

## **Fabrication of novel AIE active chemosensor for selective detection of Pd(II) and picric acid: extended detection in HeLa cancer cell line**

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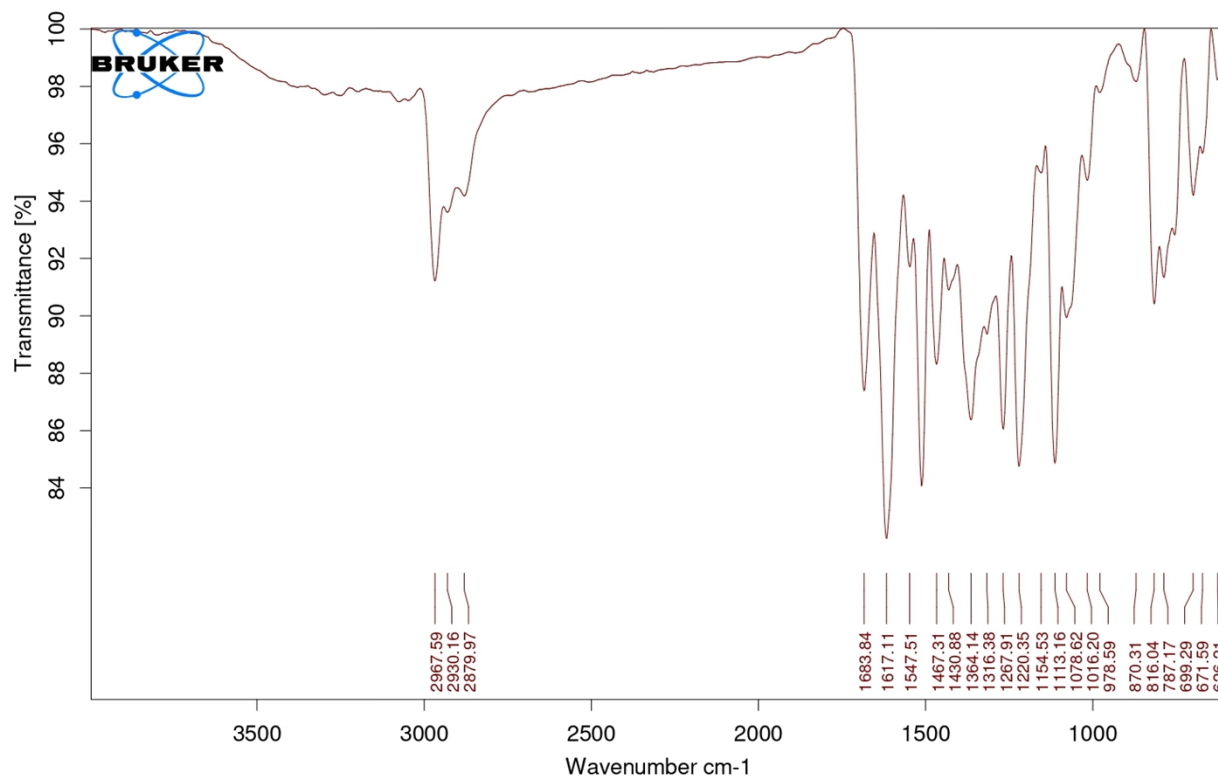
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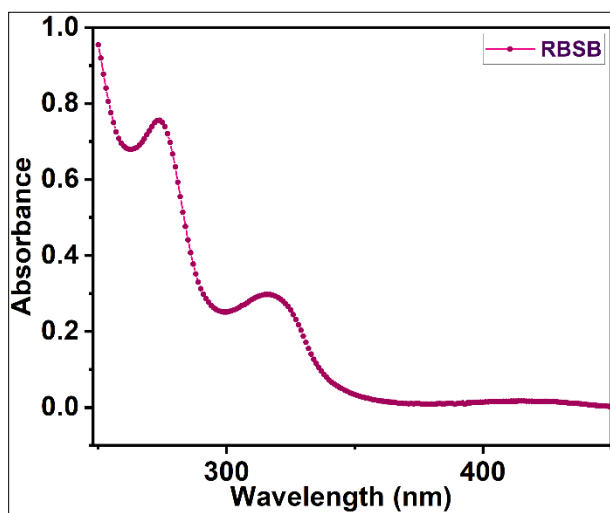
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**Fig. S1:** FTIR spectrum RBSB



