# **Supplementary Information**

# Chiral Ultrasmall Nickel Hydroxide Nanoparticles Enable Enantioselective Magnetic Resonance Imaging of Hepatocellular Carcinoma and Lung Metastases

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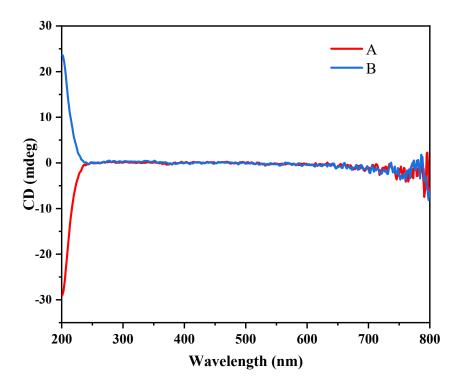
## **Experimental Section**

### Chemicals

Sodium hydroxide (NaOH, AR,96%), nickel chloride hexahydrate (NiCl<sub>2</sub>·6H<sub>2</sub>O,AR), D-ASP (99%), L-ASP (99%) and DL-ASP (99%) were all purchased from Sigma-Aldrich (St. Louis, Missouri, USA). All aqueous solutions in the experiment were prepared using ultrapure water ( $\geq$ 18.25 M $\Omega$ , Milli-Q, Millipore).

#### Instruments

Transmission electron microscopy (TEM) was performed with a JEOL JEM-2100 (Hitachi, Tokyo, Japan) microscope operating at 200kV. The CD signals were characterized by a CHIRASCAN CD spectrometer from Applied Photophysics (Surrey, UK) with an optical path length of 1 cm. X-ray photoelectron spectra (XPS) were acquired with a Kratos Analytical Axis Ultra system (Manchester, UK). Powder X-ray powder diffraction (XRD) patterns were acquired by a Bruker D8 (Germany) equipped with a Cu Ka radiation source (40 kV, 40 mA) at a rate of 6°/min. UV-visible (UV-vis) absorption spectra were determined with a Shimadzu UV-vis 3101 spectroscope. A Magnetom skyra 3.0T(SIEMENS,Lanzhou,China) was used to obtain T1 values and T1-weighted images. Magnetic resonance imaging (MRI) processing software (SIEMENS, Lanzhou, China) was used to process the contrast ratio of the original T1-weighted images. Size distributions were determined using a Malvern Zetasizer (Malvern, UK). Confocal images were obtained using an Olympus FV1000 laser scanning confocal microscope (Tokyo, Japan).



**Figure S1.** CD spectrum of the precursor solution. A is the precursor solution of D-Aspartic acid and nickel chloride without added NaOH, while B is the precursor solution of L-Aspartic acid and nickel chloride without added NaOH.

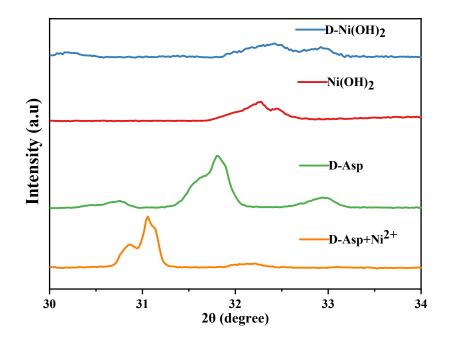


Figure S2. The XRD spectrum of chiral  $Ni(OH)_2$  NPs,  $Ni(OH)_2$  without chiral ligands, D-Aspartic acid and D-Aspartic acid- $Ni^{2+}$  complexes with 20 ranging from 30 degree to 34 degree.

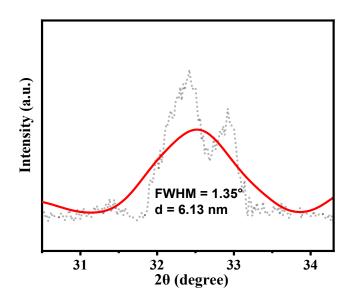


Figure S3. Grain size of  $D-Ni(OH)_2$  NPs calculated using the Scherrer formula from the XRD spectrum.

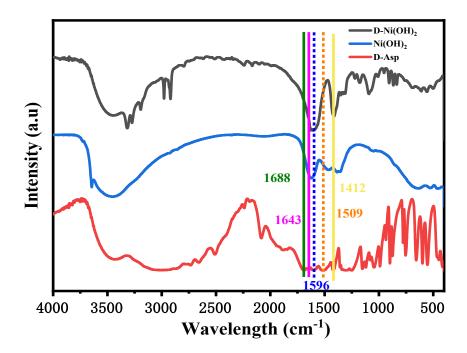


Figure S4. The infrared spectrum of chiral  $Ni(OH)_2$  NPs,  $Ni(OH)_2$  and D-Aspartic acid.

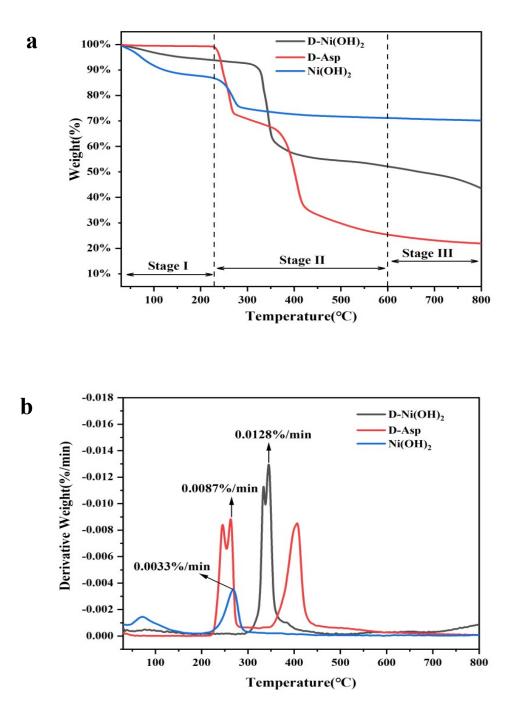


Figure S5. (a)Thermogravimetric curves of  $D-Ni(OH)_2$  NPs, D-Aspartic acid and  $Ni(OH)_2$ . (b) Differential thermogravimetric (DTG) curves of D- $Ni(OH)_2$  NPs, D-Aspartic acid and  $Ni(OH)_2$ .

