

Ba₄FeCuS₆: a new mixed metal sulfide with a pseudo-zero-dimensional structure containing rare CuS₃ units

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Electronic Supplementary Information (ESI)

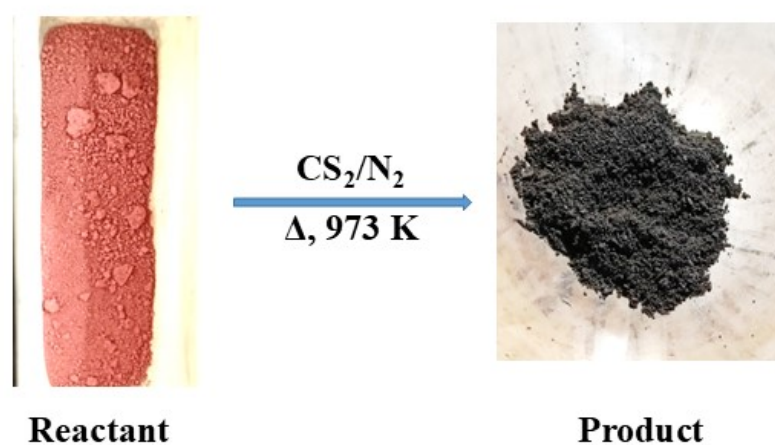


Fig. SI1: The polycrystalline Ba₄FeCuS₆ sample before and after reaction by the CS₂ method.

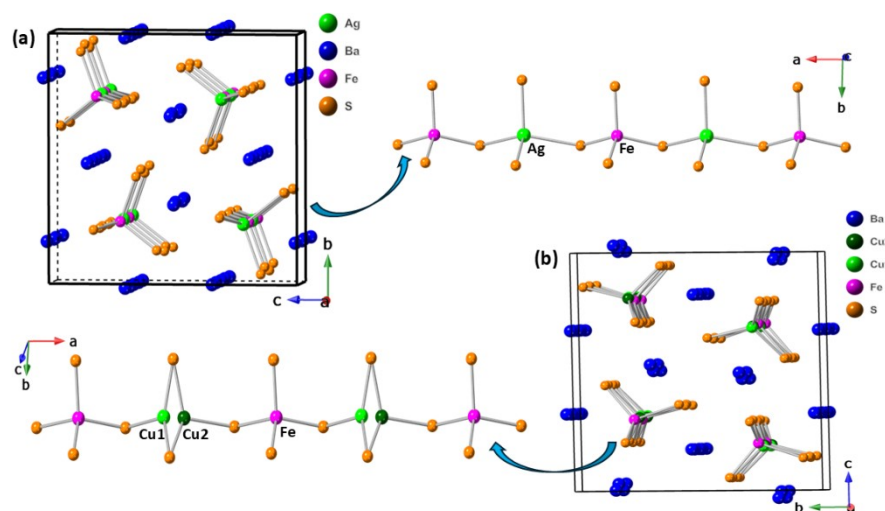


Fig. SI2: The structure comparison of (a) Ba₄FeAgS₆ (Ag site was fully occupied)¹ and (b) Ba₄FeCuS₆ (Cu site was split into two sites; Cu1 and Cu2).

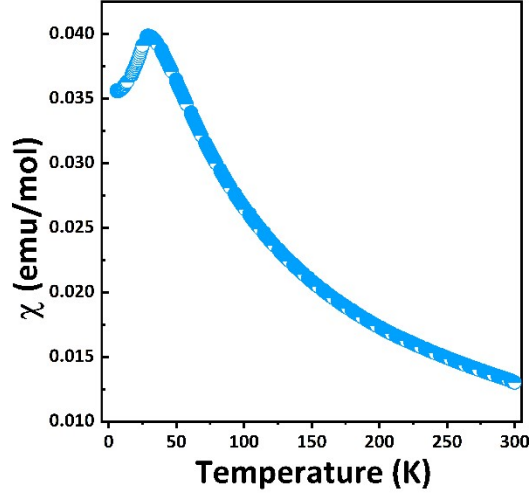


Fig. SI3: The temperature dependence (FW) of the molar magnetic susceptibility (χ_{mol}) of the polycrystalline $\text{Ba}_4\text{FeCuS}_6$ sample under an applied magnetic field of 1000 Oe (0.1 T).

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ba1	0.0175(3)	0.0076(3)	0.0133(3)	-0.0003(3)	0.0043(3)	0.0002(2)
Ba2	0.0178(3)	0.0076(3)	0.0128(3)	0.0000(3)	0.0045(3)	-0.0003(2)
Ba3	0.0203(4)	0.0103(4)	0.0191(30)	-0.0003(3)	0.0076(3)	0.0008(3)
Ba4	0.0175(4)	0.0115(4)	0.0173(3)	-0.0007(3)	0.0050(3)	-0.0022(3)
Fe1	0.0167(8)	0.0059(8)	0.0117(7)	0.0005(6)	0.0020(6)	-0.0006(6)
Cu1	0.0440(15)	0.0123(9)	0.0186(8)	-0.0035(9)	0.0126(9)	-0.0015(7)
Cu2	0.0440(15)	0.0123(9)	0.0186(8)	-0.0035(9)	0.0126(9)	-0.0015(7)
S1	0.0205(15)	0.0099(13)	0.0104(12)	-0.0009(11)	0.0038(11)	-0.0028(10)
S2	0.0198(15)	0.0087(13)	0.0157(13)	-0.0003(11)	0.0054(11)	-0.0004(11)
S3	0.0188(14)	0.0101(13)	0.0126(12)	0.0012(11)	0.0039(11)	0.0007(10)
S4	0.0195(15)	0.0098(13)	0.0134(13)	0.0007(11)	0.0045(11)	-0.0001(10)
S5	0.0204(14)	0.0068(13)	0.0133(13)	0.0000(11)	0.0028(11)	-0.0004(10)
S6	0.0294(17)	0.0097(14)	0.0180(14)	0.0025(13)	0.0093(13)	0.0019(11)

