

Supporting Information

Rapid synthesis of sea urchin-like $\text{Ni(OH)}_2@ \text{Ni(Fe)OOH}$ electrocatalysts for the oxygen evolution reaction

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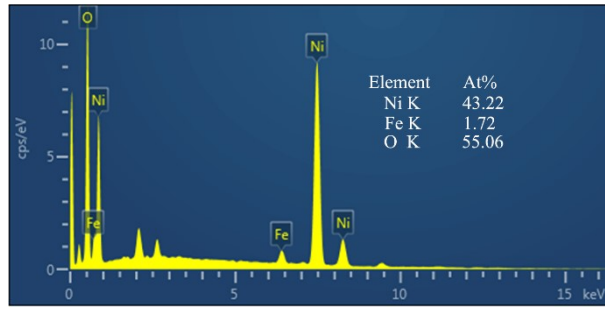


Fig. S1 SEM-EDX spectrum of Ni(OH)₂@Ni(Fe)OOH-100.

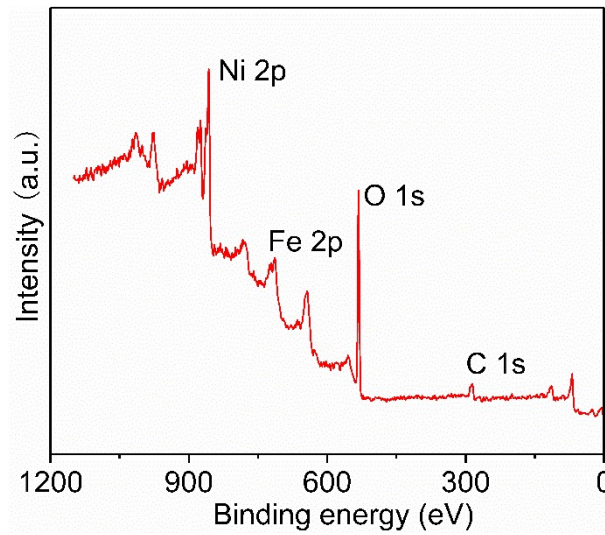


Fig. S2 XPS survey spectrum of Ni(OH)₂@Ni(Fe)OOH-100.

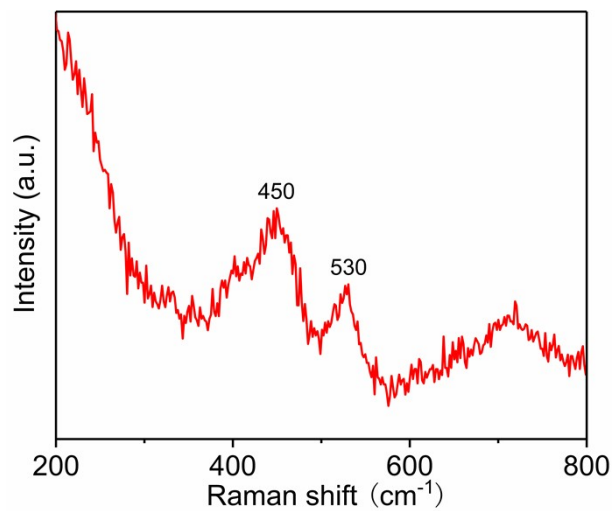


Fig. S3 Raman spectrum of Ni(OH)₂@Ni(Fe)OOH-100.

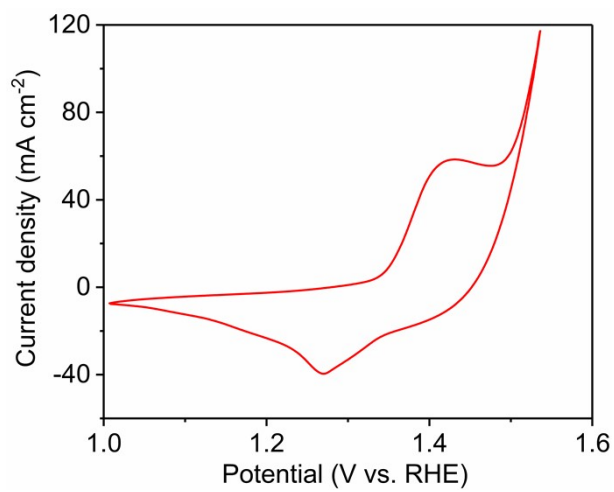


Fig. S4 The cycling voltammetry curve (the 20th cycle) of Ni(OH)₂@Ni(Fe)OOH-100 in 1 M KOH at the scan rate of 5 mV s⁻¹ after iR-correction.

