

Supporting information for

**Rate constants for H-Atom abstraction by  $\text{HOO}^\bullet$  from  
H-donor compounds of Antioxidant Relevance**

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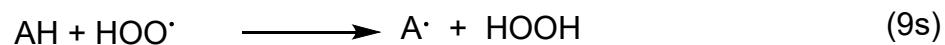
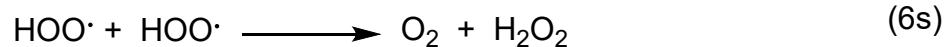
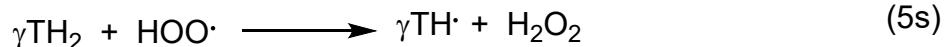
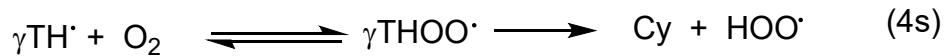
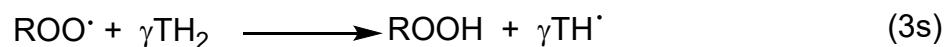
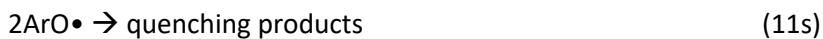
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## 1s. Antioxidants in acetonitrile at 37 °C

Phenolic antioxidants ( $\text{ArOH}$ ) exert their full action in apolar media. In polar media, i.e., solvents ( $S$ ) with the ability to form H-bonding to phenols,  $\text{ArOH}-S$ , they have a lower ability to capture free radicals. Organic matter ( $\gamma\text{-terpinene}$ ,  $\gamma\text{TH}_2$ , in our case) is subject to a process of peroxidation caused by the air oxygen which forms terpenyl peroxy radicals,  $\gamma\text{THOO}\cdot$  and then  $\text{HOO}\cdot$ , see the paper.

These species self-quench to non-radical products or are quenched by  $\text{ArOH}$



$$\text{AH} = 1 - 16$$

**Scheme 1s.** Autoxidation of  $\gamma\text{-terpinene}$  ( $\gamma\text{TH}_2$ ) to *p*-cymene (Cy) and  $\text{H}_2\text{O}_2$ . The reaction steps are numbered with an “s” because these reactions are reported in the current Supporting Information file. The rate constant  $k_9$  (see step 9s) is reported in the old literature as  $k_{\text{inh}}$ . We prefer to keep the same nomenclature, therefore instead of talking about  $k_9$  we called it  $k_{\text{inh}}$ .

## 2s. Apolar Solvents

In *apolar* solvents, the main quenching reactions are reactions 9s and 10s only, because the self-quenching of two  $\text{HOO}\cdot$  is negligible because the concentration of  $\text{HOO}\cdot$  in inhibited peroxidation is very low. The steady-state concentration of  $\text{HOO}\cdot$  is given by eq. 12s,

$$nk_{\text{inh}} [\text{ArOH}] [\text{HOO}\cdot] = R_i \quad (12s)$$

$$[\text{HOO}\cdot] = R_i / nk_{\text{inh}} [\text{ArOH}] \quad (13s)$$

where  $R_i$  is the rate of initiation and  $n$  the number of  $\text{HOO}\cdot$  captured by one molecule of  $\text{ArOH}$  (typically 2.0). Equation 12s expresses the condition for having steady state for  $\text{HOO}\cdot$ ,

rate of formation of  $\text{HOO}\bullet$  = rate of quenching of  $\text{HOO}\bullet$  by  $\text{ArOH}$

The rate-law for the inhibited peroxidation is then

$$R_p = k_5 [\gamma\text{TH}_2][\text{HOO}\bullet] = k_5[\gamma\text{TH}_2] \text{R}_i / n k_{inh} [\text{ArOH}] \quad (14s)$$

$k_5$  is the propagation rate constant,  $R_p$  the rate of peroxidation.

### 3s. Polar solvents and Calculations

In *polar* solvents (e.g. acetonitrile), the quenching reactions are now reactions 6s, 9s, 10s (reaction 11s is negligible since  $\text{ArO}\bullet$  has a low concentration). Reaction 6s must be included, this time, because the  $\text{HOO}\bullet$  concentration is higher, and this is because the quenching reaction 10s is slow. Again, we must equal  $\text{R}_i$  to the sum of reactions 6s and 9s (eq. 10s is included through the  $n$  value since it is a fast process while reaction 6s is the slow step)

$$\text{R}_i = 2k_6 [\text{HOO}\bullet]^2 + nk_{inh}[\text{ArOH}] [\text{HOO}\bullet] \quad (15s)$$

Equation 15s is a second-degree equation with one variable ( $\text{HOO}\bullet$ )

$$2k_6 [\text{HOO}\bullet]^2 + \alpha[\text{HOO}\bullet] - \text{R}_i = 0 \quad (16s)$$

$\alpha = nk_9[\text{ArOH}]_0$   $k_9$  is called  $k_{inh}$  in the paper

The solution to eq. 16s is given by  $[\text{HOO}\bullet] = (-\alpha + (\alpha^2 + 8 \text{R}_i k_6)^{0.5}) / 4k_6$

Then the peroxidation rate in the presence of inhibitors is calculated

$$R_p = k_5 [\gamma\text{TH}_2][\text{HOO}\bullet] = k_5 [\text{TH}_2] (-\alpha + (\alpha^2 + 8 \text{R}_i k_6)^{0.5}) / 4k_6 \quad (17s)$$

In the *absence* of inhibitors, the peroxidation rate is indicated with  $R_0$  and its value is given by

$$R_0 = k_5[\text{TH}_2] \text{R}_i^{0.5} / (2k_6)^{0.5} \quad (18s)$$

$$\text{Therefore, } R_0/R_p = (8 k_6 \times \text{R}_i)^{0.5} / (-\alpha + (\alpha^2 + 8 k_6 \text{R}_i)^{0.5}) \quad (19s)$$

$$\text{And } R_p/R_0 = (-\alpha + (\alpha^2 + 8 k_6 \text{R}_i)^{0.5}) / (8 k_6 \text{R}_i)^{0.5} \quad (20s)$$

$$\text{Now, } R_0/R_p - R_p/R_0 =$$

$$= (8 k_6 \text{R}_i)^{0.5} / (-\alpha + (\alpha^2 + 8 k_6 \text{R}_i)^{0.5}) - (-\alpha + (\alpha^2 + 8 k_6 \text{R}_i)^{0.5}) / (8 k_6 \text{R}_i)^{0.5} \quad (21s)$$

$$= (8 k_6 \text{R}_i - (-\alpha + (\alpha^2 + 8 k_6 \text{R}_i)^{0.5})^2) / ((8 k_6 \text{R}_i)^{0.5} (-\alpha + (\alpha^2 + 8 k_6 \text{R}_i)^{0.5})) \quad (22s)$$

$$= (2 \alpha (\alpha^2 + 8 k_6 \text{R}_i)^{0.5}) - 2 \alpha^2 / (8 k_6 \text{R}_i)^{0.5} (-\alpha + (\alpha^2 + 8 k_6 \text{R}_i)^{0.5}) \quad (23s)$$

$$= \alpha ((\alpha^2 + 8 k_6 \text{R}_i)^{0.5} - \alpha) / ((2 k_6 \text{R}_i)^{0.5} (-\alpha + (\alpha^2 + 8 k_6 \text{R}_i)^{0.5})) \quad (24s)$$

$$= \alpha / (2 k_6 \text{R}_i)^{0.5} = \quad (25s)$$

$$= nk_{inh} [ArOH]_0 / (2k_6 \times Ri)^{0.5} \quad (26s)$$

Let's pose  $\Gamma_{inh} = R_p/R_p - R_p/R_0 =$

$$= nk_{inh} [ArOH]_0 / (2k_6 \times Ri)^{0.5} \quad (27s)$$

Then

$$nk_{inh} = \Gamma_{inh} (2k_6 \times Ri)^{0.5} / [ArOH]_0 \quad (28s)$$

#### 4s. Calculations of the Rate Constant relative to PMHC as an example

We must first determine the initial rate of peroxidation with and without antioxidant

$$R_p = \frac{\Delta [Cy]_{inh}}{\Delta t}$$

where  $\Delta [Cy]_{inh}$  is the increase in [Cy] in the first 150 – 250 min of kinetics and in the presence of antioxidant. In the absence of antioxidant,

$$R_{p0} = \frac{\Delta [Cy]_0}{\Delta t}$$

where  $\Delta [Cy]_0$  is the increase in the Cy concentration in the first 100 – 150 min. The average of  $R_p^0$  was calculated to be  $10.1 \times 10^{-6} \pm 4.0 \times 10^{-6}$  M/s. The standard acetonitrile solution used in the experiments was 0.945 M in  $\gamma$ TH<sub>2</sub>,  $6.0 \times 10^{-2}$  M in AIBN and  $6.54 \times 10^{-2}$  M in mesitylene used as an internal standard.

The term  $(2k_6 \times Ri)^{1/2} = (2k_6 \times 2ek_d \times [AIBN])^{1/2}$

$[AIBN] = 6.0 \times 10^{-2}$  M in all experiments,  $2ek_d$  in acetonitrile at 37 °C is about  $1.0 \times 10^{-6}$  s<sup>-1</sup>. Therefore,  $Ri = 2ek_d[AIBN] = 6.0 \times 10^{-8}$  M/s. The rate constant  $2k_6$  is  $8.2 \times 10^7$  M<sup>-1</sup>s<sup>-1</sup> in acetonitrile at 50 °C and probably  $4.1 \times 10^7$  M<sup>-1</sup>s<sup>-1</sup> at 37 °C (reaction rates double for every 10 degrees rise in temperature). Thus  $(Ri \times 2k_6)^{1/2}$  is about 1.6 and the calculation of  $nk_9$  has been done with this number

$$\begin{aligned} nk_{inh} &= \Gamma_{inh} (2k_6 \times Ri)^{0.5} / [ArOH]_0 \\ &= 1.6 \Gamma_{inh} / [ArOH]_0 \end{aligned}$$

Example: **PMHC** (2,2,5,7,8-pentamethyl-6-chromanol)

A solution containing  $6.0 \times 10^{-2}$  M AIBN, 0.946 M (purified)  $\gamma$ TH<sub>2</sub> and  $6.54 \times 10^{-2}$  M mesitylene was prepared in acetonitrile. PMHC was added at a final concentration of  $9.74 \times 10^{-5}$  M,  $1.95 \times 10^{-4}$  M,  $3.9 \times 10^{-4}$  M and  $5.85 \times 10^{-4}$  M. Three aliquots of 400  $\mu$ L of the first solution ( $[PMHC] = 9.74 \times 10^{-5}$  M)

were added to three vials of 5.2 mL. The vials were kept open to the air for 20 min, thereafter they were capped and placed in a shaker at 37 °C with or without shaking. At given time intervals, aliquots of the above solution (2 - 3  $\mu$ L) were withdrawn with a syringe and immediately diluted with acetonitrile 1:50 v/v, then, 1  $\mu$ L of the final solution was injected in a HP5890A GC-MS instrument. The area of the peak of Cy was divided by the area of the mesitylene peak and the ratios A<sub>1</sub>, A<sub>2</sub> and A<sub>3</sub> are reported in Table 1. These operations were repeated for each [PMHC].