Table SI1: Atomic displacement parameter (\AA^2) for monoclinic $\text{Ba}_4\text{FeCuS}_6$

Table SI2: Geometric parameters (\AA , $^\circ$) for monoclinic $\text{Ba}_4\text{FeCuS}_6$

Ba1–S4 ⁱ	3.107(3)	Ba3–S1	3.310(3)
Ba1–S1 ⁱⁱ	3.172(3)	Ba3–S4 ^{vi}	3.323(3)
Ba1–S4	3.174(3)	Ba3–Cu2 ^{ix}	3.602(18)
Ba1–S6	3.186(3)	Ba3–Fe1	3.6553(18)
Ba1–S5 ⁱⁱ	3.200(3)	Ba3–Cu1 ^{ix}	3.712(2)
Ba1–S3 ⁱⁱⁱ	3.201(3)	Ba3–Cu2	3.81(2)
Ba1–S2	3.311(3)	Ba4–Cu2 ^{vi}	3.035(19)
Ba1–Cu2 ^{iv}	3.64(2)	Ba4–S3	3.138(3)
Ba1–Cu1 ⁱⁱ	3.651(2)	Ba4–S4	3.175(3)
Ba1–Fe1 ⁱⁱ	3.8467(18)	Ba4–S6	3.204(3)

Ba1–Ba2 ^{iv}	4.2791(11)	Ba4–S2 ^x	3.209(3)
Ba1–Ba2 ^v	4.4089(11)	Ba4–S5 ^v	3.254(3)
Ba2–S3 ^{vi}	3.121(3)	Ba4–S1 ^{xi}	3.311(3)
Ba2–S4 ⁱⁱⁱ	3.129(3)	Ba4–S2 ^{vii}	3.345(3)
Ba2–S6	3.150(3)	Ba4–Cu1 ^{vi}	3.550(2)
Ba2–S1 ^v	3.197(3)	Ba4–Fe1 ^{xi}	3.5521(18)
Ba2–S2 ^{vii}	3.236(3)	Ba4–Fe1 ^v	3.7794(18)
Ba2–S1 ^{viii}	3.236(3)	Fe1–S1	2.272(3)
Ba2–S5	3.236(3)	Fe1–S6 ^{vii}	2.282(4)
Ba2–Cu1	3.650(2)	Fe1–S2 ^{vii}	2.287(3)
Ba2–Cu2 ^{iv}	3.67(2)	Fe1–S5	2.296(3)
Ba2–Fe1	3.8767(18)	Cu1–Cu2	0.88(3)
Ba2–Fe1 ^v	3.8858(18)	Cu1–S4 ^{vi}	2.280(3)
Ba3–S3	3.087(3)	Cu1–S3 ^{vi}	2.290(4)
Ba3–Cu1	3.236(2)	Cu1–S5	2.332(4)
Ba3–S2 ^{vii}	3.247(3)	Cu2–S3 ^{vi}	2.311(19)
Ba3–S6 ^{vi}	3.294(3)	Cu2–S4 ^{vi}	2.315(19)
Ba3–S3 ^{vi}	3.303(3)	Cu2–S6 ^{xii}	2.45(3)
Ba3–S5 ^{ix}	3.303(3)		
S4 ⁱ –Ba1–S1 ⁱⁱ	85.92(8)	Cu1 ^{vi} –Ba4–Fe1 ^{xi}	87.85(5)
S4 ⁱ –Ba1–S4	80.98(9)	Cu2 ^{vi} –Ba4–Fe1 ^v	136.6(4)
S1 ⁱⁱ –Ba1–S4	79.43(8)	S3–Ba4–Fe1 ^v	133.74(6)
S4 ⁱ –Ba1–S6	133.11(8)	S4–Ba4–Fe1 ^v	93.80(6)
S1 ⁱⁱ –Ba1–S6	123.23(9)	S6–Ba4–Fe1 ^v	37.00(6)
S4–Ba1–S6	71.15(8)	S2 ^x –Ba4–Fe1 ^v	127.42(6)
S4 ⁱ –Ba1–S5 ⁱⁱ	77.10(8)	S5 ^v –Ba4–Fe1 ^v	37.16(6)
S1 ⁱⁱ –Ba1–S5 ⁱⁱ	71.30(8)	S1 ^{xi} –Ba4–Fe1 ^v	91.81(6)
S4–Ba1–S5 ⁱⁱ	144.32(8)	S2 ^{vii} –Ba4–Fe1 ^v	74.71(6)
S6–Ba1–S5 ⁱⁱ	142.88(8)	Cu1 ^{vi} –Ba4–Fe1 ^v	127.54(5)
S4 ⁱ –Ba1–S3 ⁱⁱⁱ	88.04(8)	Fe1 ^{xi} –Ba4–Fe1 ^v	127.73(5)
S1 ⁱⁱ –Ba1–S3 ⁱⁱⁱ	153.35(8)	Cu2 ^{vi} –Ba4–Ba3 ^{vi}	60.9(5)
S4–Ba1–S3 ⁱⁱⁱ	125.13(8)	S3–Ba4–Ba3 ^{vi}	50.78(6)
S6–Ba1–S3 ⁱⁱⁱ	78.82(8)	S4–Ba4–Ba3 ^{vi}	51.05(6)
S5 ⁱⁱ –Ba1–S3 ⁱⁱⁱ	82.05(8)	S6–Ba4–Ba3 ^{vi}	50.43(6)
S4 ⁱ –Ba1–S2	155.46(8)	S2 ^x –Ba4–Ba3 ^{vi}	140.03(6)
S1 ⁱⁱ –Ba1–S2	88.29(8)	S5 ^v –Ba4–Ba3 ^{vi}	124.52(6)
S4–Ba1–S2	121.30(8)	S1 ^{xi} –Ba4–Ba3 ^{vi}	136.42(6)
S6–Ba1–S2	68.94(8)	S2 ^{vii} –Ba4–Ba3 ^{vi}	87.04(6)
S5 ⁱⁱ –Ba1–S2	78.46(8)	Cu1 ^{vi} –Ba4–Ba3 ^{vi}	48.28(4)
S3 ⁱⁱⁱ –Ba1–S2	86.52(8)	Fe1 ^{xi} –Ba4–Ba3 ^{vi}	136.11(3)
S4 ⁱ –Ba1–Cu2 ^{iv}	107.9(4)	Fe1 ^v –Ba4–Ba3 ^{vi}	87.43(3)
S1 ⁱⁱ –Ba1–Cu2 ^{iv}	164.2(4)	S1–Fe1–S6 ^{vii}	106.19(13)
S4–Ba1–Cu2 ^{iv}	94.9(3)	S1–Fe1–S2 ^{vii}	108.59(12)
S6–Ba1–Cu2 ^{iv}	41.3(4)	S6 ^{vii} –Fe1–S2 ^{vii}	107.25(13)
S5 ⁱⁱ –Ba1–Cu2 ^{iv}	118.5(3)	S1–Fe1–S5	108.78(12)
S3 ⁱⁱⁱ –Ba1–Cu2 ^{iv}	38.8(3)	S6 ^{vii} –Fe1–S5	116.45(13)
S2–Ba1–Cu2 ^{iv}	82.1(3)	S2 ^{vii} –Fe1–S5	109.32(13)
S4 ⁱ –Ba1–Cu1 ⁱⁱ	38.37(6)	S1–Fe1–Ba4 ^{xi}	65.03(9)