**Fig. S2:** UV spectrum RBSB

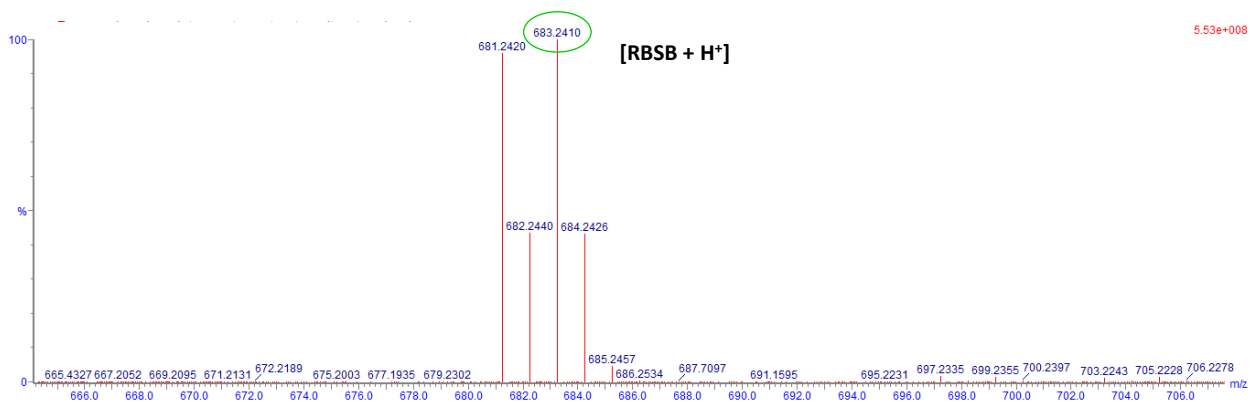


Fig. S3: Mass spectra of RBSB

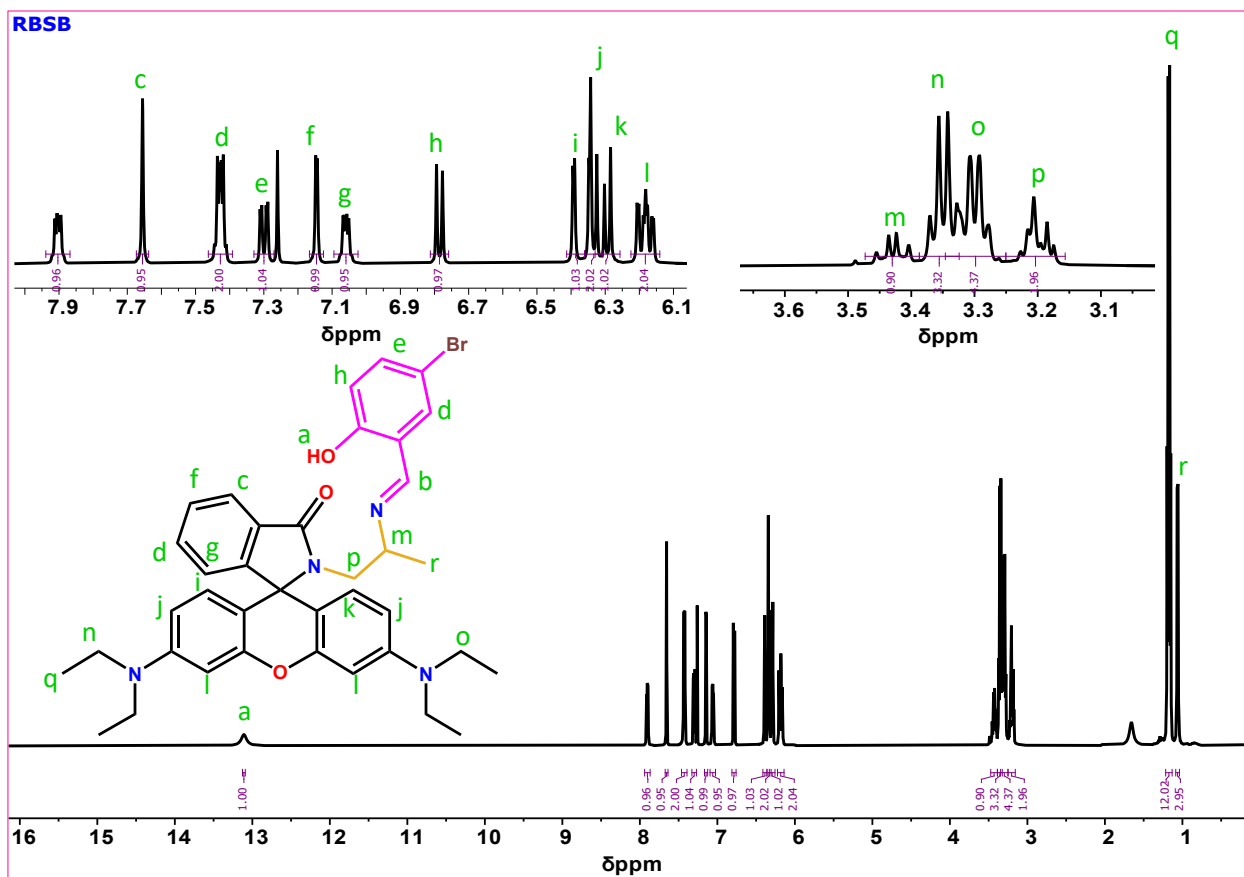


Fig. S4: <sup>1</sup>H NMR spectrum of RBSB

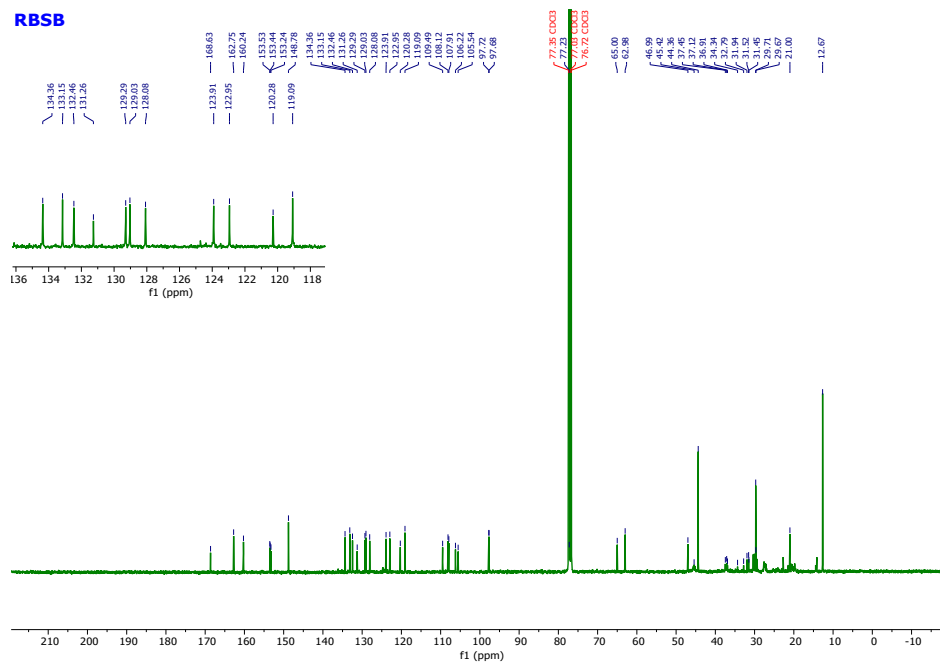


Fig. S5:  $^{13}\text{C}$  NMR spectrum of RBSB

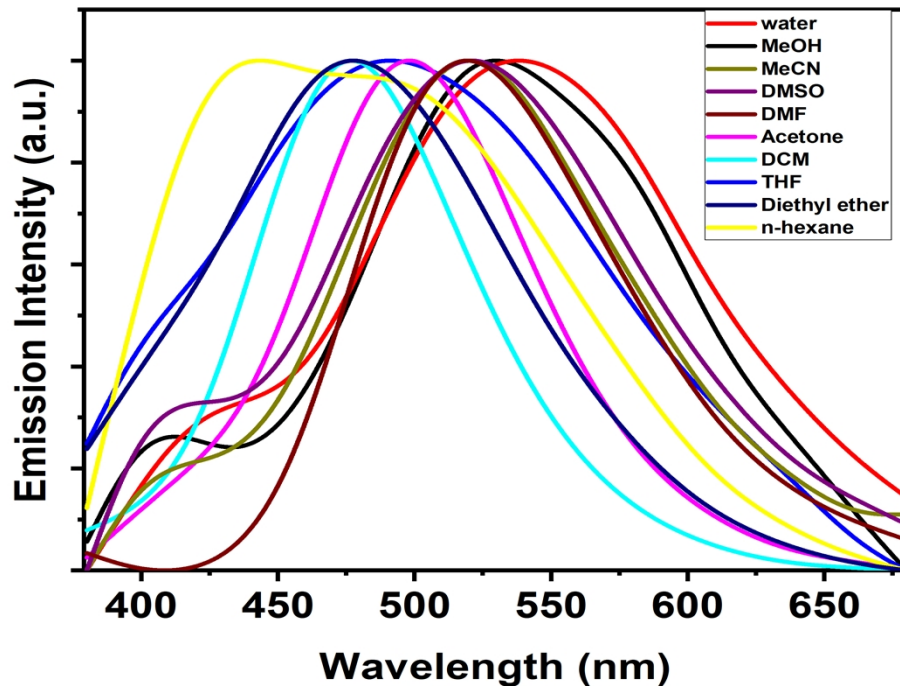


Fig. S6: Emission intensity of RBSB ( $2\ \mu\text{M}$ ) in different solvents with polarity difference

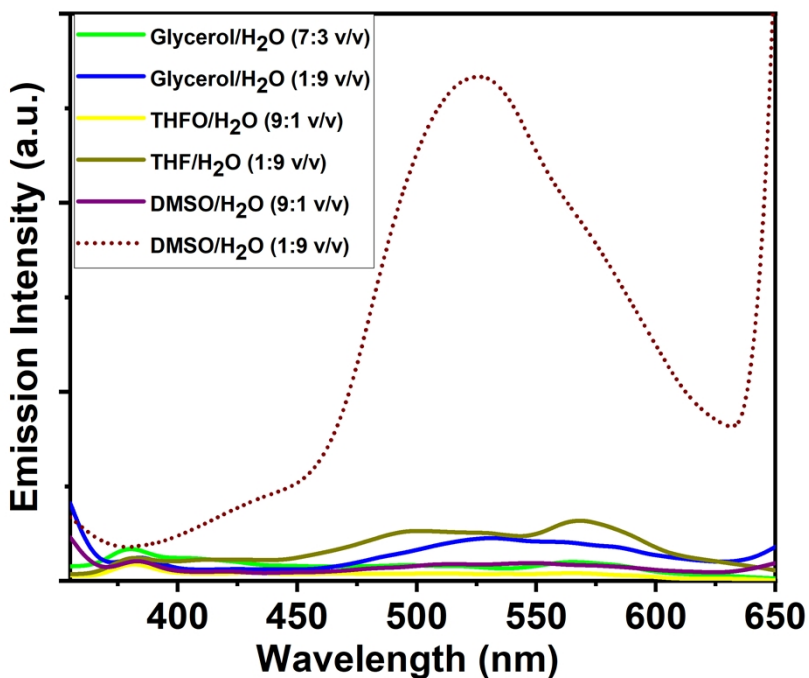


Fig. S7: The emission intensity enhancement of the probe **RBSB** in glycerol-water, THF-water, DMSO-water AIE active solvents indicating high AIE active performance in 1:9 DMSO water medium