**Table 1s.** Ratios of the peak area for pCy divided by the peak area of mesitylene.

[PMHC] = 9.74x10 <sup>-5</sup> M	Time/min	A1	A2	A3
	0	1.45	1.43	1.47
	114	1.92	1.95	1.955
	160	2.149	2.274	2.31
	255	2.84	3.16	3.14

$$\text{Initial rate, } R_{p1} = (1.92 - 1.45) \times 6.54 \times 10^{-2} / (60 \times 114) = 4.49 \times 10^{-6} \text{ M/s}$$

$$R_{p2} = (1.95 - 1.43) \times 6.54 \times 10^{-2} / (60 \times 114) = 4.97 \times 10^{-6} \text{ M/s}$$

$$R_{p3} = (1.955 - 1.47) \times 6.54 \times 10^{-2} / (60 \times 114) = 4.64 \times 10^{-6} \text{ M/s}$$

where  $6.54 \times 10^{-2}$  M is the concentration of mesitylene.  $R_p^0$  was determined from kinetics without PMHC. The average value was  $10.1 \times 10^{-6} \pm 4.0 \times 10^{-6}$  M/s, see **Table 4s**. In the presence of PMHC,  $R_p < 10.1 \times 10^{-6}$  M/s. Now, it is possible to calculate the rate constants nk<sub>9</sub>.

$$\Gamma_{inh-1} = R_p^0 / R_{inh}^0 - R_{inh}^0 / R_p^0 = 10.1 / 4.49 - 4.49 / 10.1 = 1.805$$

$$nk_{inh-1} = 1.805 \times 1.6 / 9.74 \times 10^{-5} = \mathbf{3.0 \times 10^4 \text{ M}^{-1}\text{s}^{-1}}$$

$$\Gamma_{inh-2} = R_p^0 / R_{inh}^0 - R_{inh}^0 / R_p^0 = 10.1 / 4.97 - 4.97 / 10.1 = 1.540$$

$$nk_{inh-2} = 1.540 \times 1.6 / 9.74 \times 10^{-5} = \mathbf{2.5 \times 10^4 \text{ M}^{-1}\text{s}^{-1}}$$

$$\Gamma_{inh-3} = R_p^0 / R_{inh}^0 - R_{inh}^0 / R_p^0 = 10.1 / 4.64 - 4.64 / 10.1 = 1.717$$

$$nk_{inh-3} = 1.717 \times 1.6 / 9.74 \times 10^{-5} = \mathbf{2.8 \times 10^4 \text{ M}^{-1}\text{s}^{-1}}$$

**Table 2s.** Ratios of the peak area for pCy divided by the peak area of mesitylene.

[PMHC] = 1.95x10 <sup>-4</sup> M	Time/min	A1	A2	A3
	0	1.35	1.40	1.39
	93	1.51	1.525	1.50
	178	1.659	1.71	1.72
	272	2.268	2.29	2.27

$$R_{p1} = (1.51-1.35) \times 6.54 \times 10^{-2} / (60 \times 93) = 1.9 \times 10^{-6} \text{ M/s}$$

$$R_{p2} = (1.525-1.40) \times 6.54 \times 10^{-2} / (60 \times 93) = 1.52 \times 10^{-6} \text{ M/s}$$

$$R_{p3} = (1.50-1.39) \times 6.54 \times 10^{-2} / (60 \times 93) = 1.29 \times 10^{-6} \text{ M/s}$$

$$\Gamma_{inh-1} = R_p^0 / R_{inh}^0 - R_{inh}^0 / R_p^0 = 10.1/1.9 - 1.9/10.1 = 5.128$$

$$nk_{inh-1} = 5.128 \times 1.6 / 1.95 \times 10^{-4} = \mathbf{4.2 \times 10^4 \text{ M}^{-1}\text{s}^{-1}}$$

$$\Gamma_{inh-2} = R_p^0 / R_{inh}^0 - R_{inh}^0 / R_p^0 = 10.1/1.52 - 1.52/10.1 = 6.494$$

$$nk_{inh-2} = 6.494 \times 1.6 / 1.95 \times 10^{-4} = \mathbf{5.3 \times 10^4 \text{ M}^{-1}\text{s}^{-1}}$$

$$\Gamma_{inh-3} = R_p^0 / R_{inh}^0 - R_{inh}^0 / R_p^0 = 10.1/1.29 - 1.29/10.1 = 7.702$$

$$nk_{inh-3} = 7.702 \times 1.6 / 1.95 \times 10^{-4} = \mathbf{6.3 \times 10^4 \text{ M}^{-1}\text{s}^{-1}}$$

### 1. Inhibition Rate Constants $nk_{inh}$ in Acetonitrile at 37 °C

**Table 3s.** Rate constants of PMHC as a function of [PMHC]. Initial rates.

[PMHC]	$nk_{inh}/10^4 \text{ M}^{-1}\text{s}^{-1}$	Initial Rate M/s
$9.74 \times 10^{-5}$	3.0	$4.49 \times 10^{-6}$
"	2.5	$4.97 \times 10^{-6}$
"	2.8	$4.64 \times 10^{-6}$
$1.95 \times 10^{-4}$	4.2	$1.90 \times 10^{-6}$
"	5.3	$1.52 \times 10^{-6}$
"	6.3	$1.29 \times 10^{-6}$
$3.9 \times 10^{-4}$	3.1	$1.31 \times 10^{-6}$
"	3.8	$1.07 \times 10^{-6}$
"	3.1	$1.31 \times 10^{-6}$
$5.85 \times 10^{-4}$	2.7	$1.0 \times 10^{-6}$
"	3.0	$0.91 \times 10^{-6}$
Average $\mathbf{3.6 \pm 1.2}$		

### 4-Methylcatechol

[4MeCat] / mM	Initial Rate $\times 10^6$ M/s	$nk_{inh}$ , rate constants / $\text{M}^{-1}\text{s}^{-1}$
$9.62 \times 10^{-2}$	$3.5 \pm 0.5$	$4.2 \times 10^4$
0.162	$1.56 \pm 0.5$	$6.2 \times 10^4$
0.206	$1.03 \pm 0.5$	$7.5 \times 10^4$
0.326	$0.55 \pm 0.5$	$9.0 \times 10^4$
Average $\mathbf{7 \times 10^4 \pm 2 \times 10^4}$		

### Quercetin

[quercetin] / mM	Initial Rate $\times 10^6$ M/s	$nk_{inh}$ , rate constants / $\text{M}^{-1}\text{s}^{-1}$
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8.91E-02	7 ± 0.7	1.3x10 <sup>4</sup>
1.71E-01	4 ± 0.7	2.0x10 <sup>4</sup>
2.40E-01	2.4 ± 0.7	4.0x10 <sup>4</sup>
3.43E-01	1.2 ± 0.7	3.9x10 <sup>4</sup>
5.14E-01	0.49 ± 0.20	6.4x10 <sup>4</sup>
		Average <b>3.5x10<sup>4</sup> ± 2.0x10<sup>4</sup></b>

### Catechol

[catechol] / mM	Initial Rate x 10 <sup>6</sup> M/s	nk <sub>inh</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
1.06E-01	6 ± 0.8	1.6x10 <sup>4</sup>
2.07E-01	3.3 ± 0.8	2.1x10 <sup>4</sup>
3.38E-01	2.6 ± 0.8	1.7x10 <sup>4</sup>
5.80E-01	2.14 ± 0.8	1.2x10 <sup>4</sup>
		Average <b>1.7 x10<sup>4</sup> ± 0.4x10<sup>4</sup></b>

### Caffeic acid

[caffeic acid] / mM	Initial Rate x 10 <sup>6</sup> M/s	nk <sub>9</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
1.19E-01	7 ± 0.7	1.0x10 <sup>4</sup>
2.39E-01	4.5 ± 0.7	3.8x10 <sup>4</sup>
4.77E-01	2.5 ± 0.7	1.3x10 <sup>4</sup>
5.97E-01	2.1 ± 0.7	1.2x10 <sup>4</sup>
7.16E-01	1.9 ± 0.7	1.1x10 <sup>4</sup>
		Average <b>1.15x10<sup>4</sup> ± 0.13x10<sup>4</sup></b>

### di-Hydrocaffeic acid

[di-hydrocaffeic acid] / mM	Initial Rate x 10 <sup>6</sup> M/s	nk <sub>9</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
0.120	4.52 ± 0.4	2.4x10 <sup>4</sup>
0.48	2.5 ± 0.4	1.3x10 <sup>4</sup>
0.721	2.3 ± 0.4	0.92x10 <sup>4</sup>
		Average 1.5x10 <sup>4</sup> ± 0.8x10 <sup>4</sup>

### Ascorbyl palmitate

[ascorbyl palmitate] / mM	Initial Rate x 10 <sup>6</sup> M/s	nk <sub>inh</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
0.418	0.4 ± 0.3	10.0x10 <sup>4</sup>
0.224	1.0 ± 0.7	7.1 x10 <sup>4</sup>
0.596	0.2 ± 0.1	13.6x10 <sup>4</sup>
0.294	0.8 ± 0.7	6.8x10 <sup>4</sup>
0.0735	2.5 ± 0.7	8.3x10 <sup>4</sup>
		Average <b>9.2x10<sup>4</sup> ± 2.5x10<sup>4</sup></b>

