

Ionic liquid mediated Pd-free biaryls synthesis catalysed by *in-situ* generated nickel nano particles

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General Information: All starting materials and solvents were purchased from common commercial sources and were used without additional purification [(NiCl₂.6H₂O, CAS No 7791-20-0, Merk), (1-methyl imidazole , CAS No 616-47-7, Sigma-Aldrich), (NaBH₄, CAS No 16940-66-2, Sigma-Aldrich), (1-chlorooctane, CAS No CAS RN: 111-85-3, TCI), (NaOH palletes, CAS No 1310-73-2 , Merk), toluene CAS No 108-88-3, Merk). To maintain palladium free environment, each and every apparatus such as glassware, spatula, round bottomed flask, magnetic bead used for the reaction is subjected to acid wash and oven dried. All reactions were carried out under atmospheric conditions and reported yields are all isolated yields. Aluminium sheets pre-coated with silica gel 60F254 (Merck) were used for thin layer chromatography (TLC) to visualize products under 254 nm UV light. Column chromatography was executed on silica gel (120–230 mesh) to obtain isolated yields. All ¹H-NMR spectra were recorded on 500 MHz and ¹³C on 125 MHz spectrometer using TMS as an internal standard. Chemical shifts are reported in parts per million (ppm, δ). The Transmission Electron Microscope (TEM) analysis was performed by using JEOL JEM-F200 TEM model, Scanning Electron Microscopes (SEM) was performed by using FESEM, JEOL, JSM-7600F, X-Ray Photoelectron Spectrometer (XPS) analysis was performed by using ESCALAB Xi+ model, X-ray Diffraction (XRD) analysis was performed using Bruker D8 Advance model. Both ¹HNMR and ¹³CNMR NMR Spectrometer were recorded on Bruker Advance 500MHz. Mass spectra are recorded on Thermo-Scientific Mass Spectrometer TSQ Endura model.

1. Synthesis of IL: 1-Methyl-3-octyl imidazolium chloride [OMIM][Cl]:

Equimolar mixture of 1-methylimidazole (10 g, 121.80 mmol) and 1-chlorooctane (18.11 g, 121.80 mmol) was stirred at 80 °C for near 10-15 h. After the accomplishment of the reaction, two distinct layers of liquids were found to be observed and IL layer was collected from that. Subsequently repeated wash of that layer was performed with ethyl acetate to remove excess 1-chlorooctane and residual viscous layer therefore obtained was [OMIM][Cl]¹.

2. General experimental procedure for Suzuki reaction:

In a round bottomed flask, a mixture comprising with aryl halide (0.5 mmol, 1 equiv.), arylboronic acid (0.75 mmol, 1.5 equiv), nickel precursor (5 mol%), NaBH₄ (0.25 mmol, 0.5 equiv.), base (1 mmol, 2 equiv.) and IL (0.25 mmol, 0.5 equiv.) was stirred in toluene (1.5ml) for 5 h at 110 °C as mention in **Scheme 4** in the manuscript. Consequently, the reaction solution was extracted three times with ethyl acetate (3 × 10 mL). The reaction mixture were purified by silica gel column chromatography using n-hexane/ethyl acetate (9:1 v/v) to get the desired

coupling products. The progress of the reaction was monitored by TLC. The products were characterized by ^1H NMR and LC-MS.

3. Additional analyses data

3.1 Average particle size

distribution from HRTEM image

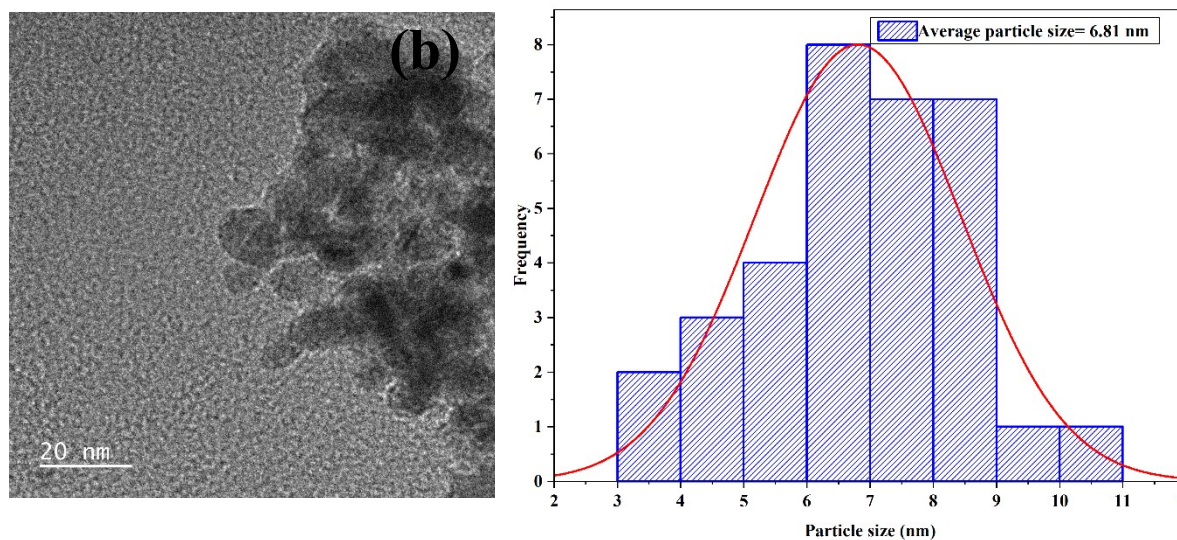


Figure S1: (a) High resolution transmission electron microscope (HRTEM) image of Ni NPs.

(b) Average particle size distribution curve

3.2 d-spacing calculation of Ni NPs from SAED pattern

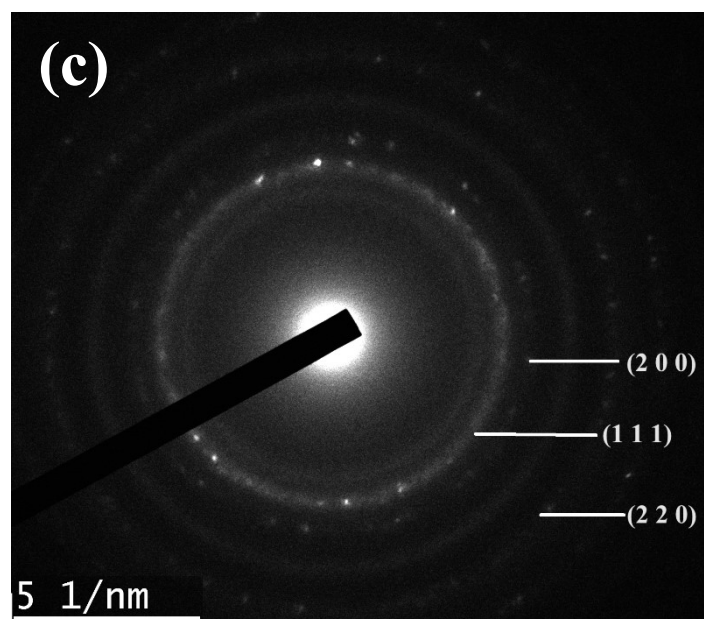


Fig. S2: (c) SAED pattern of Ni NPs.

Table S1: d-spacing calculation of Ni NPs from SAED pattern.

Sl. No.	1/D (nm ⁻¹)	1/r (nm ⁻¹)	r (nm)	d-spacing (Å)	(h k l)
1	9.127	4.5635	0.21913	2.19	(1 1 1)
2	10.587	5.2935	0.188911	1.88	(2 0 0)
3	15.53	7.765	0.128783	1.28	(2 2 0)

3.3: EDAX spectrum of the sample

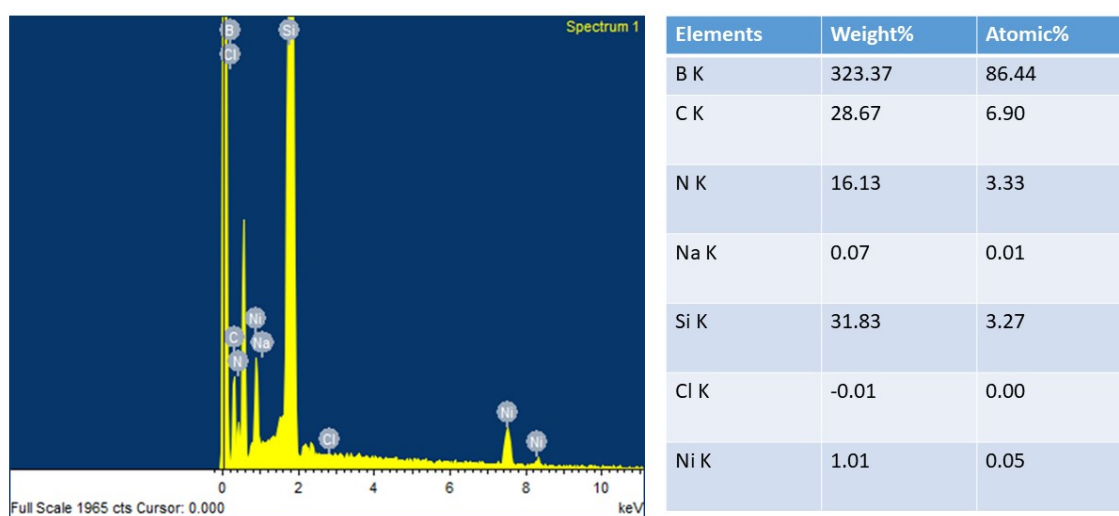


Fig. S3: EDAX spectrum of the sample.

3.4 Elemental mapping of the sample

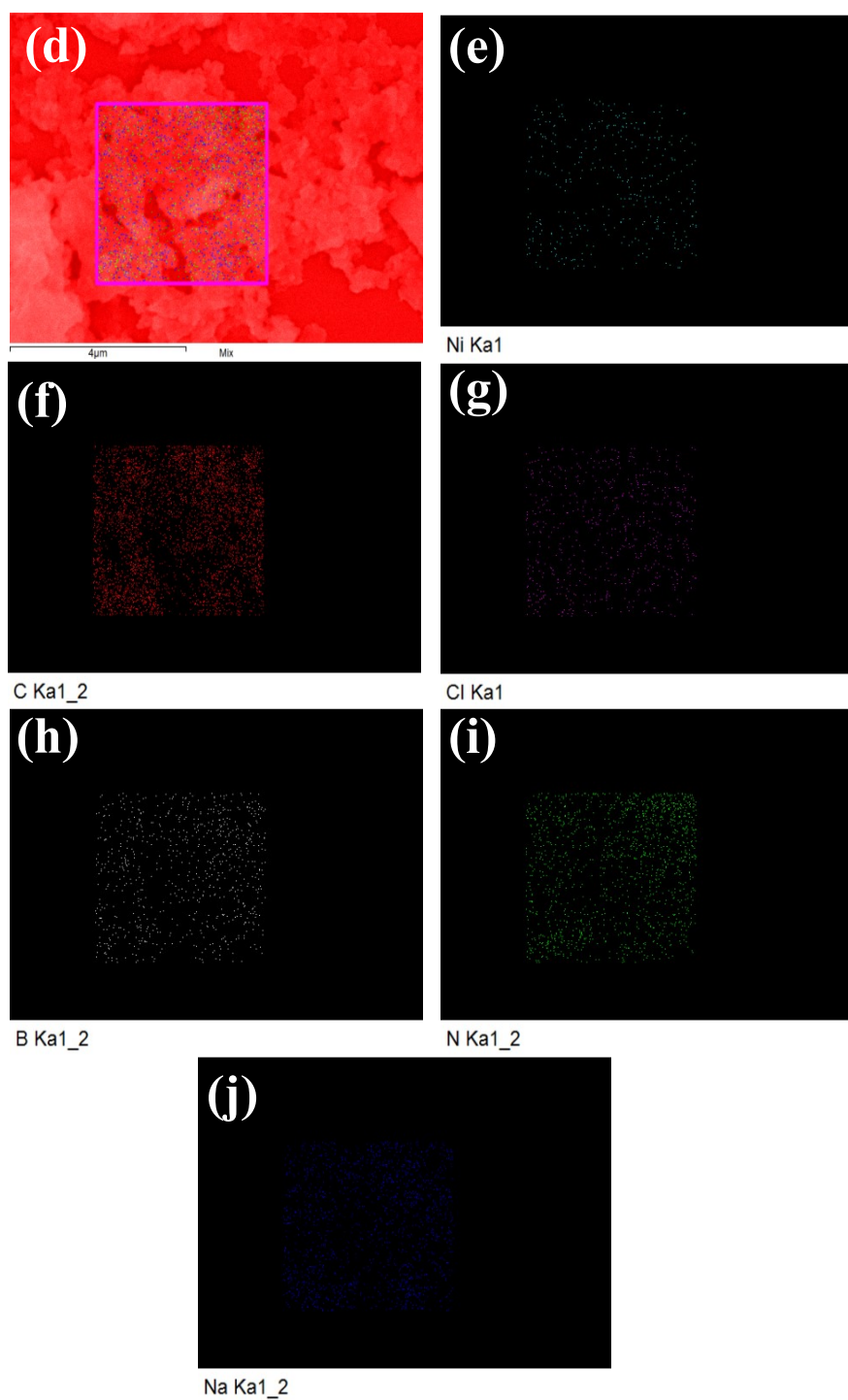


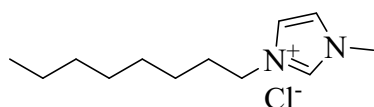
Figure S4: (d) Elemental mapping of the sample. (e), (f), (g), (h), (i) and (j) are showing elemental mappings of Ni, C, Cl, B, N & Na present in the sample respectively.

4. Salient features of the developed methodology compared to literature report:
 A comparative study showing the advantages of this protocol over other reported Pd-free methods

Literature	Catalyst	Reaction conditions	Yield (%)	Substrate scope
Wu <i>et al.</i> ⁶ (2011)	Phosphine Dendrimer stabilized Nickel nanoparticles	THF/60 °C/36 h	99	29 examples
Wang <i>et Al</i> ⁷ (2014)	Ni(TFA) ₂ /Ligand (PPh ₃)	Imidazolium based ionic liquid : water (3:1)/ 80 °C / 18 h	96.7	25 examples
Boit <i>et Al</i> ⁸ (2017)	Ni(cod) ₂ (5 mol%)/N-heterocyclic carbene (NHC) ligand (10 mol%)	Toluene/120 °C /16 h (Prolonged time)	94	27 examples
Key <i>et al.</i> ⁹ (2019)	SiO ₂ supported nickel catalyst ([[(2,2':6',2''-terpyridine-4'-benzoic acid)Ni(II)]Cl ₂)]	Dioxane/115 °C /24 h	80	20 examples
Goldfogel <i>et al.</i> ¹⁰ (2022)	NiCl ₂ .6H ₂ O/ ligand (1'-bis(diphenylphosphino)ferrrocene (DPPF))	Dimethylacetamide (DMAc) :methanol (9:1)/ 80 °C 24 h	80	43 examples
This work	NiCl ₂ .6H ₂ O (5 mol%)	Ionic liquid/NaBH ₄ /NaOH/ 110 °C/5 h	95	42 examples

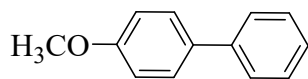
Characterization of the compounds:

1-methyl-3-octylimidazolium chloride



¹H NMR (500 MHz, CDCl₃) δ 10.58 (s, 1H), 7.62 (d, *J* = 12.2 Hz, 1H), 7.41 (s, 1H), 4.27 (d, *J* = 7.4 Hz, 2H), 4.09 (s, 3H), 1.94 – 1.71 (m, 2H), 1.24 (dd, *J* = 22.0, 15.3 Hz, 10H), 0.83 (t, *J* = 6.8 Hz, 3H).
¹³C NMR (125 MHz, CDCl₃) δ 138.00, 123.62, 121.78, 50.05, 36.54, 31.60, 30.28, 28.92, 26.20, 22.50, 13.98.