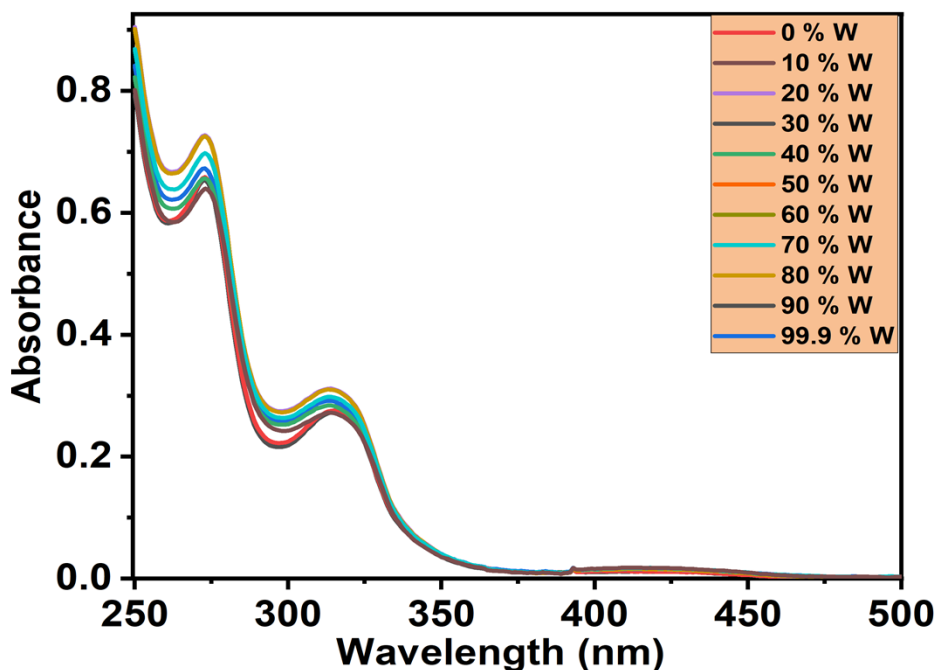
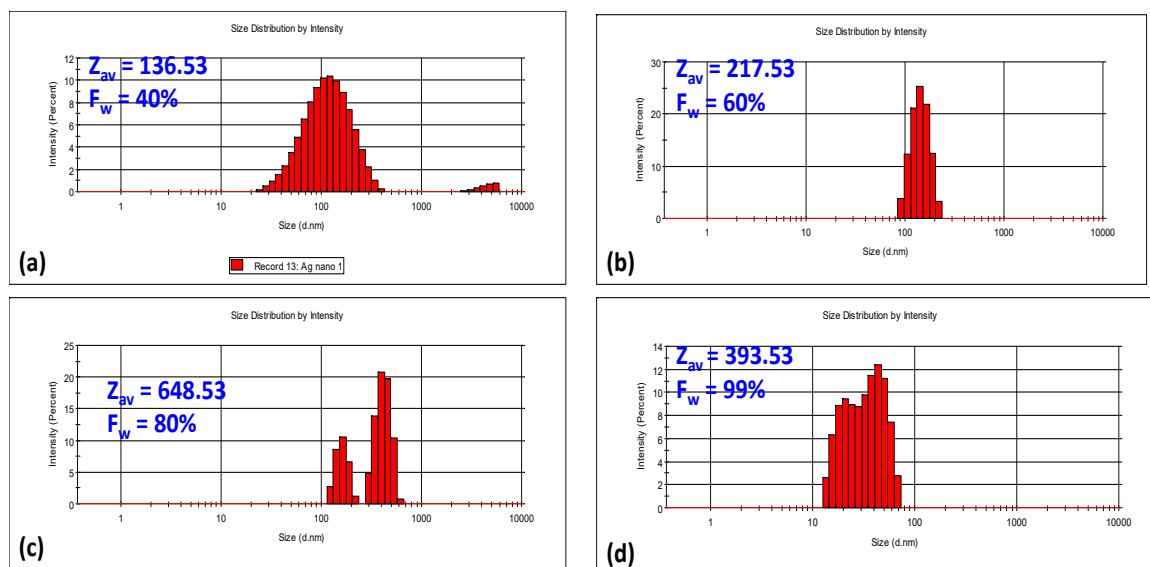
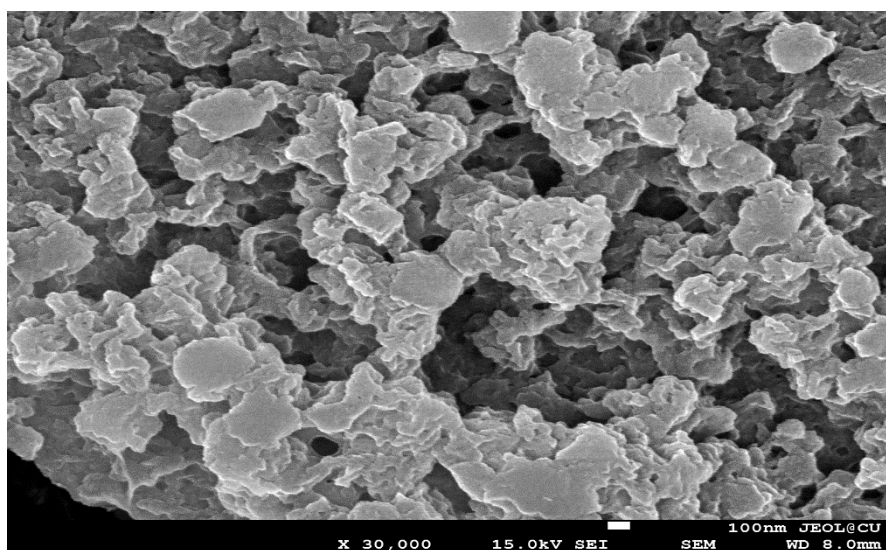


Fig. S8: Change of absorbance spectra of **RBSB** upon increasing volume percentages of water from 0% to 90% in DMSO-water AIE active medium.

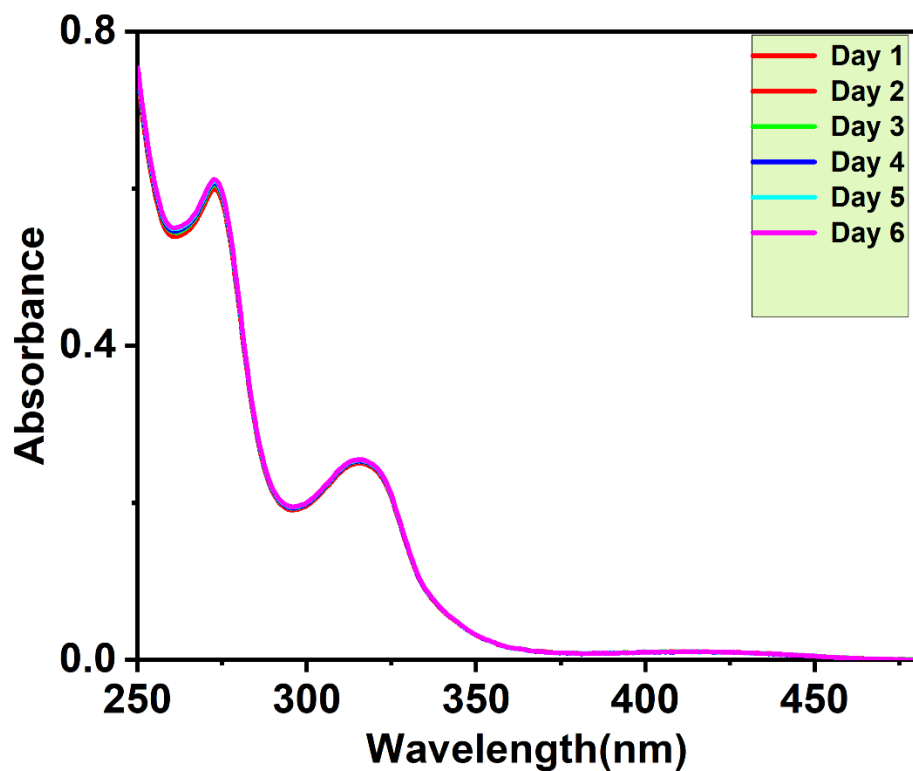


**Fig. S9:** Observation of the particle size change by enhancement of the water fraction with the help of DLS measurement in DMSO – H<sub>2</sub>O medium

