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## Comparative Study of Luminescent Cd-MOF and Cd-MOF@HNT Nanomaterial for detection of Hydroxyl-Functionalized Nitroaromatic compounds

Kushal Arya,<sup>a</sup> Sanjay Mehra,<sup>b</sup> Ajay Kumar,<sup>a,c</sup> Vaneet Saini,<sup>a</sup> Arvind Kumar,<sup>b</sup> Surinder Kumar Mehta,<sup>a</sup> Ramesh Kataria<sup>a</sup>\*

<sup>a</sup>Department of Chemistry, Centre for Advanced studies in Chemistry, Panjab University, Chandigarh, India-160014.

<sup>b</sup>CSIR-Central Salt and Marine Chemicals Research Institute, Council of Scientific and Industrial Research, Bhavnagar, Gujarat, India-364002.

<sup>c</sup>University Center for Research and Development, Chandigarh University, Mohali, Punjab-140301, India

Email: <u>rkataria@pu.ac.in</u>

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Figure S3. TGA analysis of PUC-10.



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Figure S6. Elemental mapping (a) and EDX spectrum (b) of PUC-10.



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Figure S10. Elemental mapping (a) and EDX spectrum (b) of PUC-10@HNT.



Figure S11. Mapping of elemental distribution of Cd (a), C (b), N (c), O (d), Al (e) and Si (f) of PUC-10@HNT.



Figure S12. High-resolution XPS spectrum of Cd 3d (a), C 1s (b), N 1s (c) and O 1s (d) elements present in PUC-10.



**Figure S13.** High-resolution XPS spectrum of Cd 3d (a), C 1s (b), N 1s (c) and O 1s (d) elements present in PUC-10 after treatment with TNP.



**Figure S14.** High-resolution XPS spectrum of Cd 3d (a), N 1s (b), O 1s (c) and C 1s (d) elements present in PUC-10 after treatment with DNP.



**Figure S15.** High-resolution XPS spectrum of Cd 3d (a), C 1s (b), N 1s (c) and O 1s (d) elements present in PUC-10 after treatment with 4-NP.



**Figure S16.** High resolution XPS spectrum of Cd 3d (a), N 1s (b), Si 2p (c), O 1s (d), C 1s (e) and Al 2p (f) elements present in PUC-10@HNT after treatment with TNP.



Figure S17. High-resolution XPS spectrum of Cd 3d (a), C 1s (b), O 1s (c), N 1s (d), Si 2p (e) and Al 2p (f) elements present in PUC-10@HNT after treatment with DNP.



Figure S18. High-resolution XPS spectrum of Cd 3d (a), C 1s (b), N 1s (c), O 1s (d), Si 2p (e) and Al 2p (f) elements present in PUC-10@HNT after treatment with 4-NP.



**Figure S19**. Emission spectra of PUC-10@HNT at different excitation wavelengths ranging from 300 to 350 nm.



**Figure S20**. Emission spectra of PUC-10@HNT and both the ligands  $L_1$  and  $H_2TP$  used in the synthesis of PUC-10



Figure S21. Excitation and emission spectra of PUC-10@HNT.



**Figure S22**. Emission spectra (a) and normalized fluorescence intensity (b) of PUC-10 at different pH values ranging from 2 to 12.



**Figure S23**. Emission spectra (a) and normalized fluorescence intensity (b) of PUC-10@HNT in different solvents.



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Figure S25. Stern-Volmer plot of PUC-10@HNT in the presence of 0-20  $\mu$ M concentration of TNP.



Figure S26. Stern-Volmer plot of PUC-10@HNT in presence of 0-20  $\mu$ M concentration of DNP.



Figure S27. Stern-Volmer plot of PUC-10@HNT in the presence of 0-20  $\mu$ M concentration of 4-NP.



Figure S28. Calibration curve of PUC-10@HNT on addition of TNP (0-20 µM).



Figure S29. Calibration curve of PUC-10@HNT on the addition of DNP (0-20  $\mu$ M).



Figure S30. Calibration curve of PUC-10@HNT on the addition of 4-NP (0-20 µM)



**Figure S31**. The emission spectrum of PUC-10 before and after incremental addition of TNP from 0 to 66.6  $\mu$ M concentration (a) and its Calibration curve (b).