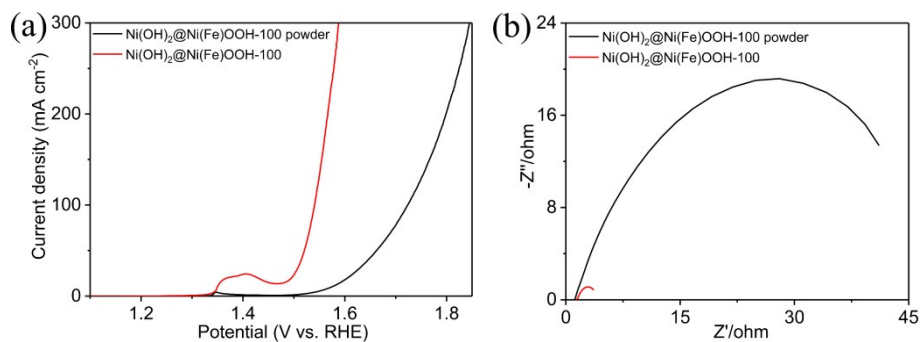


Fig. S5 (a) Polarization curves and (b) EIS spectra of $\text{Ni(OH)}_2@Ni(Fe)OOH-100$ and $\text{Ni(OH)}_2@Ni(Fe)OOH-100$ powder.

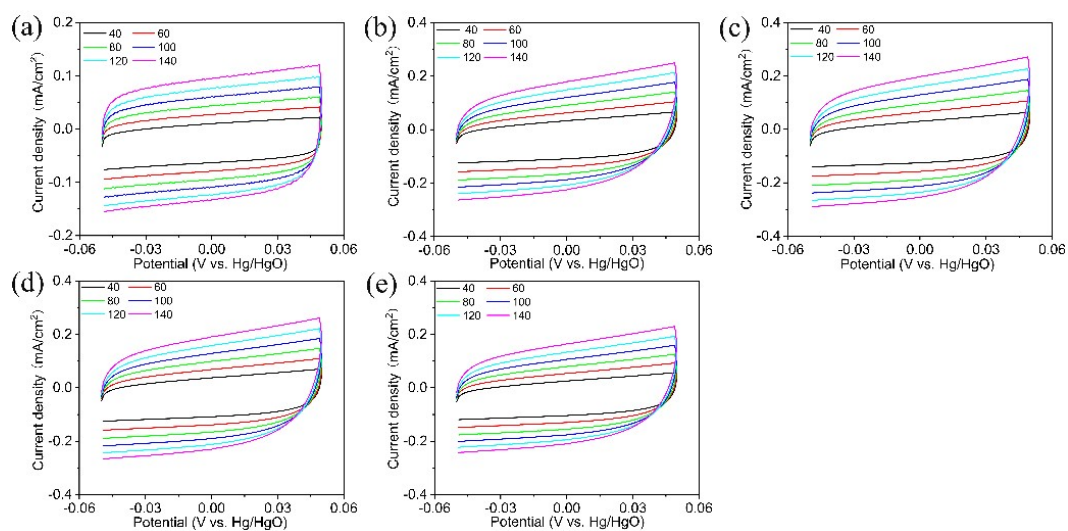


Fig. S6 The CV curves of (a) Ni(OH)_2 , (b) $\text{Ni(OH)}_2@Ni(Fe)OOH-25$, (c) $\text{Ni(OH)}_2@Ni(Fe)OOH-50$, (d) $\text{Ni(OH)}_2@Ni(Fe)OOH-75$ and (e) $\text{Ni(OH)}_2@Ni(Fe)OOH-100$ with different scan rates to evaluate the ECSA for OER process.