S1 ⁱⁱ -Ba1-Cu1 ⁱⁱ	79.64(7)	S6 ^{vii} -Fe1-Ba4 ^{xi}	79.04(10)
S4-Ba1-Cu1 ⁱⁱ	116.55(6)	S2 ^{vii} -Fe1-Ba4 ^{xi}	62.30(9)
S6-Ba1-Cu1 ⁱⁱ	157.10(8)	S5-Fe1-Ba4 ^{xi}	164.48(10)
S5 ⁱⁱ -Ba1-Cu1 ⁱⁱ	39.10(6)	S1-Fe1-Ba3	62.90(9)
S3 ⁱⁱⁱ -Ba1-Cu1 ⁱⁱ	79.55(7)	S6 ^{vii} -Fe1-Ba3	156.14(11)
S2-Ba1-Cu1 ⁱⁱ	117.10(6)	S2 ^{vii} -Fe1-Ba3	61.23(8)
Cu2 ^{iv} -Ba1-Cu1 ⁱⁱ	115.9(4)	S5-Fe1-Ba3	87.41(9)
S4 ⁱ -Ba1-Fe1 ⁱⁱ	87.35(6)	Ba4 ^{xi} -Fe1-Ba3	77.10(4)
S1 ⁱⁱ -Ba1-Fe1 ⁱⁱ	36.18(6)	S1-Fe1-Ba4 ^{vii}	127.63(10)
S4-Ba1-Fe1 ⁱⁱ	115.31(6)	S6 ^{vii} -Fe1-Ba4 ^{vii}	57.66(9)
S6-Ba1-Fe1 ⁱⁱ	138.48(7)	S2 ^{vii} -Fe1-Ba4 ^{vii}	123.68(9)
S5 ⁱⁱ -Ba1-Fe1 ⁱⁱ	36.59(6)	S5-Fe1-Ba4 ^{vii}	58.88(8)
S3 ⁱⁱⁱ -Ba1-Fe1 ⁱⁱ	117.65(6)	Ba4 ^{xi} -Fe1-Ba4 ^{vii}	136.49(5)
S2-Ba1-Fe1 ⁱⁱ	74.16(6)	Ba3-Fe1-Ba4 ^{vii}	146.14(5)
Cu2 ^{iv} -Ba1-Fe1 ⁱⁱ	148.3(3)	S1-Fe1-Ba1 ^{xiii}	55.51(8)
Cu1 ⁱⁱ -Ba1-Fe1 ⁱⁱ	60.30(5)	S6 ^{vii} -Fe1-Ba1 ^{xiii}	113.15(10)
S4 ⁱ -Ba1-Ba2 ^{iv}	46.89(6)	S2 ^{vii} -Fe1-Ba1 ^{xiii}	139.29(10)
S1 ⁱⁱ -Ba1-Ba2 ^{iv}	132.79(6)	S5-Fe1-Ba1 ^{xiii}	56.18(8)
S4-Ba1-Ba2 ^{iv}	92.08(6)	Ba4 ^{xi} -Fe1-Ba1 ^{xiii}	120.45(5)
S6-Ba1-Ba2 ^{iv}	96.38(6)	Ba3-Fe1-Ba1 ^{xiii}	79.27(4)
S5 ⁱⁱ -Ba1-Ba2 ^{iv}	93.08(6)	Ba4 ^{vii} -Fe1-Ba1 ^{xiii}	83.79(4)
S3 ⁱⁱⁱ -Ba1-Ba2 ^{iv}	46.62(5)	S1-Fe1-Ba2	141.00(10)
S2-Ba1-Ba2 ^{iv}	133.13(6)	S6 ^{vii} -Fe1-Ba2	112.63(10)
Cu2 ^{iv} -Ba1-Ba2 ^{iv}	61.6(4)	S2 ^{vii} -Fe1-Ba2	56.49(9)
Cu1 ⁱⁱ -Ba1-Ba2 ^{iv}	62.87(4)	S5-Fe1-Ba2	56.49(8)
Fe1 ⁱⁱ -Ba1-Ba2 ^{iv}	123.09(3)	Ba4 ^{xi} -Fe1-Ba2	118.46(5)
S4 ⁱ -Ba1-Ba2 ^v	132.35(6)	Ba3-Fe1-Ba2	79.58(4)
S1 ⁱⁱ -Ba1-Ba2 ^v	46.43(6)	Ba4 ^{vii} -Fe1-Ba2	78.83(3)
S4-Ba1-Ba2 ^v	88.72(6)	Ba1 ^{xiii} -Fe1-Ba2	109.58(4)
S6-Ba1-Ba2 ^v	84.80(6)	S1-Fe1-Ba2 ^{vii}	55.35(8)
S5 ⁱⁱ -Ba1-Ba2 ^v	85.70(6)	S6 ^{vii} -Fe1-Ba2 ^{vii}	54.16(9)
S3 ⁱⁱⁱ -Ba1-Ba2 ^v	133.41(6)	S2 ^{vii} -Fe1-Ba2 ^{vii}	136.87(10)
S2-Ba1-Ba2 ^v	46.94(5)	S5-Fe1-Ba2 ^{vii}	113.77(9)
Cu2 ^{iv} -Ba1-Ba2 ^v	119.3(4)	Ba4 ^{xi} -Fe1-Ba2 ^{vii}	75.30(4)
Cu1 ⁱⁱ -Ba1-Ba2 ^v	115.88(4)	Ba3-Fe1-Ba2 ^{vii}	118.18(4)
Fe1 ⁱⁱ -Ba1-Ba2 ^v	55.66(3)	Ba4 ^{vii} -Fe1-Ba2 ^{vii}	81.99(4)
Ba2 ^{iv} -Ba1-Ba2 ^v	178.74(3)	Ba1 ^{xiii} -Fe1-Ba2 ^{vii}	69.52(3)
S3 ^{vi} -Ba2-S4 ⁱⁱⁱ	89.09(8)	Ba2-Fe1-Ba2 ^{vii}	160.75(5)
S3 ^{vi} -Ba2-S6	84.75(8)	Cu2-Cu1-S4 ^{vi}	81.2(12)
S4 ⁱⁱⁱ -Ba2-S6	78.75(9)	Cu2-Cu1-S3 ^{vi}	80.2(12)
S3 ^{vi} -Ba2-S1 ^v	154.64(8)	S4 ^{vi} -Cu1-S3 ^{vi}	116.08(13)
S4 ⁱⁱⁱ -Ba2-S1 ^v	88.34(8)	Cu2-Cu1-S5	137.3(12)
S6-Ba2-S1 ^v	70.01(8)	S4 ^{vi} -Cu1-S5	116.96(14)
S3 ^{vi} -Ba2-S2 ^{vii}	83.37(8)	S3 ^{vi} -Cu1-S5	117.87(14)
S4 ⁱⁱⁱ -Ba2-S2 ^{vii}	156.79(8)	Cu2-Cu1-Ba3	125.2(12)
S6-Ba2-S2 ^{vii}	78.72(8)	S4 ^{vi} -Cu1-Ba3	71.73(9)
S1 ^v -Ba2-S2 ^{vii}	89.19(8)	S3 ^{vi} -Cu1-Ba3	71.08(9)
S3 ^{vi} -Ba2-S1 ^{viii}	121.96(8)	S5-Cu1-Ba3	97.48(11)