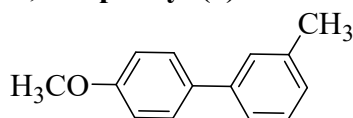
4-methoxy-1,1'-biphenyl (1)



White solid, mp- 49-51°C, ¹H NMR (500 MHz, CDCl₃) δ 7.43 (dd, *J* = 12.4, 8.0 Hz, 4H), 7.30 (t, *J* = 7.7 Hz, 2H), 7.19 (t, *J* = 7.4 Hz, 1H), 6.86 (d, *J* = 8.8 Hz, 2H), 3.71 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 159.23, 140.86, 133.78, 128.83, 128.22, 126.80, 126.76, 114.30, 77.44, 77.18, 76.93, 55.38, 1.16. 19. MS (ESI) *m/z*: [M+H]⁺ calcd for C₁₃H₁₂O 185.09; Found 185.10

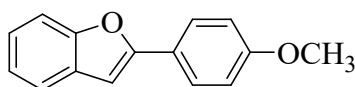
4'-methoxy-3-methyl-1,1'-biphenyl (4)



White solid, mp- 72-74°C, ¹H NMR (500 MHz, CDCl₃) δ 7.46 – 7.44 (m, 2H), 7.30 – 7.19 (m, 6H), 6.91 (d, 7.9 Hz, 2H), 3.78 (s, 3H), 2.34 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 159.62, 142.09, 140.31, 128.65, 128.18, 127.58, 127.28, 123.90, 122.89, 114.14, 55.46, 20.27. MS (ESI) *m/z*: [M+H]⁺ calcd for C₁₄H₁₄O 199.10; Found 199.12

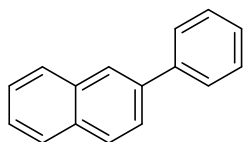
2-(4-methoxyphenyl)benzofuran (22)



White solid, mp- 146-148 °C ¹H NMR (500 MHz, CDCl₃) δ 7.83 (d, *J* = 8.7 Hz, 2H), 7.57 – 7.54 (m, 3H), 7.01 (d, *J* = 8.6 Hz, 3H), 6.92 (s, 1H), 3.89 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 164.81, 161.79, 159.39, 130.26, 126.43, 123.76, 117.41, 114.29, 111.45, 111.00, 109.84, 99.68, 55.45. MS (ESI) *m/z*: [M+H]⁺ calcd for C₁₅H₁₂O₂ 225.08; Found 225.09

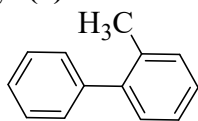
2-phenylnaphthalene (5)



White solid, mp- 94-96°C, ¹H NMR (500 MHz, CDCl₃) δ 8.08 (s, 1H), 7.96-7.89 (m, 3H), 7.77 (t, 3H), 7.53-7.40 (m, 5H).

¹³C NMR (125 MHz, CDCl₃) δ 128.88, 128.43, 128.22, 127.66, 127.45, 127.37, 126.31, 125.95, 125.82, 125.61. MS (ESI) *m/z*: [M+H]⁺ calcd for C₁₆H₁₂ 205.09; Found 205.10

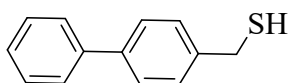
2-methyl-1,1'-biphenyl (6)



White solid, mp- 26-28 °C, ¹H NMR (500 MHz, CDCl₃) δ 7.76 – 7.72 (m, 4H), 7.57-7.55 (m, 5H), 2.29 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 142.42, 140.53, 139.36, 132.19, 131.19, 129.62, 127.80, 126.96, 19.16. MS (ESI) m/z: [M+H]⁺ calcd for C₁₃H₁₂ 169.09; Found 169.10

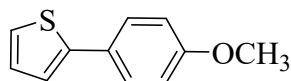
1,1'-biphenyl]-4-ylmethanethiol (14)



White solid, mp- 63-65 °C, ¹H NMR (500 MHz, CDCl₃) δ 7.20 – 7.10 (m, 9H), 3.45 (m, 2H), 1.18 (s, 1H).

¹³C NMR (125 MHz, CDCl₃) δ 140.32, 138.63, 136.40, 135.01, 132.87, 130.29, 129.40, 128.67, 77.30, 77.05, 76.80, 34.89. MS (ESI) m/z: [M+H]⁺ calcd for C₁₃H₁₂S 201.06; Found 201.08

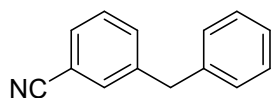
2-(4-methoxyphenyl)thiophene (21)



White solid, mp- 112-114°C, ¹H NMR (500 MHz, CDCl₃) δ 7.31 – 7.19 (m, 3H), 6.71 (d, J = 8.6 Hz, 2H), 3.71 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 158.72, 132.24, 130.77, 115.75, 112.83, 55.44. MS (ESI) m/z: [M+H]⁺ calcd for C₁₃H₁₂O 191.04; Found 191.06

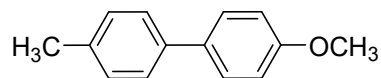
3-benzylbenzonitrile (13)



White solid, mp- 103-105 °C, ¹H NMR (500 MHz, CDCl₃) δ 7.75 (s, 1H), 7.56-7.42 (m, 5H), 7.19 – 6.80 (m, 12H), 4.11 (d, J = 6.2 Hz, 5H).

¹³C NMR (125 MHz, CDCl₃) δ 141.39, 138.50, 133.28, 130.78, 127.25, 117.45, 111.79, 29.70. MS (ESI) m/z: [M+H]⁺ calcd for C₁₄H₁₁N 194.09; Found 194.10

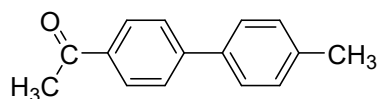
4-methoxy-4'-methyl-1,1'-biphenyl (7)



White solid, mp- 73-76 °C, ¹H NMR (600 MHz, CHLOROFORM-D) δ 7.50 (d, J = 36.4 Hz, 4H), 7.26 (d, J = 20.7 Hz, 2H), 6.97 (s, 2H), 3.85 (s, 3H), 2.38 (s, 3H).

¹³C NMR (150 MHz, CDCl₃) δ 158.99, 138.03, 136.43, 133.82, 129.51, 128.02, 126.65, 114.22, 55.42, 21.12. MS (ESI) m/z: [M+H]⁺ calcd for C₁₄H₁₄O 199.10; Found 199.11

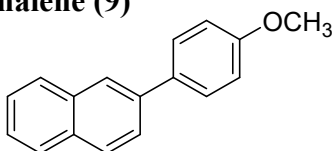
1-(4'-methyl-[1,1'-biphenyl]-4-yl)ethan-1-one (10)



White solid, mp- 111-114 °C, ¹H NMR (600 MHz, CHLOROFORM-D) δ 8.03 (d, J = 8.0 Hz, 2H), 7.68 (d, J = 8.0 Hz, 2H), 7.62 (d, J = 7.5 Hz, 2H), 7.43 (dt, J = 42.1, 7.4 Hz, 3H), 2.63 (s, 3H), 1.57 (s, 9H).

¹³C NMR (150 MHz, CHLOROFORM-D) δ 188.98, 145.15, 141.34, 136.79, 129.03, 129.00, 127.36, 127.31, 26.76. MS (ESI) m/z: [M+H]⁺ calcd for C₁₅H₁₅O 211.10; Found 211.11

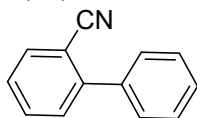
2-(4-methoxyphenyl)naphthalene (9)



White solid, mp- 139-141 °C, ¹H NMR (600 MHz, CHLOROFORM-D) δ 7.52-7.39 (d, J = 8.2 Hz, 9H), 6.97 (d, J = 8.2 Hz, 2H), 3.85 (s, 3H).

¹³C NMR (150 MHz, CHLOROFORM-D) δ 159.49, 139.82, 132.58, 131.86, 128.38, 128.06, 120.86, 114.41, 55.45. MS (ESI) m/z: [M+H]⁺ calcd for C₁₇H₁₄O 235.10; Found 235.11

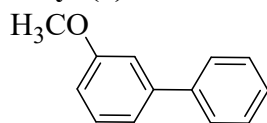
1,1'-biphenyl]-2-carbonitrile (11)



White solid, mp- 90-94 °C, ¹H NMR (500 MHz, CDCl₃) δ 7.77 (d, J = 7.8 Hz, 1H), 7.71-7.64 (m, 2H), 7.57-7.45 (m, 5H), 7.23 (d, J = 7.4 Hz, 1H).

¹³C NMR (125 MHz, CDCl₃) δ 134.03, 133.83, 132.86, 130.49, 130.07, 129.46, 128.76, 127.53, 127.12, 125.88. MS (ESI) m/z: [M+H]⁺ calcd for C₁₃H₉N 180.07; Found 185.10

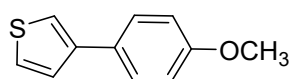
3-methoxy-1,1'-biphenyl (8)



Pale yellow solid, mp- 49-52 °C, ¹H NMR (500 MHz, CDCl₃) δ 7.65 (d, *J* = 6.7 Hz, 2H), 7.50 – 7.19 (m, 5H), 6.96 (d, *J* = 5.5 Hz, 2H), 3.91 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 160.00, 142.82, 141.15, 129.78, 128.77, 127.45, 127.24, 119.73, 112.96, 112.73, 55.32. MS (ESI) *m/z*: [M+H]⁺ calcd for C₁₃H₁₂O 185.09; Found 185.11

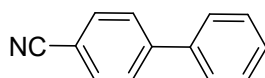
3-(4-methoxyphenyl)thiophene (18)



White solid, mp- 112-115 °C, ¹H NMR (500 MHz, CDCl₃) δ 7.75-7.66 (m, 3H), 7.51 (d, *J* = 8.0 Hz, 2H), 6.99 (d, *J* = 8.1 Hz, 2H), 3.87 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 156.15, 141.75, 132.93, 127.74, 124.12, 114.18, 55.45. MS (ESI) *m/z*: [M+H]⁺ calcd for C₁₃H₁₀OS 191.04; Found 191.05

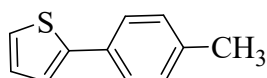
1,1'-biphenyl]-4-carbonitrile (16)



White solid, mp- 91-93 °C, ¹H NMR (500 MHz, CDCl₃) δ 7.45 – 7.34 (m, 6H), 7.22 (s, 1H), 7.16 (d, *J* = 7.8 Hz, 2H).

¹³C NMR (125 MHz, CDCl₃) δ 148.42, 142.10, 132.22, 126.57, 124.13, 120.96, 112.16. MS (ESI) *m/z*: [M+H]⁺ calcd for C₁₃H₉N 180.07; Found 180.10

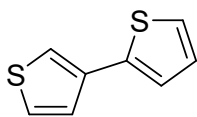
2-(p-tolyl)thiophene (19)



White solid, mp- 90-93 °C, ¹H NMR (500 MHz, DMSO) δ 7.42 – 7.21 (m, 4H), 6.96 (d, *J* = 8.2 Hz, 3H), 2.51 (s, 3H).

¹³C NMR (125 MHz, DMSO) δ 152.65, 132.57, 130.13, 129.42, 127.98, 125.91, 122.40, 24.14. MS (ESI) *m/z*: [M+H]⁺ calcd for C₁₁H₁₀S 175.05; Found 175.06

2,3'-bithiophene (17)

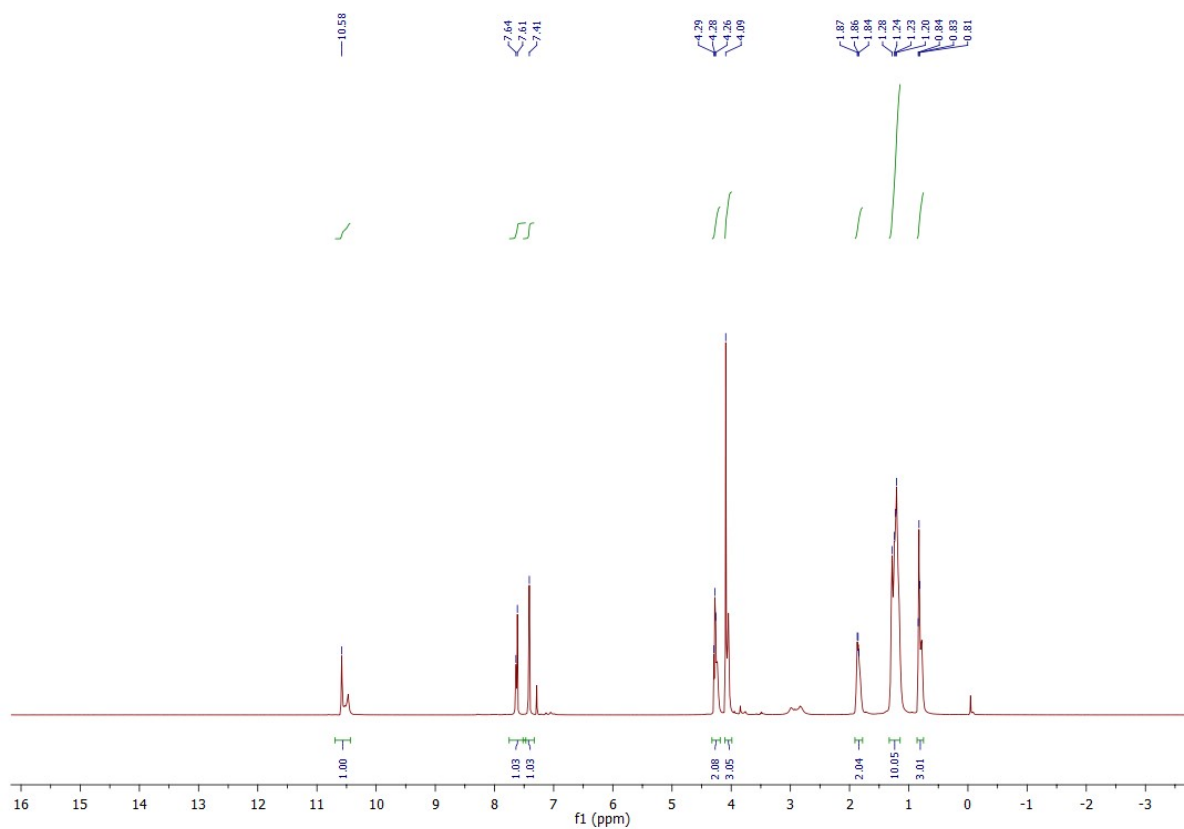


White solid, mp- 130-133 °C, ^1H NMR (500 MHz, CDCl_3) δ 8.11 (d, $J = 8.7$ Hz, 1H), 7.76-7.74 (m, 1H), 7.57 – 7.55 (m, 1H), 7.41 (d, $J=8.8\text{Hz}$, 2H), 6.82 (d, $J=8.8\text{Hz}$, 1H).

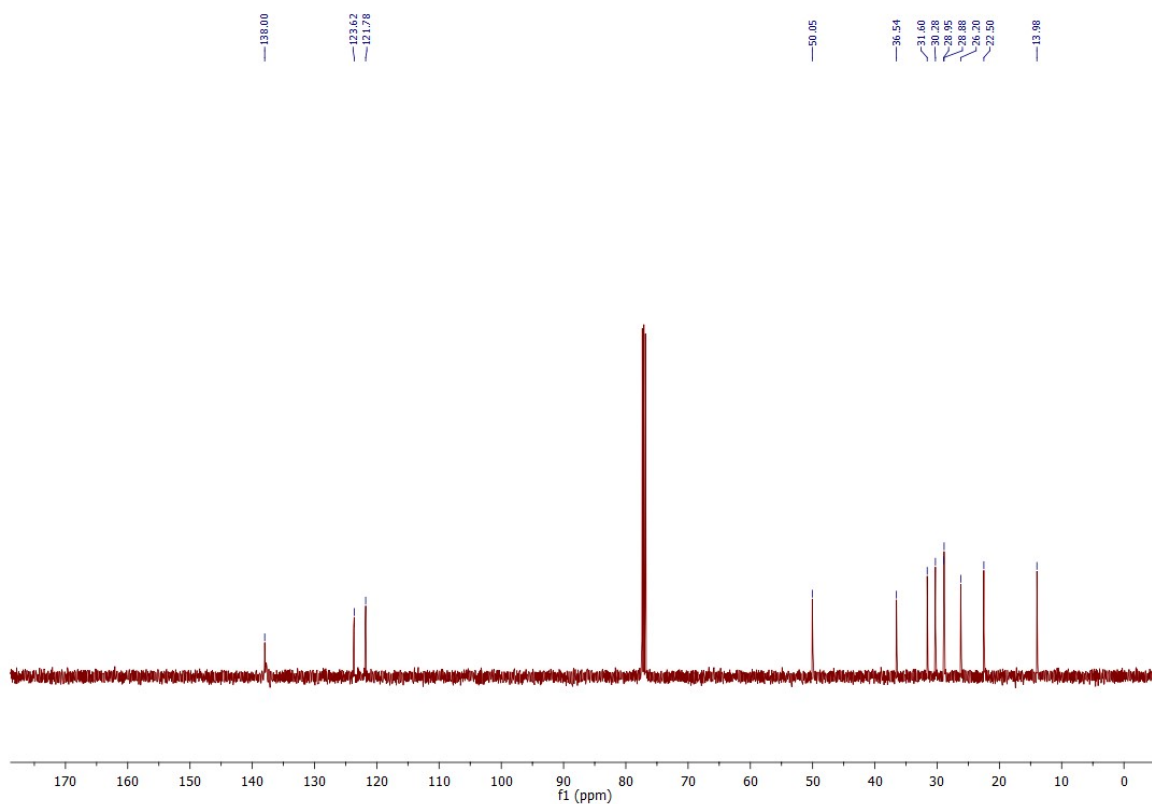
^{13}C NMR (125 MHz, CDCl_3) δ 140.67, 137.87, 130.78, 130.51, 121.67.

