

## Electronic Supplementary Information

# Construction of Metal–Organic Frameworks ZIF-8-derived Heterostructure ZnO/BiVO<sub>4</sub> for Enhanced Photocatalytic Degradation of Carbamazepine

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## Section 1: Experimental

### *Chemicals and materials*

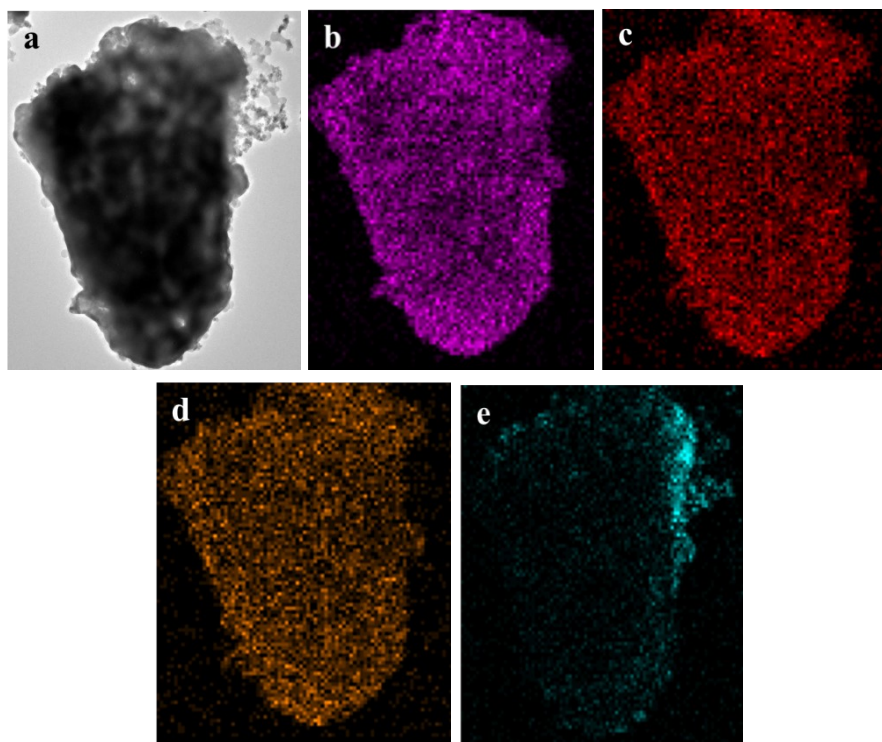
Carbamazepine (CBZ),  $\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$ ,  $\text{NH}_4\text{VO}_3$ , and  $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$  were obtained from Aladdin Chemistry Co., Ltd. (Shanghai, China).  $\text{HNO}_3$  and 2-methylimidazole were purchased from Sinopharm Chemical Reagent Co. (Shanghai, China). Methanol was obtained from Fuyu Fine Chemical Co. (Tianjin, China). All chemical reagents were of analytical grade and were used directly without further treatment.

### *Characterization*

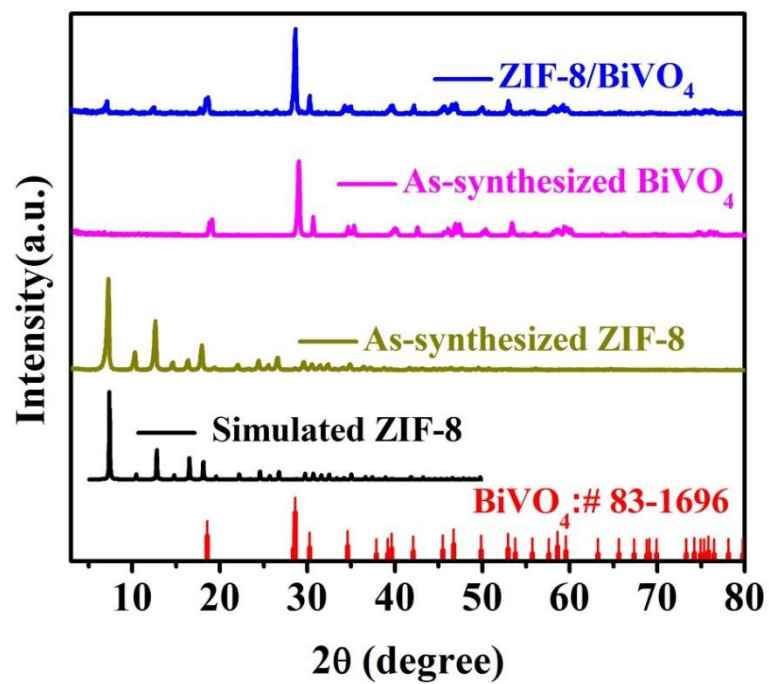
The X-ray diffraction (XRD) patterns of the as-prepared samples were obtained using a Rigaku Ultima IV diffractometer with  $\text{Cu } K_\alpha$  radiation. Transmission electron microscopy (TEM) was performed using a FEI Tecnai G2F20 electron microscope (Hillsboro, OR, USA). X-ray photoelectron spectroscopy (XPS) was performed using a Kratos XSAM 800 photoelectron spectrometer (Manchester, UK).  $\text{N}_2$  adsorption-desorption experiment was performed on 3-Flex automated gas sorption analyzer at 77 K. The surface area and pore-size-distribution curves were calculated using Brunauer-Emmett-Teller (BET) model and DFT method, respectively. Photoluminescence (PL) was measured using FL7000 (Hitachi, Japan). UV-vis diffuse reflectance spectra (UV-vis DRS) were recorded using a UV-3600 instrument (Shimadzu, Japan). The electrochemical impedance spectra (EIS) and photocurrent were measured on an electrochemical workstation (CHI660E, China) using a three-electrode cell system.