S4 ⁱⁱⁱ –Ba2–S1 ^{viii}	79.13(8)	Cu2–Cu1–Ba4 ^{vi}	48.4(12)
S6–Ba2–S1 ^{viii}	144.72(8)	S4 ^{vi} –Cu1–Ba4 ^{vi}	61.52(9)
S1 ^v –Ba2–S1 ^{viii}	82.22(8)	S3 ^{vi} –Cu1–Ba4 ^{vi}	60.54(9)
S2 ^{vii} –Ba2–S1 ^{viii}	123.33(8)	S5–Cu1–Ba4 ^{vi}	174.25(12)
S3 ^{vi} –Ba2–S5	77.02(8)	Ba3–Cu1–Ba4 ^{vi}	76.77(4)
S4 ⁱⁱⁱ –Ba2–S5	128.96(8)	Cu2–Cu1–Ba2	114.3(12)
S6–Ba2–S5	145.69(8)	S4 ^{vi} –Cu1–Ba2	160.11(12)
S1 ^v –Ba2–S5	123.11(8)	S3 ^{vi} –Cu1–Ba2	58.12(8)
S2 ^{vii} –Ba2–S5	70.57(8)	S5–Cu1–Ba2	60.86(9)
S1 ^{viii} –Ba2–S5	68.39(7)	Ba3–Cu1–Ba2	88.70(6)
S3 ^{vi} –Ba2–Cu1	38.55(6)	Ba4 ^{vi} –Cu1–Ba2	118.42(5)
S4 ⁱⁱⁱ –Ba2–Cu1	108.85(7)	Cu2–Cu1–Ba1 ^{xiii}	116.4(12)
S6–Ba2–Cu1	120.42(7)	S4 ^{vi} –Cu1–Ba1 ^{xiii}	57.80(9)
S1 ^v –Ba2–Cu1	160.91(7)	S3 ^{vi} –Cu1–Ba1 ^{xiii}	158.62(12)
S2 ^{vii} –Ba2–Cu1	78.23(7)	S5–Cu1–Ba1 ^{xiii}	59.94(9)
S1 ^{viii} –Ba2–Cu1	92.68(6)	Ba3–Cu1–Ba1 ^{xiii}	87.86(5)
S5–Ba2–Cu1	39.00(6)	Ba4 ^{vi} –Cu1–Ba1 ^{xiii}	119.21(5)
S3 ^{vi} –Ba2–Cu2 ^{iv}	78.8(3)	Ba2–Cu1–Ba1 ^{xiii}	119.59(6)
S4 ⁱⁱⁱ –Ba2–Cu2 ^{iv}	38.8(3)	Cu2–Cu1–Ba3 ^{viii}	76.0(12)
S6–Ba2–Cu2 ^{iv}	41.1(4)	S4 ^{vi} –Cu1–Ba3 ^{viii}	116.61(10)
S1 ^v –Ba2–Cu2 ^{iv}	83.5(3)	S3 ^{vi} –Cu1–Ba3 ^{viii}	116.88(10)
S2 ^{vii} –Ba2–Cu2 ^{iv}	118.1(3)	S5–Cu1–Ba3 ^{viii}	61.33(8)
S1 ^{viii} –Ba2–Cu2 ^{iv}	116.4(3)	Ba3–Cu1–Ba3 ^{viii}	158.80(8)
S5–Ba2–Cu2 ^{iv}	153.1(4)	Ba4 ^{vi} –Cu1–Ba3 ^{viii}	124.42(7)
Cu1–Ba2–Cu2 ^{iv}	115.0(3)	Ba2–Cu1–Ba3 ^{viii}	80.62(4)
S3 ^{vi} –Ba2–Fe1	86.56(6)	Ba1 ^{xiii} –Cu1–Ba3 ^{viii}	81.70(4)
S4 ⁱⁱⁱ –Ba2–Fe1	165.23(7)	Cu1–Cu2–S3 ^{vi}	77.6(13)
S6–Ba2–Fe1	114.82(7)	Cu1–Cu2–S4 ^{vi}	76.7(13)
S1 ^v –Ba2–Fe1	101.54(6)	S3 ^{vi} –Cu2–S4 ^{vi}	113.9(8)
S2 ^{vii} –Ba2–Fe1	36.11(6)	Cu1–Cu2–S6 ^{xii}	152.7(15)
S1 ^{viii} –Ba2–Fe1	91.23(6)	S3 ^{vi} –Cu2–S6 ^{xii}	116.9(9)
S5–Ba2–Fe1	36.27(6)	S4 ^{vi} –Cu2–S6 ^{xii}	113.5(9)
Cu1–Ba2–Fe1	60.02(5)	Cu1–Cu2–Ba4 ^{vi}	119.0(14)
Cu2 ^{iv} –Ba2–Fe1	152.4(3)	S3 ^{vi} –Cu2–Ba4 ^{vi}	70.4(5)
S3 ^{vi} –Ba2–Fe1 ^v	119.06(6)	S4 ^{vi} –Cu2–Ba4 ^{vi}	71.4(5)
S4 ⁱⁱⁱ –Ba2–Fe1 ^v	90.43(6)	S6 ^{xii} –Cu2–Ba4 ^{vi}	88.2(7)
S6–Ba2–Fe1 ^v	35.96(6)	Cu1–Cu2–Ba3 ^{viii}	90.2(12)
S1 ^v –Ba2–Fe1 ^v	35.77(6)	S3 ^{vi} –Cu2–Ba3 ^{viii}	120.3(6)
S2 ^{vii} –Ba2–Fe1 ^v	74.40(6)	S4 ^{vi} –Cu2–Ba3 ^{viii}	119.5(7)
S1 ^{viii} –Ba2–Fe1 ^v	117.65(6)	S6 ^{xii} –Cu2–Ba3 ^{viii}	62.6(4)
S5–Ba2–Fe1 ^v	139.22(6)	Ba4 ^{vi} –Cu2–Ba3 ^{viii}	150.8(9)
Cu1–Ba2–Fe1 ^v	147.03(4)	Cu1–Cu2–Ba1 ^{xii}	121.1(14)
Cu2 ^{iv} –Ba2–Fe1 ^v	64.6(4)	S3 ^{vi} –Cu2–Ba1 ^{xii}	60.2(4)
Fe1–Ba2–Fe1 ^v	104.017(18)	S4 ^{vi} –Cu2–Ba1 ^{xii}	155.7(9)
S3 ^{vi} –Ba2–Ba1 ^{xii}	48.19(6)	S6 ^{xii} –Cu2–Ba1 ^{xii}	59.3(5)
S4 ⁱⁱⁱ –Ba2–Ba1 ^{xii}	46.46(6)	Ba4 ^{vi} –Cu2–Ba1 ^{xii}	84.9(5)
S6–Ba2–Ba1 ^{xii}	96.29(7)	Ba3 ^{viii} –Cu2–Ba1 ^{xii}	79.3(4)
S1 ^v –Ba2–Ba1 ^{xii}	134.80(6)	Cu1–Cu2–Ba2 ^{xii}	118.6(14)