MS (ESI) m/z: $[\text{M}+\text{H}]^+$ calcd for $\text{C}_8\text{H}_6\text{S}_2$ 167.10; Found 167.11

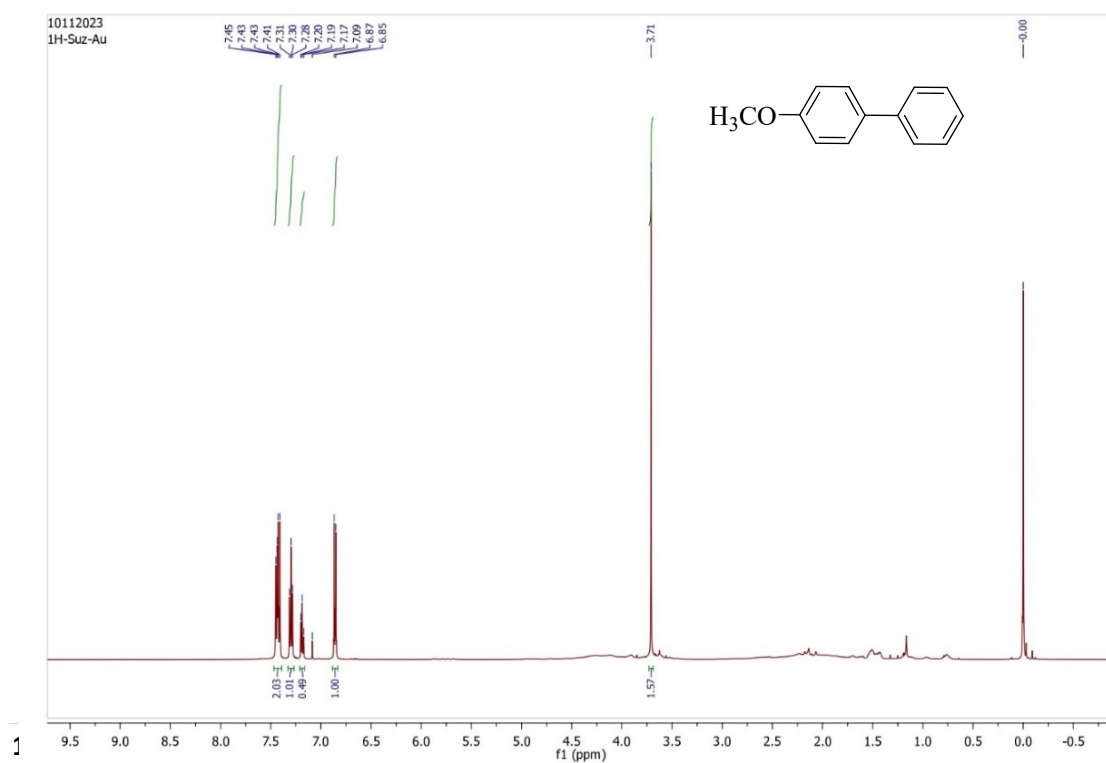
^1H NMR Spectrum of 1-methyl-3-octylimidazolium chloride



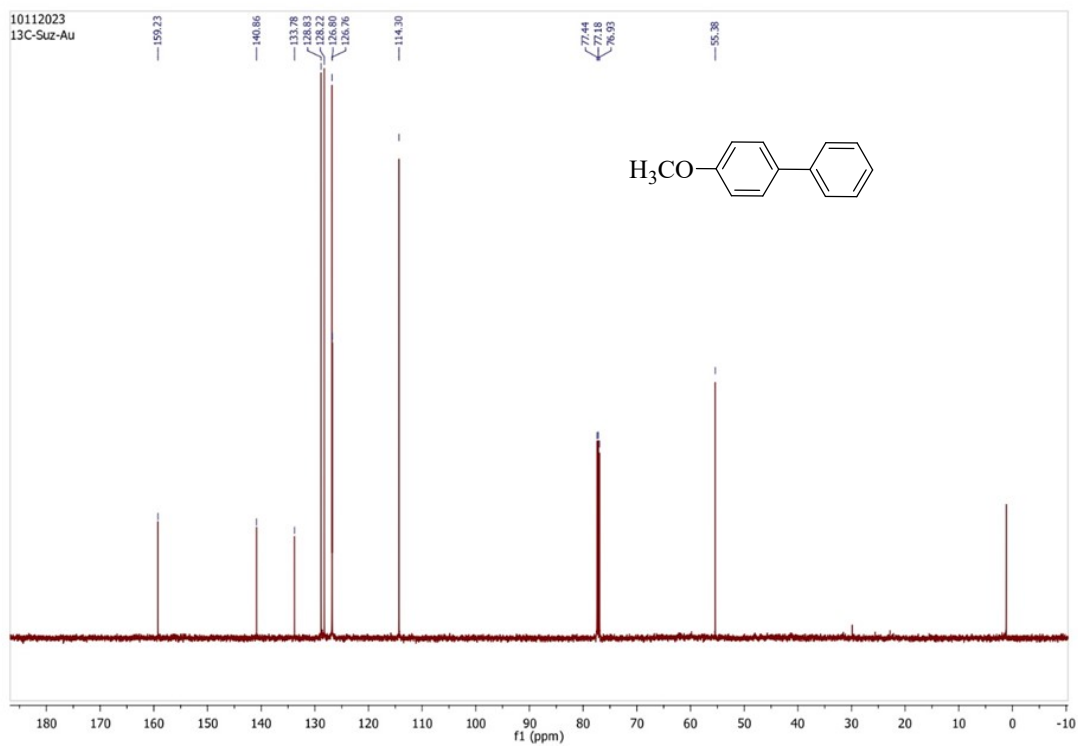
¹³C NMR Spectrum of 1-methyl-3-octylimidazolium chloride



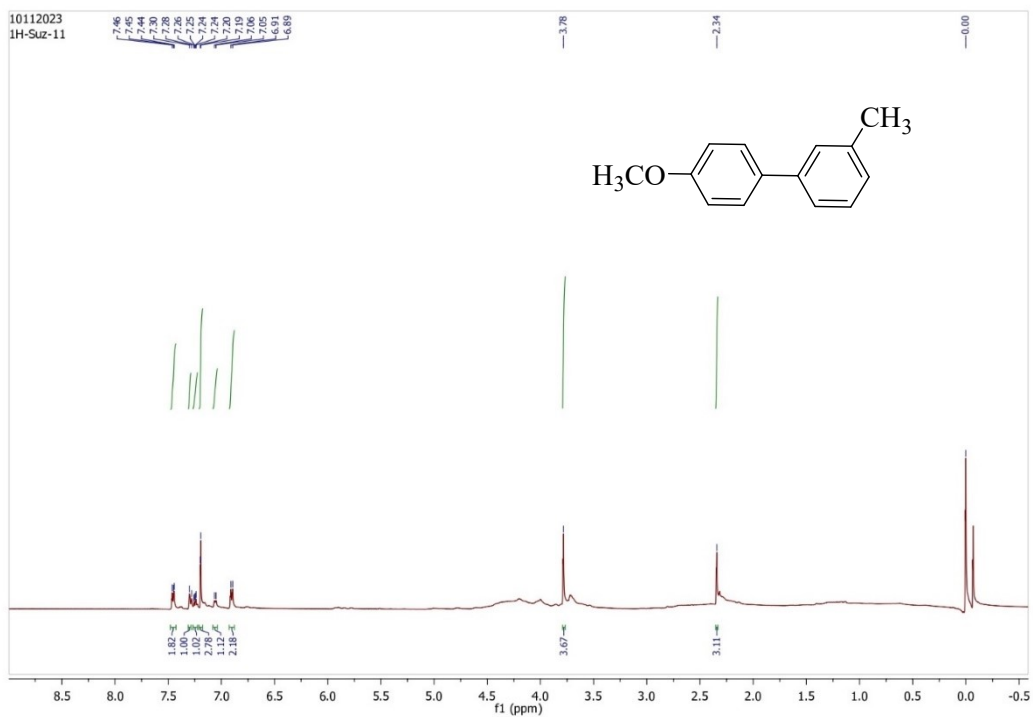
¹H NMR Spectrum of 4-methoxy-1, 1'-biphenyl (1)



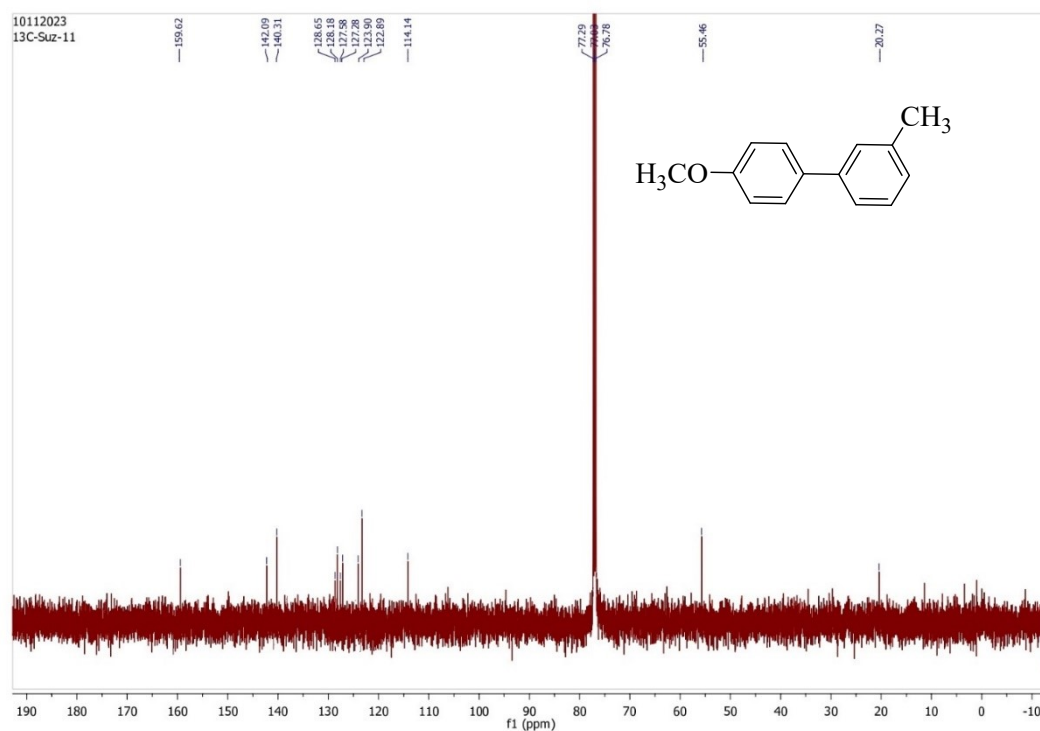
¹³C NMR Spectrum of 4-methoxy-1,1'-biphenyl (1)



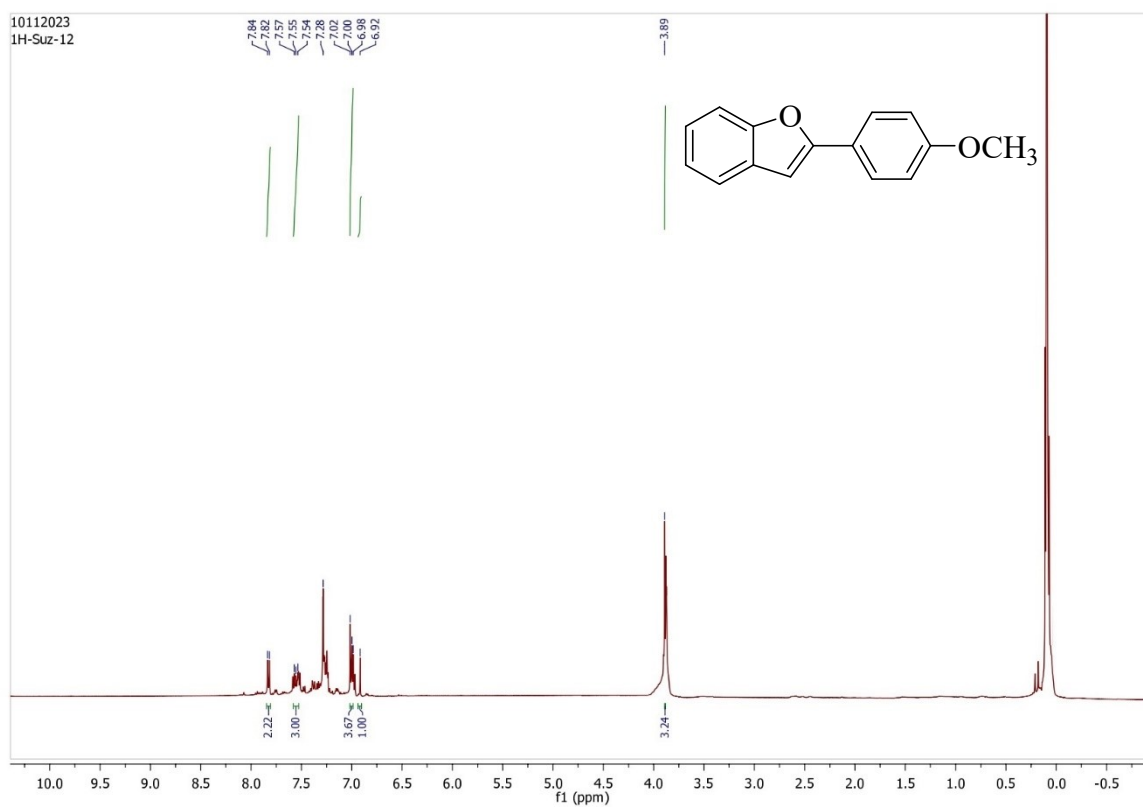
¹H NMR Spectrum of 4'-methoxy-3-methyl-1,1'-biphenyl (4)



