Supporting Information

Non-noble metal Li, Mn co-doped Ni₃S₂ electrocatalyst for glycerol oxidation synergistic coupling to promote hydrogen evolution reaction

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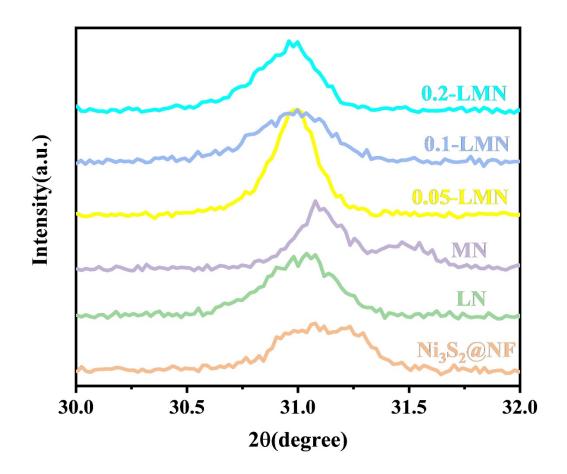


Fig. S1 n-LMN,LN,MN, and Ni $_3S_2@NF$ highest diffraction peak angle shift.

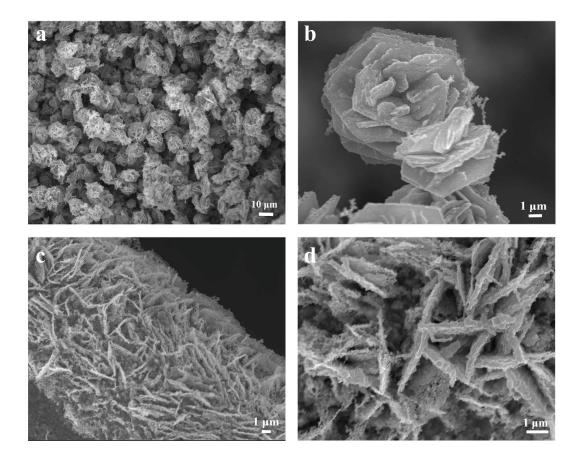


Fig. S2 The SEM images of a, b MN and c, d LN at different multiples were obtained.

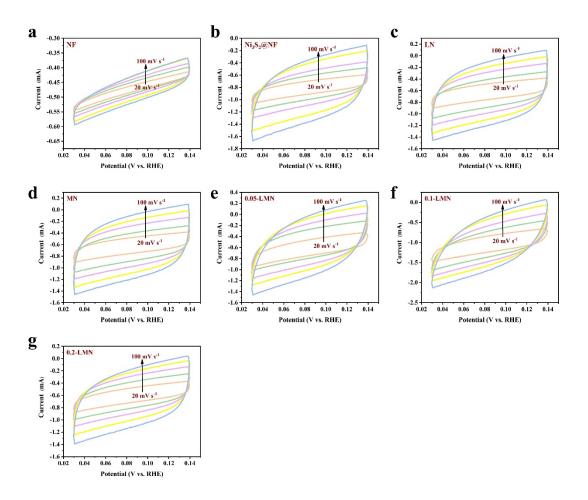


Fig. S3 Cyclic voltammetry curves of different materials between 0.03-

0.138 V vs. RHE.

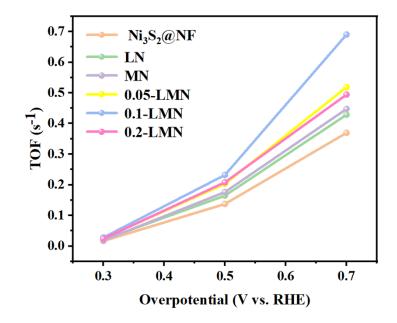


Fig. S4 The TOF of HER in Ni₃S₂@NF, LN, MN, 0.05-LMN, 0.1-LMN,

and 0.2-LMN at different overpotentials.

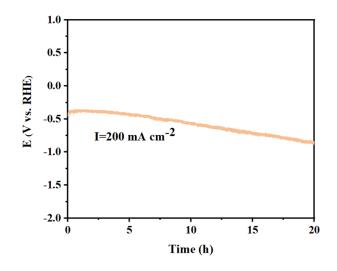


Fig. S5 The stability of 0.1-LMN catalyzed HER at a current density of

200 mA cm⁻².

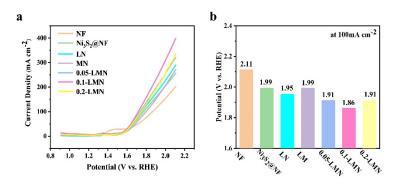


Fig. S6 The (a) LSV curve of OER catalyzed by different materials; (b) Overpotential comparison at a current density of 100 mA cm⁻².