### Luteolin

[luteolin] / mM	Initial Rate x 10 <sup>6</sup> M/s	nk <sub>9</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
0.0916	6.4 ± 0.3	1.6 x10 <sup>4</sup>
0.174	6.3 ± 0.3	0.90x10 <sup>4</sup>
0.521	1.9 ± 0.3	1.6x10 <sup>4</sup>
		Average 1.4x10 <sup>4</sup> ± 0.2 x10 <sup>4</sup>

### 6-Methoxyluteolin (nepetin)

[6-methoxyluteolin] / mM	Initial Rate x 10 <sup>6</sup> M/s	nk <sub>9</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
0.0556	6.3 ± 0.3	2.8 x10 <sup>4</sup>
0.156	3.7 ± 0.3	2.4 x10 <sup>4</sup>
0.487	1.3 ± 0.3	2.5 x10 <sup>4</sup>
		Average 2.6x10 <sup>4</sup> ± 0.3 x10 <sup>4</sup>

**(+)-Catechin hydrate • H<sub>2</sub>O.** We determined via NMR that in our sample the water content was 6 mol H<sub>2</sub>O per mol of catechin, i.e. (+)-catechin • 6H<sub>2</sub>O

[catechin] / mM	Initial Rate x 10 <sup>6</sup> M/s	nk <sub>inh</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
0.166	4.3 ± 0.5	1.9 x10 <sup>4</sup>
0.066	3.6 ± 0.5	5.9 x10 <sup>4</sup>
0.400	2.9 ± 0.5	1.3 x10 <sup>4</sup>
0.249	2.9 ± 0.5	2.1x10 <sup>4</sup>
0.125	4.4 ± 0.5	2.4x10 <sup>4</sup>
		Average 2.7x10 <sup>4</sup> ± 0.4 x10 <sup>4</sup>

### Protocatechuic acid

[protocatechuic acid] / mM	Initial Rate x 10 <sup>6</sup> M/s	Nk <sub>inh</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
0.345	9.6 ± 0.9	0.5x10 <sup>3</sup>
0.645	5.35 ± 0.8	3.4 x10 <sup>3</sup>
0.115	9.3 ± 0.8	2.3 x10 <sup>3</sup>
		Average 2.8x10 <sup>3</sup> ± 0.8 x10 <sup>3</sup>

### 3-Hydroxyflavone

[3-hydroxyflavone] / mM	Initial Rate x 10 <sup>6</sup> M/s	Nk <sub>inh</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
0.354	6.3 ± 0.5	4.4 x10 <sup>3</sup>
0.637	7.8 ± 0.5	1.3 x10 <sup>3</sup>
		Average 2.9x10 <sup>3</sup> ± 1.0 x10 <sup>3</sup>

### Galangin

[galangin] / mM	Initial Rate x 10 <sup>6</sup> M/s	Nk <sub>inh</sub> , rate constants /M <sup>-1</sup> s <sup>-1</sup>
0.115	9.6± 0.5	1.4 x10 <sup>3</sup>

0.604	$8.1 \pm 0.5$	$1.2 \times 10^3$
		Average $1.3 \times 10^3 \pm 0.1 \times 10^3$

#### 4-Methoxyphenol

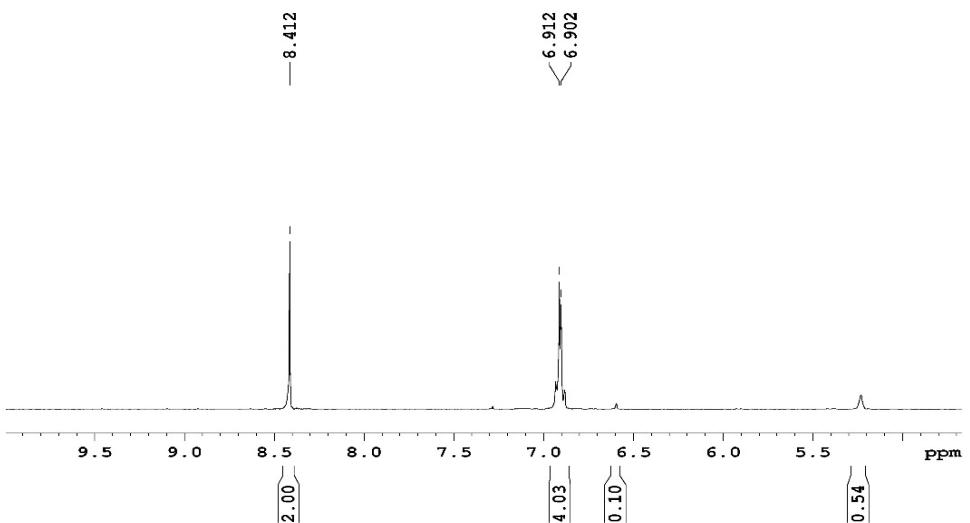
[4-methoxyphenol] / mM	Initial Rate x $10^6$ M/s	$n k_{inh}$ , rate constants /M $^{-1}$ s $^{-1}$
0.0958	$7.5 \pm 0.7$	$1.0 \times 10^4$
0.383	$6.0 \pm 0.7$	$4.6 \times 10^3$
0.192	$4.0 \pm 0.7$	$1.8 \times 10^4$
		Average $1.1 \times 10^4 \pm 0.6 \times 10^4$

#### Phenol

[phenol] / mM	Initial Rate x $10^6$ M/s	$n k_{inh}$ , rate constants /M $^{-1}$ s $^{-1}$
0.720	$8.0 \pm 0.8$	$1.0 \times 10^3$
0.720	$7.9 \pm 0.8$	$1.1 \times 10^3$
1.077	$7.1 \pm 0.8$	$1.1 \times 10^3$
		Average $1.1 \times 10^3 \pm 0.1 \times 10^3$

#### Pinobanksin

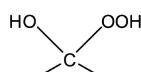
[pinobanksin] / mM	Initial Rate x $10^6$ M/s	$n k_{inh}$ , rate constants /M $^{-1}$ s $^{-1}$
0.1113	no effect	---
0.223	$9.8 - 10.1$	$\leq 10^2$



**Figure 1s.**  $^1\text{H}$  NMR of the solution of  $\gamma\text{-TH}_2$  ( $0.898 \text{ M}$ ) + AIBN ( $6.00 \times 10^{-2} \text{ M}$ ) + mesitylene ( $2.18 \times 10^{-2} \text{ M}$ ) in acetonitrile after 24 hours of reaction at  $37^\circ\text{C}$ . The spectrum was done by mixing  $50 \mu\text{L}$  of the final solution to  $500 \mu\text{L}$  of  $\text{CDCl}_3$ . The singlet at  $8.41 \text{ ppm}$  was due to  $\text{HOOH}$  while the aromatic moiety of  $p$ -cymene gave rise to the multiplet at  $6.91 \text{ ppm}$ . The broad signal at  $5.25 \text{ ppm}$  was assigned to the protons of the residual  $\gamma\text{-TH}_2$ . Finally, the peak at  $6.6 \text{ ppm}$  was due to mesitylene (the internal standard). The peaks of Cy and  $\text{HOOH}$  were in the area ratio of  $4:2$  corresponding to a molar ratio of  $1:1$ .

The final products of reaction 1 were identified by NMR to be  $\text{H}_2\text{O}_2$  and Cy in  $1:1$  mole ratio. The proton signal of  $\text{H}_2\text{O}_2$  (in  $\text{CDCl}_3 + \text{CH}_3\text{CN}$   $10:1$  v/v) appeared, rather variably, at a chemical shift of *ca.*  $8.41 \text{ ppm}$  as a sharp singlet that disappeared after treatment with  $\text{D}_2\text{O}$  or methanol- $d_4$ . Occasionally, a weaker singlet, shifted by  $\pm 0.02 \text{ ppm}$  from the  $\text{H}_2\text{O}_2$  main peak, was observed. Most likely, this peak was generated by the mono-deuterated form, HOOD.<sup>1</sup> Addition of a genuine sample of  $\text{H}_2\text{O}_2$  increased the area of the main peak and hence confirmed the presence of  $\text{H}_2\text{O}_2$  in the reaction mixture. The aromatic signals of Cy appeared at *ca.*  $6.91 \text{ ppm}$ . The area of the Cy peak was in a  $2:1$  ratio with that of  $\text{H}_2\text{O}_2$  i.e., in a molar ratio of  $1:1$  as dictated by reaction 1. The presence of Cy was confirmed by comparing the spectrum of the mixture with that of an authentic sample of Cy.

In acetone- $d_6$ , the proton spectrum of  $H_2O_2$  showed one singlet at ca. 9.50 ppm due to  $H_2O_2$  and two more peaks at 10.09 ppm and 5.04 ppm both growing over time at the expense of  $H_2O_2$ . These two peaks disappeared after treatment with heavy water,  $D_2O$ . The carbon spectrum showed the presence of a peak at 101.05 ppm which disappeared in the DEPT-135 spectrum, proof that this resonance was generated by a quaternary C atom. When we changed the solvent to acetonitrile- $d_3$ ,  $CH_3OH$  (external lock with  $D_2O$  in a coaxial insert) or chloroform- $d_1$ , all these peaks, except for the  $H_2O_2$  signal, did not appear. Hence, we think that these resonances are due to 2-



hydroperoxypropan-2-ol- $d_6$  produced by the reaction of  $H_2O_2$  with  $(CD_3)_2CO$ .<sup>2</sup> As a further proof, the spectra of  $H_2O_2$  in a mix  $(CH_3)_2CO/(CD_3)_2CO$  3:1 v/v showed an additional proton peak at 1.33 ppm for the methyl groups in  $(CH_3)_2C(OOH)(OH)$  and the corresponding carbon peak at 24.8 ppm. Finally, we observed that reaction 1 was faster in acetone than in other solvents, most likely because of the presence of this hydroperoxide.

