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Electronic Supplementary Information

Deprotonation of –NH Proton from Pyrrole Moiety Enables Concentration Dependent Colorimetric and Fluorescence Quenching of Silver (I) ions

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Figure S1. ¹H NMR spectra of ligand (1) in CDCl₃.



Figure S2. ¹³C NMR spectra of ligand (1) in CDCl₃.





Figure S3. HRMS spectra of ligand (1) in CH₃CN.



Figure S4 Steady state fluorescence of ligand (1) 5.07 μ M and its spiked samples containing biological sample (BS).



Figure S5: Steady state fluorescence of ligand (1) 5.07 μ M and its spiked samples containing environmental sample (ES).



Figure S6: Steady state fluorescence of ligand (1) 5.07 μ M and its spiked samples containing industrial sample (IS).



Figure S7: Fluorescence emission response of 1 with successive additions of Ag^+ and EDTA.



Figure S8: Ligand (1)-coated silica plates as a tool for *point-of-care-testing* devices prompting nakedeye detection showing the color change in the ligand (1) upon interaction with Ag^+ ions under sunlight.

Table S1: Comparison of limit of detection, binding constant, stoichiometry, and applications

 of ligand (1) with similar data reported for other chemosensors.

Name	1 /2	LOD	K _a	Stoichiometry Ligand:metal	Applications
1- Benzylidenethio semicarbazide	2	0.2 μM		1:1	Water samples [1]
Thiosemicarbazi de	2	6.13 nM		1:2	Teststrips,Watersamples,ImagingMCF7 cells [2]
Isocyanate naphthalene based derivative	2	2.87 nM		1:2	Real sample analysis, Test strips and Bioimaging [3]

Semicarbazone based Schiff	2	7.7 μM			Water samples [4]
base					
Thiosemicarbaz one based Schiff	1	2.2 μΜ	2.8×10 ⁹ M ²	1:2	Real sample analysis [5]
base	2	16	-		
~ 1 1	2	1.6 µM			
Carbazole- Rhodanine based derivative	2	12.8 nM	3.8×10 ⁴ M ⁻¹	1:1	Imaging HeLa cells [6]
Octapamine					
based Schiff	2	1.49 µM	7.00×10^4	1:1	NR [7]
base					
Thiourea derivative	2	8×10 ⁻⁷ M	1.8×10 ⁸ M ⁻²	1:2	NR [8]
p-aminoben zoic acid and p- amin ophenol based carbon dots	2	1.4 μM	NR	NR	Real sample analysis [9]
1	1	3.08 nM	1.73×10 ⁶ M ⁻¹	2:1	Test strips, water analysis, and imaging of onion epidermal cells
	2	0.57 nM	1.73×10 ⁶ M ⁻¹		(this work)

1=UV and 2=fluorescence method, NR=not reported

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