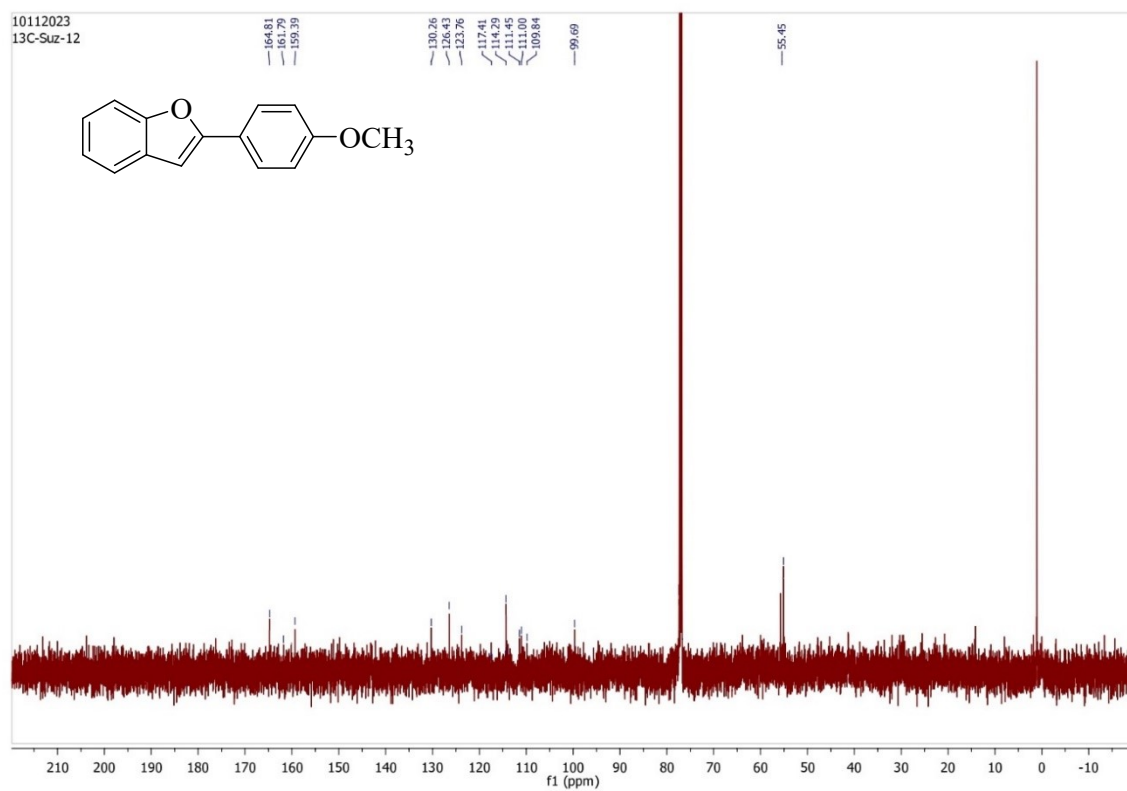
¹³C NMR Spectrum of 4'-methoxy-3-methyl-1,1'-biphenyl (4)



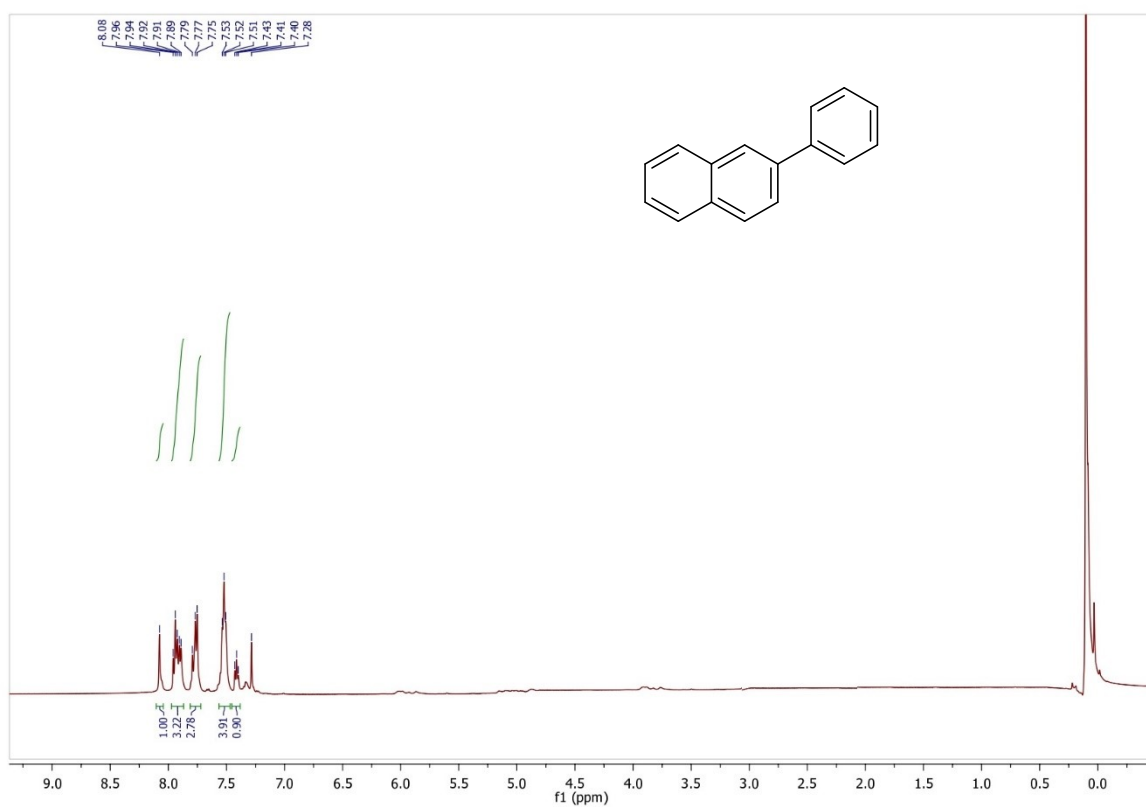
¹H NMR Spectrum of 2-(4-methoxyphenyl)benzofuran (22)



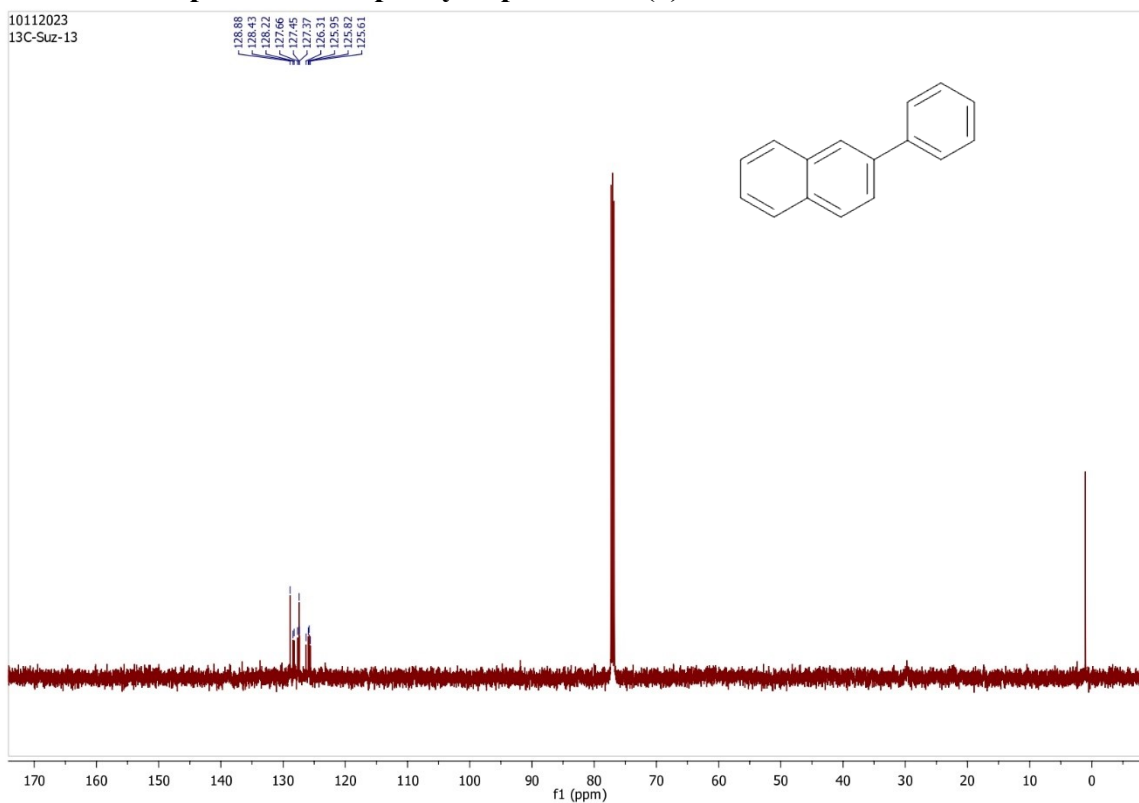
¹³C NMR Spectrum of 2-(4-methoxyphenyl)benzofuran (22)



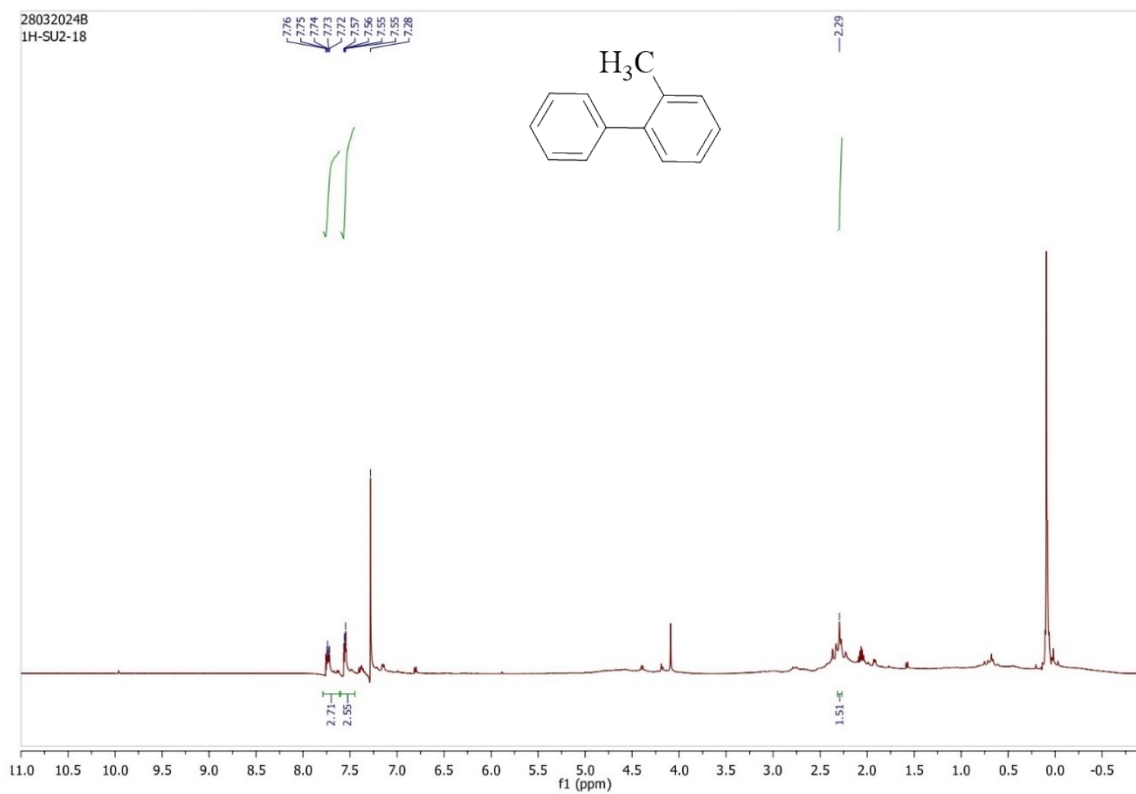
¹H NMR Spectrum of 2-phenylnaphthalene (5)



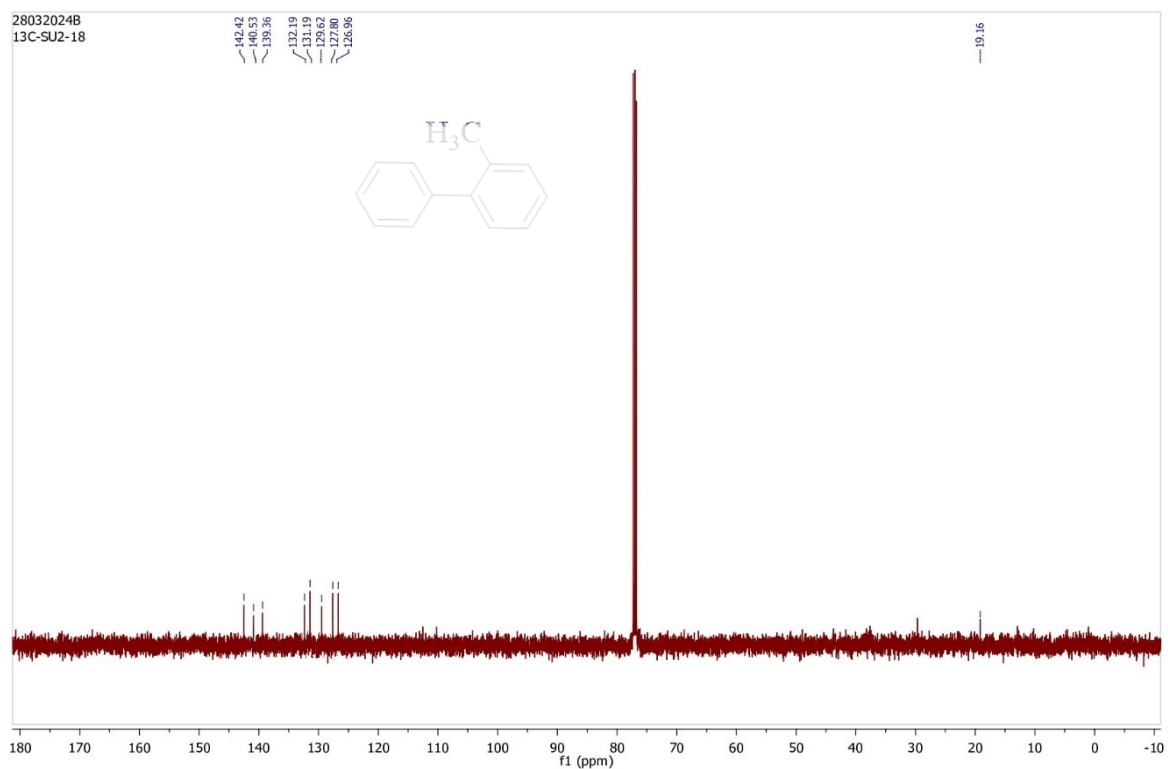
¹³C NMR Spectrum of 2-phenylnaphthalene (5)



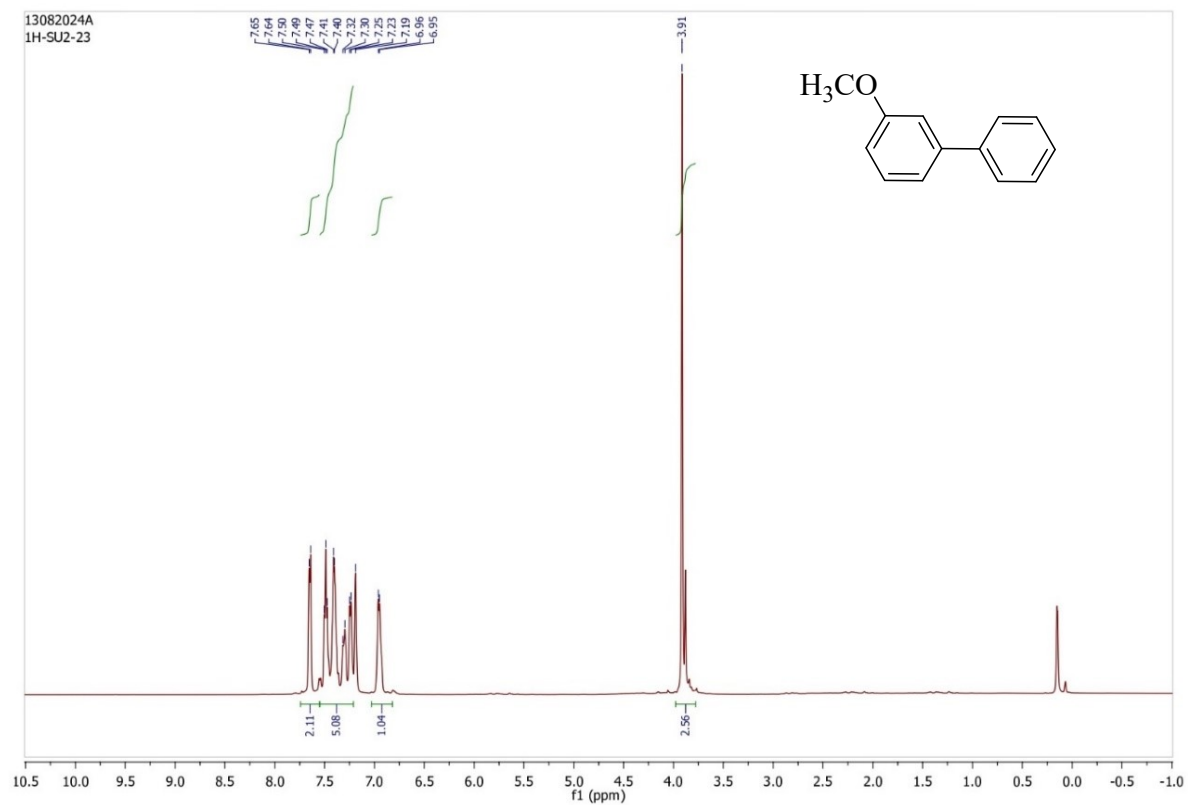
¹H NMR Spectrum of 2-methyl-1,1'-biphenyl (6)