**Figure S32**. The emission spectrum of PUC-10 before and after incremental addition of DNP from 0 to 66.6  $\mu$ M concentration (a) and its Calibration curve (b).



**Figure S33**. The emission spectrum of PUC-10 before and after incremental addition of 4-NP from 0 to 66.6  $\mu$ M concentration (a) and its Calibration curve (b).



**Figure S34**. The stern-Volmer plot of PUC-10 in the presence of 0-66.6  $\mu$ M concentration of TNP.

![](_page_16_Figure_3.jpeg)

Figure S35. Stern-Volmer plot of PUC-10 in the presence of 0-66.6 µM concentration of DNP.

![](_page_17_Figure_0.jpeg)

Figure S36. Stern-Volmer plot of PUC-10 in the presence of 0-66.6  $\mu$ M concentration of 4-NP.

![](_page_17_Figure_2.jpeg)

Figure S37. Calibration curve of PUC-10@HNT on addition of TNP (0-20  $\mu$ M) using real water sample (tap water) as solvent.

![](_page_18_Figure_0.jpeg)

Figure S38. Calibration curve of PUC-10@HNT on the addition of DNP (0-20  $\mu$ M) using real water sample (tap water) as solvent.

![](_page_18_Figure_2.jpeg)

Figure S39. Calibration curve of PUC-10@HNT on the addition of 4-NP (0-20  $\mu$ M) using real water sample (tap water) as solvent.

![](_page_19_Figure_0.jpeg)

Figure S40. Zeta potential analysis curve of PUC-10@HNT before (a) and after the addition of TNP (b), DNP (c) and 4-NP (d).

![](_page_19_Figure_2.jpeg)

Figure S41. Recyclability of PUC-10@HNT with TNP (pink), DNP (orange) and 4-NP (green).

Identification Code	PUC-10
CCDC No.	2402328
Empirical formula	$C_{24}H_{21}CdN_5O_9$
Formula weight	635.86 g/mol
Temperature/K	293(2)
Crystal system	monoclinic
Space group	<i>P2/c</i>
a/Å	25.1793(9)
b/Å	6.27470(10)
c/Å	20.2026(5)
α/°	90
β/°	103.253(3)
$\gamma/^{\circ}$	90
Volume (Å <sup>3</sup> )	3106.85(15)
Z	4
$D_{calc}$ (g/cm <sup>3</sup> )	1.359
μ (mm <sup>-1</sup> )	0.754
F(000)	1280.0
Crystal size/mm <sup>3</sup>	0.009×0.007×0.006
Radiation	Mo Ka ( $\lambda = 0.71073$ )
Reflections collected	28608
Independent reflections	67474 [R <sub>int</sub> = 0.056, R <sub>sigma</sub> =
Data/restraints/parameters	0.0552]
Goodness-of-fit on F <sup>2</sup>	6474/6/355
Final R indexes [all data]	0.993

 Table S1. Crystal data and Structure Refinement for PUC-10.

	PUC-10				PUC-10@HNT			
Reflection number	20	Miller Indices	FWHM (β)	Crystallite size (nm)	20	Miller Indices	FWHM (β)	Crystallite size (nm)
1								
	7.186	(200)	0.09722	81.87549	7.186	(200)	0.19658	40.49209
2								
	10.769	(002)	0.11645	68.52297	10.769	(002)	0.20292	39.32338
3								
	17.986	(012)	0.13158	61.12753	17.986	(012)	0.21325	37.71705
4								
	21.621	(600)	0.13435	60.19966	21.621	(600)	0.1904	42.47807
5								
	25.275	(700)	0.14954	54.44381	25.275	(700)	0.19868	40.97809
	Average size = 65.2290 nm				Average size = 40.19773 nm			

 Table S2. Crystallographic features for PUC-10 and PUC-10@HNT.

Table S3. Compositional ratios of all the elements present in PUC-10 and PUC-10@HNT

Sample	Cd (at%)	O (at%)	N (at%)	C (at%)	Si (at%)	Al (at%)
PUC-10	2.11	17.21	25.59	55.09	-	-
PUC-10@HNT	2.01	18.81	20.72	56.12	2.47	0.5