(1) Double Water Peaks in Deuterated NMR Solvents (Sigma-Aldrich)

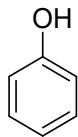
[https://www.sigmaaldrich.com/content/dam/sigma-aldrich/docs/Aldrich/General\\_Information/double\\_water\\_peaks.pdf](https://www.sigmaaldrich.com/content/dam/sigma-aldrich/docs/Aldrich/General_Information/double_water_peaks.pdf)

(2) Sauer, M. C. V.; Edwards, J. O. *J. Phys. Chem.* **1971**, 75, 3004-3011.

**Table 4s.** Initial rates in the absence of inhibitor  $R_p^0$  (in  $10^{-5} M/s$  units) measured in vials of 5.2 ml containing aliquots of the standard solution at 37 °C. The average value was calculated using the data 5.2/200+400+600+800+shaken 5.2/200 yielding  $1.01 \pm 0.4$ .

ml vial volume/ $\mu L$ solution								
5.2/200	5.2/400	5.2/600	5.2/800	18.0/720	18.0/1650		shaken/40.0/650	shaken/5.2/200
0.99 ± 0.08	1.82 ± 0.10	0.42 ± 0.09	0.84 ± 0.07	2.63 ± 0.20	0.56 ± 0.06		1.07 ± 0.05	1.49 ± 0.08
0.92 ± 0.05	1.71 ± 0.12	0.76 ± 0.06	0.57 ± 0.08	2.53 ± 0.06	0.13 ± 0.07		2.06 ± 0.05	1.19 ± 0.10
1.17 ± 0.1	1.96 ± 0.10	0.77 ± 0.06	1.01 ± 0.06	2.73 ± 0.12	0.42 ± 0.08		2.09 ± 0.05	1.27 ± 0.06
1.03 ± 0.03	0.43 ± 0.10		0.37 ± 0.02	2.80 ± 0.05	0.62 ± 0.07		1.53 ± 0.03	1.20 ± 0.10
0.99 ± 0.05	0.38 ± 0.08		0.63 ± 0.03				1.85 ± 0.02	0.39 ± 0.09
1.14 ± 0.04	0.35 ± 0.08		0.49 ± 0.03					
1.04 ± 0.08	1.05 ± 0.05							
1.31 ± 0.07	1.10 ± 0.15							
1.11 ± 0.10	1.41 ± 0.16							
1.22 ± 0.10	1.61 ± 0.07							
1.15 ± 0.04								
1.12 ± 0.08								
0.99 ± 0.05								
averages								
<b>1.091</b>	<b>1.182</b>	<b>0.65</b>	<b>0.65</b>	<b>2.67</b>	<b>0.43</b>		<b>1.72</b>	<b>1.11</b>
± 0.109	± 0.62	± 0.20	± 0.24	± 0.12	± 0.22		± 0.43	± 0.42

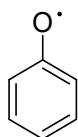
B3LYP/6-31+g(d,p) SRCF=(PCM, solvent =acetonitrile) optimized structures and energies



Sum of electronic and thermal Enthalpies: -307.389836

0 1

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C	1.21954500	0.25282800	0.00000000
C	1.22276100	-1.14384200	0.00000000
C	0.02053400	-1.86102500	0.00000000
C	-1.19174000	-1.16367100	0.00000000
C	-1.20929700	0.23439900	0.00000000
H	2.14769200	0.81581900	0.00000000
H	2.17186400	-1.67225900	0.00000000
H	0.02872700	-2.94642800	0.00000000
H	-2.13318800	-1.70550200	0.00000000
H	-2.15369100	0.77268600	0.00000000
O	0.05143300	2.31251500	0.00000000
H	-0.84368300	2.68135200	0.00000000



Sum of electronic and thermal Enthalpies: -306.760624

0 2

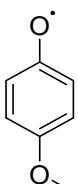
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C	-1.24271500	0.29086100	0.00000000
C	-1.22825500	-1.08778900	0.00000000
C	0.00002600	-1.78644600	0.00000000
C	1.22824300	-1.08784600	0.00000000
O	0.00004400	2.30913600	0.00000000
H	-2.17160200	0.85269800	0.00000000
H	-2.16007400	-1.64481500	0.00000000
H	0.00004100	-2.87185300	0.00000000
H	2.16004800	-1.64489900	0.00000000
H	2.17165700	0.85246700	0.00000000



Sum of electronic and thermal Enthalpies: -421.883095

0 1

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C	1.52916200	0.91323100	0.00000000
C	1.30743500	-0.46047200	0.00000000
C	0.00000000	-0.97256700	0.00000000
C	-1.08365300	-0.08805300	0.00000000
C	-0.85696100	1.29606100	0.00000000
H	2.54048100	1.30776500	0.00000000
H	2.14378200	-1.15256700	0.00000000
O	-0.11001300	-2.34162100	0.00000000
H	-2.10488400	-0.44919900	0.00000000
H	-1.70600700	1.97475300	0.00000000
O	0.72565400	3.14809700	0.00000000
H	-0.09558600	3.65971800	0.00000000
C	-1.41827700	-2.91344300	0.00000000
H	-1.27152400	-3.99373000	0.00000000
H	-1.97957000	-2.62162600	0.89540000
H	-1.97957000	-2.62162600	-0.89540000

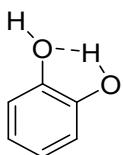


Sum of electronic and thermal Enthalpies: -421.263950

0 2

C	0.87729000	-1.34402300	0.00000000
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C	-1.56162500	-0.94687200	0.00000000
C	-1.33166000	0.40366200	0.00000000
C	0.00000000	0.90677100	0.00000000
C	1.10183800	0.01533200	0.00000000
O	-0.66460100	-3.14765200	0.00000000
O	0.10796800	2.24606300	0.00000000
C	1.40935400	2.86110900	0.00000000
H	-2.57240000	-1.34268500	0.00000000
H	-2.14855800	1.11844200	0.00000000
H	2.11666200	0.39414000	0.00000000
H	1.70918900	-2.04149800	0.00000000
H	1.22074700	3.93332600	0.00000000

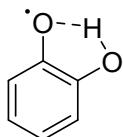
H        1.96741700    2.58137900    0.89789900  
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Sum of electronic and thermal Enthalpies: -382.612745

0 1

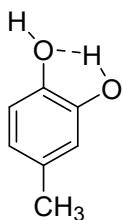
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C        0.00000000    0.86255900    0.00000000  
C        -0.84506100    -0.26077100    0.00000000  
C        -0.29534800    -1.54147400    0.00000000  
C        1.09494000    -1.70574800    0.00000000  
C        1.93377500    -0.58825000    0.00000000  
O        -0.64365400    2.07908700    0.00000000  
O        -2.20710100    -0.10570900    0.00000000  
H        -0.96316500    -2.39750100    0.00000000  
H        1.51604800    -2.70622100    0.00000000  
H        3.01235400    -0.70967000    0.00000000  
H        2.02718900    1.57661100    0.00000000  
H        -0.00774300    2.80802300    0.00000000  
H        -2.41334400    0.84246000    0.00000000



Sum of electronic and thermal Enthalpies: -381.994869

0 2

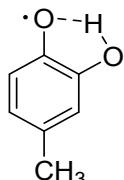
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C        -0.35793200    -1.55629000    0.00000000  
C        -0.93046700    -0.23259400    0.00000000  
C        0.00000000    0.90266800    0.00000000  
C        1.37985700    0.71489400    0.00000000  
C        1.87572700    -0.58725200    0.00000000  
O        -2.16910900    0.01999500    0.00000000  
O        -0.54566800    2.12436200    0.00000000  
H        2.04381200    1.57288900    0.00000000  
H        2.94917700    -0.74664900    0.00000000  
H        1.43825000    -2.71301900    0.00000000  
H        -1.03736200    -2.40257400    0.00000000  
H        -1.51767900    1.98957800    0.00000000



Sum of electronic and thermal Enthalpies: -421.903823

0 1

C	1.15696000	1.08942300	0.00000000
C	1.18255100	-0.31659500	0.00000000
C	0.00000000	-1.05124600	0.00000000
C	-1.24404100	-0.39750800	0.00000000
C	-1.28167800	0.99302900	0.00000000
C	-0.08903500	1.72830900	0.00000000
O	-0.06090700	-2.42682500	0.00000000
O	-2.41329400	-1.11736600	0.00000000
H	-2.24554300	1.49305800	0.00000000
H	-0.13671400	2.81359700	0.00000000
C	2.44872300	1.87556200	0.00000000
H	2.13406900	-0.84438100	0.00000000
H	0.82596100	-2.81317900	0.00000000
H	-2.19604600	-2.06285400	0.00000000
H	2.25439800	2.95170600	0.00000000
H	3.05830100	1.64487500	0.88155100
H	3.05830100	1.64487500	-0.88155100

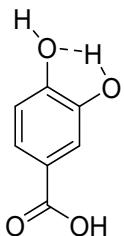


Sum of electronic and thermal Enthalpies: -421.290101

0 2

C	0.13349400	-1.69997200	0.00000000
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C	-1.17985600	0.36636300	0.00000000
C	0.00000000	1.09331700	0.00000000
C	1.31132500	0.43803100	0.00000000
C	1.31719600	-1.00176500	0.00000000
C	-2.40911600	-1.83363400	0.00000000
O	0.00281200	2.43520500	0.00000000
O	2.34306300	1.17042200	0.00000000
H	2.27814600	-1.50634400	0.00000000
H	0.14890500	-2.78582900	0.00000000
H	-2.13186000	0.88877900	0.00000000
H	0.94661500	2.70299100	0.00000000
H	-2.21400400	-2.90826800	0.00000000

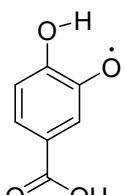
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Sum of electronic and thermal Enthalpies: -571.189004