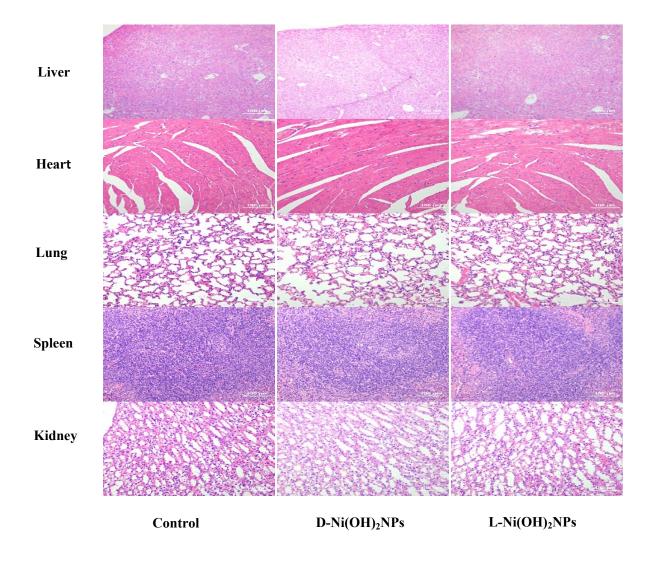
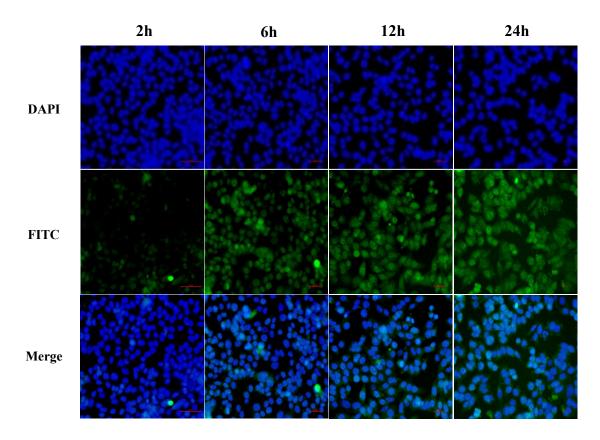
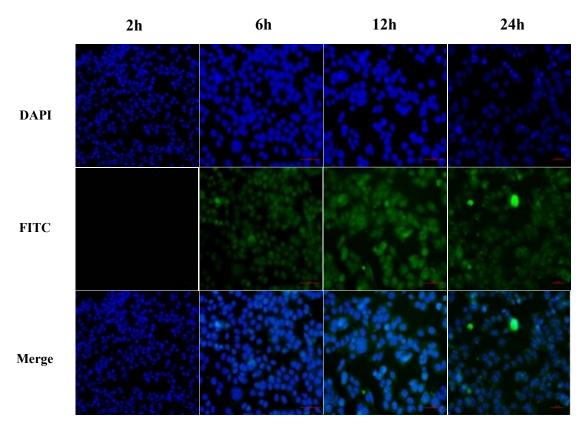


Figure S6. After 72 hours of injecting healthy mice with PBS,  $D-Ni(OH)_2$  NPs, and  $L-Ni(OH)_2$  NPs (20 mg/kg), HE images of different organs (liver, heart, lung, spleen, and kidney).

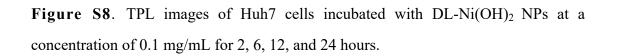


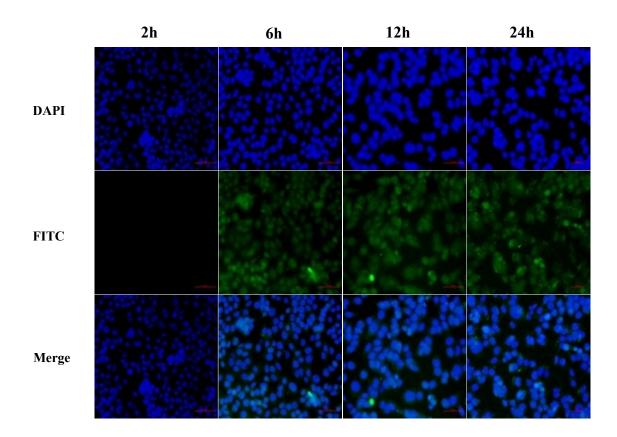
D-Ni(OH)<sub>2</sub>NPs

Figure S7. TPL images of Huh7 cells incubated with  $D-Ni(OH)_2$  NPs at a concentration of 0.1 mg/mL for 2, 6, 12, and 24 hours.



DL-Ni(OH)<sub>2</sub>NPs





L-Ni(OH)<sub>2</sub>NPs

Figure S9. TPL images of Huh7 cells incubated with  $L-Ni(OH)_2$  NPs at a concentration of 0.1 mg/mL for 2, 6, 12, and 24 hours.