

## Electronic Supplementary Information

### ***In Situ* Synthesis of NiCo<sub>2</sub>O<sub>4</sub>/Carbon Nanocomposites: Play of Carbon Content and Symmetric/Asymmetric Device Configuration on Supercapacitor Performance**

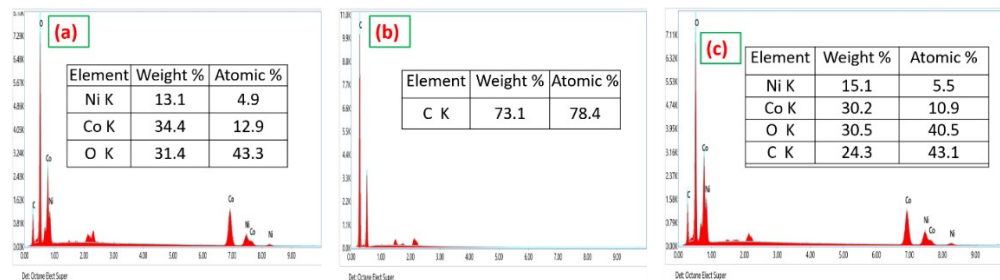
Raji Yuvaraja,<sup>a</sup> Sankar Sarathkumar,<sup>a</sup> Venkatesan Gowsalya,<sup>a</sup> Sorna Pandian Anitha Juliet,<sup>a</sup> Selvakumar Veeralakshmi,<sup>b</sup> Siva Kalaiselvam,<sup>b</sup> Shamima Hussain,<sup>c</sup> Selvan Nehru<sup>\*a</sup>

<sup>a</sup>Department of Physical Chemistry, University of Madras, Guindy Campus, Chennai - 600025, Tamil Nadu, India

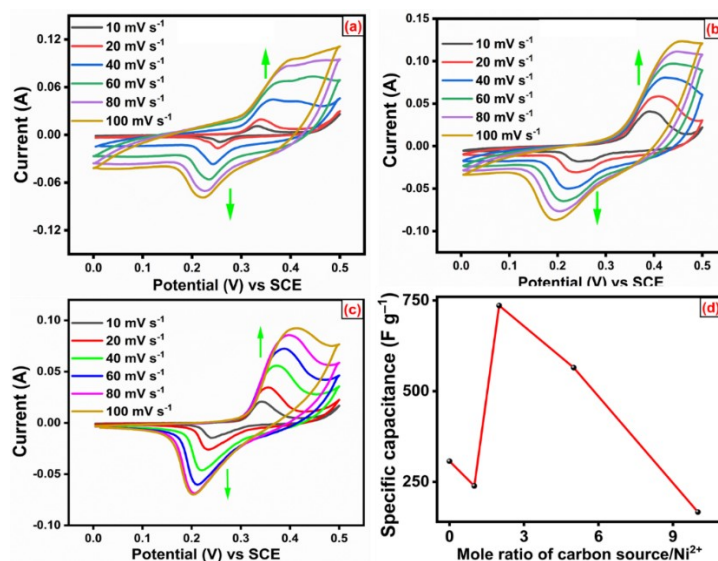
<sup>b</sup>Centre for Industrial Safety, Anna University, Chennai - 600025, Tamil Nadu, India

<sup>c</sup>UGC-DAE Consortium for Scientific Research, Kalpakkam Node, Kokilamedu - 603104, Tamil Nadu, India

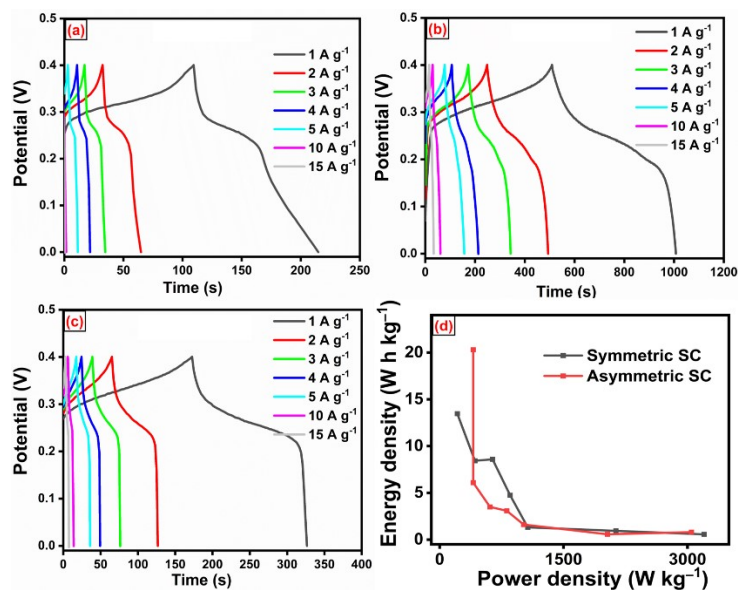
\*Corresponding author's email: [nehruchem@gmail.com](mailto:nehruchem@gmail.com)



**Fig. S1** EDS analysis of the elements present in nanomaterials: (a) NiCo<sub>2</sub>O<sub>4</sub>; (b) CNS and (c) NiCo<sub>2</sub>O<sub>4</sub>/C (D2).