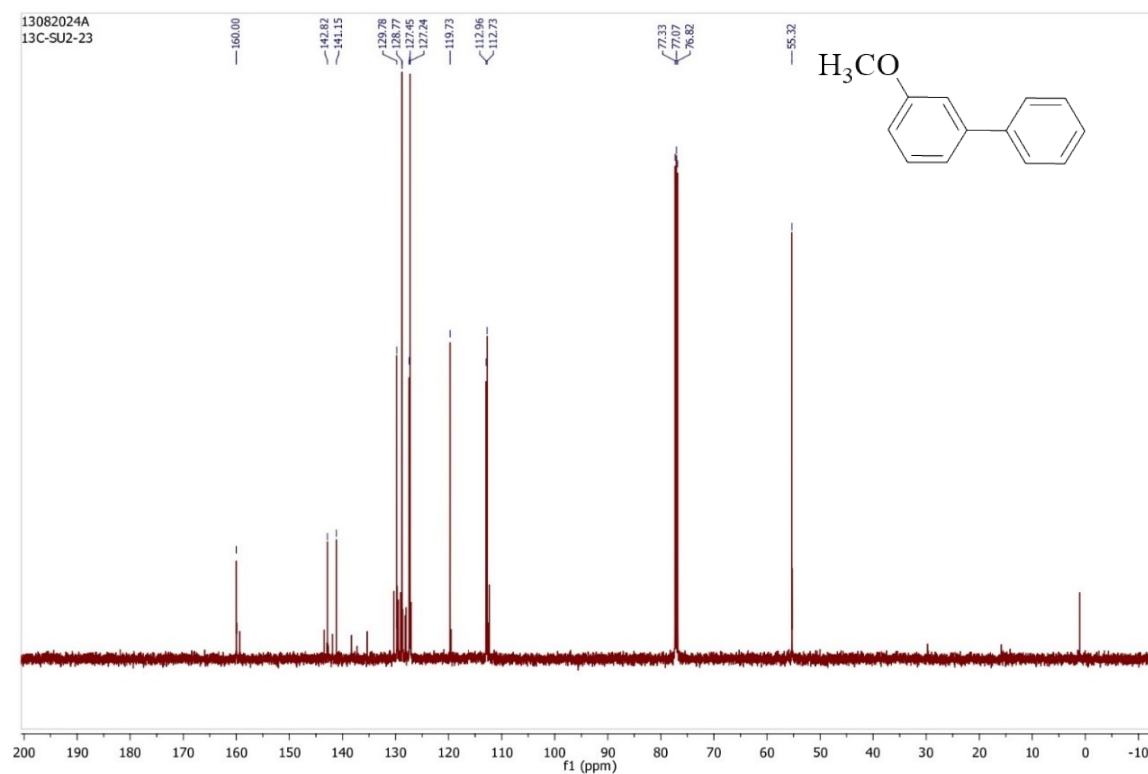
¹³C NMR Spectrum of 2-methyl-1,1'-biphenyl (6)



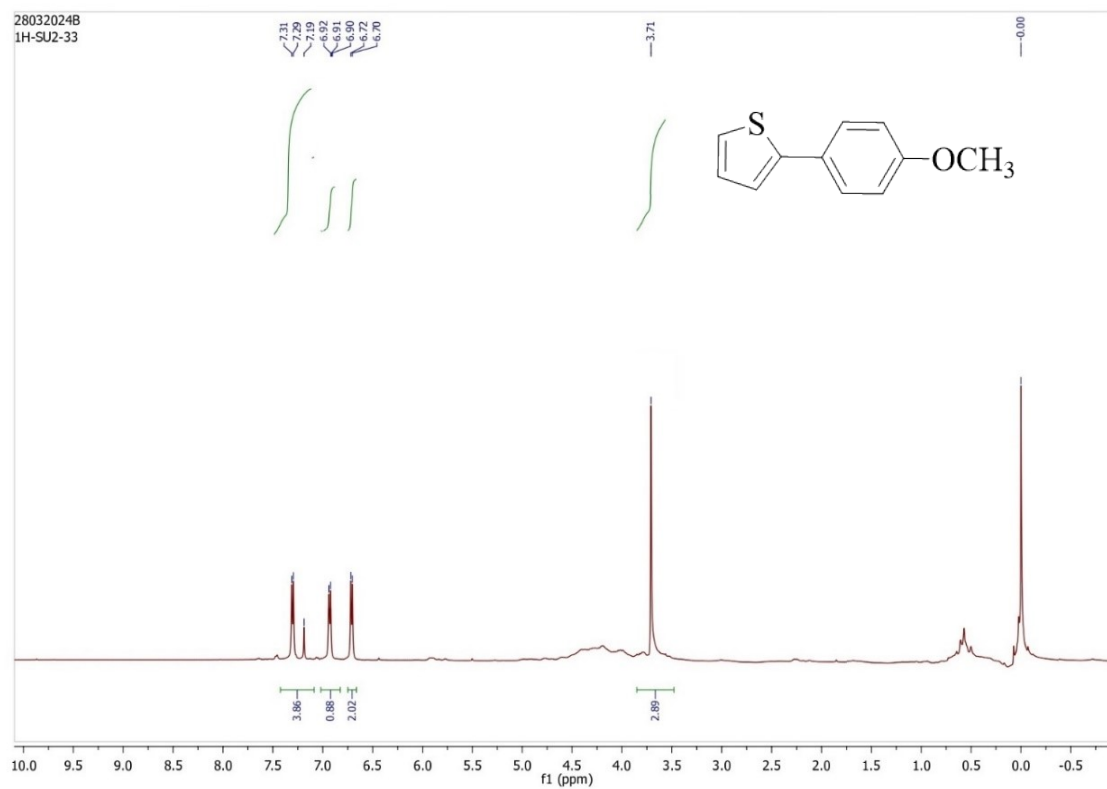
¹H NMR Spectrum of 3-methoxy-1,1'-biphenyl (8)



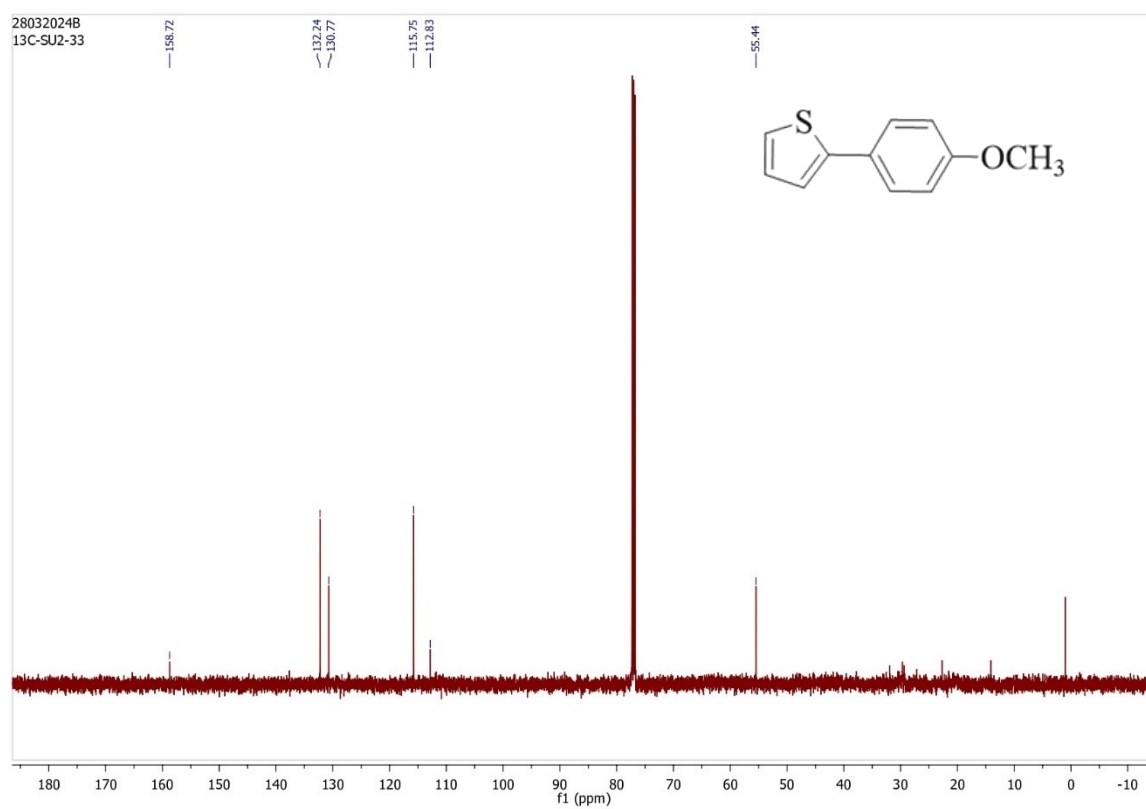
¹³C NMR Spectrum of 3-methoxy-1,1'-biphenyl (8)



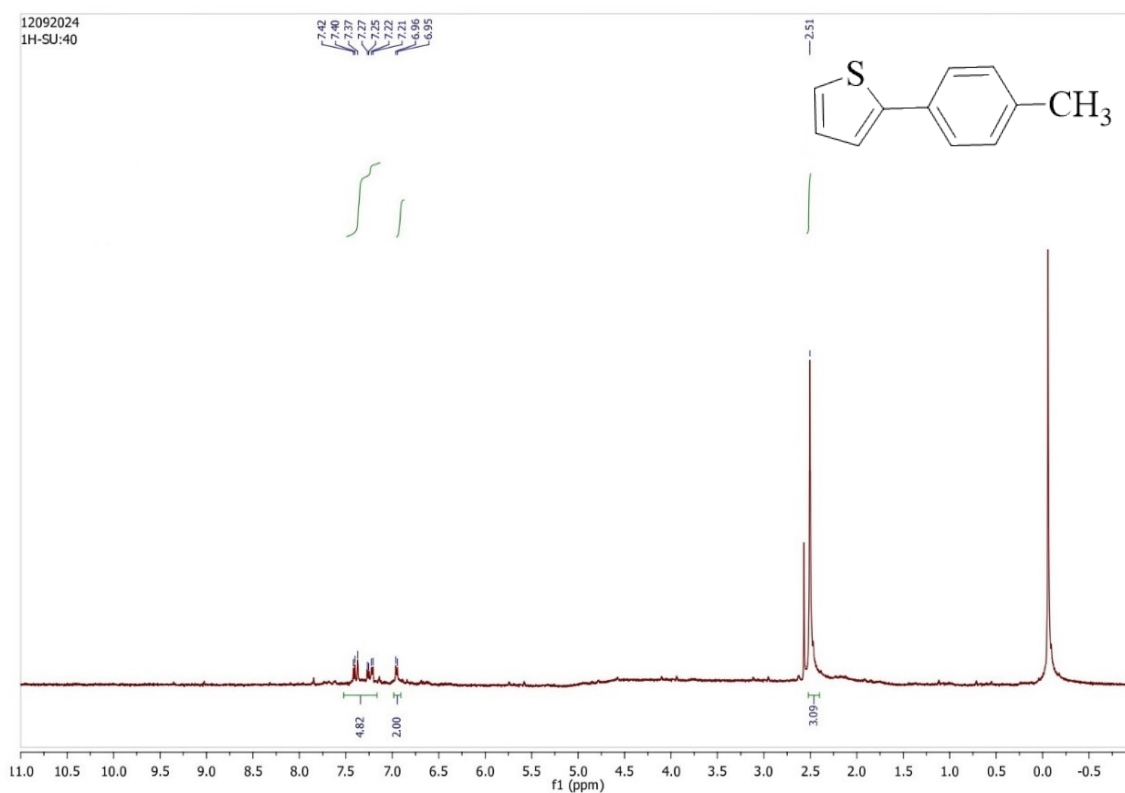
¹H NMR Spectrum of 2-(4-methoxyphenyl)thiophene (21)



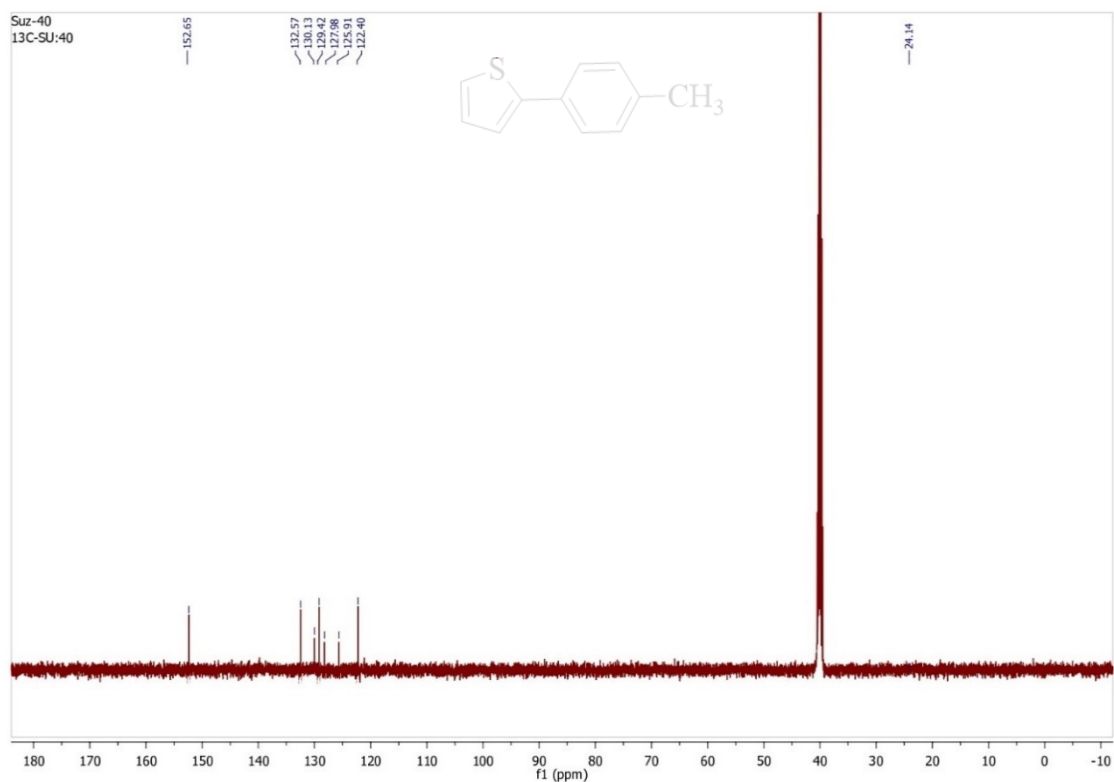
¹³C NMR Spectrum of 2-(4-methoxyphenyl)thiophen (21)