The intermediates of CBZ were identified by a Shimadzu 20AD LC coupled with AB Sciex 5600 hybrid quadrupole TOF-MS/MS in the positive electrospray ionization mode with a scan range of  $m/z$  50~500. Chromatographic separation was determined with a C18 column (4.6×150 mm, 5  $\mu\text{m}$ ). The flow rate was 0.3 mL/min, and the injection volume was 2  $\mu\text{L}$ .

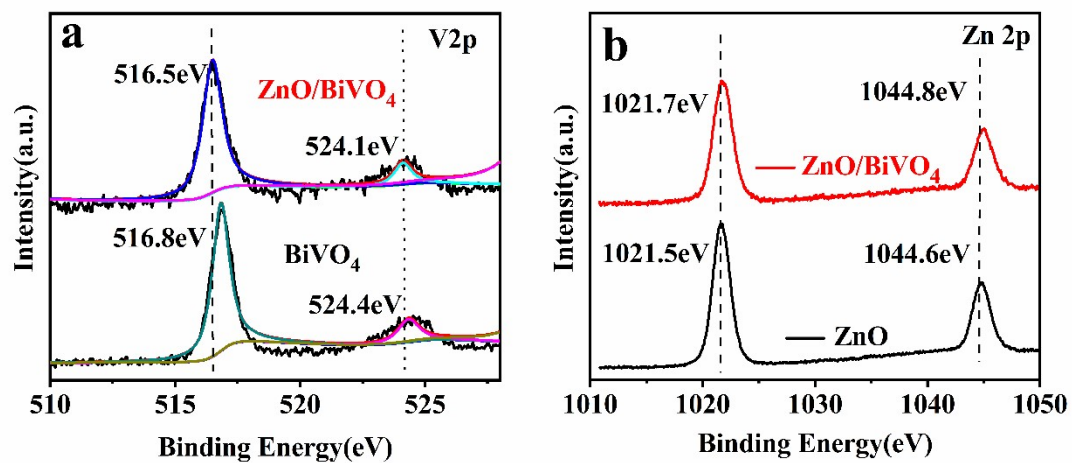
## Section 2: Figures



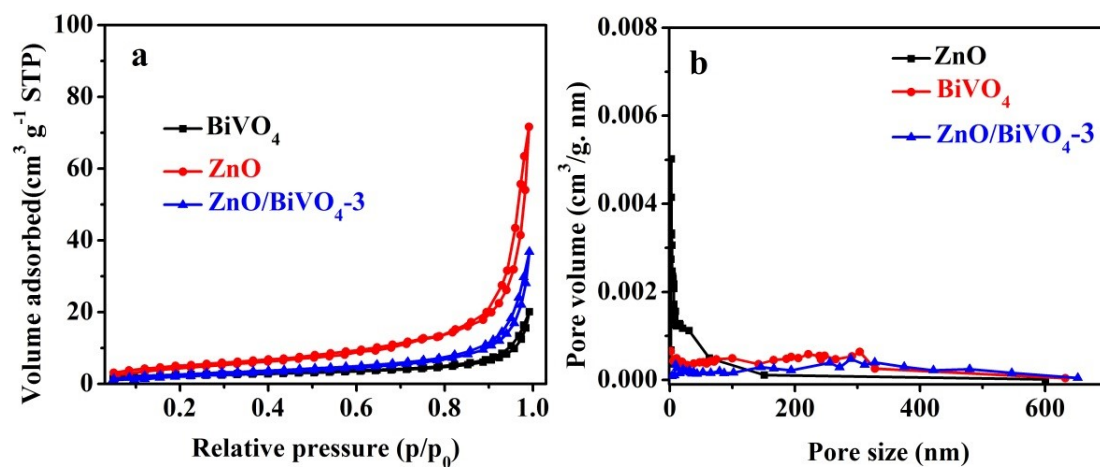
**Figure S1.** Elemental mapping images of ZnO/BiVO<sub>4</sub> structures: (a) the test area and element distribution of (b) Bi, (c) O, (d) V, and (e) Zn.