**Fig. S2** CV curves at different scan rates: (a) NiCo<sub>2</sub>O<sub>4</sub>/C (D1); (b) NiCo<sub>2</sub>O<sub>4</sub>/C (D5); (c) NiCo<sub>2</sub>O<sub>4</sub>/C (D10) and (d) Impact of carbon mole ratio with respect to Ni<sup>2+</sup> in NiCo<sub>2</sub>O<sub>4</sub>/C (Dx) based nanocomposites.



**Fig. S3** GCD curves at different current densities: (a) NiCo<sub>2</sub>O<sub>4</sub>/C (D1); (b) NiCo<sub>2</sub>O<sub>4</sub>/C (D5); (c) NiCo<sub>2</sub>O<sub>4</sub>/C (D10) and (d) Ragone plot of symmetric and asymmetric SCs.

**Table S1** Three electrode specific capacitance of NiCo<sub>2</sub>O<sub>4</sub>, CNS and NiCo<sub>2</sub>O<sub>4</sub>/C based nanocomposites at current density 1 A g<sup>-1</sup> using 3 M KOH electrolyte.

S. No.	Electrode Materials	Specific capacitance (F g <sup>-1</sup> )
1	NiCo <sub>2</sub> O <sub>4</sub>	307
2	CNS	52
3	NiCo <sub>2</sub> O <sub>4</sub> /C (D1)	239
4	NiCo <sub>2</sub> O <sub>4</sub> /C (D2)	736
5	NiCo <sub>2</sub> O <sub>4</sub> /C (D5)	565
6	NiCo <sub>2</sub> O <sub>4</sub> /C (D10)	167

**Table S2** Comparison of the supercapacitor performance of NiCo<sub>2</sub>O<sub>4</sub>/C (D2) nanocomposite with similar and some advanced electrode materials.

S. No.	Material	Methodology	Electrolyte	Specific capacitance (F g <sup>-1</sup> or *F cm <sup>-2</sup> )	Cyclic stability (capacity retention %)	Ref.
1	Nitrogen-doped carbon capsules@NiO/NiCo <sub>2</sub> O <sub>4</sub>	<i>In situ</i> calcination	3 M KOH	659	93.5% after 8000 cycles	1
2	Carbon/NiCo <sub>2</sub> O <sub>4</sub> composite	Hydrothermal	3 M KOH	204.3	90.35% after 3000 cycles	2
3	NiCo <sub>2</sub> O <sub>4</sub> /carbon-active composite	Hydrothermal	6 M KOH	273.5	96% after 3000 cycles	3
4	Carbon nanotube@NiCo <sub>2</sub> O <sub>4</sub>	One-pot co-precipitation	6 M KOH	210	92.70% after 2500 cycles	4
5	Porous marigold micro-flower like NiCo <sub>2</sub> O <sub>4</sub>	Chemical bath deposition	6 M KOH	530	90.5% after 3000 cycles.	5
6	Submicron-sized NiCo <sub>2</sub> O <sub>4</sub>	Sol-gel method	1 M KOH	217	96.3% after 600 cycles	6
7	NiCo <sub>2</sub> O <sub>4</sub> @g-C <sub>3</sub> N <sub>4</sub> (C)	Hydrothermal	3 M KOH	325.7	93.6% after 2000 cycles	7
8	NiCo <sub>2</sub> O <sub>4</sub> /carbon cloth	Hydrothermal	6 M KOH	249.7	63.3% after 1000 cycles	8
9	NiCo <sub>2</sub> O <sub>4</sub> nanospheres	Laser ablation in liquid and hydrothermal	1 M KOH	299.7	90.4% after 10,000 cycles	9
10	Polypyrrole-decorated SrFeO <sub>3-δ</sub> perovskites on carbon cloth	Electrochemical deposition	6 M KOH	421	63.6% after 3000 cycles	10
11	Fe-substituted SrCoO <sub>3</sub> perovskites	Solid-state sintering	1 M NaOH	527	85.7% after 5000 cycles	11
12	Fe <sub>3</sub> Mo <sub>3</sub> C/Mo <sub>2</sub> C@ carbon nanotubes	Hydrothermal	1 M KOH	202.3	73.9 % after 4000 cycles	12
13	Cobalt vanadate on CoO urchin-like microspheres	Multi-step process	3 M KOH	*7.58	84.6% after 5000 cycles	13
14	NiCo <sub>2</sub> O <sub>4</sub> /C (D2)	<i>In situ</i> hydrothermal	3 M KOH	736	84.9% after 1000 cycles	<i>This report</i>

## References

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