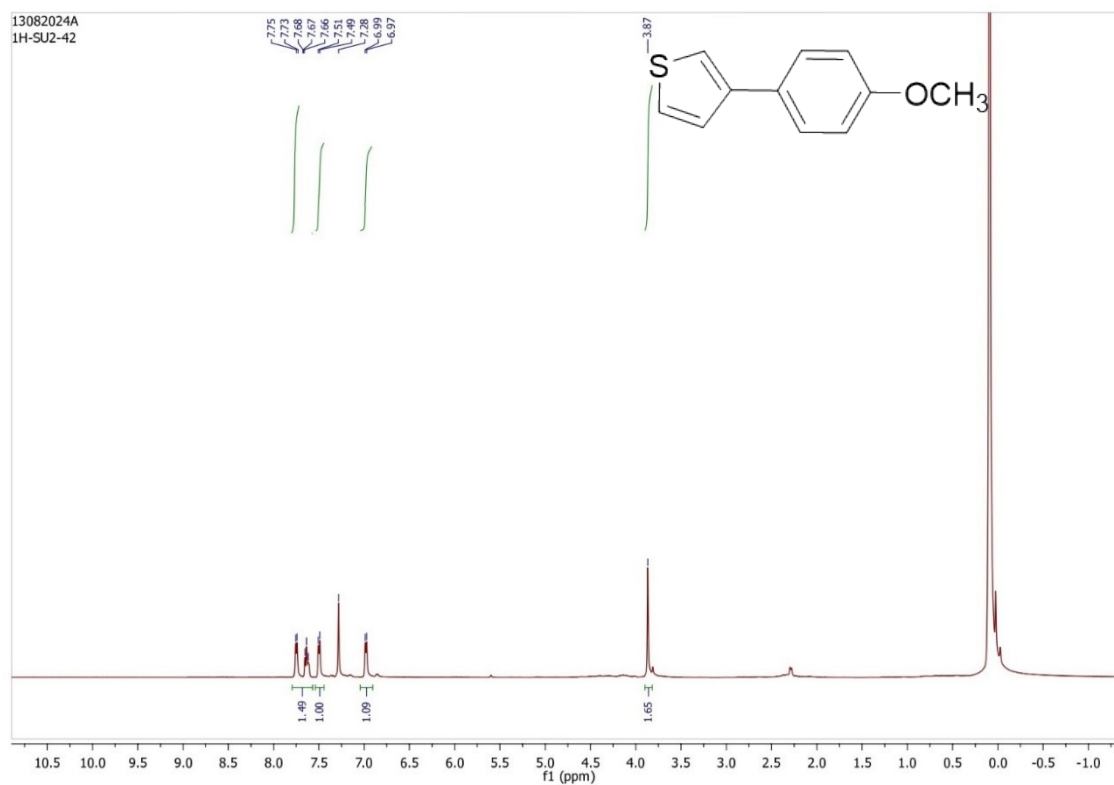
¹H NMR Spectrum of 2-(p-tolyl)thiophene (19)



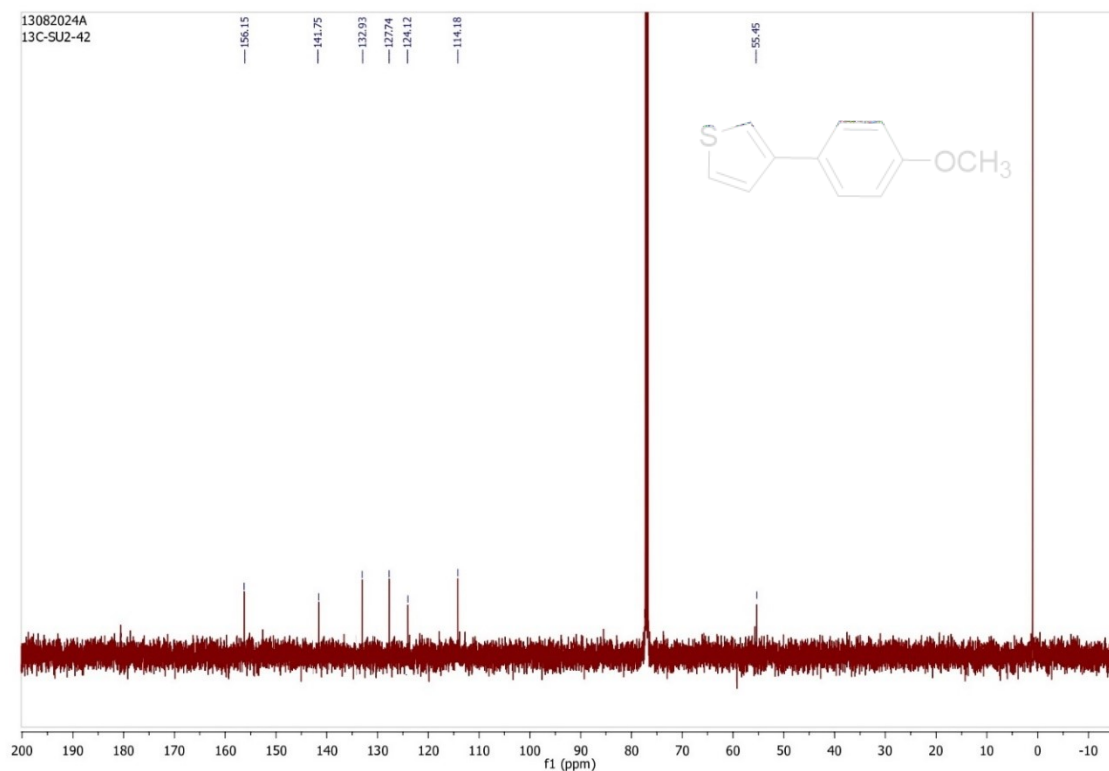
¹³C NMR Spectrum of 2-(p-tolyl)thiophene (19)



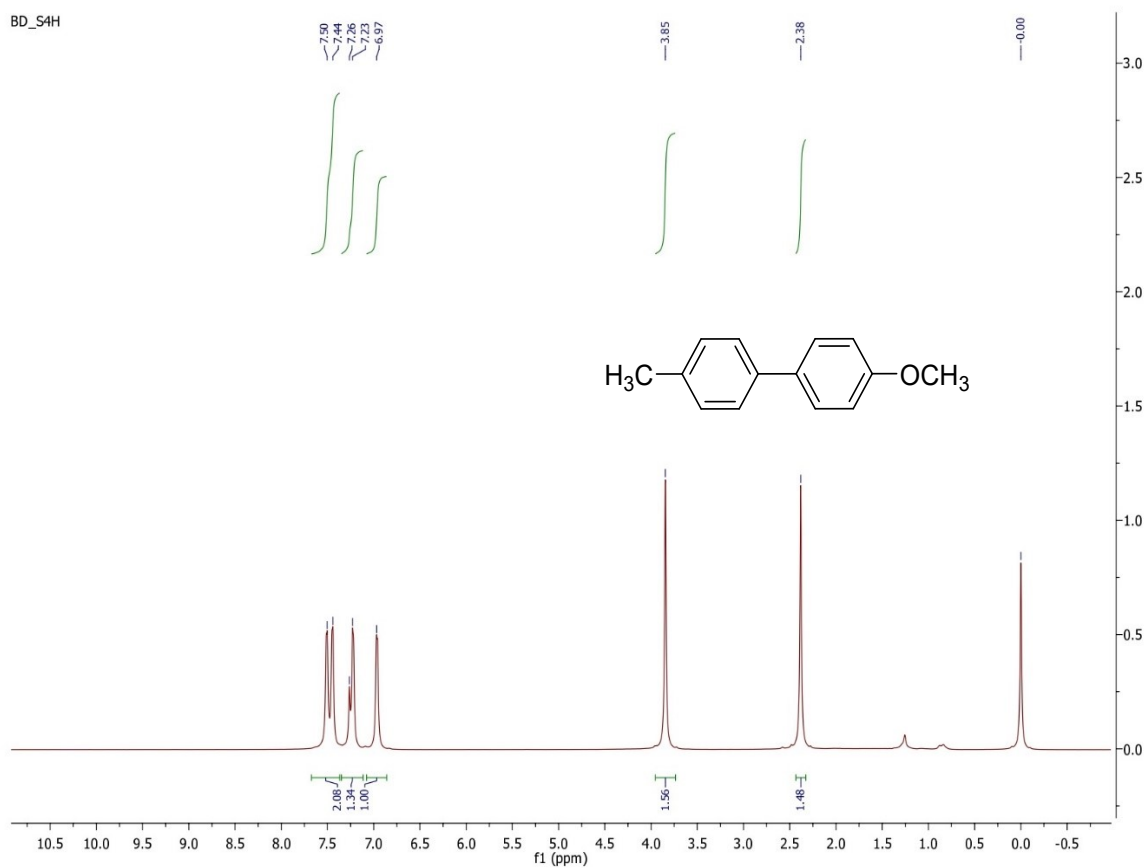
¹H NMR Spectrum of 3-(4-methoxyphenyl)thiophene (21)



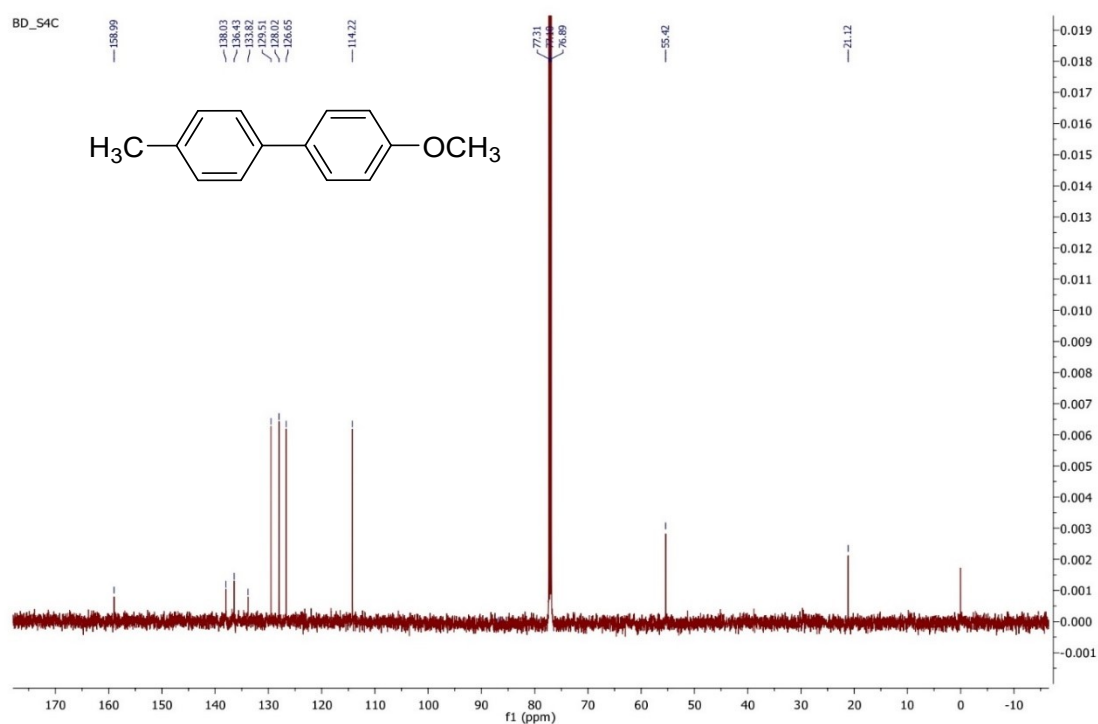
¹³C NMR Spectrum of 3-(4-methoxyphenyl)thiophene (21)



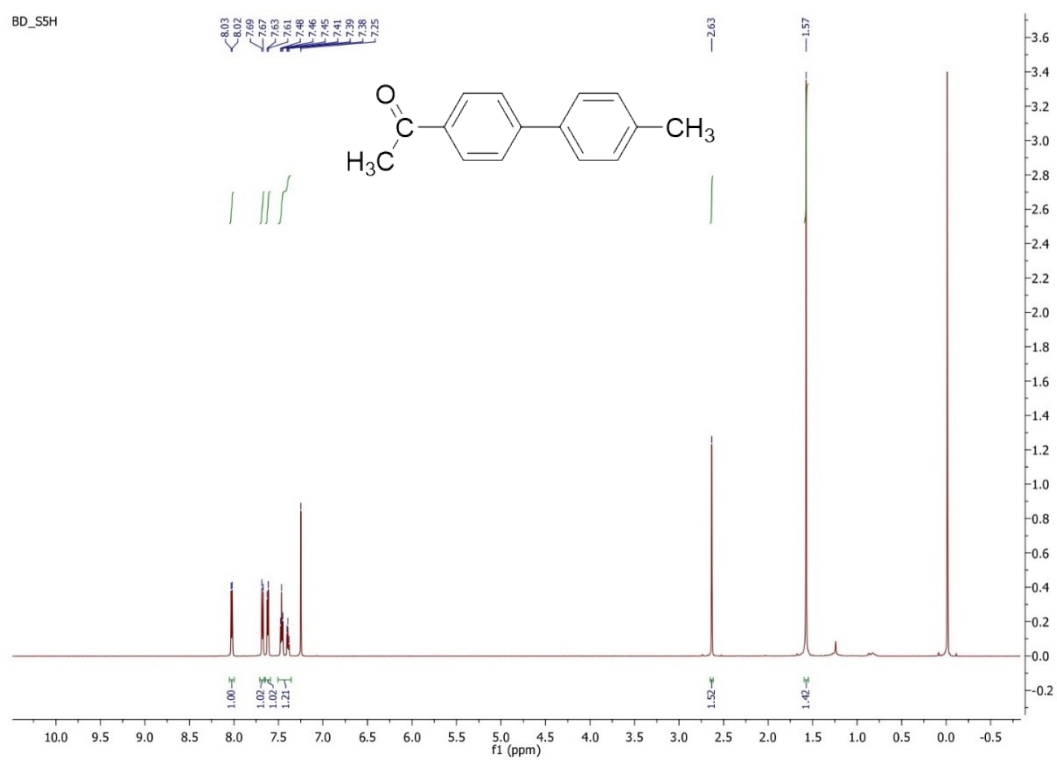
¹H NMR Spectrum of 4-methoxy-4'-methyl-1,1'-biphenyl (7)



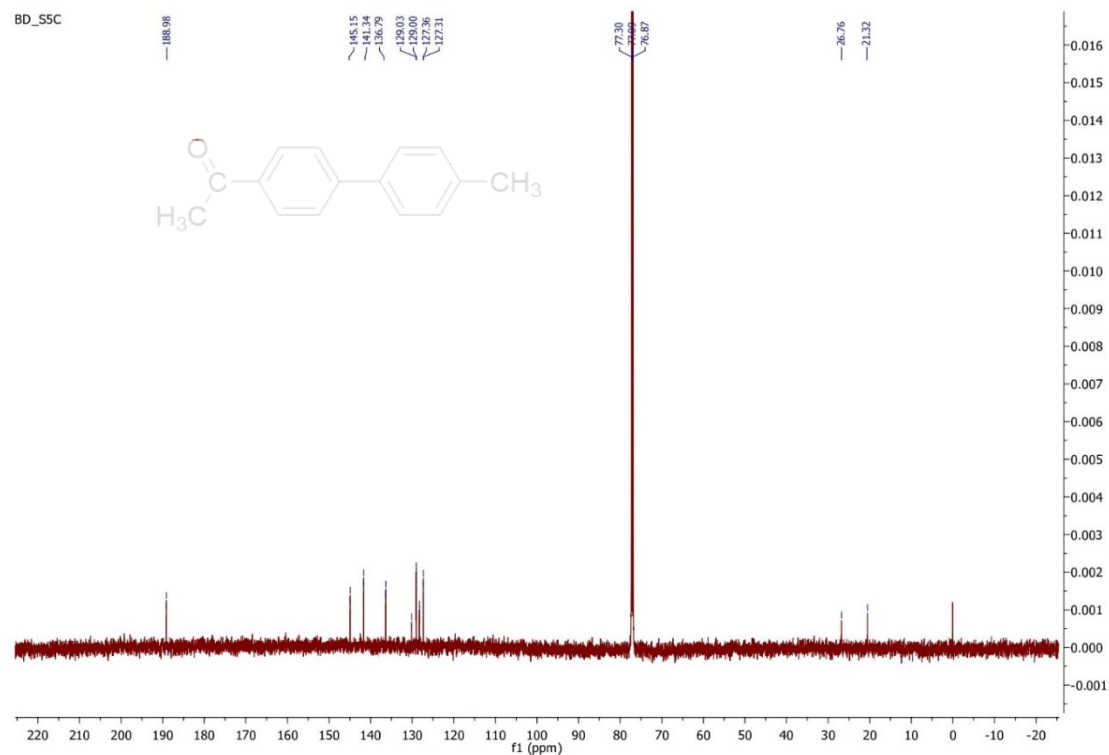
¹³C NMR Spectrum of 4-methoxy-4'-methyl-1,1'-biphenyl (7)