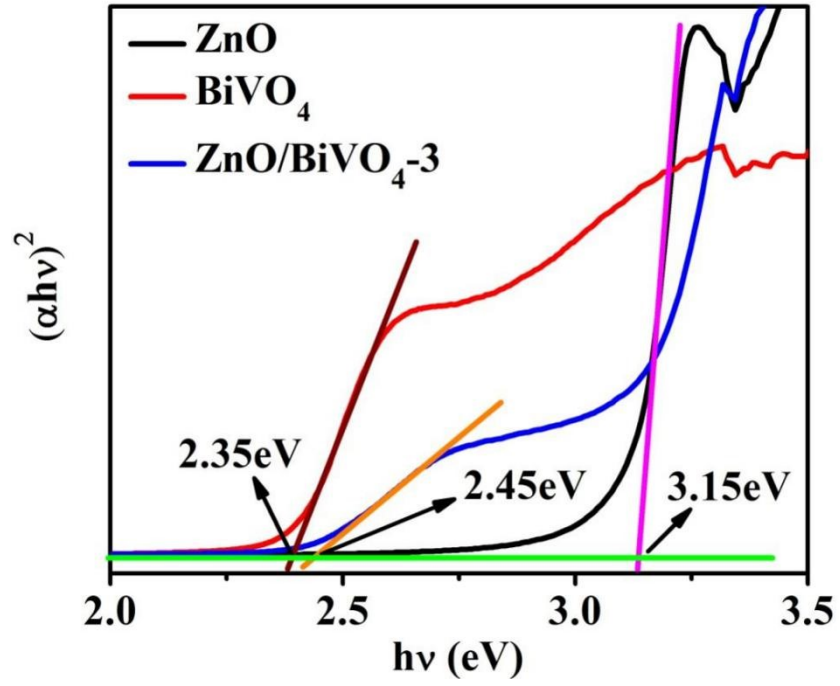
**Figure S2.** XRD patterns of ZIF-8 and ZIF-8-BiVO<sub>4</sub>.



**Figure S3.** XPS spectrum of the as-prepared samples. (a) V 2p and (b) Zn 2p.

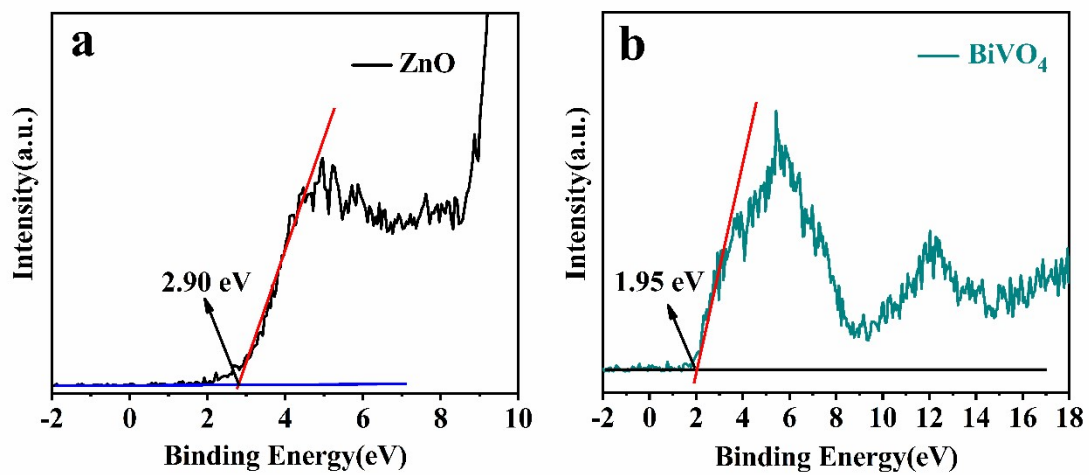


**Figure S4.** (a) N<sub>2</sub> adsorption-desorption isotherms and (b) pore size distributions of ZnO, BiVO<sub>4</sub> and ZnO/BiVO<sub>4</sub>-3 samples.