0 1

C 0.00000000 0.91863800 0.00000000  
C 0.99344500 -0.07874600 0.00000000  
C 0.63031100 -1.41663200 0.00000000  
C -0.73200900 -1.78069900 0.00000000  
C -1.71697800 -0.79231600 0.00000000  
C -1.35528700 0.55368800 0.00000000  
O 1.50889800 -2.47048500 0.00000000  
O -1.09364200 -3.09149200 0.00000000  
H -2.75987400 -1.09127700 0.00000000  
H -2.12318100 1.31740500 0.00000000  
C 0.43563000 2.33074000 0.00000000  
H 2.04014200 0.20868700 0.00000000  
H 2.42728500 -2.16573700 0.00000000  
H -0.29368400 -3.64246400 0.00000000  
O -0.59246200 3.21331200 0.00000000  
O 1.60112900 2.70569100 0.00000000  
H -0.21275100 4.10914400 0.00000000

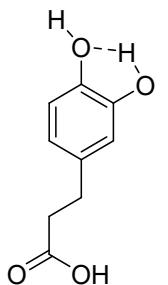


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0 2

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C 0.78632000 -1.43744200 0.00000000  
C -0.61654400 -1.86513300 0.00000000  
C -1.65708700 -0.93876600 0.00000000  
C -1.35123200 0.41711000 0.00000000  
O 1.68046700 -2.32974500 0.00000000  
O -0.84599700 -3.17723900 0.00000000  
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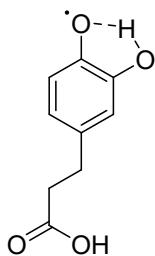
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O	-0.77862700	3.11575100	0.00000000
O	1.45120000	2.79160900	0.00000000
H	-0.48993300	4.04498700	0.00000000



Sum of electronic and thermal Enthalpies: -649.761437

0 1

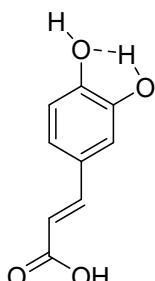
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C	-0.93578400	0.87560100	-0.40512300
C	-2.28677100	0.80301900	-0.07317100
C	-2.88786400	-0.43926200	0.18827100
C	-2.12119700	-1.59890600	0.11108100
C	-0.76438500	-1.52176200	-0.22290100
O	-3.12468600	1.89105000	0.01867300
O	-4.21900800	-0.51819000	0.51042100
H	-2.59439800	-2.55607800	0.30785600
H	-0.18088300	-2.43628700	-0.28422000
C	1.32305500	-0.20002800	-0.81506500
H	-0.49088700	1.84732600	-0.60952200
H	-2.65170800	2.71121300	-0.17979600
H	-4.59304400	0.37699100	0.51188300
H	1.64364200	-1.11894400	-1.31620300
H	1.50016900	0.62094400	-1.51759600
C	2.18966500	0.01617000	0.43577300
C	3.66778200	0.10785200	0.14007800
H	1.90046700	0.93500100	0.96040300
H	2.04552000	-0.79648700	1.15795800
O	4.39418700	0.28220400	1.26789300
O	4.18662700	0.03938100	-0.96016100
H	5.33444300	0.33537500	1.02074500



Sum of electronic and thermal Enthalpies: -649.145630

0 2

C	-0.19939600	-0.09791400	-0.49262000
C	-1.02971100	1.01128700	-0.29496900
C	-2.36848600	0.81875900	0.02353400
C	-2.94456600	-0.52229200	0.15723100
C	-2.05644100	-1.63632900	-0.05775400
C	-0.73776700	-1.41924200	-0.36991800
O	-3.20059900	1.85001700	0.22034300
O	-4.17230000	-0.61583100	0.44579400
H	-2.46512000	-2.63789600	0.03009200
H	-0.07782200	-2.26607700	-0.53637900
C	1.26375100	0.08581100	-0.81250300
H	-0.64125000	2.02076100	-0.38942100
H	-4.07595500	1.45713600	0.42584900
H	1.57158600	-0.65073300	-1.56235300
H	1.42885200	1.07651700	-1.24490200
C	2.14822900	-0.07474500	0.43692500
C	3.62176700	0.08210900	0.14055800
H	1.88139200	0.66155500	1.20398000
H	2.00686200	-1.05770900	0.90128700
O	4.37004200	-0.05727700	1.25668500
O	4.11291200	0.30563000	-0.95095500
H	5.30672000	0.05146600	1.01400000

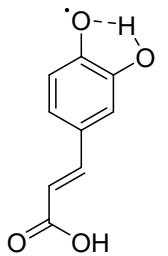


Sum of electronic and thermal Enthalpies: -648.563853

0 1

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C	-0.10566200	-0.47584700	-0.00001400
C	-0.76729600	0.77399600	0.00002400
C	-2.15129200	0.83308500	0.00001100
C	-2.91926300	-0.35202800	-0.00001000

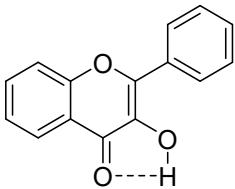
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C	1.34506800	-0.60353300	0.00000600
C	2.26696300	0.38640700	-0.00003100
C	3.69854300	0.07373800	0.00000600
O	4.20740400	-1.04209400	0.00009500
O	-2.88714000	1.99089600	0.00003800
O	-4.27880700	-0.29116800	-0.00002700
O	4.45957600	1.20021800	-0.00007100
H	-2.87867100	-2.49307900	-0.00004100
H	-0.39373700	-2.61619800	-0.00003100
H	-0.20129400	1.70060800	0.00003000
H	-2.31657500	2.77237000	0.00002500
H	-4.55506900	0.63992200	0.00015200
H	1.72298800	-1.62463900	0.00004300
H	2.00257500	1.43815000	-0.00007600
H	5.39376500	0.92898100	-0.00003800



Sum of electronic and thermal Enthalpies: -647.946192

0 2

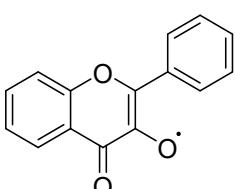
C	0.94591500	1.65057800	0.00025600
C	0.15122200	0.45074000	-0.00006200
C	0.79234900	-0.80923100	-0.00027000
C	2.16964600	-0.87121900	-0.00010500
C	3.00396300	0.34369200	0.00012100
C	2.31299900	1.61237100	0.00038800
C	-1.28756000	0.60213600	-0.00022200
C	-2.21150700	-0.39158400	0.00007600
C	-3.65008400	-0.07438400	-0.00006400
O	-4.14129700	1.04597500	-0.00077500
O	2.82531300	-2.03969200	-0.00026600
O	4.25239500	0.20387900	0.00005600
O	-4.40606800	-1.19722900	0.00077300
H	2.91106400	2.51752000	0.00065100
H	0.43165700	2.60691800	0.00045000
H	0.22426400	-1.73252600	-0.00053700
H	3.78098400	-1.82051300	-0.00026700
H	-1.66121200	1.62374400	-0.00045800
H	-1.94814600	-1.44317400	0.00048600
H	-5.34302200	-0.93401700	0.00066200



Sum of electronic and thermal Enthalpies: -803.125400

0 1

O	0.81812200	-0.50606000	0.00000000
C	0.33485500	-1.77776600	0.00000000
C	-1.04437700	-2.04513800	0.00000000
C	-1.96368500	-0.91939700	0.00000000
C	-1.35999100	0.40552100	0.00000000
C	0.00000000	0.59222100	0.00000000
O	-3.21179300	-1.00609900	0.00000000
O	-2.25114600	1.43062600	0.00000000
C	0.75786200	1.85223000	0.00000000
C	1.27989600	-2.81340500	0.00000000
C	0.83041200	-4.12626800	0.00000000
C	-0.54980500	-4.41887700	0.00000000
C	-1.47566400	-3.38896400	0.00000000
H	2.33709600	-2.57128200	0.00000000
H	1.55320700	-4.93607400	0.00000000
H	-0.88240900	-5.45168000	0.00000000
H	-2.54180800	-3.58895000	0.00000000
H	-3.12814800	0.98967600	0.00000000
C	2.16961700	1.81626700	0.00000000
C	2.91195600	2.99555500	0.00000000
C	2.26801300	4.23738500	0.00000000
C	0.87134600	4.28570900	0.00000000
C	0.11844500	3.11073500	0.00000000
H	2.68360900	0.86312200	0.00000000
H	3.99656300	2.94335800	0.00000000
H	2.84832200	5.15514200	0.00000000
H	0.35942800	5.24337300	0.00000000
H	-0.96059600	3.17071000	0.00000000

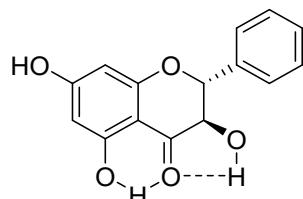


Sum of electronic and thermal Enthalpies: -802.498975

0 2

O	0.75094600	-0.49997300	0.00000000
C	0.22910700	-1.76846200	0.00000000

C	-1.15239000	-2.00085000	0.00000000
C	-2.07280800	-0.86147000	0.00000000
C	-1.44584400	0.52325400	0.00000000
C	0.00000000	0.62099000	0.00000000
O	-3.29823600	-0.98027600	0.00000000
O	-2.19217200	1.52512000	0.00000000
C	0.79845300	1.83577400	0.00000000
C	1.15481400	-2.81462900	0.00000000
C	0.68063700	-4.12266400	0.00000000
C	-0.70139400	-4.38640000	0.00000000
C	-1.60706300	-3.33491000	0.00000000
H	2.21600100	-2.59181000	0.00000000
H	1.38885500	-4.94499700	0.00000000
H	-1.05676100	-5.41142400	0.00000000
H	-2.67678800	-3.51395800	0.00000000
C	2.21433100	1.73583400	0.00000000
C	3.00637600	2.87762400	0.00000000
C	2.41561200	4.14806500	0.00000000
C	1.02144600	4.26305400	0.00000000
C	0.21525100	3.12761000	0.00000000
H	2.68417100	0.76044200	0.00000000
H	4.08739900	2.77933600	0.00000000
H	3.03721900	5.03823200	0.00000000
H	0.55704700	5.24426100	0.00000000
H	-0.86059900	3.22403300	0.00000000