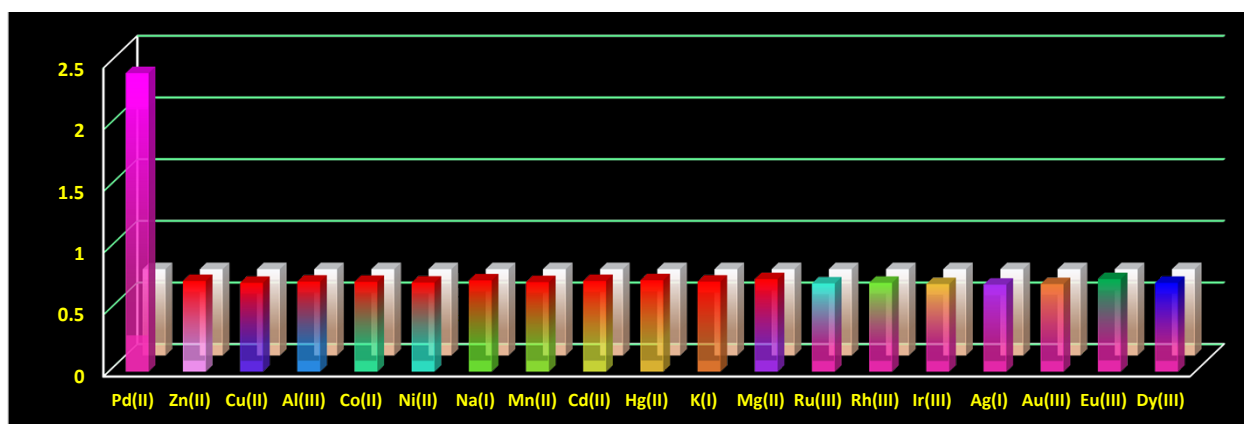


**Fig. S10.** SEM image of aggregated **RBSB** particle as a result of AIE behavior with a scale bar of (I) 1 mM and (II) 100 nm in DMSO – H<sub>2</sub>O (1:9 v/v) medium





**Fig. S11:** Change of Absorbance of RBSB ( $2 \times 10^{-6}$  M) as a function of time in DMSO – H<sub>2</sub>O (1:9 v/v) solvent system



**Fig. S12:** Emission intensity change of RBSB ( $3 \times 10^{-6}$  M) in the presence (five equivalent) of several cations including Pd(II) in DMSO-water (1:9) AIE medium over excitation at 326 nm.

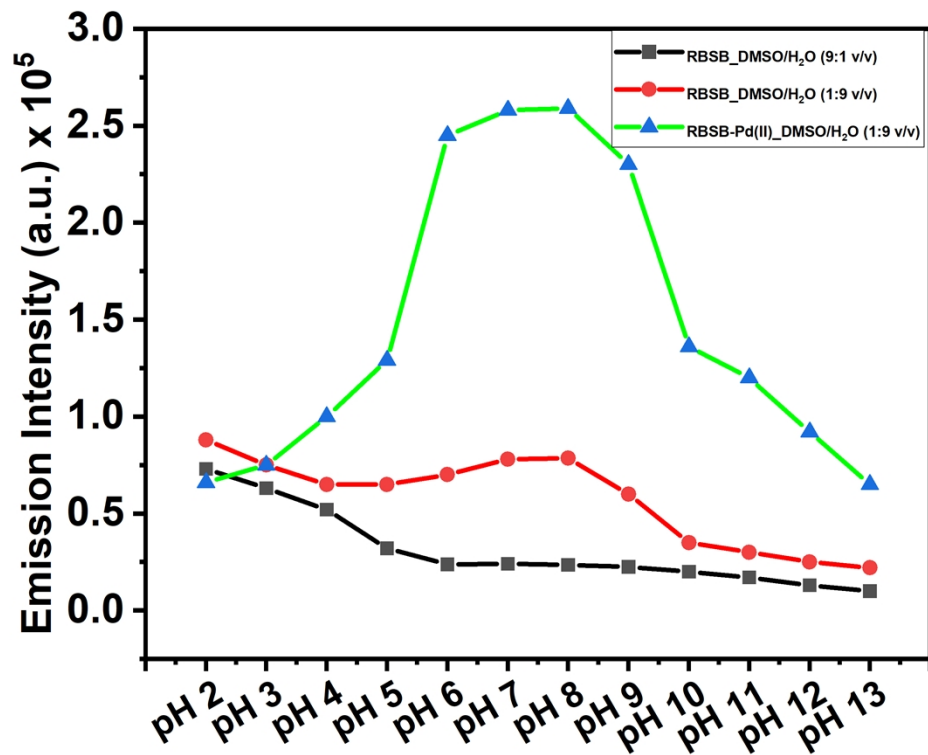


Fig. S13: Demonstration of pH effect on the emission intensity of the probe in the presence and the absence of target analyte in 9:1 DMSO-water and 1:9 DMSO water(AIE active) medium.

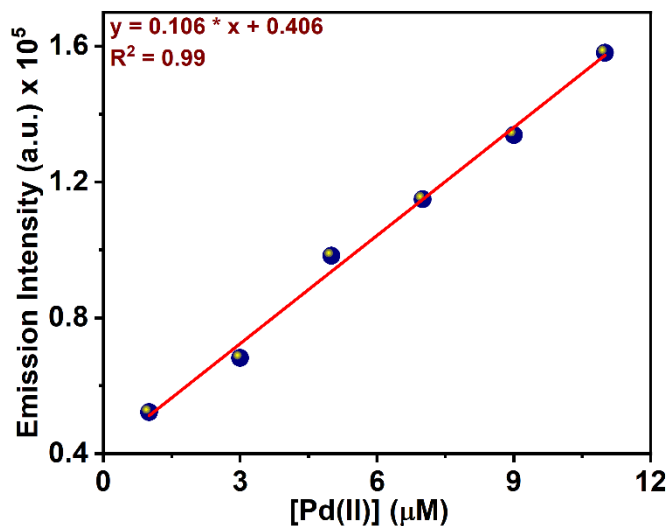


Fig. S14: Determination of LOD value during detection of Pd(II) by the probe **RBSB** in DMSO–H<sub>2</sub>O AIE medium

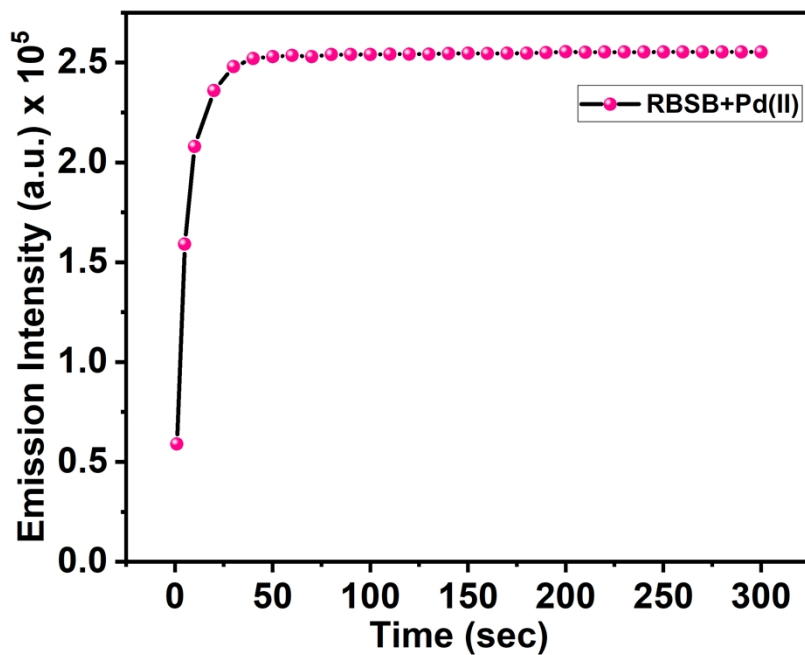


Fig. 15: Effect of response time on the emission intensity of the probe in the presence of Pd(II)