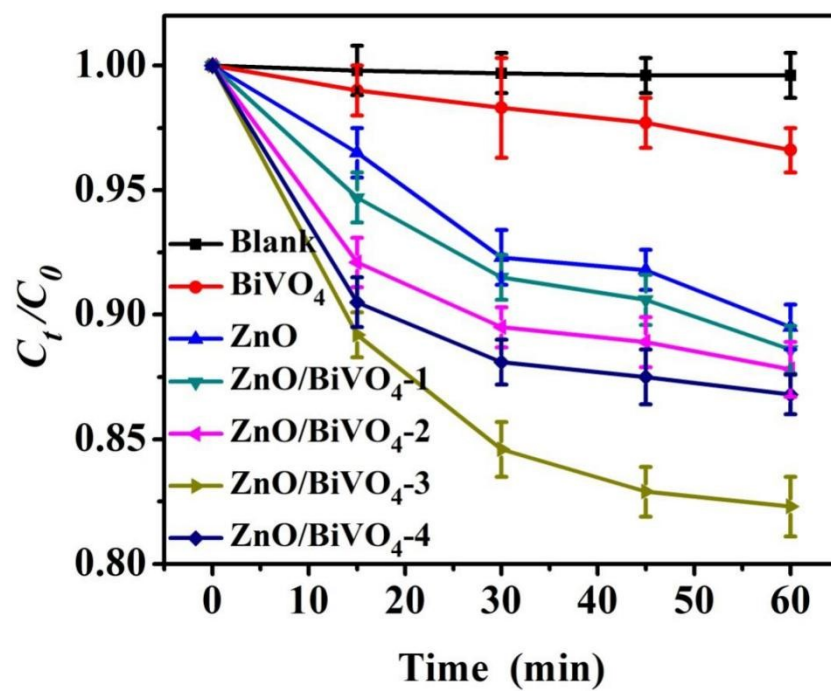


**Figure S5.** Tauc plots of prepared samples.

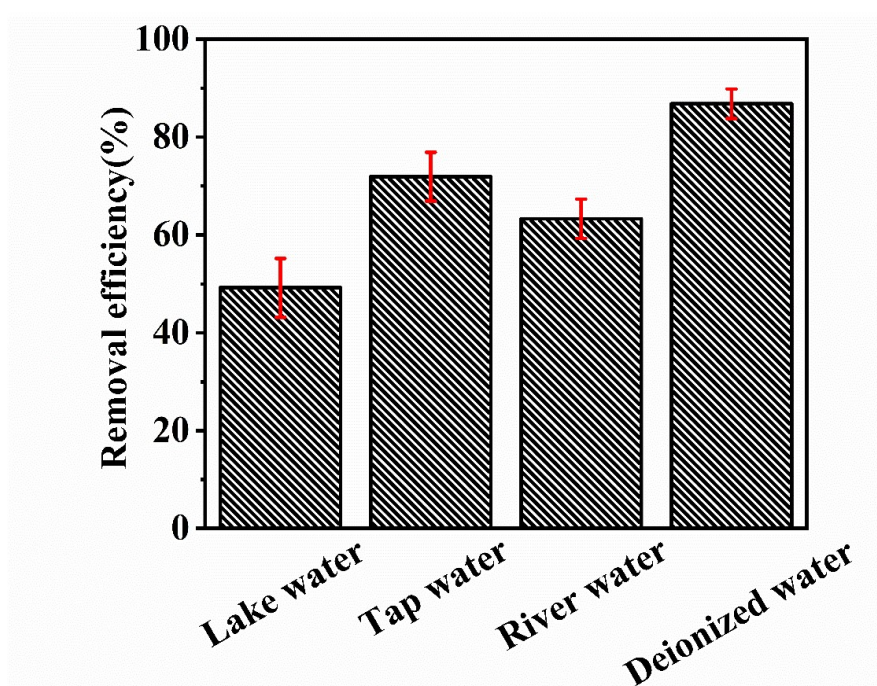




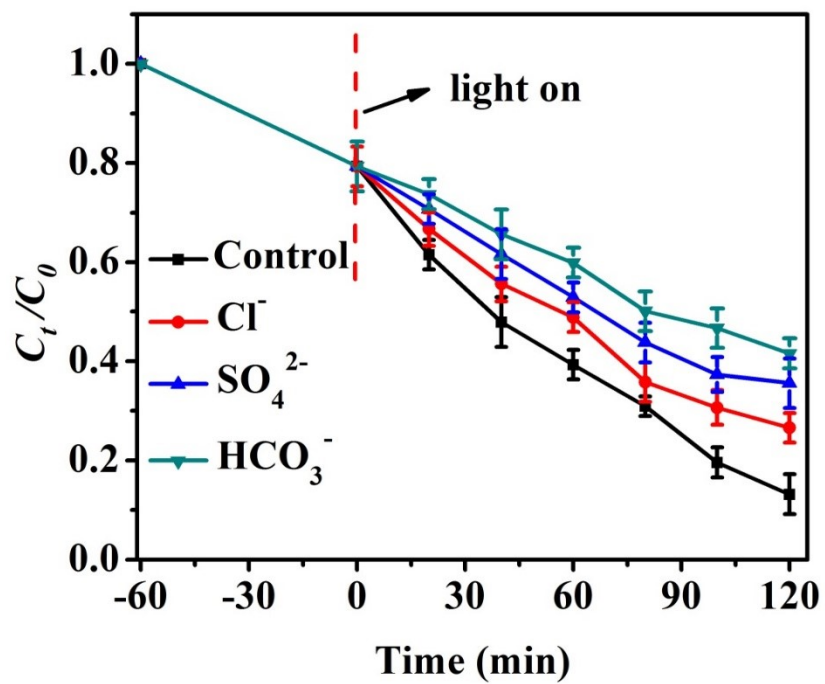