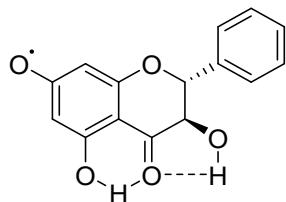


Sum of electronic and thermal Enthalpies: -954.760668

0 1

O	0.00000000	0.88792300	0.00000000
C	1.27783100	0.42391000	0.00000000
C	1.53003500	-0.96208300	0.00000000
C	0.42583500	-1.87322800	0.00000000
C	-0.90551600	-1.29684000	0.00000000
C	-1.09665300	0.06188700	0.00000000
O	0.54666700	-3.13433300	0.00000000
O	-1.92475600	-2.19549500	0.00000000
C	-2.35895600	0.81465900	0.00000000
C	2.30152900	1.36545400	0.00000000
C	3.61752400	0.89353800	0.00000000
C	3.91998000	-0.48123000	0.00000000

C 2.88374200 -1.40492100 0.00000000  
 H 2.08823300 2.42715200 0.00000000  
 O 4.60018100 1.83310600 0.00000000  
 H 4.94889600 -0.82624700 0.00000000  
 O 3.15923900 -2.72592700 0.00000000  
 C -2.32945400 2.22653600 0.00000000  
 C -3.51240700 2.96300800 0.00000000  
 C -4.75101400 2.31304900 0.00000000  
 C -4.79280800 0.91611200 0.00000000  
 C -3.61441000 0.16890100 0.00000000  
 H -1.37881100 2.74498000 0.00000000  
 H -3.46550400 4.04782800 0.00000000  
 H -5.67158400 2.88887300 0.00000000  
 H -5.74806100 0.39976800 0.00000000  
 H -3.66956700 -0.91038100 0.00000000  
 H 5.47414100 1.41557900 0.00000000  
 H 2.29249300 -3.20803800 0.00000000  
 H -1.49241700 -3.07422900 0.00000000

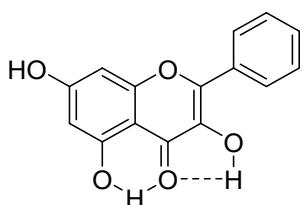


Sum of electronic and thermal Enthalpies: -954.119187

0 2

O 0.00000000 0.90812100 0.00000000  
 C 1.29262400 0.47983800 0.00000000  
 C 1.59691800 -0.89431400 0.00000000  
 C 0.52267300 -1.84179500 0.00000000  
 C -0.82108800 -1.30455500 0.00000000  
 C -1.06402700 0.06038200 0.00000000  
 O 0.67546400 -3.09830000 0.00000000  
 O -1.80487500 -2.22208600 0.00000000  
 C -2.34877500 0.76070600 0.00000000  
 C 2.28141200 1.44833400 0.00000000  
 C 3.68284000 1.03899200 0.00000000  
 C 3.99024900 -0.37378700 0.00000000  
 C 2.98706000 -1.30724300 0.00000000  
 H 2.03843300 2.50431800 0.00000000  
 O 4.58223300 1.92143100 0.00000000  
 H 5.03027800 -0.67918500 0.00000000  
 O 3.26716400 -2.62556300 0.00000000  
 C -2.36690900 2.17473600 0.00000000  
 C -3.57394100 2.86703600 0.00000000

C	-4.78810100	2.17066100	0.00000000
C	-4.78297600	0.77289300	0.00000000
C	-3.58051700	0.06780400	0.00000000
H	-1.43527600	2.72639900	0.00000000
H	-3.56757700	3.95246700	0.00000000
H	-5.72840400	2.71337400	0.00000000
H	-5.71982400	0.22464300	0.00000000
H	-3.59881800	-1.01232200	0.00000000
H	2.40941300	-3.12325700	0.00000000
H	-1.35275900	-3.09338700	0.00000000

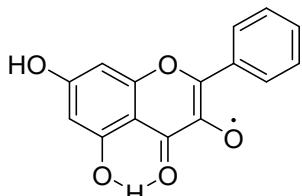


Sum of electronic and thermal Enthalpies: -953.582895

0 1

C	-3.37710300	-0.17030400	1.26037500
C	-2.47345400	-0.24148500	0.19325300
C	-2.92855800	-0.67292100	-1.06212800
C	-4.26675700	-1.02712200	-1.24353600
C	-5.16659500	-0.94705500	-0.17421800
C	-4.71979900	-0.51681200	1.07781800
C	-1.03238100	0.15695700	0.40007800
O	-0.19943600	-0.95800100	-0.02657300
C	1.14945000	-0.78113900	0.00005500
C	1.72180000	0.52004900	-0.03072600
C	0.87721000	1.67030900	-0.16262000
C	-0.60889500	1.41104900	-0.38173500
C	3.14239700	0.63935800	0.01646900
C	3.94921700	-0.49323000	0.05663000
C	3.34020800	-1.75615300	0.05980000
C	1.94460900	-1.91524600	0.04019800
O	1.28529300	2.85143100	-0.13481000
O	-1.37846100	2.52281800	0.02282700
O	4.07113800	-2.89760900	0.09910900
O	3.72229900	1.85469000	0.00468300
H	1.50234500	-2.90349100	0.07178900
H	5.02856500	-0.38689900	0.08271000
H	-2.23668900	-0.74255500	-1.89639900
H	-4.60760500	-1.36438500	-2.21795100
H	-6.20774900	-1.22085900	-0.31731200
H	-5.41090400	-0.45530200	1.91317400
H	-3.03175900	0.15767000	2.23704000

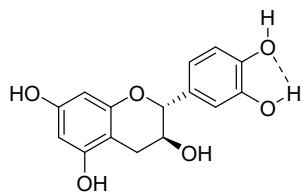
H	5.01924900	-2.69962400	0.12285400
H	3.00155400	2.53160300	-0.02803500
H	-0.78180600	3.29215500	0.00386900
H	-0.84573700	0.33129400	1.46729400
H	-0.74421100	1.21623500	-1.45919300



Sum of electronic and thermal Enthalpies: -1031.153668

0 2

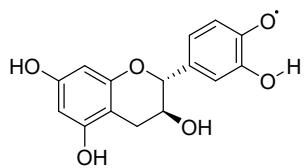
C	-3.34381500	-0.14110800	1.26779300
C	-2.44865900	-0.22247200	0.19434200
C	-2.91694000	-0.64090400	-1.06044700
C	-4.26168200	-0.97256600	-1.23499200
C	-5.15346100	-0.88251100	-0.15977400
C	-4.69276300	-0.46518200	1.09155500
C	-1.00068100	0.14992600	0.39618400
O	-0.18725900	-0.97546700	-0.04120400
C	1.16513200	-0.83887000	-0.00192700
C	1.77627400	0.45695700	-0.03763900
C	0.92727500	1.63163200	-0.16219100
C	-0.55561100	1.39155800	-0.39595500
C	3.20441100	0.57846100	0.02341800
C	3.99193600	-0.55583100	0.06208800
C	3.38055700	-1.86965700	0.06195000
C	1.93368200	-1.97827100	0.03965300
O	1.35882300	2.79380600	-0.10273600
O	-1.31973000	2.51281300	-0.02181300
O	4.09798800	-2.90693900	0.09722800
O	3.79356900	1.78943400	0.02299600
H	1.47942400	-2.96154800	0.06538100
H	5.07272800	-0.48182200	0.09286600
H	-2.23163700	-0.71883700	-1.89937600
H	-4.61374100	-1.30015600	-2.20862800
H	-6.19971400	-1.13872800	-0.29780400
H	-5.37787000	-0.39633500	1.93116400
H	-2.98742800	0.17691100	2.24370600
H	3.09112000	2.47992800	0.00039400
H	-0.73374300	3.28846400	-0.05798400
H	-0.80748200	0.32001500	1.46244200
H	-0.67271400	1.17596400	-1.47227600



Sum of electronic and thermal Enthalpies: -1031.153668

0 1

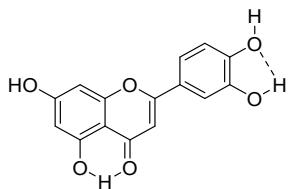
C	-2.73659300	0.15760000	1.03444200
C	-1.77094700	0.02599200	0.02679100
C	-2.16021500	-0.45739700	-1.22924100
C	-3.49089900	-0.80052800	-1.47842900
C	-4.44973300	-0.65868900	-0.47622700
C	-4.06647700	-0.17508400	0.78488100
C	-0.33907200	0.41530600	0.30470700
O	0.48741100	-0.73634400	0.00398700
C	1.85032700	-0.57573100	0.04704100
C	2.45033800	0.69325700	-0.00967800
C	1.60940900	1.94601700	-0.12901300
C	0.16400900	1.61908000	-0.51843800
C	3.85525000	0.73342200	0.03374100
C	4.63216600	-0.42563200	0.10613200
C	3.98978900	-1.66624700	0.14206500
C	2.59778600	-1.75465600	0.11911300
O	-0.68538200	2.75461700	-0.38078100
O	4.68878000	-2.84427400	0.21163600
O	4.54572000	1.91824300	0.00232900
H	2.09936000	-2.71621300	0.15969400
H	5.71426700	-0.34450400	0.13238300
H	-1.42500200	-0.57809500	-2.01825100
H	-3.79794900	-1.17891600	-2.44843100
O	-5.75251300	-0.99506600	-0.72880700
O	-5.08022600	-0.07572600	1.70880800
H	-2.45444700	0.52245500	2.01922000
H	5.64020500	-2.66807200	0.22558400
H	3.94054700	2.67160500	-0.03586500
H	-0.62675400	3.09145900	0.52567000
H	-0.22501100	0.64823800	1.37350300
H	0.11424200	1.35047200	-1.57846200
H	1.59467500	2.49574500	0.82473900
H	2.01722400	2.63115700	-0.88345700
H	-6.27847700	-0.83459500	0.07070000
H	-4.75402100	0.27739700	2.54825600