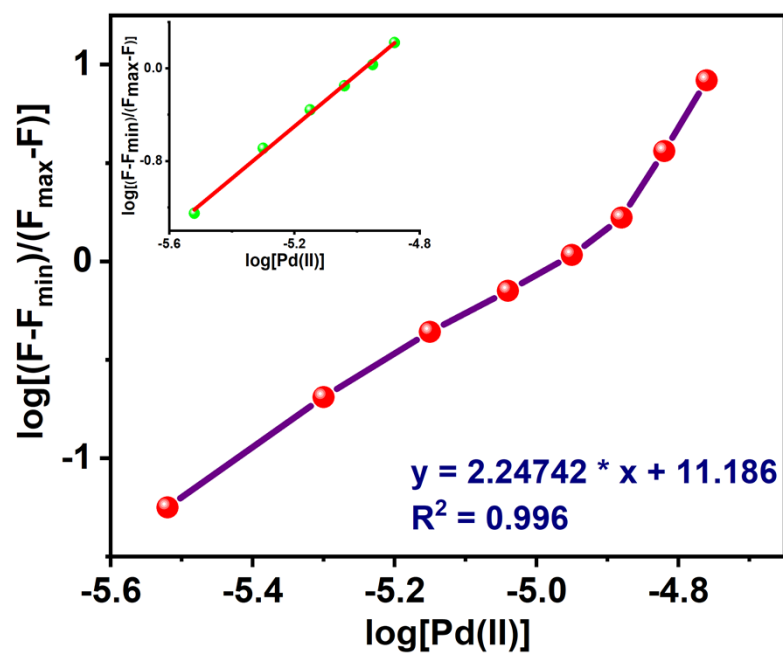
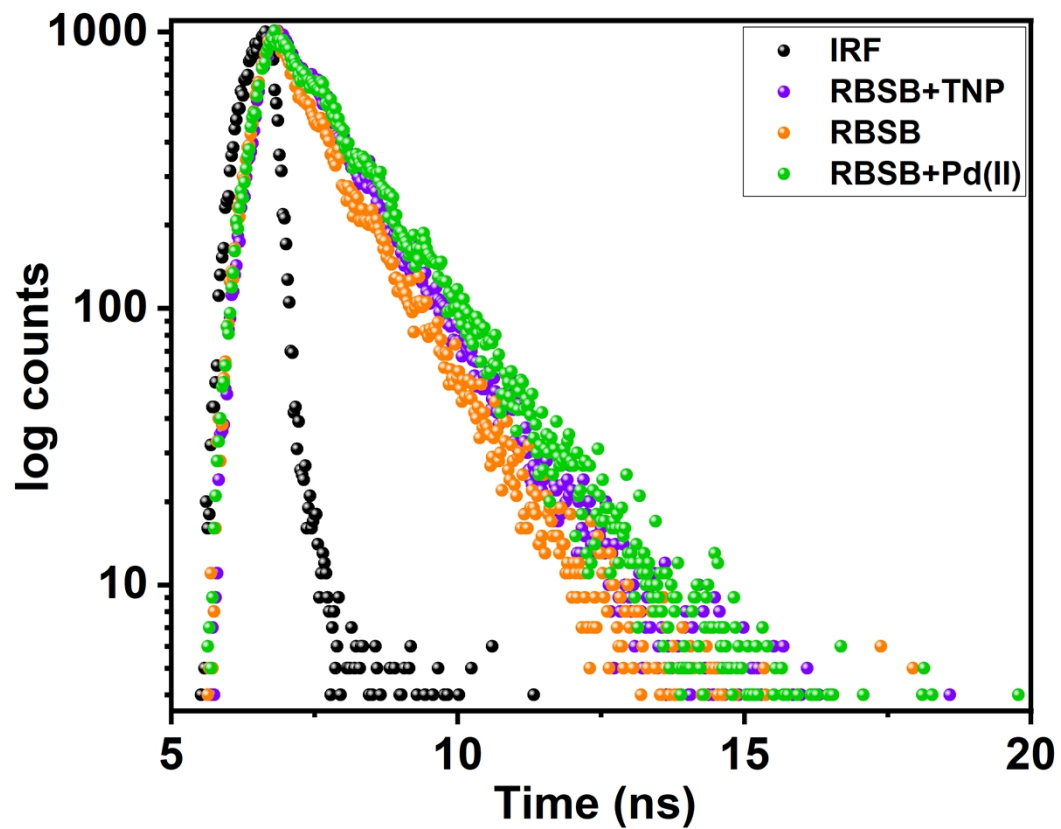


Fig. S16: Binding constant determination plot for **RBSB**-Pd(II) adduct in AIE active medium



**Fig. S17:** Lifetime measurement of RBSB in the presence and the absence of Pd(II) and TNP in AIE active medium.

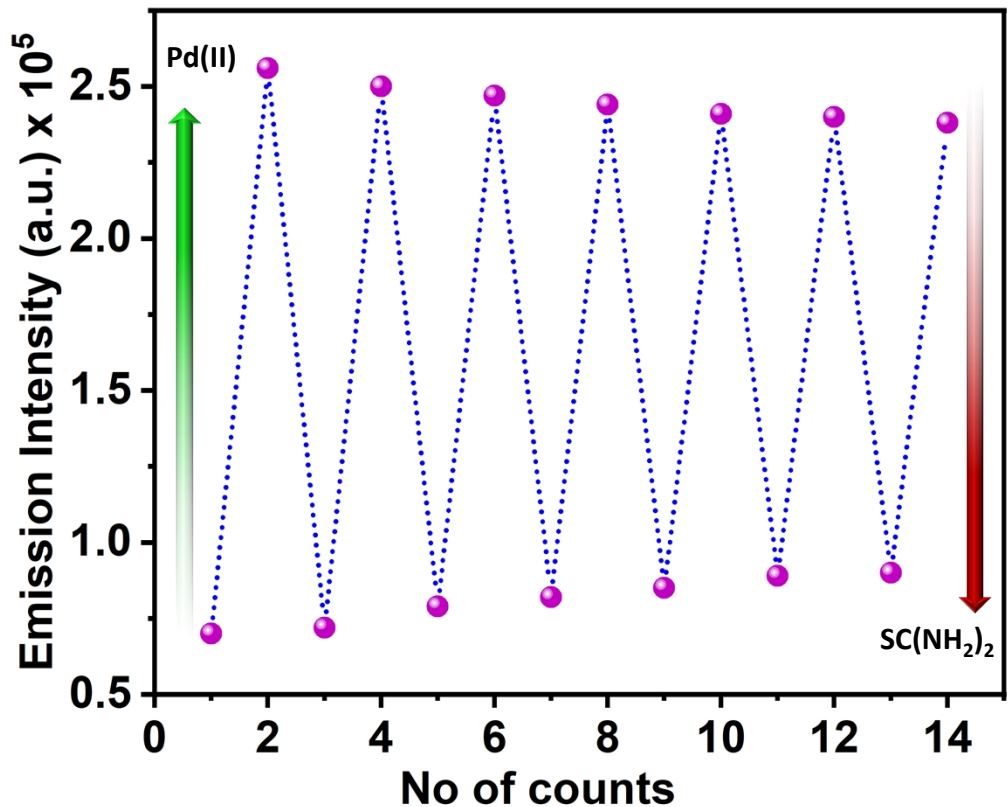


Fig. S18: The variation of fluorescence intensity of **RBSB** after alternative addition Pd(II) and thiourea, disclosing the existence of reversibility.