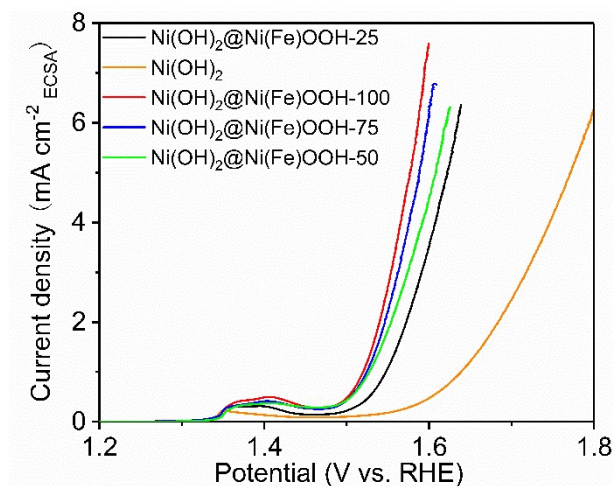


Fig. S7 ECSA-normalized polarization curves for Ni(OH)₂, (b) Ni(OH)₂@Ni(Fe)OOH-25, (c) Ni(OH)₂@Ni(Fe)OOH-50, (d) Ni(OH)₂@Ni(Fe)OOH-75 and (e) Ni(OH)₂@Ni(Fe)OOH-100.

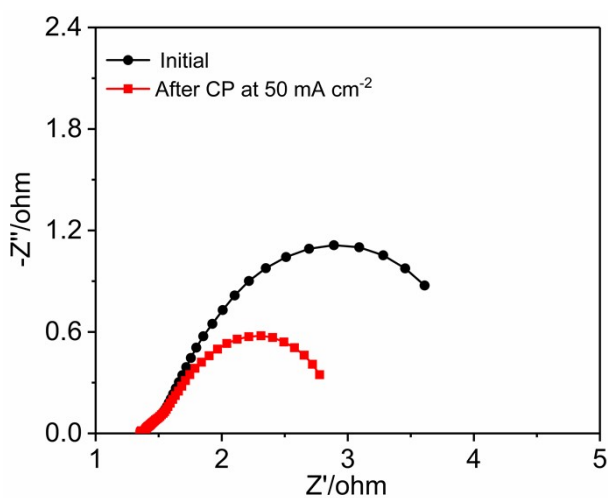


Fig. S8 EIS spectra of Ni(OH)₂@Ni(Fe)OOH-100 before and after stability test for 15 h.

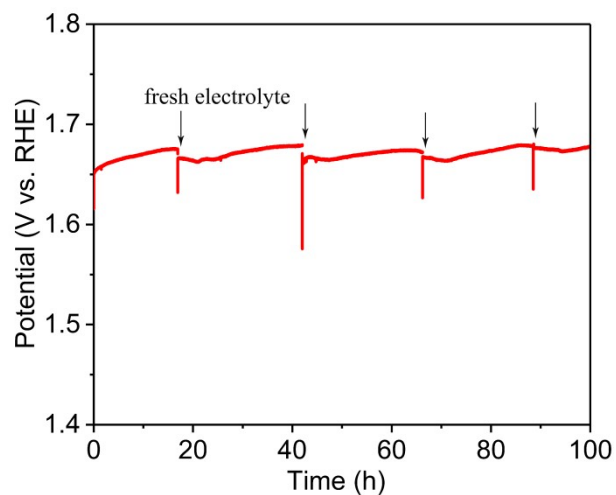


Fig. S9 Stability test of Ni(OH)₂@Ni(Fe)OOH-100 at 100 mA cm⁻² for 100 hours.

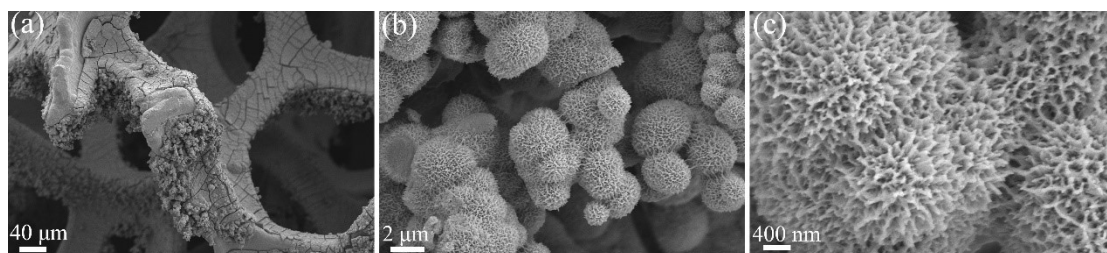


Fig. S10 SEM images of Ni(OH)₂@Ni(Fe)OOH-100 after stability test for 15 h.