Sum of electronic and thermal Enthalpies: -1030.536076

0 2

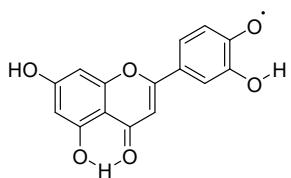
C	-2.79877200	0.30509700	1.01400100
C	-1.80506000	-0.02018700	0.09021000
C	-2.13275200	-0.72804900	-1.10919800
C	-3.42306300	-1.10033100	-1.38757400
C	-4.48715200	-0.78113100	-0.46874600
C	-4.11596600	-0.05649200	0.75084500
C	-0.37771500	0.40035700	0.34616900
O	0.45276700	-0.74674200	0.09025900
C	1.81645500	-0.56445300	0.09513900
C	2.39255700	0.70966500	-0.02812100
C	1.52662700	1.94183300	-0.17919700
C	0.08492600	1.57025500	-0.54296400
C	3.79748700	0.77231800	-0.01496300
C	4.59241300	-0.37174700	0.08947300
C	3.97094500	-1.61924500	0.19183300
C	2.58007300	-1.72902600	0.20179100
O	-0.81379300	2.65206200	-0.31553900
O	4.68909300	-2.78221300	0.29766700
O	4.46940600	1.96375900	-0.10938900
H	2.09671800	-2.69452500	0.29492700
H	5.67334400	-0.27367000	0.09148900
H	-1.33587300	-0.97860300	-1.80137200
H	-3.67832700	-1.63896600	-2.29437900
O	-5.70355600	-1.07887800	-0.63456600
O	-5.10634300	0.23318100	1.60244900
H	-2.56037400	0.83612400	1.92994800
H	5.63791600	-2.59172100	0.29013900
H	3.85389300	2.70858900	-0.14808000
H	-0.57198400	3.38697500	-0.89654400
H	-0.26525700	0.69598400	1.39751600
H	0.03334400	1.24457600	-1.59091900
H	1.49760200	2.52577700	0.75142900
H	1.91875300	2.60292300	-0.96401700
H	-5.92637300	-0.12599800	1.20063100



Sum of electronic and thermal Enthalpies: -1028.812150

0 1

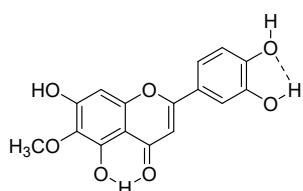
O	0.40327200	-0.74919800	0.00000000
C	-0.50219400	-1.77628200	0.00000000
C	-1.88274800	-1.50711200	0.00000000
C	-2.35097400	-0.13502800	0.00000000
C	-1.32785800	0.87737800	0.00000000
C	0.00000000	0.55187800	0.00000000
O	-3.58640600	0.14005900	0.00000000
C	1.13454000	1.48560900	0.00000000
C	0.01183900	-3.06659000	0.00000000
C	-0.90042200	-4.12818300	0.00000000
C	-2.28832300	-3.91581300	0.00000000
C	-2.77780200	-2.61243600	0.00000000
H	1.07973300	-3.24662800	0.00000000
O	-0.37190900	-5.38206900	0.00000000
H	-2.98260600	-4.75026800	0.00000000
O	-4.10791400	-2.39415900	0.00000000
C	2.45694400	1.00676400	0.00000000
C	3.53492800	1.89055700	0.00000000
C	3.31418400	3.26615100	0.00000000
C	1.99242900	3.75632600	0.00000000
C	0.91815000	2.88058000	0.00000000
H	2.64679400	-0.05886100	0.00000000
H	4.55525600	1.52169100	0.00000000
O	4.37021700	4.12352400	0.00000000
O	1.88454400	5.12336300	0.00000000
H	-0.08486400	3.29262100	0.00000000
H	-1.07473800	-6.04819500	0.00000000
H	-4.22535600	-1.39988800	0.00000000
H	4.03977600	5.03677900	0.00000000
H	0.95901300	5.40585700	0.00000000
H	-1.64359700	1.91195100	0.00000000



Sum of electronic and thermal Enthalpies: -1028.192342

0 2

O	0.45446800	-0.68978600	0.00000000
C	-0.41208400	-1.75018700	0.00000000
C	-1.80410500	-1.53278000	0.00000000
C	-2.32646400	-0.18379900	0.00000000
C	-1.33877500	0.87199300	0.00000000
C	0.00000000	0.59351600	0.00000000
O	-3.56624800	0.05309400	0.00000000
C	1.08799700	1.57438100	0.00000000
C	0.14982700	-3.01863100	0.00000000
C	-0.72223900	-4.11449200	0.00000000
C	-2.11755200	-3.95516300	0.00000000
C	-2.65709800	-2.67281000	0.00000000
H	1.22362100	-3.15846900	0.00000000
O	-0.14760500	-5.34537800	0.00000000
H	-2.77851500	-4.81588600	0.00000000
O	-3.99365300	-2.50785100	0.00000000
C	2.45111800	1.11953600	0.00000000
C	3.49820400	2.00059400	0.00000000
C	3.26028300	3.42399100	0.00000000
C	1.85589400	3.85927900	0.00000000
C	0.81023800	2.95462500	0.00000000
H	2.64540000	0.05474800	0.00000000
H	4.52710100	1.65668300	0.00000000
O	4.15537300	4.30674000	0.00000000
O	1.63870000	5.17927300	0.00000000
H	-0.20309300	3.33637100	0.00000000
H	-0.82382700	-6.03889300	0.00000000
H	-4.15679000	-1.52239000	0.00000000
H	2.52171700	5.60590200	0.00000000
H	-1.69536800	1.89289200	0.00000000

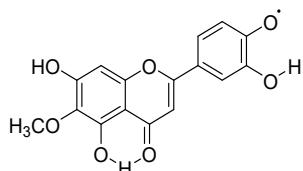


Sum of electronic and thermal Enthalpies: -1143.297935

0 1

C	-3.61385400	0.21802700	0.00000000
C	-2.35281600	0.84870100	0.00000000
C	-2.29657200	2.25366500	0.00000000
C	-3.46827100	3.01093000	0.00000000
C	-4.70922700	2.37759200	0.00000000
C	-4.77877300	0.97042500	0.00000000
C	-1.11706800	0.05123000	0.00000000

O 0.00000000 0.82417100 0.00000000  
 C 1.25174200 0.26582400 0.00000000  
 C 1.41948100 -1.12900900 0.00000000  
 C 0.24626700 -1.98401500 0.00000000  
 C -1.02546900 -1.31294700 0.00000000  
 C 2.73725300 -1.67539300 0.00000000  
 C 3.83586100 -0.81190100 0.00000000  
 C 3.60923100 0.58711100 0.00000000  
 C 2.33119200 1.14009200 0.00000000  
 O 0.34575900 -3.24753600 0.00000000  
 O 4.68001800 1.42030500 0.00000000  
 O 2.89352200 -3.01920800 0.00000000  
 O -5.91509000 3.02001600 0.00000000  
 O -5.98910700 0.33551100 0.00000000  
 H 2.19173700 2.21411200 0.00000000  
 O 5.17157300 -1.12354400 0.00000000  
 H -1.33997200 2.76013500 0.00000000  
 H -3.41650100 4.09609700 0.00000000  
 H -3.71344600 -0.86081900 0.00000000  
 H 5.48587000 0.87392200 0.00000000  
 H 1.96327700 -3.39860200 0.00000000  
 H -5.80378500 3.98151800 0.00000000  
 H -6.69590600 1.00035200 0.00000000  
 H -1.91616800 -1.92639000 0.00000000  
 C 5.65682200 -2.47906300 0.00000000  
 H 6.74274000 -2.38368300 0.00000000  
 H 5.32697800 -3.01098300 0.89400400  
 H 5.32697800 -3.01098300 -0.89400400

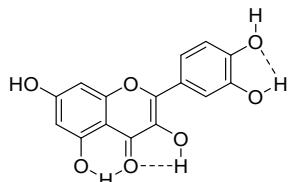


Sum of electronic and thermal Enthalpies: -1142.678393

0 2

C -3.63571500 0.42087800 0.00000000  
 C -2.34605400 0.98542900 0.00000000  
 C -2.19099000 2.41389700 0.00000000  
 C -3.27431700 3.25010600 0.00000000  
 C -4.61454900 2.71534200 0.00000000  
 C -4.74161800 1.25085400 0.00000000  
 C -1.15711500 0.13039300 0.00000000  
 O 0.00000000 0.84133300 0.00000000  
 C 1.21745500 0.21320800 0.00000000  
 C 1.30714400 -1.19158400 0.00000000

C	0.09212000	-1.98129100	0.00000000
C	-1.14498100	-1.23757000	0.00000000
C	2.59324600	-1.81023200	0.00000000
C	3.73731300	-1.00882400	0.00000000
C	3.58814700	0.40197200	0.00000000
C	2.34266500	1.02541200	0.00000000
O	0.11225100	-3.24533700	0.00000000
O	4.70109600	1.17332700	0.00000000
O	2.67786600	-3.15975400	0.00000000
O	-5.66767000	3.40266400	0.00000000
O	-5.98585200	0.75854500	0.00000000
H	2.26335900	2.10540800	0.00000000
O	5.05185800	-1.39195700	0.00000000
H	-1.19157500	2.82958000	0.00000000
H	-3.15684800	4.32860200	0.00000000
H	-3.79282800	-0.65063100	0.00000000
H	5.47718200	0.58503500	0.00000000
H	1.73326200	-3.49387600	0.00000000
H	-6.58983300	1.53122500	0.00000000
H	-2.06739900	-1.80175900	0.00000000
C	5.46631800	-2.77176800	0.00000000
H	6.55552500	-2.73195900	0.00000000
H	5.10918200	-3.28476700	0.89435800
H	5.10918200	-3.28476700	-0.89435800

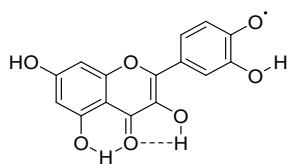


Sum of electronic and thermal Enthalpies: -1104.031250

0 1

C	-0.64524600	2.91809700	0.00000000
C	-1.00796000	1.55290800	0.00000000
C	-2.37823800	1.21925700	0.00000000
C	-3.35342400	2.21436100	0.00000000
C	-2.98353300	3.55878800	0.00000000
C	-1.62112100	3.90599900	0.00000000
C	0.00000000	0.48829400	0.00000000
O	-0.56542500	-0.76220200	0.00000000
C	0.16797900	-1.90760100	0.00000000
C	1.57473800	-1.84964200	0.00000000
C	2.22168600	-0.57120800	0.00000000
C	1.36899100	0.59858300	0.00000000
C	2.30320000	-3.07273300	0.00000000
C	1.62981900	-4.28719900	0.00000000