**Figure S6.** The VB-XPS of as-prepared (a) ZnO and (b) BiVO<sub>4</sub>.



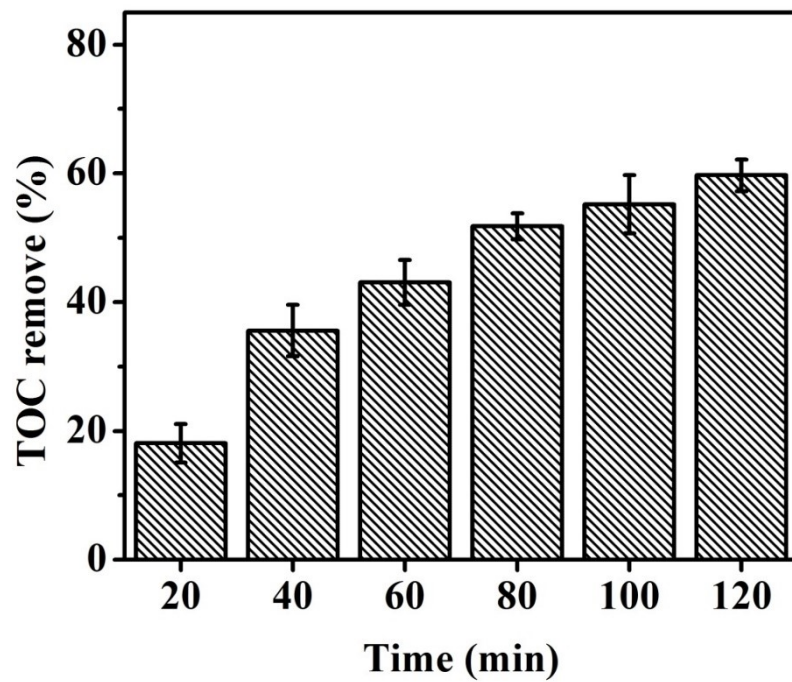
**Figure S7.** The adsorption of CBZ on as-prepared ZnO, BiVO<sub>4</sub> and ZnO/BiVO<sub>4</sub> composites.



