are drawn at 99 % probability, all others - at 50 %. The hydrogen bonds are dashed blue.

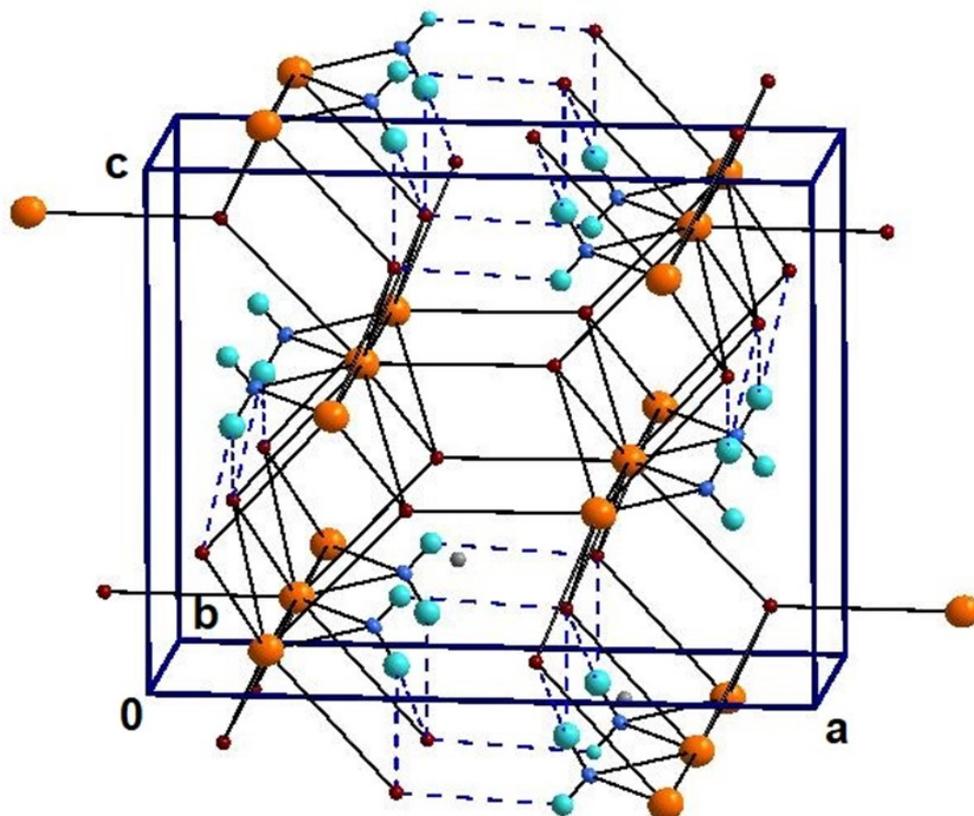


Fig. S2. Packing diagram of **Br1**. The Eu ellipsoids are drawn at 99 % probability, all others - at 50 %. The hydrogen bonds are dashed blue.