S2 ^{vii} –Ba2–Ba1 ^{xii}	131.51(6)	S3 ^{vi} –Cu2–Ba2 ^{xii}	155.1(8)
S1 ^{viii} –Ba2–Ba1 ^{xii}	87.95(6)	S4 ^{vi} –Cu2–Ba2 ^{xii}	57.8(4)
S5–Ba2–Ba1 ^{xii}	92.88(6)	S6 ^{xii} –Cu2–Ba2 ^{xii}	57.9(4)
Cu1–Ba2–Ba1 ^{xii}	62.87(4)	Ba4 ^{vi} –Cu2–Ba2 ^{xii}	84.8(5)
Cu2 ^{iv} –Ba2–Ba1 ^{xii}	61.9(4)	Ba3 ^{viii} –Cu2–Ba2 ^{xii}	80.5(4)
Fe1–Ba2–Ba1 ^{xii}	122.77(3)	Ba1 ^{xii} –Cu2–Ba2 ^{xii}	116.4(7)
Fe1 ^v –Ba2–Ba1 ^{xii}	126.43(3)	Cu1–Cu2–Ba3	43.9(10)
S3–Ba3–Cu1	126.29(7)	S3 ^{vi} –Cu2–Ba3	59.4(5)
S3–Ba3–S2 ^{vii}	85.62(8)	S4 ^{vi} –Cu2–Ba3	59.9(5)
Cu1–Ba3–S2 ^{vii}	84.43(7)	S6 ^{xii} –Cu2–Ba3	163.2(8)
S3–Ba3–S6 ^{vi}	82.89(8)	Ba4 ^{vi} –Cu2–Ba3	75.1(4)
Cu1–Ba3–S6 ^{vi}	98.42(7)	Ba3 ^{viii} –Cu2–Ba3	134.1(7)
S2 ^{vii} –Ba3–S6 ^{vi}	167.50(8)	Ba1 ^{xii} –Cu2–Ba3	119.6(5)
S3–Ba3–S3 ^{vi}	85.31(8)	Ba2 ^{xii} –Cu2–Ba3	117.6(5)
Cu1–Ba3–S3 ^{vi}	40.99(6)	Fe1–S1–Ba1 ^{xiii}	88.31(10)
S2 ^{vii} –Ba3–S3 ^{vi}	80.40(8)	Fe1–S1–Ba2 ^{vii}	88.88(10)
S6 ^{vi} –Ba3–S3 ^{vi}	93.72(8)	Ba1 ^{xiii} –S1–Ba2 ^{vii}	87.61(8)
S3–Ba3–S5 ^{ix}	82.13(8)	Fe1–S1–Ba2 ^{ix}	169.85(13)
Cu1–Ba3–S5 ^{ix}	150.26(7)	Ba1 ^{xiii} –S1–Ba2 ^{ix}	99.57(8)
S2 ^{vii} –Ba3–S5 ^{ix}	90.01(8)	Ba2 ^{vii} –S1–Ba2 ^{ix}	97.78(8)
S6 ^{vi} –Ba3–S5 ^{ix}	93.29(8)	Fe1–S1–Ba3	79.44(10)
S3 ^{vi} –Ba3–S5 ^{ix}	164.74(8)	Ba1 ^{xiii} –S1–Ba3	95.20(8)
S3–Ba3–S1	138.75(8)	Ba2 ^{vii} –S1–Ba3	167.87(10)
Cu1–Ba3–S1	84.10(7)	Ba2 ^{ix} –S1–Ba3	93.39(7)
S2 ^{vii} –Ba3–S1	68.74(7)	Fe1–S1–Ba4 ^{xi}	76.51(9)
S6 ^{vi} –Ba3–S1	123.57(8)	Ba1 ^{xiii} –S1–Ba4 ^{xi}	164.45(10)
S3 ^{vi} –Ba3–S1	119.51(7)	Ba2 ^{vii} –S1–Ba4 ^{xi}	88.71(8)
S5 ^{ix} –Ba3–S1	66.73(7)	Ba2 ^{ix} –S1–Ba4 ^{xi}	95.90(8)
S3–Ba3–S4 ^{vi}	140.84(8)	Ba3–S1–Ba4 ^{xi}	85.45(7)
Cu1–Ba3–S4 ^{vi}	40.64(6)	Fe1 ^v –S2–Ba4 ⁱⁱⁱ	78.57(9)
S2 ^{vii} –Ba3–S4 ^{vi}	119.63(7)	Fe1 ^v –S2–Ba2 ^v	87.40(10)
S6 ^{vi} –Ba3–S4 ^{vi}	68.00(8)	Ba4 ⁱⁱⁱ –S2–Ba2 ^v	164.47(11)
S3 ^{vi} –Ba3–S4 ^{vi}	71.62(8)	Fe1 ^v –S2–Ba3 ^v	80.65(10)
S5 ^{ix} –Ba3–S4 ^{vi}	123.62(8)	Ba4 ⁱⁱⁱ –S2–Ba3 ^v	88.19(8)
S1–Ba3–S4 ^{vi}	80.36(8)	Ba2 ^v –S2–Ba3 ^v	96.14(8)
S3–Ba3–Cu2 ^{ix}	80.4(3)	Fe1 ^v –S2–Ba1	87.33(10)
Cu1–Ba3–Cu2 ^{ix}	132.8(4)	Ba4 ⁱⁱⁱ –S2–Ba1	88.03(8)
S2 ^{vii} –Ba3–Cu2 ^{ix}	140.8(4)	Ba2 ^v –S2–Ba1	84.66(7)
S6 ^{vi} –Ba3–Cu2 ^{ix}	41.3(4)	Ba3 ^v –S2–Ba1	167.89(11)
S3 ^{vi} –Ba3–Cu2 ^{ix}	133.9(4)	Fe1 ^v –S2–Ba4 ^v	170.33(13)
S5 ^{ix} –Ba3–Cu2 ^{ix}	52.1(4)	Ba4 ⁱⁱⁱ –S2–Ba4 ^v	95.08(8)
S1–Ba3–Cu2 ^{ix}	99.2(4)	Ba2 ^v –S2–Ba4 ^v	99.65(8)
S4 ^{vi} –Ba3–Cu2 ^{ix}	93.1(4)	Ba3 ^v –S2–Ba4 ^v	91.95(8)
S3–Ba3–Fe1	123.61(6)	Ba1–S2–Ba4 ^v	99.84(8)
Cu1–Ba3–Fe1	66.00(5)	Cu1 ^{vi} –S3–Cu2 ^{vi}	22.2(6)
S2 ^{vii} –Ba3–Fe1	38.12(6)	Cu1 ^{vi} –S3–Ba3	162.59(14)
S6 ^{vi} –Ba3–Fe1	153.46(6)	Cu2 ^{vi} –S3–Ba3	164.9(5)
S3 ^{vi} –Ba3–Fe1	87.73(6)	Cu1 ^{vi} –S3–Ba2 ^{vi}	83.33(10)