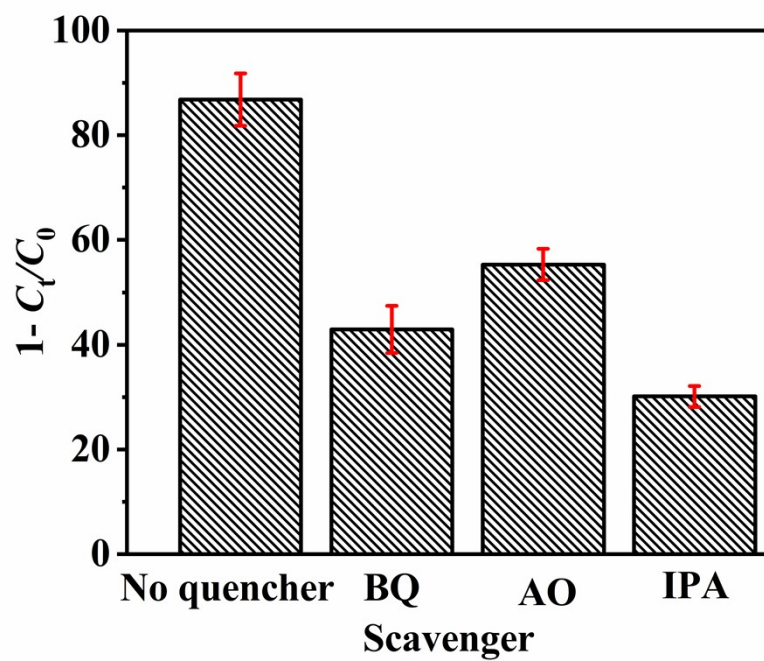
**Figure S8.** Evaluating the practical application of ZnO/BiVO<sub>4-3</sub>.



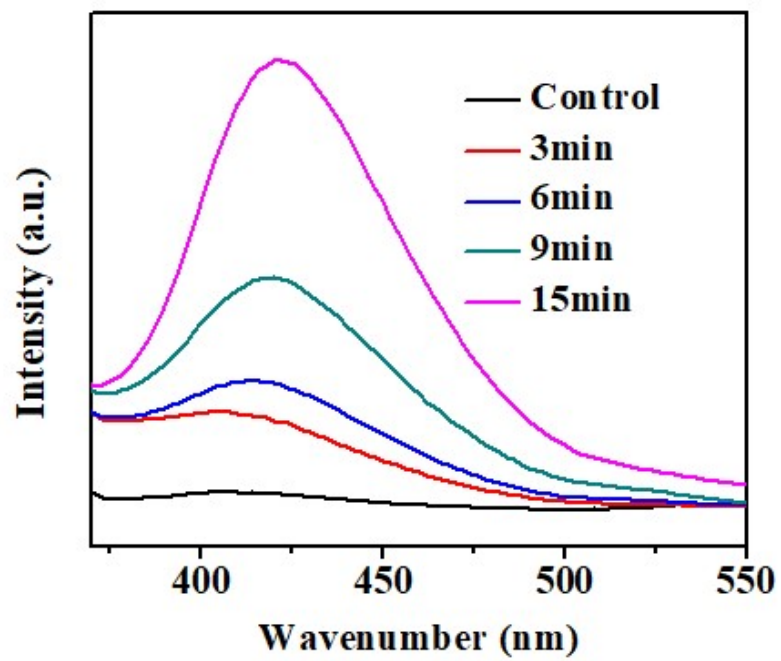
**Figure S9.** Effect of co-existing ions on the photocatalytic degradation of CBZ in the presence of  $\text{ZnO/BiVO}_4$ -3.