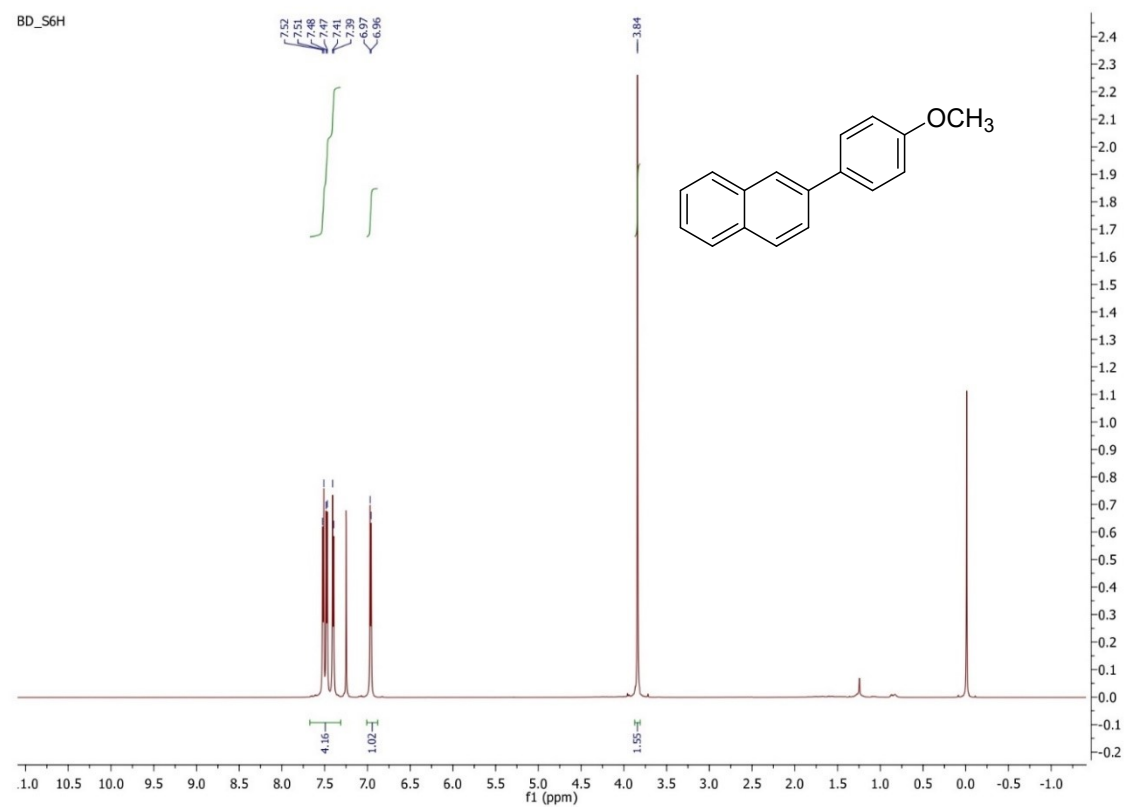
¹H NMR Spectrum of 1-(4'-methyl-[1,1'-biphenyl]-4-yl)ethan-1-one (10)



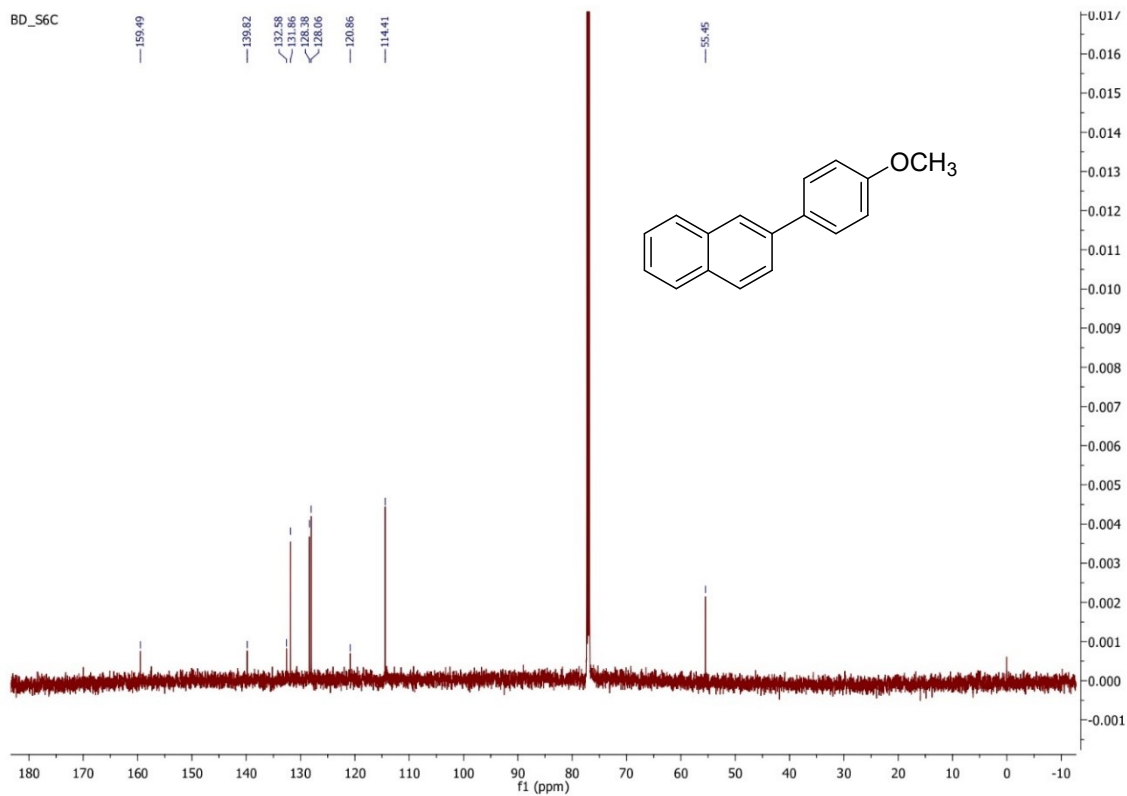
¹³C NMR Spectrum of 1-(4'-methyl-[1,1'-biphenyl]-4-yl)ethan-1-one (10)



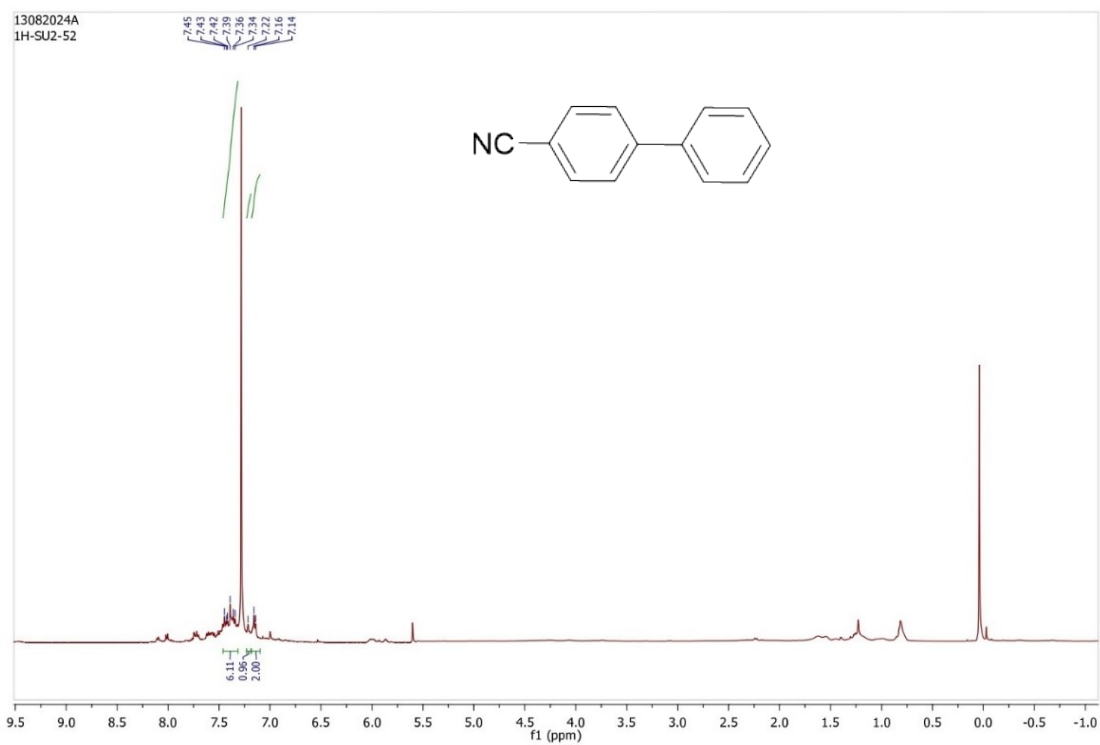
¹H NMR Spectrum of 2-(4-methoxyphenyl)naphthalene (9)



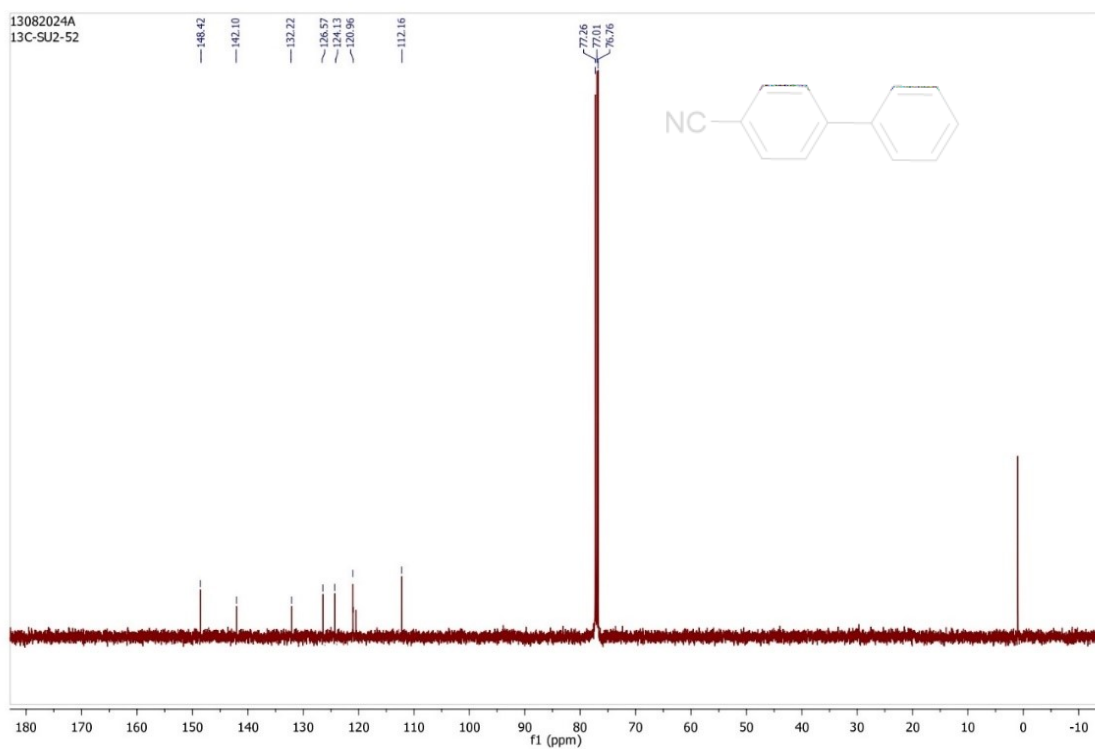
¹³C NMR Spectrum of 2-(4-methoxyphenyl)naphthalene (9)



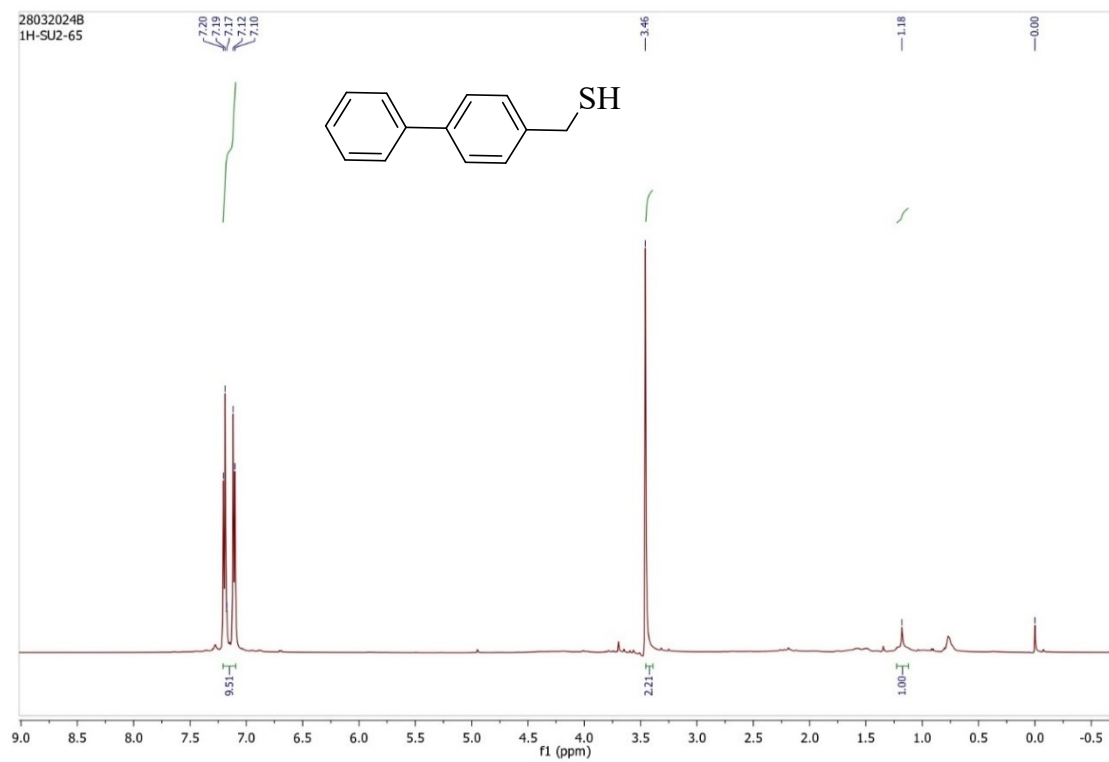
¹H NMR Spectrum of 1,1'-biphenyl]-4-carbonitrile (16)



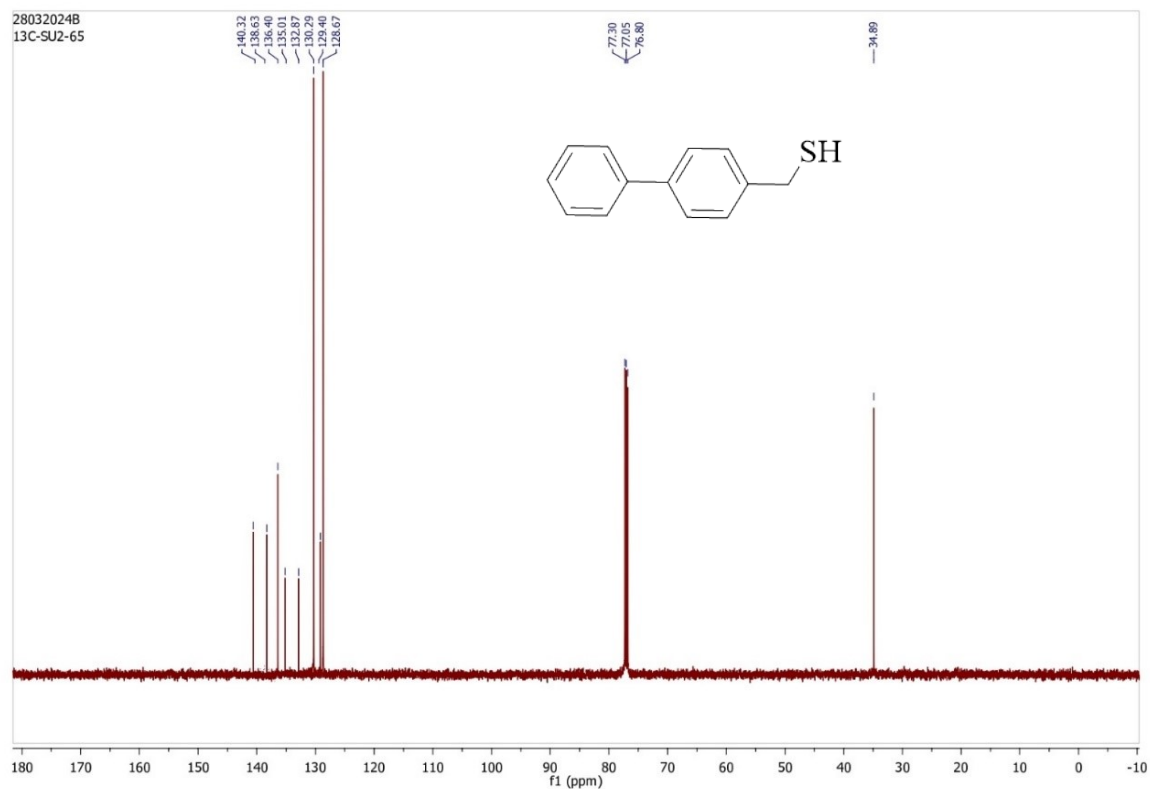
^{13}C NMR Spectrum of 1,1'-biphenyl-4-carbonitrile (16)