S5 ^{ix} –Ba3–Fe1	92.02(6)	Cu2 ^{vi} –S3–Ba2 ^{vi}	96.7(5)
S1–Ba3–Fe1	37.66(5)	Ba3–S3–Ba2 ^{vi}	98.45(9)
S4 ^{vi} –Ba3–Fe1	87.50(6)	Cu1 ^{vi} –S3–Ba4	80.00(10)
Cu2 ^{ix} –Ba3–Fe1	136.1(4)	Cu2 ^{vi} –S3–Ba4	65.7(5)
S3–Ba3–Cu1 ^{ix}	80.02(6)	Ba3–S3–Ba4	99.19(8)
Cu1–Ba3–Cu1 ^{ix}	142.76(5)	Ba2 ^{vi} –S3–Ba4	162.35(11)
S2 ^{vii} –Ba3–Cu1 ^{ix}	127.51(7)	Cu1 ^{vi} –S3–Ba1 ^x	97.43(11)
S6 ^{vi} –Ba3–Cu1 ^{ix}	55.04(7)	Cu2 ^{vi} –S3–Ba1 ^x	81.0(6)
S3 ^{vi} –Ba3–Cu1 ^{ix}	146.70(7)	Ba3–S3–Ba1 ^x	99.97(9)
S5 ^{ix} –Ba3–Cu1 ^{ix}	38.26(6)	Ba2 ^{vi} –S3–Ba1 ^x	85.19(7)
S1–Ba3–Cu1 ^{ix}	90.38(6)	Ba4–S3–Ba1 ^x	91.23(8)
S4 ^{vi} –Ba3–Cu1 ^{ix}	102.12(6)	Cu1 ^{vi} –S3–Ba3 ^{vi}	67.94(9)
Cu2 ^{ix} –Ba3–Cu1 ^{ix}	13.8(4)	Cu2 ^{vi} –S3–Ba3 ^{vi}	83.6(6)
Fe1–Ba3–Cu1 ^{ix}	125.21(5)	Ba3–S3–Ba3 ^{vi}	94.68(8)
S3–Ba3–Cu2	120.9(3)	Ba2 ^{vi} –S3–Ba3 ^{vi}	97.27(8)
Cu1–Ba3–Cu2	10.9(3)	Ba4–S3–Ba3 ^{vi}	81.83(7)
S2 ^{vii} –Ba3–Cu2	93.3(3)	Ba1 ^x –S3–Ba3 ^{vi}	164.63(11)
S6 ^{vi} –Ba3–Cu2	88.4(3)	Cu1 ^{vi} –S4–Cu2 ^{vi}	22.2(6)
S3 ^{vi} –Ba3–Cu2	37.0(3)	Cu1 ^{vi} –S4–Ba1 ⁱ	83.83(10)
S5 ^{ix} –Ba3–Cu2	156.9(3)	Cu2 ^{vi} –S4–Ba1 ⁱ	97.9(5)
S1–Ba3–Cu2	93.2(3)	Cu1 ^{vi} –S4–Ba2 ^x	99.71(11)
S4 ^{vi} –Ba3–Cu2	37.0(3)	Cu2 ^{vi} –S4–Ba2 ^x	83.5(6)
Cu2 ^{ix} –Ba3–Cu2	125.2(6)	Ba1 ⁱ –S4–Ba2 ^x	86.65(8)
Fe1–Ba3–Cu2	76.9(3)	Cu1 ^{vi} –S4–Ba1	158.38(14)
Cu1 ^{ix} –Ba3–Cu2	137.0(3)	Cu2 ^{vi} –S4–Ba1	162.6(5)
Cu2 ^{vi} –Ba4–S3	43.9(4)	Ba1 ⁱ –S4–Ba1	99.02(9)
Cu2 ^{vi} –Ba4–S4	43.7(4)	Ba2 ^x –S4–Ba1	101.85(9)
S3–Ba4–S4	75.78(8)	Cu1 ^{vi} –S4–Ba4	79.34(10)
Cu2 ^{vi} –Ba4–S6	105.6(5)	Cu2 ^{vi} –S4–Ba4	64.9(5)
S3–Ba4–S6	98.76(8)	Ba1 ⁱ –S4–Ba4	162.75(11)
S4–Ba4–S6	70.92(8)	Ba2 ^x –S4–Ba4	92.42(8)
Cu2 ^{vi} –Ba4–S2 ^x	94.1(4)	Ba1–S4–Ba4	98.03(8)
S3–Ba4–S2 ^x	89.36(8)	Cu1 ^{vi} –S4–Ba3 ^{vi}	67.62(9)
S4–Ba4–S2 ^x	130.58(8)	Cu2 ^{vi} –S4–Ba3 ^{vi}	83.1(6)
S6–Ba4–S2 ^x	158.47(8)	Ba1 ⁱ –S4–Ba3 ^{vi}	96.18(8)
Cu2 ^{vi} –Ba4–S5 ^v	155.4(4)	Ba2 ^x –S4–Ba3 ^{vi}	166.52(11)
S3–Ba4–S5 ^v	160.21(8)	Ba1–S4–Ba3 ^{vi}	90.76(8)
S4–Ba4–S5 ^v	117.53(8)	Ba4–S4–Ba3 ^{vi}	80.97(7)
S6–Ba4–S5 ^v	74.12(8)	Fe1–S5–Cu1	109.10(13)
S2 ^x –Ba4–S5 ^v	91.58(8)	Fe1–S5–Ba1 ^{xiii}	87.22(10)
Cu2 ^{vi} –Ba4–S1 ^{xi}	92.5(4)	Cu1–S5–Ba1 ^{xiii}	80.96(9)
S3–Ba4–S1 ^{xi}	130.93(8)	Fe1–S5–Ba2	87.24(10)
S4–Ba4–S1 ^{xi}	85.61(8)	Cu1–S5–Ba2	80.14(10)
S6–Ba4–S1 ^{xi}	117.48(8)	Ba1 ^{xiii} –S5–Ba2	157.34(10)
S2 ^x –Ba4–S1 ^{xi}	69.17(8)	Fe1–S5–Ba4 ^{vii}	83.95(10)
S5 ^v –Ba4–S1 ^{xi}	67.28(7)	Cu1–S5–Ba4 ^{vii}	166.36(13)
Cu2 ^{vi} –Ba4–S2 ^{vii}	127.1(4)	Ba1 ^{xiii} –S5–Ba4 ^{vii}	104.21(9)
S3–Ba4–S2 ^{vii}	83.18(8)	Ba2–S5–Ba4 ^{vii}	97.03(8)