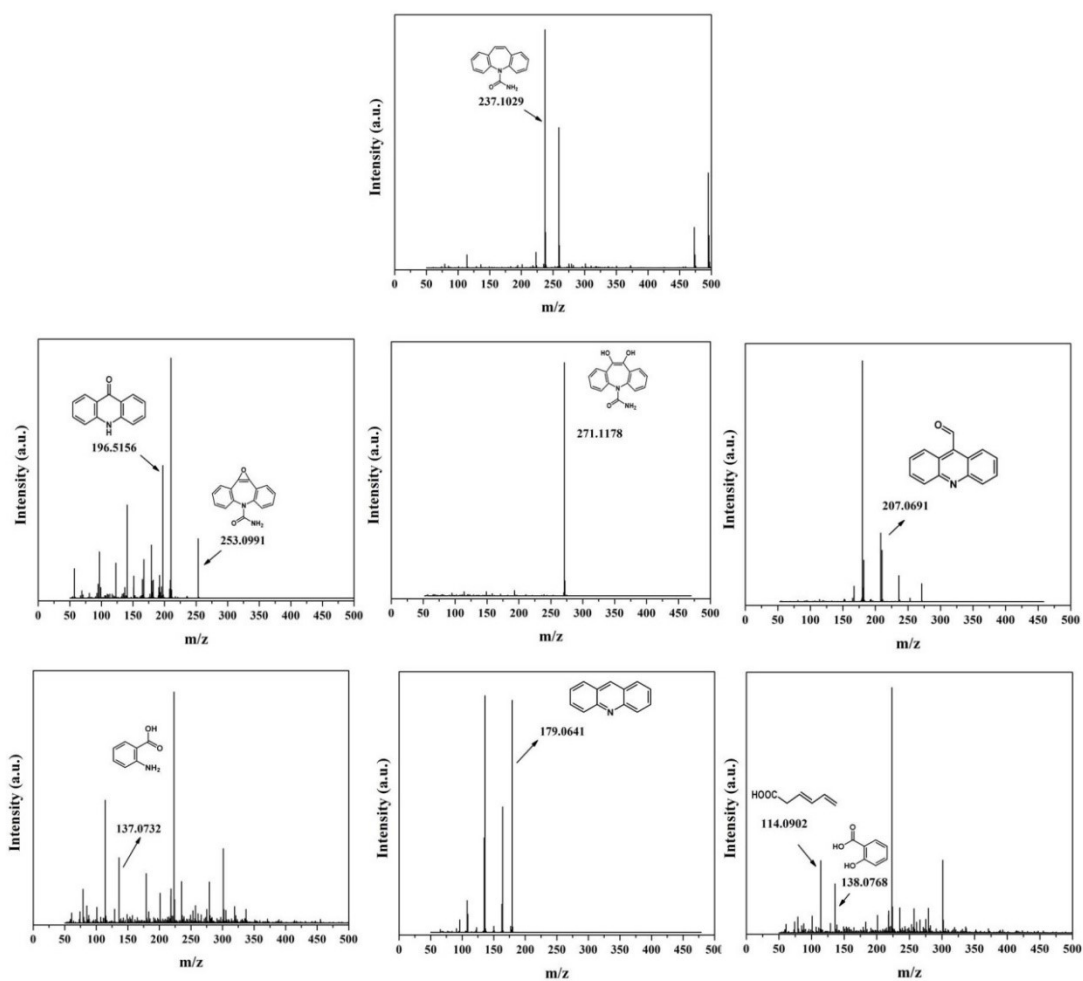
**Figure S10.** The total organic carbon (TOC) removal efficiency of ZnO/BiVO<sub>4</sub>-3.