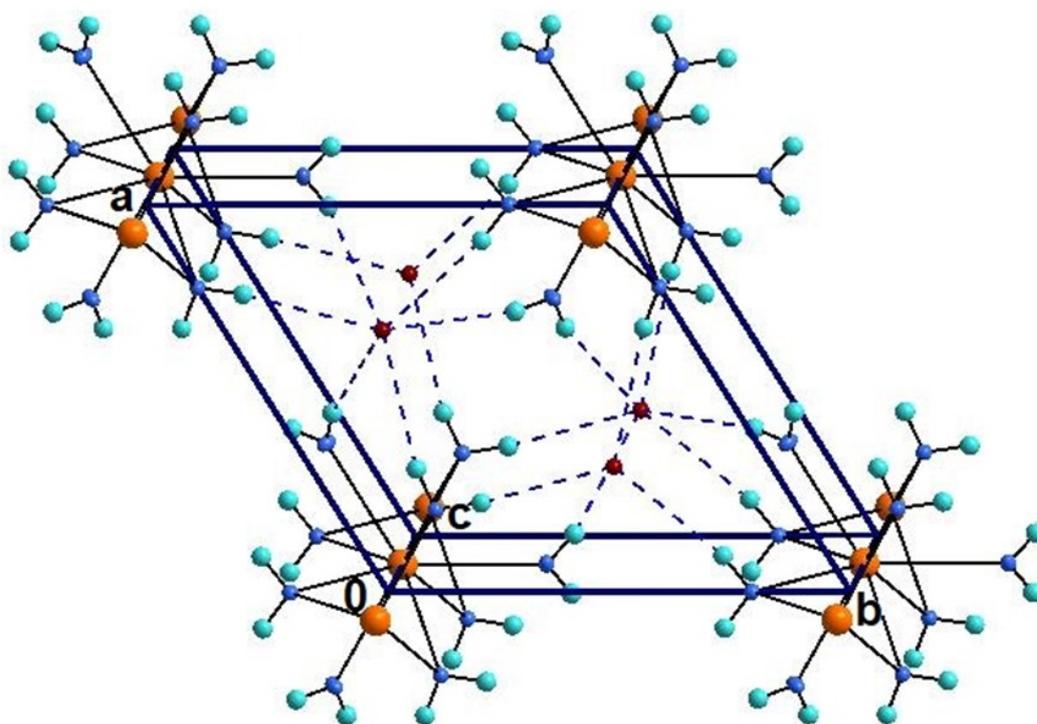


Fig. S3. Packing diagram of **Br6**. The Eu ellipsoids are drawn at 99 % probability, all others - at 50 %. The hydrogen bonds are dashed blue.

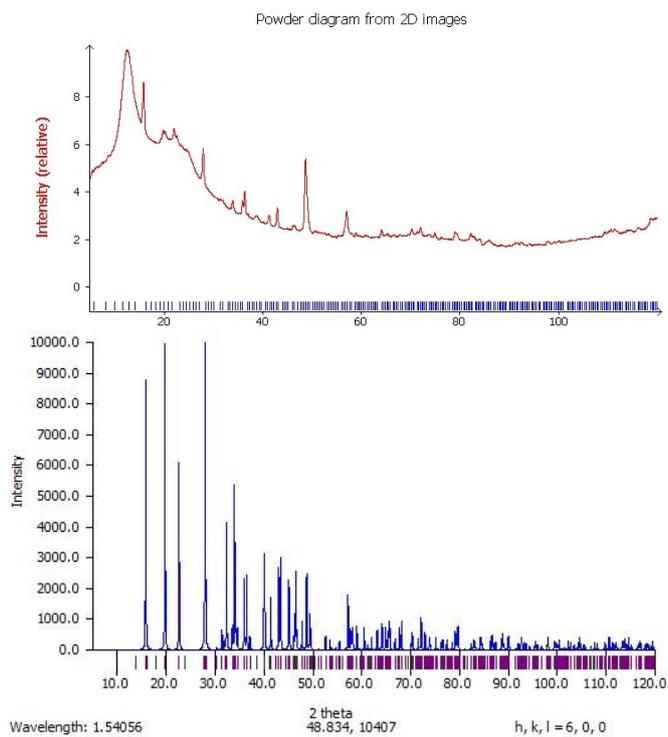


Fig. S4. Experimental<sup>1</sup> (upper diagram) and simulated<sup>2</sup> (lower diagram) powder diffractogram of **Cl2**.