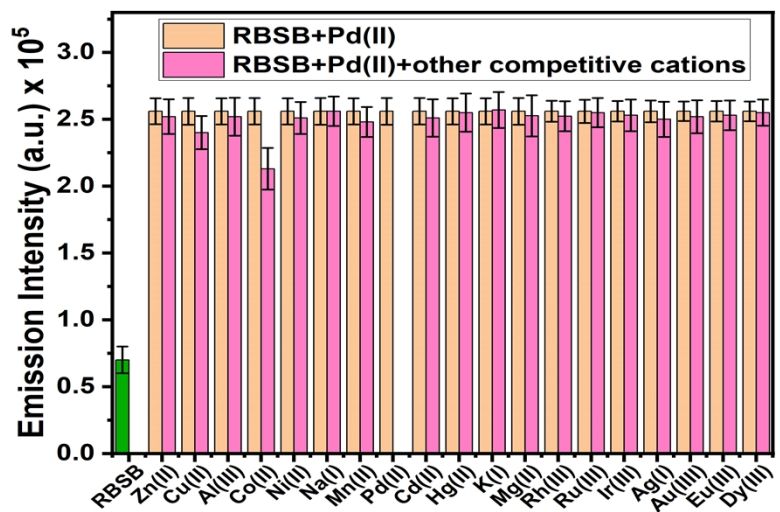
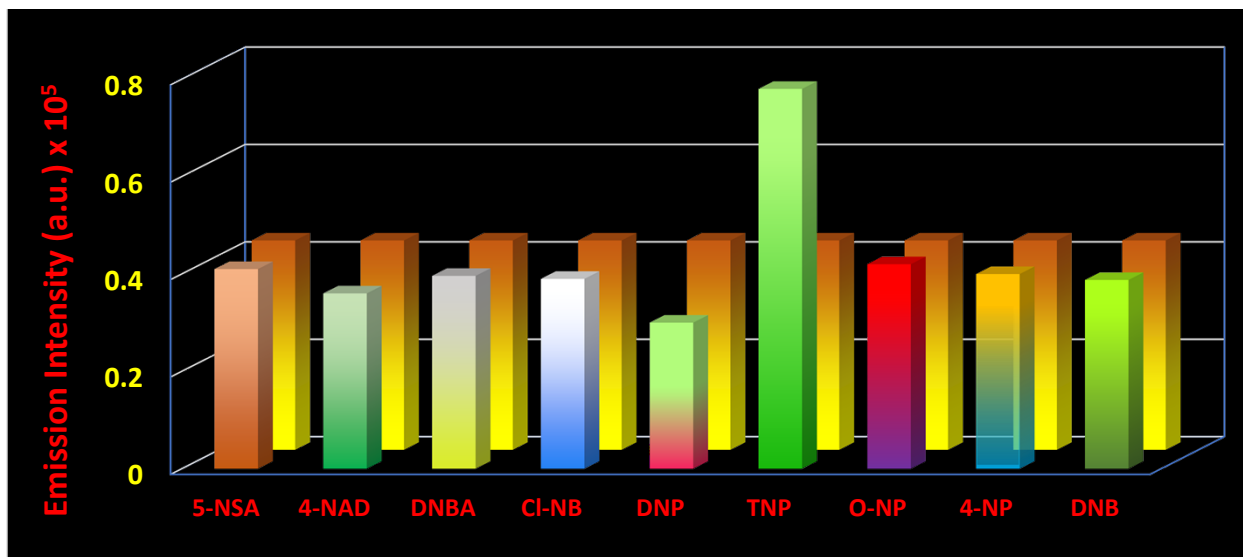
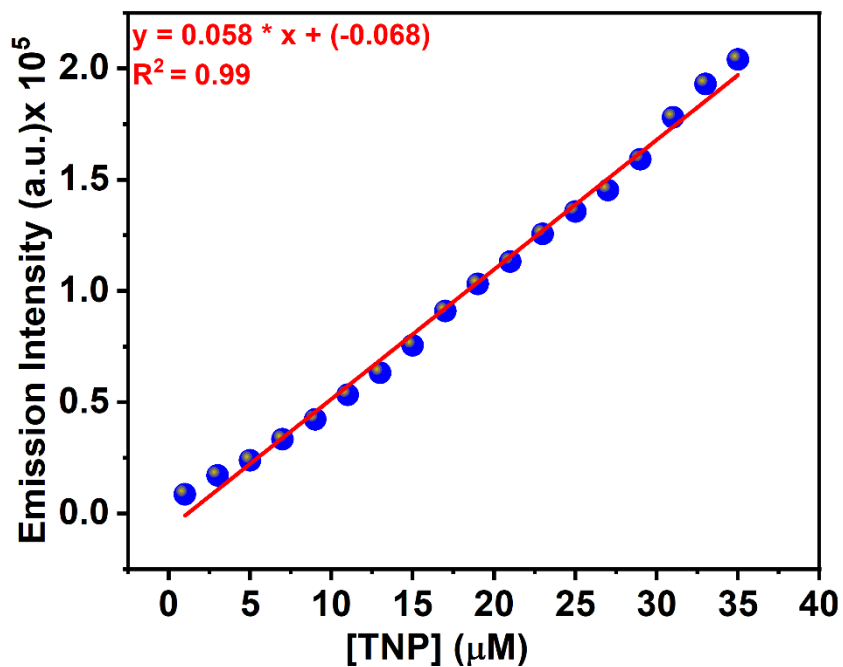


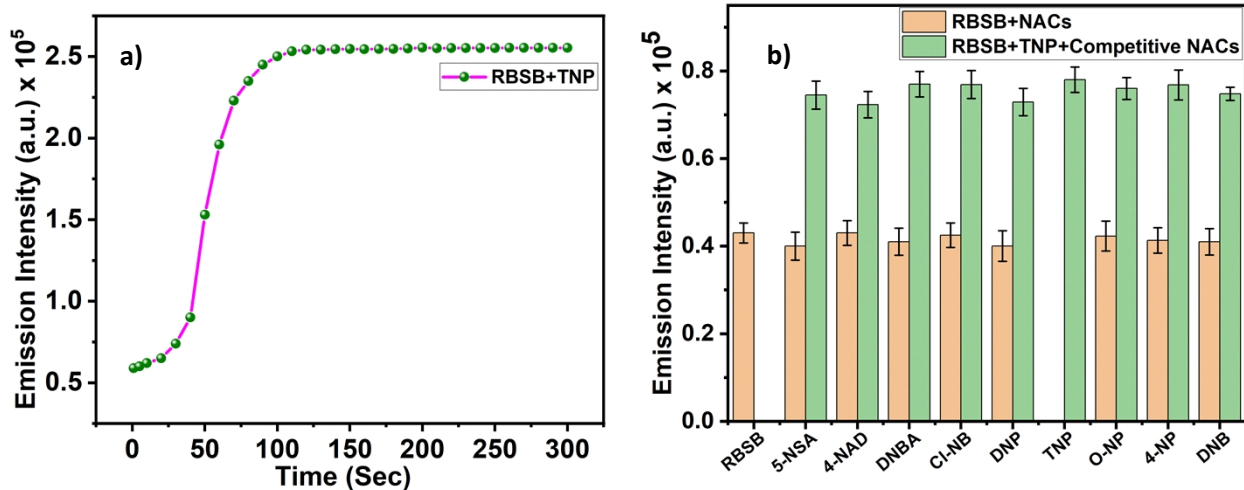
Fig. S19: Change of Emission Intensity of **RBSB** – Pd(II) adduct in the presence of several competitive metal ions to check the competitive ion effect in DMSO-water AIE active medium.