**Figure S11.** Effects of different scavengers to photocatalytic degradation of CBZ by ZnO/BiVO<sub>4</sub>-3.

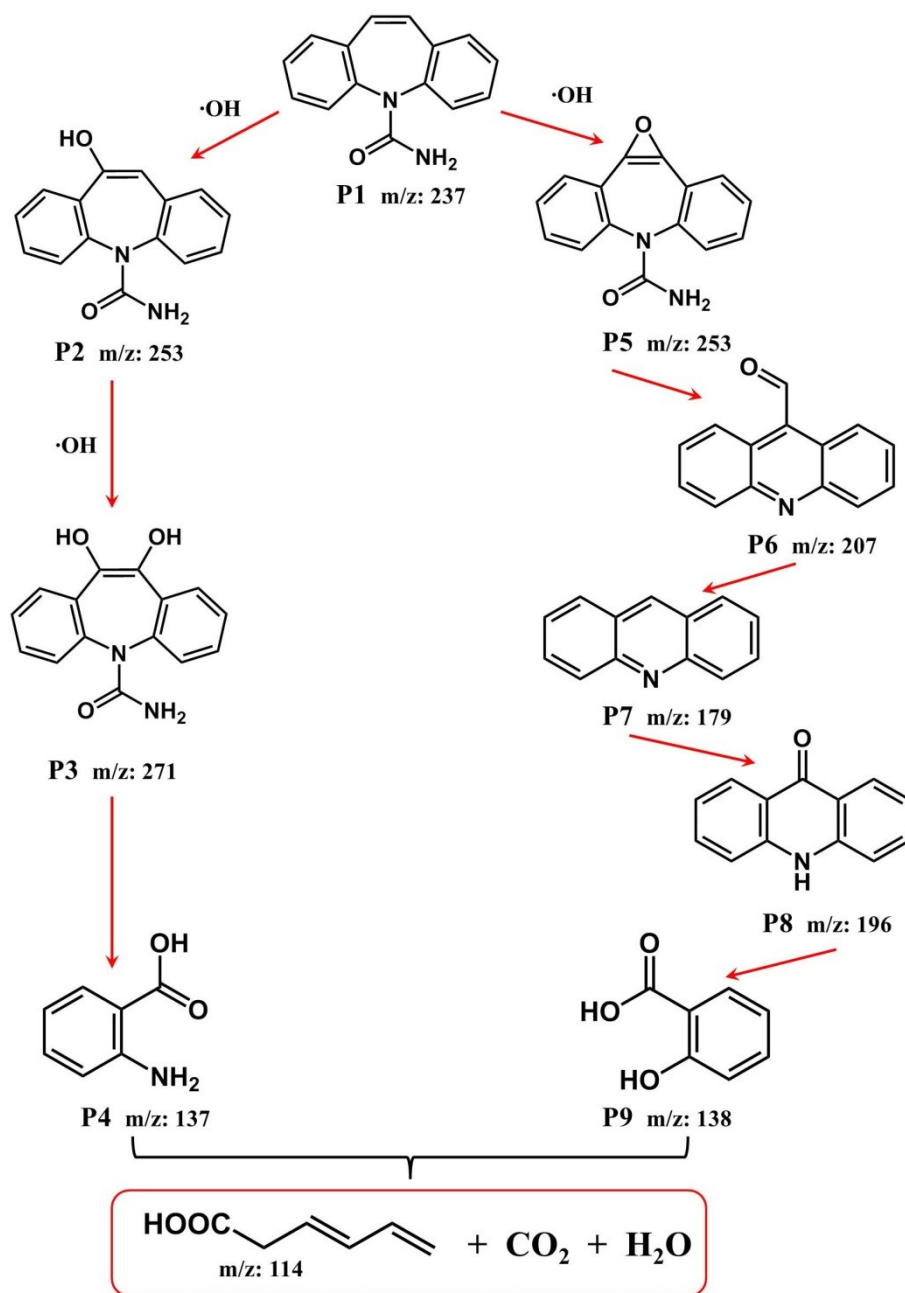


**Figure S12.** PL spectra of  $\bullet\text{OH}$  generated in the ZnO/BiVO<sub>4</sub>-3 aqueous reaction system.



**Figure S13.** The CBZ and its intermediate product in the degradation process.





**Figure S14.** Degradation pathways of CBZ over ZnO/BiVO<sub>4-3</sub> composite.

### Section 3: Tables

**Table S1.** Specific surface area and porosity of the samples

Samples	Specific surface area (m <sup>2</sup> /g)	Pore volume (cm <sup>3</sup> /g)	Average pore diameter (nm)
ZnO	67.26	0.116	26.57
BiVO <sub>4</sub>	13.17	0.194	186.83
ZnO/BiVO <sub>4</sub> -3	30.82	0.151	125.62

**Table S2.** Comparison of CBZ degradation efficiency for different catalysts.

Catalyst	Reaction condition	Reaction time (min)	Rate constant (min <sup>-1</sup> )	Referenc e
Fe-TiO <sub>2</sub>	[catalyst] = 0.5 g/L [CBZ] = 9 mg/L	150	0.0129	[1]
BiOBr/CeO <sub>2</sub>	[catalyst] = 0.8 g/L [CBZ] = 5 mg/L	120	0.0163	[2]
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub> (0.3)/MIL-101(Cr)	[catalyst] = 0.2 g/L [CBZ] = 30 mg/L	180	0.0152	[3]
TiO <sub>2</sub> /rGO	[catalyst] = 0.05 g/L [CBZ] = 4.5 mg/L	60	0.0156	[4]
Fe <sub>3</sub> O <sub>4</sub> /BiOBr/BC	[catalyst] = 1 g/L [CBZ] = 10 mg/L	180	0.0177	[5]
Ag/AgCl/BiVO <sub>4</sub>	[catalyst] = 1.0 g/L [CBZ] = 10 mg/L	240	0.0108	[6]
ZnO/BiVO <sub>4</sub> -3	[catalyst] = 0.8 g/L [CBZ] = 10 mg/L	120	0.0145	This work

#### Section 4: References

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- [2] L.L. Liang, S.W. Gao, J.C. Zhu, L.J. Wang, Y.N. Xiong, X.F. Xia and L.W. Yang, *Chem. Eng. J.*, 2020, **391**, 123599.
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- [5] S. Li, Z.W. Wang, X.T. Zhao, X. Yang, G.W. Liang and X.Y. Xie, *Chem. Eng. J.*, 2019, **360**, 600–611.
- [6] G. Yentür and M. Dükkancı, *Ultrason. Sonochem.*, 2021, **78**, 105749.