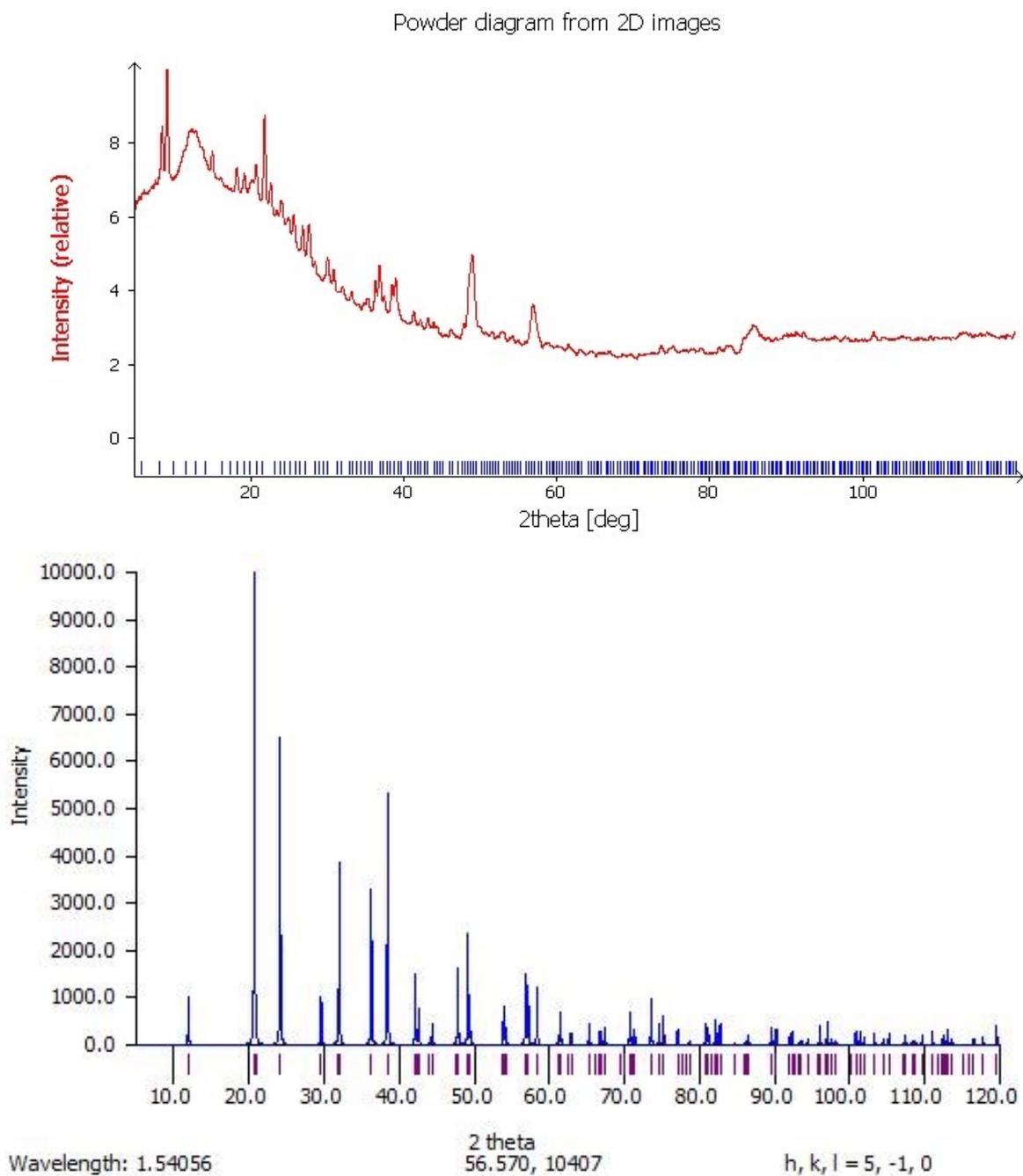


Fig. S5. Experimental<sup>1</sup> (upper diagram) and simulated<sup>2</sup> (lower diagram) powder diffractogram of **I6**.

<sup>1</sup> Rigaku Oxford Diffraction, (2023), CrysAlisPro Software system, version 1.171.42.50a, Rigaku Corporation, Wroclaw, Poland.

<sup>2</sup> Mercury 4.0: from visualization to analysis, design and prediction C. F. Macrae, I. Sovago, S. J. Cottrell, P. T. A. Galek, P. McCabe, E. Pidcock, M. Platings, G. P. Shields, J. S. Stevens, M. Towler and P. A. Wood, *J. Appl. Cryst.*, 2020, **53**, 226-235,