C	0.22233700	-4.29369000	0.00000000
C	-0.52634300	-3.11291600	0.00000000
O	3.48031100	-0.41552600	0.00000000
O	-0.48118100	-5.45771100	0.00000000
O	3.65293100	-3.05134200	0.00000000
O	-3.85840700	4.61074900	0.00000000
O	-1.23991500	5.22064100	0.00000000
H	-1.60895600	-3.13738400	0.00000000
H	2.19402300	-5.21406900	0.00000000
H	-2.68608400	0.18198400	0.00000000
H	-4.40565800	1.94416300	0.00000000
H	0.39180700	3.22211200	0.00000000
H	0.11556200	-6.22045000	0.00000000
H	3.93180400	-2.09914900	0.00000000
H	-4.77738200	4.30692300	0.00000000
H	-2.03303000	5.77955300	0.00000000
O	2.02023200	1.79331700	0.00000000
H	2.97224100	1.56511400	0.00000000

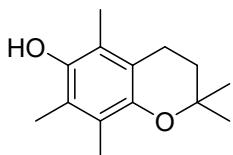


Sum of electronic and thermal Enthalpies: -1103.414263

0 2

C	-0.45926200	2.99271500	0.00000000
C	-0.91670400	1.65389500	0.00000000
C	-2.33458300	1.39345100	0.00000000
C	-3.25097300	2.40804900	0.00000000
C	-2.82472700	3.78648700	0.00000000
C	-1.37700400	4.02423600	0.00000000
C	0.00000000	0.53236000	0.00000000
O	-0.63962300	-0.68337600	0.00000000
C	0.02246600	-1.87163200	0.00000000
C	1.43314600	-1.90500200	0.00000000
C	2.16176600	-0.67649600	0.00000000
C	1.38178700	0.55443100	0.00000000
C	2.08209700	-3.17451400	0.00000000
C	1.33157400	-4.34172000	0.00000000
C	-0.07319600	-4.25643800	0.00000000
C	-0.74531500	-3.02841100	0.00000000
O	3.42270100	-0.58787500	0.00000000
O	-0.84965200	-5.36907900	0.00000000
O	3.42828500	-3.24318000	0.00000000
O	-3.59458500	4.78264400	0.00000000

O	-0.98568500	5.30851600	0.00000000
H	-1.82724300	-2.98465900	0.00000000
H	1.83389500	-5.30345500	0.00000000
H	-2.67526800	0.36628500	0.00000000
H	-4.31695400	2.20515300	0.00000000
H	0.59555900	3.22611700	0.00000000
H	-0.30571200	-6.17079500	0.00000000
H	3.77702600	-2.31669800	0.00000000
H	-1.80714800	5.84369800	0.00000000
O	2.10548800	1.69015900	0.00000000
H	3.04399500	1.40343400	0.00000000

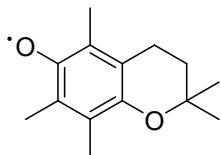


Sum of electronic and thermal Enthalpies: -695.731512

0 1

C	2.68810300	0.10094700	0.05878800
O	1.55528000	0.96752700	-0.23628300
C	0.27520200	0.45406500	-0.13194300
C	0.01594400	-0.92724700	-0.12469300
C	1.16793500	-1.91118400	-0.22362700
C	2.46854400	-1.23286600	-0.66432000
C	-1.31936200	-1.37099400	-0.02198900
C	-2.34965400	-0.41976900	0.02098100
C	-2.09636800	0.96076300	-0.01005000
C	-0.76344200	1.40420600	-0.06994300
C	-0.45262000	2.88379800	-0.08340200
C	-3.23745700	1.95214200	0.03302300
O	-3.67769700	-0.80912400	0.10391900
C	-1.67311900	-2.84263100	0.04300700
C	3.89923100	0.84269000	-0.50508000
C	2.81177000	-0.06009500	1.58104200
H	-0.79257400	-3.47213400	0.16788200
H	-2.33192000	-3.06094700	0.89387900
H	-2.18511700	-3.18552500	-0.86683400
H	-4.19879900	1.45311500	-0.09054700
H	-3.26272300	2.49367800	0.98772000
H	-3.13873800	2.70587400	-0.75578100
H	-3.74639900	-1.77097400	0.04862900
H	-1.02088600	3.41607100	0.68680500
H	0.60898500	3.06303300	0.08604800
H	-0.71831400	3.34083100	-1.04610800
H	2.43400800	-1.02919300	-1.74157800

H	3.32237100	-1.89551000	-0.48727500
H	0.92518400	-2.70523000	-0.93753500
H	1.31638900	-2.41200400	0.74304600
H	3.80026100	0.97808400	-1.58663700
H	3.99809400	1.82787600	-0.03826100
H	4.81417300	0.27511400	-0.30844300
H	3.65439200	-0.71539900	1.82574600
H	2.98545200	0.91406500	2.04855700
H	1.90725800	-0.49099300	2.01883900

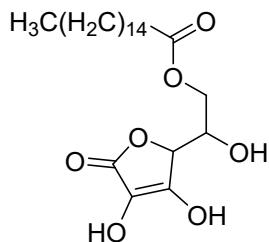


Sum of electronic and thermal Enthalpies: -695.121309

0 2

C	-2.65211500	-0.12885400	0.04434100
O	-1.45415200	-0.97094200	-0.12691400
C	-0.21591800	-0.43529900	-0.08605600
C	0.01521800	0.96595400	-0.08708400
C	-1.16202100	1.91307800	-0.17911300
C	-2.42075900	1.19992200	-0.67995000
C	1.31973900	1.42972600	-0.01197700
C	2.43586700	0.48746300	0.03132700
C	2.15186700	-0.95020500	-0.00274400
C	0.84476700	-1.39557500	-0.06018100
C	0.47316600	-2.85902900	-0.08750700
C	3.32545100	-1.89748000	0.03509100
O	3.62945000	0.90913600	0.09056800
C	1.62396200	2.90368500	0.01292500
C	-3.77556600	-0.93471800	-0.60296900
C	-2.88803700	0.03575200	1.55044100
H	1.02669300	3.42314200	0.77064100
H	2.68123100	3.06695800	0.22306900
H	1.39318000	3.37518700	-0.95167000
H	4.25632200	-1.33238400	0.08451000
H	3.27933300	-2.56211600	0.90559300
H	3.35659100	-2.53541600	-0.85588500
H	1.35435700	-3.49955300	-0.07350100
H	-0.15423700	-3.11974900	0.77206400
H	-0.11029800	-3.09838500	-0.98343800
H	-2.32672200	0.99414400	-1.75297600
H	-3.30082700	1.83731500	-0.54916600
H	-0.91928600	2.73910600	-0.85510300
H	-1.34716500	2.37383200	0.80057100

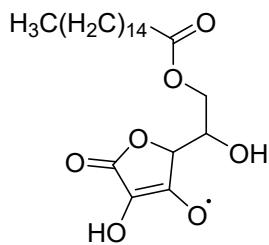
H -3.58123100 -1.08464200 -1.66913300  
 H -3.87394600 -1.91311100 -0.12316500  
 H -4.72448700 -0.40077300 -0.49483700  
 H -3.78276300 0.64251600 1.72217700  
 H -3.03889600 -0.94202400 2.01695300  
 H -2.04395700 0.52388100 2.04479900



Sum of electronic and thermal Enthalpies: -837.304830

0 1

C 2.16084500 0.85660100 0.19139200  
 C 3.00380200 -0.10427500 -0.23171800  
 C 2.36474200 -1.39631400 -0.01915400  
 O 1.12832800 -1.19522900 0.53426300  
 C 0.90148200 0.22325000 0.71925300  
 O 4.23765300 0.04833100 -0.77365400  
 O 2.80381900 -2.50433500 -0.27139900  
 C -0.38353400 0.65791000 -0.02562900  
 C -1.58719800 -0.11414100 0.48819500  
 O -2.69351400 0.18344500 -0.39116400  
 C -3.87756700 -0.39226000 -0.09519300  
 O -4.03190500 -1.13875000 0.85689400  
 O 2.27057000 2.19771200 0.19385400  
 O -0.62583800 2.04061800 0.20868800  
 C -4.94880700 0.01569800 -1.07062700  
 H 0.78102300 0.40734200 1.79316000  
 H -0.25417100 0.45676400 -1.09779400  
 H 4.60482200 -0.82869500 -0.97240500  
 H -1.83730100 0.19707600 1.50670500  
 H -1.39410300 -1.18858300 0.47594000  
 H -5.15276500 1.08580200 -0.96282300  
 H -4.61365900 -0.15486300 -2.09723700  
 H 0.09805400 2.56302300 -0.16849500  
 H -5.85924200 -0.54981500 -0.87463500  
 H 3.11185200 2.47879500 -0.20137700



Sum of electronic and thermal Enthalpies: -836.688295

0 2

C	2.15089900	0.98463400	0.17185400
C	3.05635000	-0.05291400	-0.23007600
C	2.42404500	-1.35786100	-0.01799600
O	1.18506100	-1.16102000	0.50308800
C	0.91720400	0.25628200	0.71280700
O	4.25586400	0.11868900	-0.73990800
O	2.89414500	-2.45150700	-0.26287000
C	-0.37921600	0.68737200	-0.00711900
C	-1.56619700	-0.12659700	0.48012100
O	-2.68354800	0.19661900	-0.37481700
C	-3.85334800	-0.42095800	-0.10424100
O	-3.98355400	-1.22196400	0.80591600
O	2.27596800	2.20958800	0.08246700
O	-0.62759800	2.05618000	0.28524300
C	-4.93948200	0.01517800	-1.05020500
H	0.81911300	0.41957100	1.79102400
H	-0.25694500	0.53215400	-1.08865800
H	4.66579100	-0.74544200	-0.94152000
H	-1.81049700	0.13114500	1.51470400
H	-1.35697900	-1.19632600	0.41377700
H	-5.15642200	1.07670300	-0.89451000
H	-4.61190800	-0.10641300	-2.08627100
H	0.15564900	2.57636300	0.03857800
H	-5.84004800	-0.57124900	-0.87094900