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## "Synthesis of benzylidene amino arylpyridinones and their AIE Studies: Dual metal sensing of Fe<sup>3+</sup> and Hg<sup>2+</sup>

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Figure S.1: <sup>1</sup>H NMR Spectrum of 4a



Figure S.2: <sup>13</sup>C NMR Spectrum of 4a



Figure S.3: <sup>1</sup>H NMR Spectrum of 4b



Figure S.4: <sup>13</sup>C NMR Spectrum of 4b



Figure S.5: <sup>1</sup>H NMR Spectrum of 4c



Figure S.6: <sup>13</sup>C NMR Spectrum of 4c



Figure S.7: <sup>1</sup>H NMR Spectrum of 4d



Figure S.8: <sup>13</sup>C NMR Spectrum of 4d



Figure S.9: <sup>1</sup>H NMR Spectrum of 4e



Figure S.10: <sup>13</sup>C NMR Spectrum of 4e



Figure S.11: HRMS-Spectrum of 4a



Figure S.12: HRMS-Spectrum of 4b



Figure S.13: HRMS-Spectrum of 4c



Figure S.14: HRMS-Spectrum of 4d



Figure S.15: HRMS-Sprctrum of 4e



Figure S.16: FTIR-Spectrum of 4a



Figure S.17: FTIR-Spectrum of 4b



Figure S.18: FTIR-Spectrum of 4c



Figure S.19: FTIR-Spectrum of 4d



Figure S.20: FTIR-Spectrum of 4e



Figure S.21. Uv-Vis spectra of 4e  $(5 \times 10^{-5} \text{ M})$  in Various solvents



Figure S.22. UV- Vis spectra of 4e  $(5 \times 10^{-5} \text{ M})$  in DMSO: water solvent mixtures



**Figure S.23.** UV- Vis spectra of 4e  $(5 \times 10^{-5} \text{ M})$  in different metal cations  $(300 \mu \text{l})$ 



Figure S.24: Job's plot between the Fluorescence intensity of 4e and the mole fraction of Fe<sup>3+</sup> and Hg<sup>2+</sup>



**Figures S.25.** Interference study of A-4e ( $5x10^{-5}$  M) in DMSO solvent presence of various metal ions. Blue bars represent the Fluorescence intensity of A-4e + metal cation systems; Pink bars represent the Fluorescence intensity of Hg<sup>2+</sup> + metal cation systems.



**Figures S.26.** Interference study of A-4e ( $5x10^{-5}$  M) in DMSO solvent presence of various metal ions. Blue bars represent the Fluorescence intensity of A-4e + metal cation systems; Pink bars represent the Fluorescence intensity of Fe<sup>3+</sup> + metal cation systems.



Figures S.27: (a) Cyclic voltammetry titration of A-4e with Fe<sup>3+</sup> ions (0-50 μl) (b) Cyclic voltammetry titration of A-4e with Hg<sup>2+</sup> ions (0-50 μl).



Figure S.28: Calculated HOMOs and LUMOs of 4e and  $4e+Fe^{3+}$  and  $4e+Hg^{2+}$ 

	$\lambda_{ex}$	λ <sub>em</sub>	$\tau_1$ (rel%)	$\tau_2$ (rel%)	$\tau_3$ (rel%)	T <sub>average(ns)</sub>	$x^2$
4e (Probe)	350nm	450nm	0.255636	4.77261	-	0.96	1.357502
A-4e	350nm	470nm	0.304501	1.68946	-	1.54	1.009198
A-4e+Fe <sup>3+</sup>	350nm	470nm	0.21956	1.63526	-	1.47	1.044633
A-4e+Hg <sup>2+</sup>	350nm	470nm	0.215328	1.01693	4.76184	0.89	1.137985

TCSPC – (Time correlated single photon counting) spectrometer for probe

4e, A-4e, A-4e+Fe<sup>3+</sup> and A-4e+Hg<sup>2+</sup>



Figure S.29: FTIR spectra of 4e in the presence of  $Fe^{3+}$  and  $Hg^{2+}$