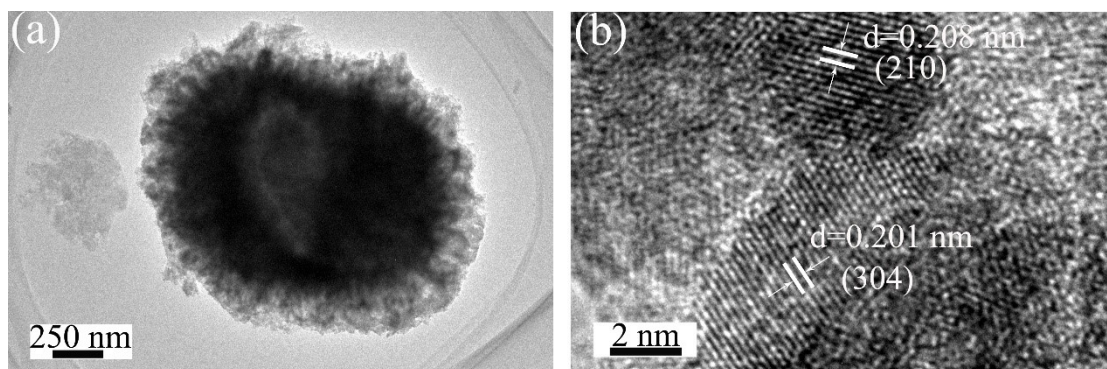


Fig. S11 (a) TEM and (b) HRTEM images of Ni(OH)₂@Ni(Fe)OOH-100 after stability test for 15 h.

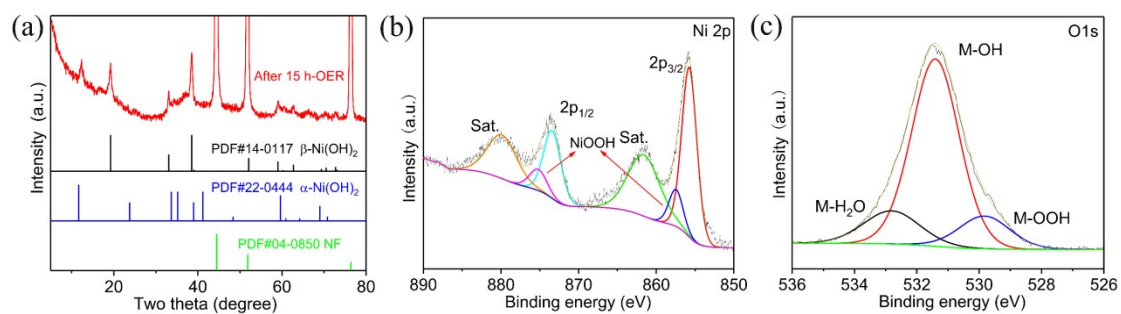


Fig. S12 (a) XRD pattern of $\text{Ni(OH)}_2@ \text{Ni(Fe)OOH-100}$ after stability test. High-resolution XPS spectra of (b) Ni 2p and (c) O 1s of $\text{Ni(OH)}_2@ \text{Ni(Fe)OOH-100}$ after stability test for 15 h.