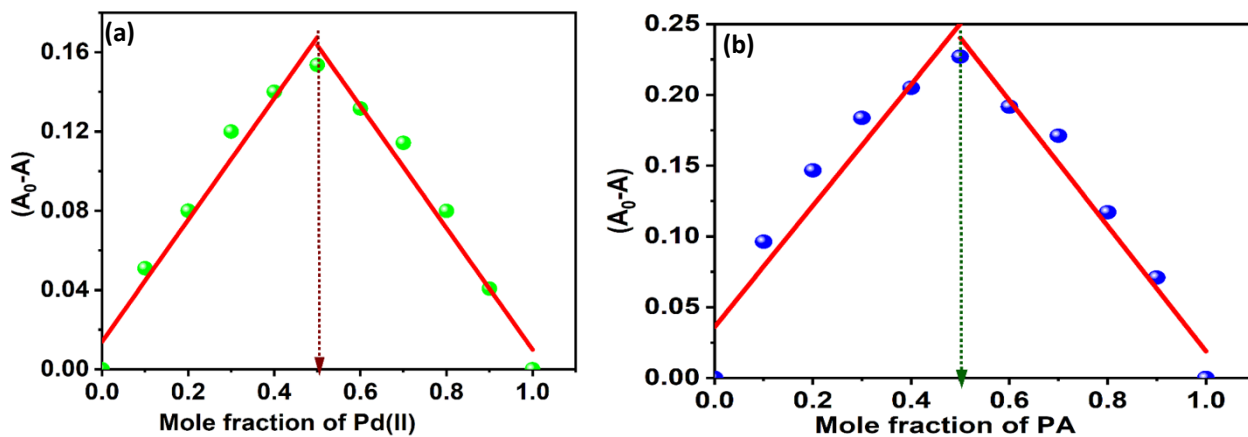
**Fig. S20:** Change of Emission intensity of **RBSB** ( $3 \times 10^{-5}$  M) after separate addition of different nitroaromatic compounds (NACs) in AIE active medium



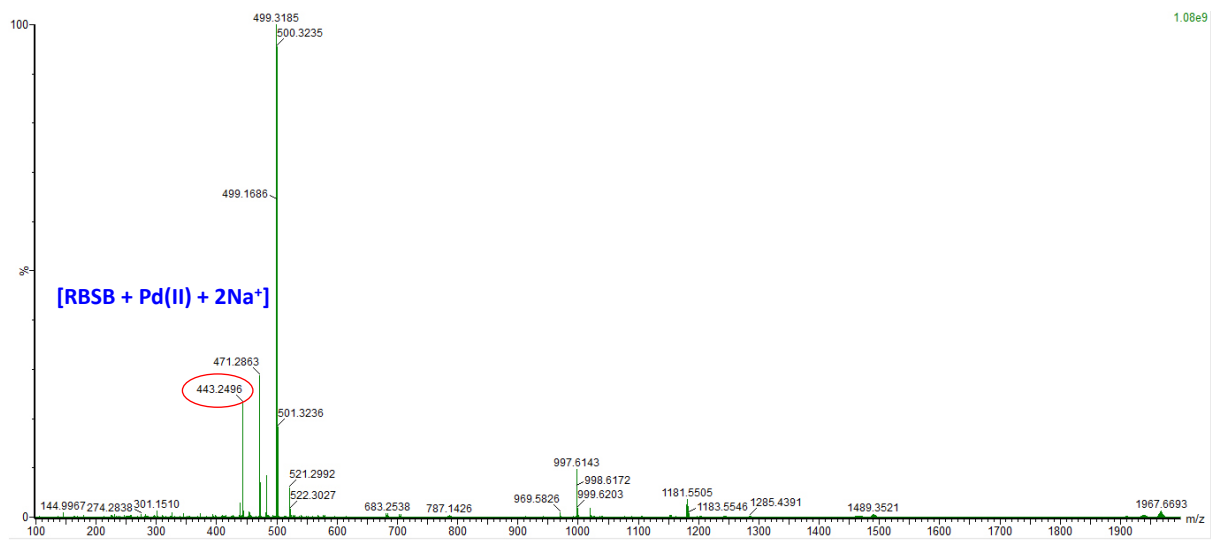
**Fig. S21:** Change of emission intensity of **RBSB** by the addition of PA to measure the limit of detection (LOD)



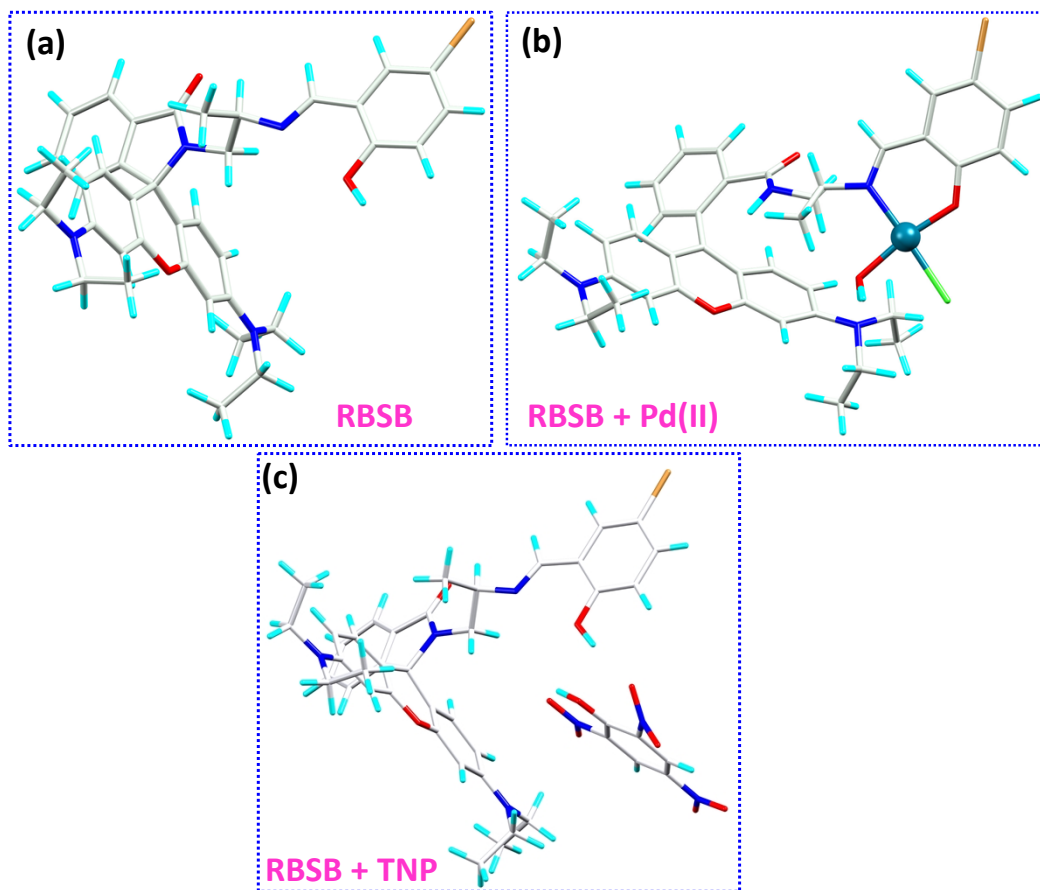
**Fig. 22:** a) Effect of response time on the emission intensity of the probe **RBSB** in the presence of TNP b) Emission intensity change of RBSB-TNP adduct with the presence of other individual NAC