S4–Ba4–S2 ^{vii}	137.41(8)	Fe1–S5–Ba3 ^{viii}	170.43(13)
S6–Ba4–S2 ^{vii}	76.38(8)	Cu1–S5–Ba3 ^{viii}	80.41(10)
S2 ^x –Ba4–S2 ^{vii}	84.92(8)	Ba1 ^{xiii} –S5–Ba3 ^{viii}	95.57(8)
S5 ^v –Ba4–S2 ^{vii}	77.22(7)	Ba2–S5–Ba3 ^{viii}	93.51(8)
S1 ^{xi} –Ba4–S2 ^{vii}	134.55(8)	Ba4 ^{vii} –S5–Ba3 ^{viii}	86.49(7)
Cu2 ^{vi} –Ba4–Cu1 ^{vi}	12.6(5)	Fe1 ^v –S6–Cu2 ^{iv}	117.5(5)
S3–Ba4–Cu1 ^{vi}	39.45(6)	Fe1 ^v –S6–Ba2	89.87(11)
S4–Ba4–Cu1 ^{vi}	39.13(6)	Cu2 ^{iv} –S6–Ba2	81.0(4)
S6–Ba4–Cu1 ^{vi}	93.99(7)	Fe1 ^v –S6–Ba1	90.50(11)
S2 ^x –Ba4–Cu1 ^{vi}	104.71(7)	Cu2 ^{iv} –S6–Ba1	79.4(4)
S5 ^v –Ba4–Cu1 ^{vi}	156.59(7)	Ba2–S6–Ba1	158.12(12)
S1 ^{xi} –Ba4–Cu1 ^{vi}	102.57(6)	Fe1 ^v –S6–Ba4	85.34(11)
S2 ^{vii} –Ba4–Cu1 ^{vi}	120.24(6)	Cu2 ^{iv} –S6–Ba4	156.8(5)
Cu2 ^{vi} –Ba4–Fe1 ^{xi}	75.3(5)	Ba2–S6–Ba4	104.64(9)
S3–Ba4–Fe1 ^{xi}	98.38(6)	Ba1–S6–Ba4	97.20(9)
S4–Ba4–Fe1 ^{xi}	96.05(6)	Fe1 ^v –S6–Ba3 ^{vi}	166.35(15)
S6–Ba4–Fe1 ^{xi}	155.06(7)	Cu2 ^{iv} –S6–Ba3 ^{vi}	76.1(5)
S2 ^x –Ba4–Fe1 ^{xi}	39.13(6)	Ba2–S6–Ba3 ^{vi}	93.67(8)
S5 ^v –Ba4–Fe1 ^{xi}	94.76(6)	Ba1–S6–Ba3 ^{vi}	91.09(8)
S1 ^{xi} –Ba4–Fe1 ^{xi}	38.46(6)	Ba4–S6–Ba3 ^{vi}	81.00(7)
S2 ^{vii} –Ba4–Fe1 ^{xi}	123.69(6)		