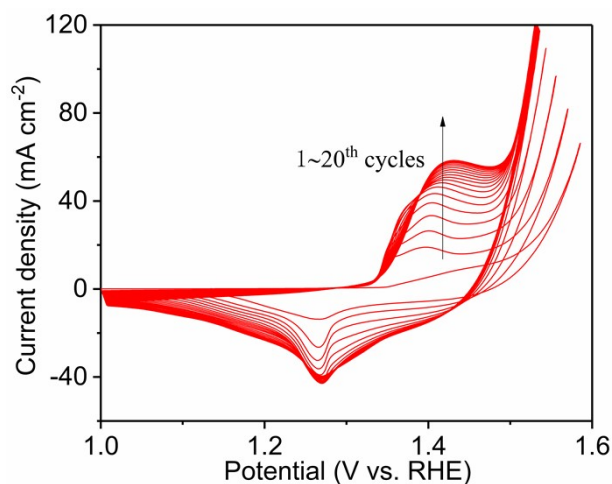


Fig. S13 20 CV cycles in the potential range of 1.0–1.6 V versus RHE at the scan rate of 5 mV s^{-1} after iR-correction.

Table S1. Comparison of OER performance of Ni(OH)₂@Ni(Fe)OOH-100 with some Ni or Fe-based electrocatalysts in 1 M KOH.

Catalysts	$\eta@$ 10 mA cm ⁻² (mV)	$\eta@$ 100 mA cm ⁻² (mV)	Tafel slope (mV dec ⁻¹)	References
Ni(OH) ₂ @Ni(Fe)OOH-100	245	310	40.7	This work
Fe-doped-Ni(OH) ₂ -40 min	248		61	[1]
Fe-NiTe-Ni ₁₂ P ₅		340	66	[2]
NiFe-LDH	247		37	[3]
Ni ₁₈ Fe ₁₂ Al ₇₀	255	345	37	[4]
CoFe ₂ O ₄	287		43	[5]
Co ₂ P-Ni ₃ S ₂ /NF		331.7	31.6	[6]
CoNiLDH/FeOOH	250		60	[7]
FeNi ₃ @NCNT	264		58.5	[8]
ZnFe ₂ O ₄ @Ni ₃ S ₂	254		39.29	[9]
NiCo ₂ O ₄ @MoS ₂ /TM	313	380	66.8	[10]
NiO-Ni ₃ Se ₄ /MXene	260		39.6	[11]
C-NiFe ₂ O ₄ @A-S-NiFe ₂ O ₄	275		76.1	[12]

Table S2. The values of R_s and R_{ct} obtained from fitted plots using equivalent circuit.

Samples	<i>Solution resistance R_s</i>	<i>Charge transfer resistance R_{ct}</i>
	(Ω)	(Ω)
Ni(OH) ₂	1.41	7.29
Ni(OH) ₂ @Ni(Fe)OOH-25	1.47	7.37
Ni(OH) ₂ @Ni(Fe)OOH-50	1.45	4.24
Ni(OH) ₂ @Ni(Fe)OOH-75	1.45	3.83
Ni(OH) ₂ @Ni(Fe)OOH-100	1.40	3.66
RuO ₂	1.31	17.88
Ni(OH) ₂ @Ni(Fe)OOH-100 powder@NF	1.15	57.12
NF	1.52	382.1
Post-Ni(OH) ₂ @Ni(Fe)OOH-100 (15 h)	1.38	2.04

Table S3. Comparison of overall water splitting performance with some reported electrocatalysts in 1 M KOH.

Catalysts	Voltage @10 mA cm ⁻² (V)	References
Ni(OH) ₂ @Ni(Fe)OOH-100 Pt/C	1.54	This work
Co/CoO/NC/CC Co/CoO/NC/CC	1.66	[13]
Ni ₂ P-Fe ₂ P/NF Ni ₂ P-Fe ₂ P/NF	1.561	[14]
W ₂ N/WC W ₂ N/WC	1.58	[15]
V-Co ₂ P ₄ O ₁₂ /CC V-Co ₂ P ₄ O ₁₂ /CC	1.60	[16]
Co ₃ S ₄ @NiFe-200/NF Co ₃ S ₄ @NiFe-200/NF	1.595	[17]
ZnCoS-NSCNT/NP ZnCoS-NSCNT/NP	1.59	[18]
MoS ₂ /NiFe-LDH MoS ₂ /NiFe-LDH	1.61	[19]
Co ₂ Mo ₃ O ₈ @NC-800/NF Pt/C	1.67	[20]
RuO ₂ /NF Pt/C@NF	1.56	[21]
Mn-NiCo ₂ S ₄ /NF Pt/C	1.59	[22]

Supplementary references

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