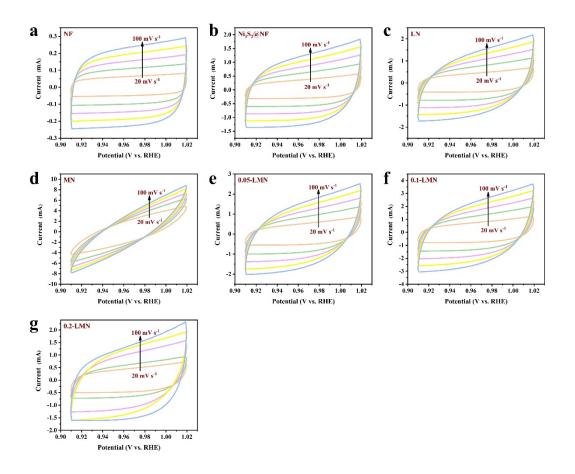


Fig. S7 Cyclic voltammetry curves of different materials between 0.909-

1.019 V vs. RHE.

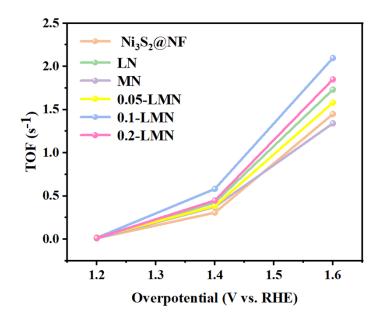


Fig. S8 The TOF of GOR in Ni₃S₂@NF, LN, MN, 0.05-LMN, 0.1-LMN,

and 0.2-LMN at different overpotentials.

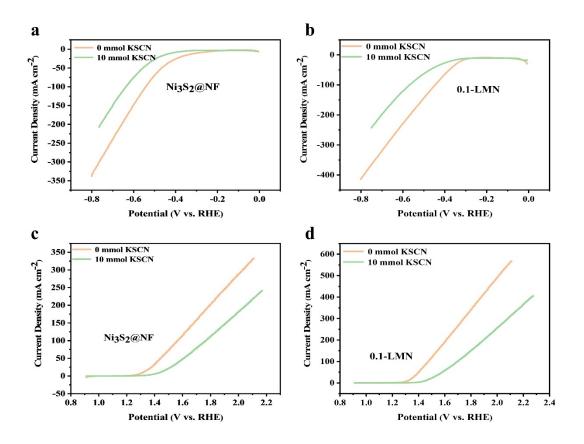


Fig. S9 The electrocatalytic performance of Ni_3S_2 @NF and 0.1-LMN

before and after poisoning test.

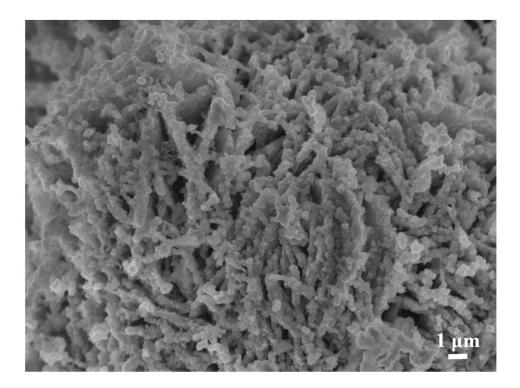


Fig. S10 SEM image of 0.1-LMN after long-term reaction.

Catalyst	Electrolyte	η ₁₀ (V vs.RHE)	Reference	
This Work	1.0 M KOH	0.131		
CoSe ₂ /c-CoP	$0.5 \text{ M H}_2 \text{SO}_4$	0.127	[1]	
Co ₁₂ @Ni ₃ S ₂ /NF	1.0 M KOH	0.297	[2]	
NiCo ₂ S ₄ NW/NF	1.0 M KOH	0.260	[3]	
THTNi 2DSP	0.5 M KOH	0.574	[4]	
Ni-Co-S/Ni	1.0 M KOH	0.371	[5]	
Ni_3S_2 nanoparticles/CNTs	1.0 M KOH	0.480	[6]	

Table S1 compares with other non-noble metal HER catalysts.

Catalyst	Electrolyte	Glycerol	Settlin-	Current/Curr-	Initial	Terminal	Refere-
		concent- ation	g time	ent density	potential	potential	nce
This Work	1.0 M KOH	0.5 M	20 h	200 mA cm^{-2}	1.54 V	1.912 V	
SnO ₂ /CoS	1.0 M KOH	0.1 M	3000 s	100 mA	1.05 V	1.24 V	[7]
NiO/NF	1.0 M KOH	0.1 M	15 h	10 mA cm^{-2}	1.35 V	1.45 V	[8]
Ni ₃ N/WO ₃	1.0 M KOH	0.1 M	24 h	10 mA cm^{-2}	1.45 V	1.56 V	[9]
Pt _{SA} -NiCo LDH/NF	1.0 M KOH	0.1 M	24 h	10 mA cm^{-2}	1.25 V	1.52 V	[10]
CoNiCuM nMo/CF	1.0 M KOH	0.1 M	25 h	10 mA cm^{-2}	1.21 V	1.26 V	[11]
Ni-Mo- N/CFC	1.0 M KOH	0.1 M	11 h	10 mA cm ⁻²	1.29 V	1.48 V	[12]

Table S2 Comparison of stability of different materials.

Reference

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