Symmetry codes: (i) $-x+1, -y+2, -z+1$; (ii) $x, y+1, z$; (iii) $x+1/2, -y+3/2, z+1/2$; (iv) $-x+3/2, y+1/2, -z+3/2$; (v) $-x+1/2, y+1/2, -z+3/2$; (vi) $-x+1, -y+1, -z+1$; (vii) $-x+1/2, y-1/2, -z+3/2$; (viii) $x+1/2, -y+1/2, z+1/2$; (ix) $x-1/2, -y+1/2, z-1/2$; (x) $x-1/2, -y+3/2, z-1/2$; (xi) $-x, -y+1, -z+1$; (xii) $-x+3/2, y-1/2, -z+3/2$; (xiii) $x, y-1, z$.

Table SI3: The fractional atomic coordinates and $U_{\text{iso}}/U_{\text{eq}}$ values for the orthorhombic $\text{Ba}_4\text{FeCuS}_6$ (space group: $Pnma$) structure.^a

Atoms	Wyckoff position	Site symmetry	SOF	x	y	z	$U_{\text{iso}}/U_{\text{eq}}$
Ba1	4c	.m.	1	0.48949(9)	0.250000	0.32485(9)	0.0160(3)
Ba2	4c	.m.	1	0.82477(11)	0.250000	0.51773(14)	0.0304(4)
Fe1	4c	.m.	0.5	0.7563(3)	0.750000	0.3061(3)	0.0677(18)
Cu1	4c	.m.	0.5	0.7563(3)	0.750000	0.3061(3)	0.0677(18)
S1	4c	.m.	1	0.6338(4)	0.750000	0.4451(4)	0.0176(10)
S2	4c	.m.	1	0.9384(4)	0.750000	0.3609(4)	0.0206(10)
S3	4c	.m.	1	0.7240(4)	0.250000	0.2068(4)	0.043(2)

^a $U_{\text{iso}}/U_{\text{eq}}$ is the one-third value of the trace of the orthogonalized U_{ij} tensor.

Table SI4: Atomic displacement parameter (\AA^2) for Orthorhombic $\text{Ba}_4\text{FeCuS}_6$ (space group: $Pnma$)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ba1	0.0084(5)	0.0244(6)	0.0153(5)	0.000	-0.0001(4)	0.000
Ba2	0.0123(6)	0.0265(7)	0.0524(9)	0.000	0.0048(5)	0.000
Fe1	0.0125(14)	0.165(6)	0.0252(17)	0.000	0.0047(12)	0.000
Cu1	0.0125(14)	0.165(6)	0.0252(17)	0.000	0.0047(12)	0.000
S1	0.0115(19)	0.028(3)	0.0134(19)	0.000	0.0012(16)	0.000
S2	0.015(2)	0.029(3)	0.018(2)	0.000	-0.0025(17)	0.000
S3	0.008(2)	0.104(7)	0.019(2)	0.000	0.0022(18)	0.000

Reference

1 G. Panigrahi, S. Yadav, S. Jana, K. V. Ramanujachary, M. K. Niranjana and J. Prakash, *Dalton Trans.*, 2023, **52**, 621–634.