**Fig. S23.** Jobs plot During sensing of a) Pd(II) b) PA by **RBSB** suggesting the 1:1 probe -analyte combination

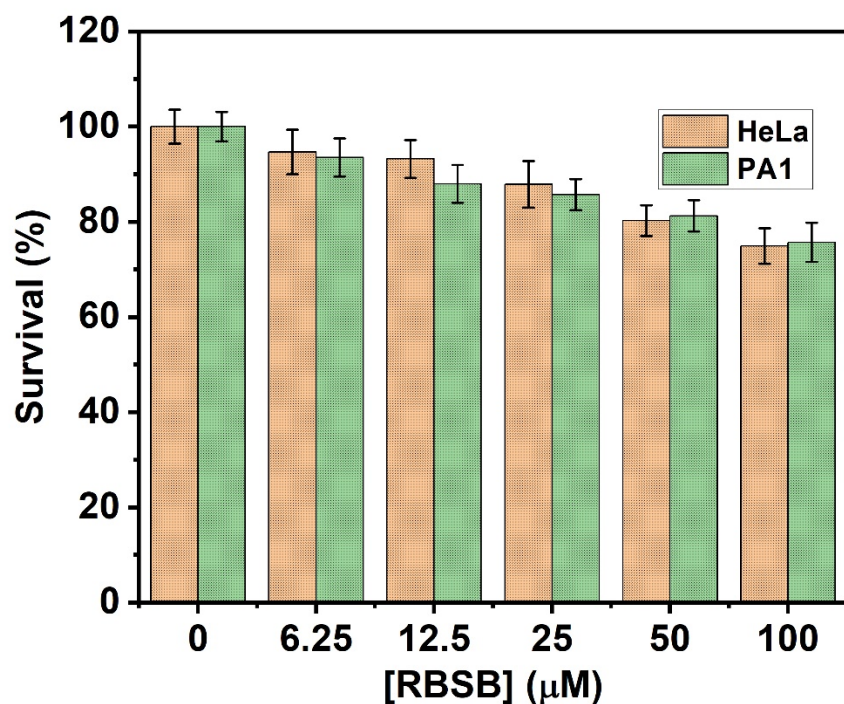


**Fig. S24:** Mass spectra of RBSB after mixing with Pd(II) in 1:1 ratio



**Fig. S25:** The DFT optimized ground state geometry of the probe and the probe-analyte complexes.





**Fig. S26:** Dose-dependent suppression of cell viability of **RBSB** on HeLa and PA1 cell line (24 hrs)

**Table S1:** Intra-molecular Hydrogen bonding parameter of **RBSB**

D-H...A	D-H (Å)	H...A (Å)	D...A (Å)	D-H...A (°)	Symmetry operation for A
O1-H1...N9	0.84	1.88	2.622(5)	147.4	x, y, z

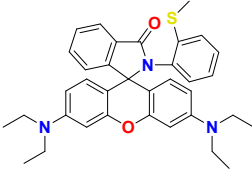
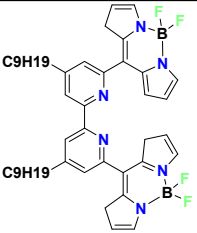
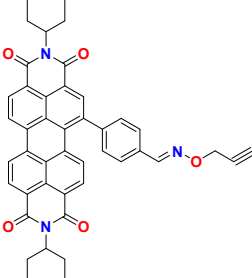
**Table S2:** C-H...O interaction parameters of **RBSB**

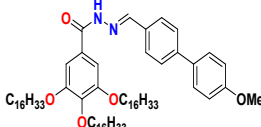
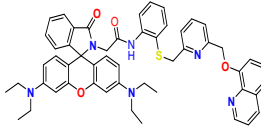
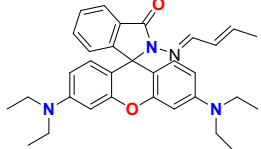
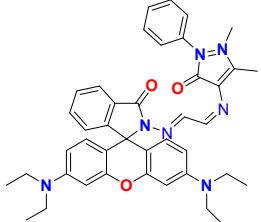
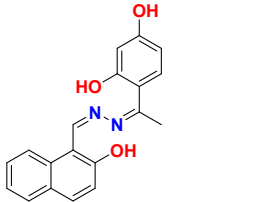
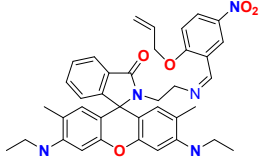
C-H...O	C-H (Å)	H...O (Å)	C...O (Å)	C-H...O (°)	Symmetry operation for O
C(8)-H(8)...O34	0.95	2.46	3.307(5)	148.6	-x+1, -y+1, -z+1

**Table S3:** Lifetime decay parameters for **RBSB** in the presence and absence of Pd(II) in AIE active medium

sample	$\alpha_1$	$\alpha_2$	$\tau_1$	$\tau_2$	$\tau_{av}$	$\chi_2$	$K_1$	$K_2$
RBSB	0.58	0.42	1.94	2.90	2.34	1.00	0.171	0.344
RBSB-Pd(II)	0.44	0.56	2.08	3.79	3.03	0.995		
RBSB-TNP	0.43	0.57	2.40	3.12	2.81	0.996		

**Table S4:** Comparative study of literature reported probes for Pd(II) detection

Compound	Solvent/ Sensing process	LOD	AIEE property	Additional analyte detection/ TNP detection	Applicati on	Refere nce
	MeCN/H <sub>2</sub> O (8:2 v/v)/ Turn on	2.4 nm	No	No/No	Paper strip	01
Al(III) MOF	H <sub>2</sub> O/Turn off	120 nm	No	No/No	Paper strip	02
	THF/ Turn off	0.97 nm	No	No/No	Paper strip, Real water, Live cell imaging	03
	DMSO-H <sub>2</sub> O (1:9 v/v)/ Turn off	$7.90 \times 10^{-8}$ M	Yes	Cu(II)/No	Live cell imaging, paper strip	04

	DMF/ Turn off	$2.30 \times 10^{-8}$ M	No	H <sub>2</sub> PO <sub>4</sub> <sup>-</sup> /No	No	05
	MeCN/H <sub>2</sub> O (5:5 v/v)/ Turn on	82 nm	No	No/No	Live cell imaging	06
	MeOH/PBS (1:1 v/v) Turn on	0.19 μM	No	No/No	Paper strip, Live cell imaging	07
	MeCN/H <sub>2</sub> O (4:1 v/v)/ Turn on	11.9 μM	No	No/No	Live cell imaging	08
	MeOH/H <sub>2</sub> O (1:1 v/v)/	$9.80 \times 10^{-7}$ M	No	Cu(II)/No	Paper strip	09
	MeCN/H <sub>2</sub> O (1:5 v/v)/ Turn on	50 nm	No	No/No	Live cell imaging	10

## References

1. M. Wang, X. Liu, H. Lu, H. Wang, and Z. Qin, *ACS Appl. Mater. Interfaces* 2015, **7**, 2, 1284–1289.
2. P. Chakraborty, A. Rana, S. Mukherjee, and S. Biswas, *Inorg. Chem.* 2023, **62**, 2, 802–809.
3. F. Xu, D. Zhang, Q. Lu, R. Zhang, and J. Xia, *Talanta*, 2023, **253**, 123967.
4. P. Sharma, S. Kaur, S. Kaur and P. Singh, *Photochem. Photobiol. Sci.*, 2020, **19**, 504.
5. S. Wu, H. Jiang, Y. Zhang, L. Wu, P. Jiang, N. Ding, H. Zhang, L. Zhao, F. Yin, and Q. Yang, *Journal of Molecular Liquids*, 2021, **327**, 114836.
6. F. K. Tang, S. M. Chan, T. Wang, C. S. Kwan, R. Huang, Z. Cai, K. and C. F. Leung, *Talanta*, 2020, **210**, 12063.

7. M. Yang, Y. Bai, W. Meng, Z. Cheng, N. Su, and B. Yang, *Inorganic Chemistry Communications*, 2014, **46**, 310–314.
8. S. Mondal, S. K. Manna, S. Pathak, A. Al Masum and S. Mukhopadhyay, *New J. Chem.*, 2019, **43**, 3513-3519.
9. A. Kumar, Virender, B. Mohan, A. A. Solovev, M. Saini, and H. K. Sharma, *Microchemical Journal*, 2022, **180**, 107561.
10. A. K. Bhanja, S. Mishra, K. D. Saha and C. Sinha, *Dalton Trans.*, 2017, **46**, 9245-9252.