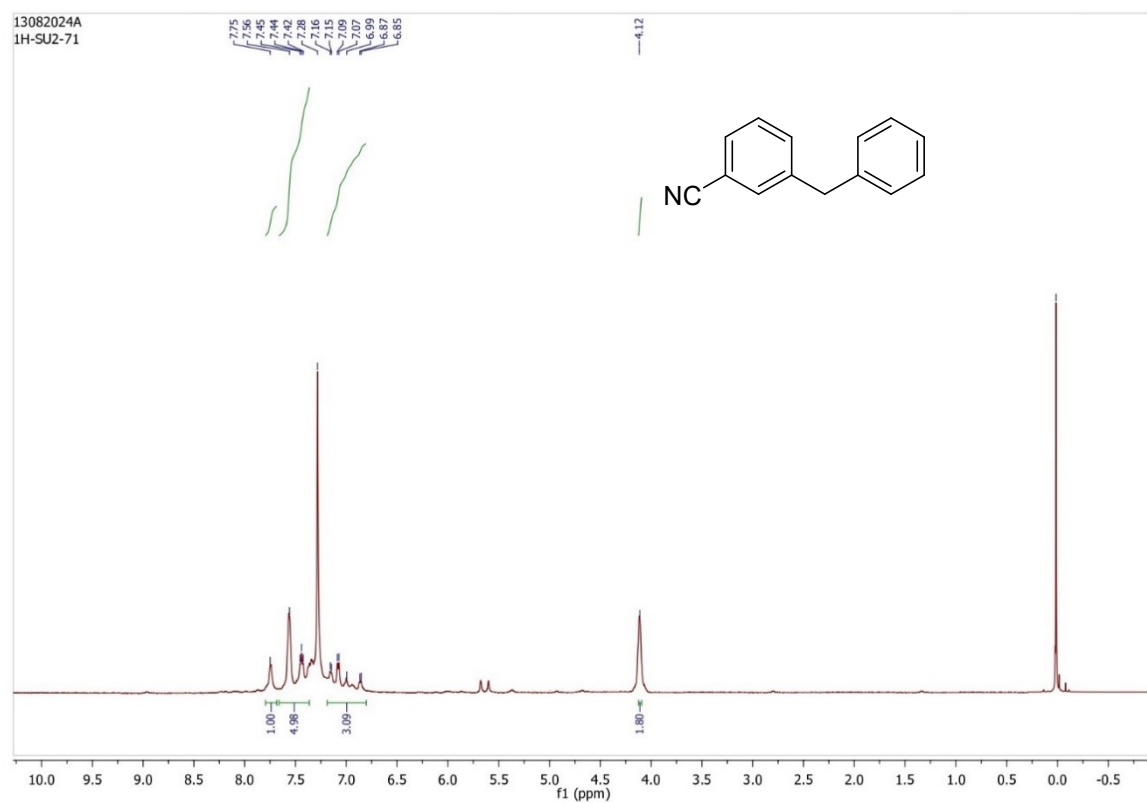
^1H NMR Spectrum of 1,1'-biphenyl-4-ylmethanethiol (14)



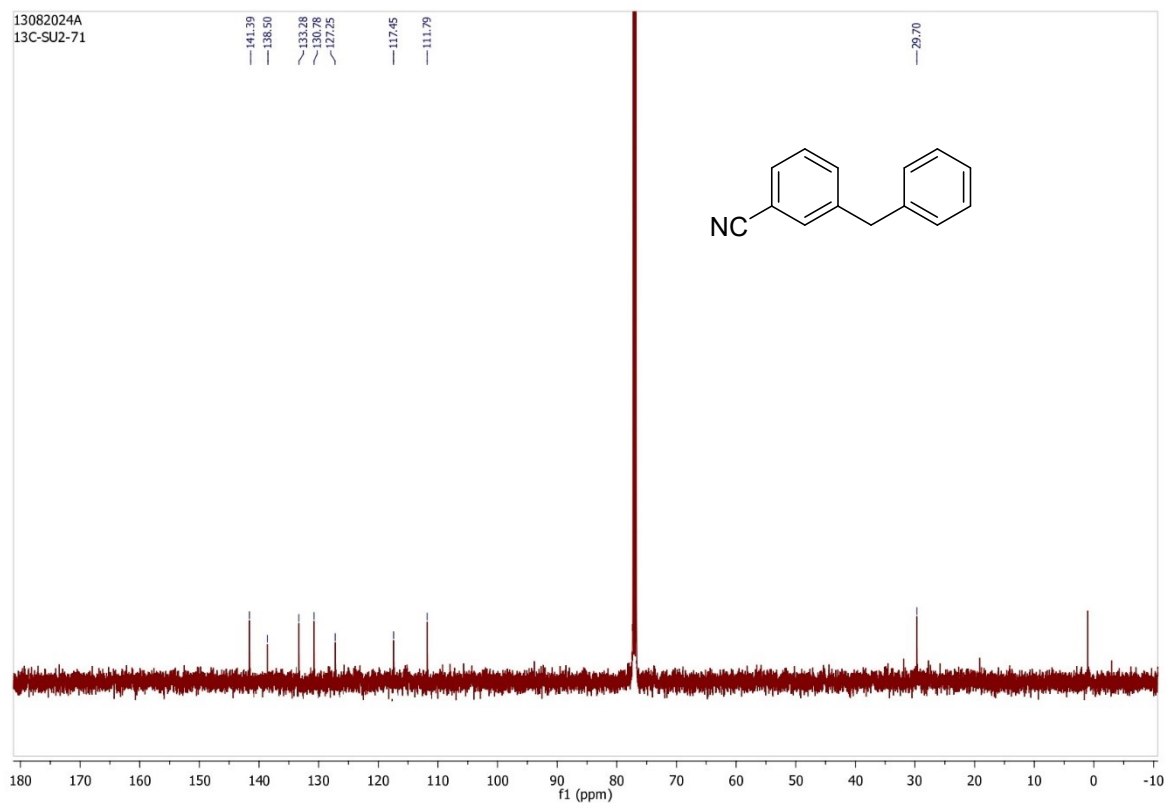
¹³C NMR Spectrum of 1,1'-biphenyl-4-ylmethanethiol (14)



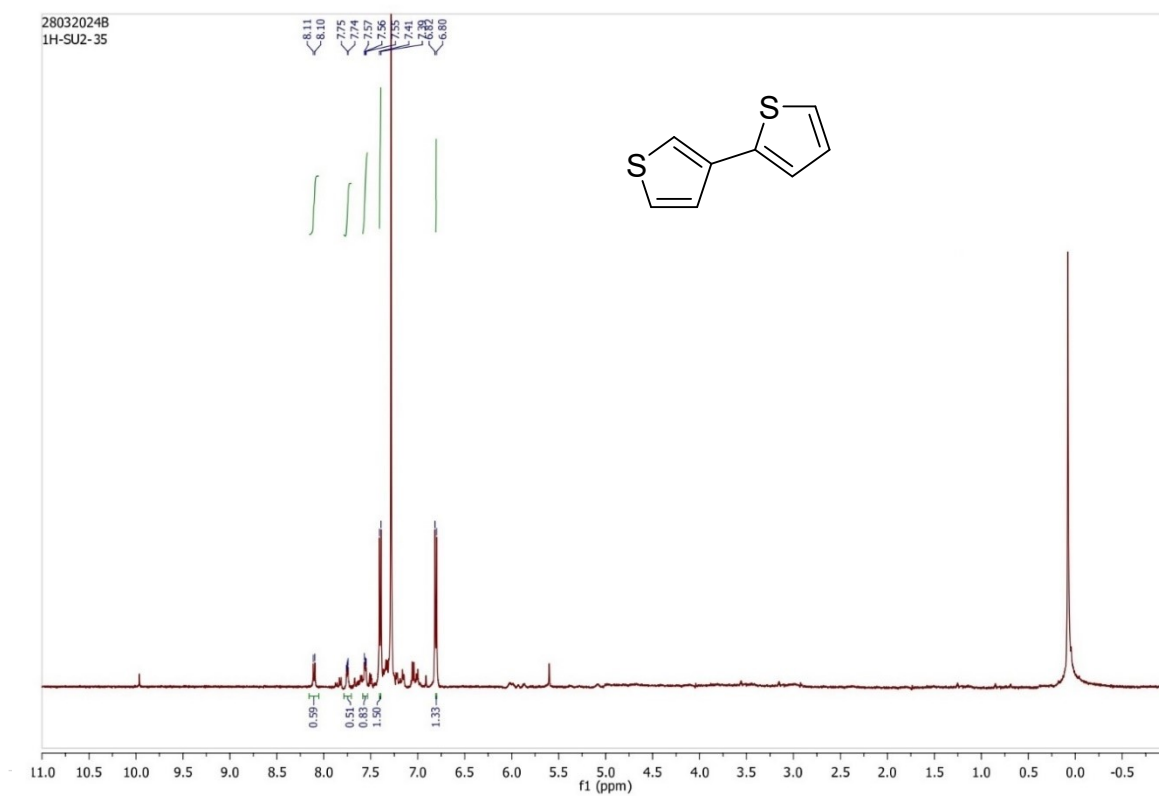
¹H NMR Spectrum of 3-benzylbenzonitrile (13)



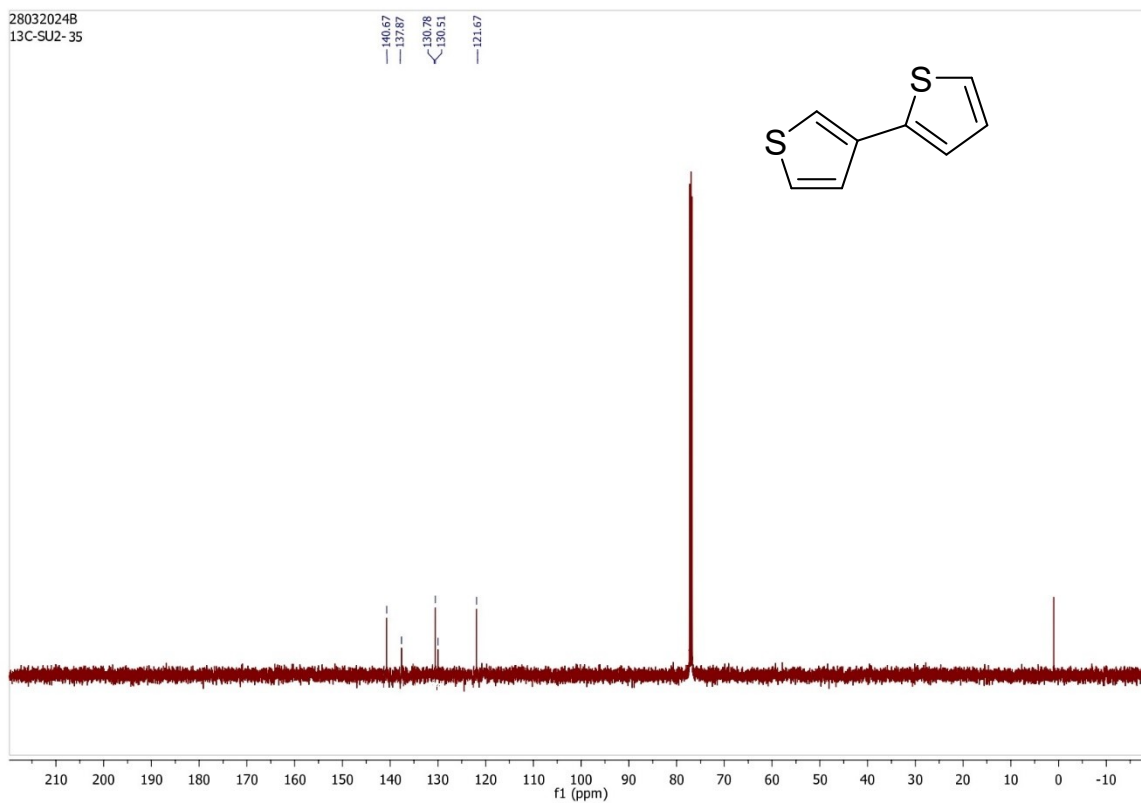
¹³C NMR Spectrum of 3-benzylbenzonitrile (13)



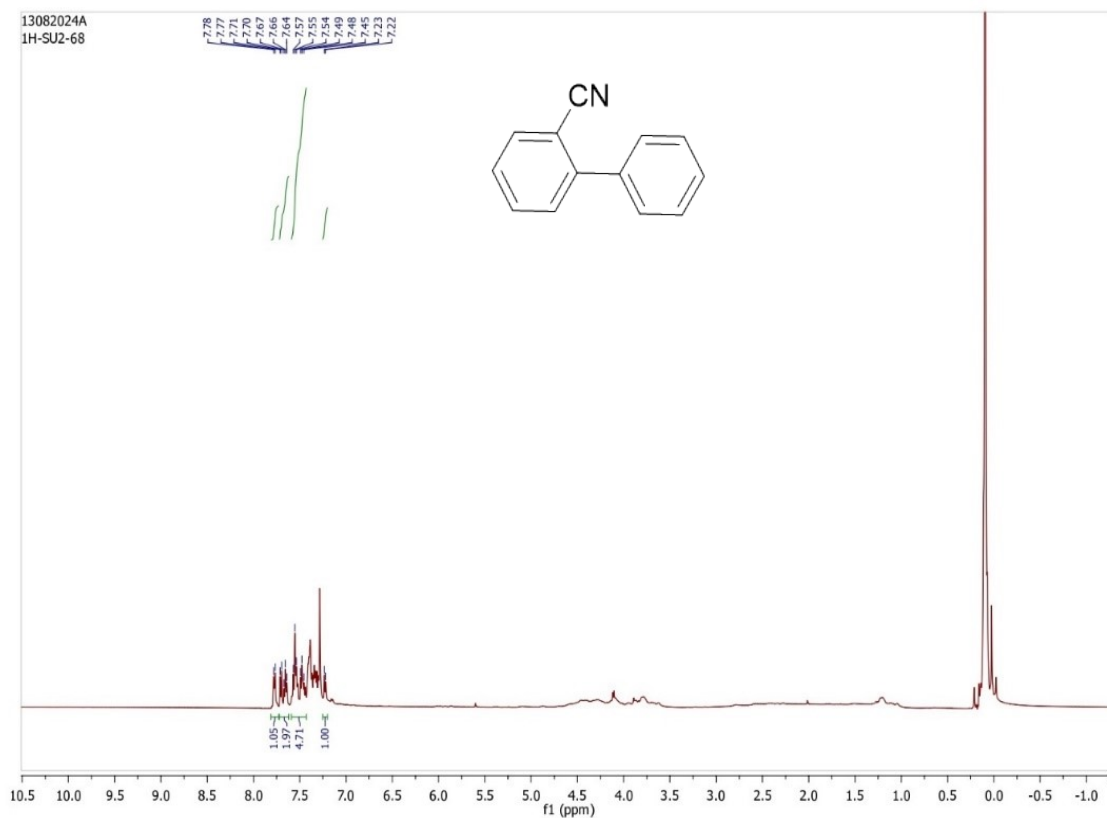
¹H NMR Spectrum of 2,3'-bithiophene (17)



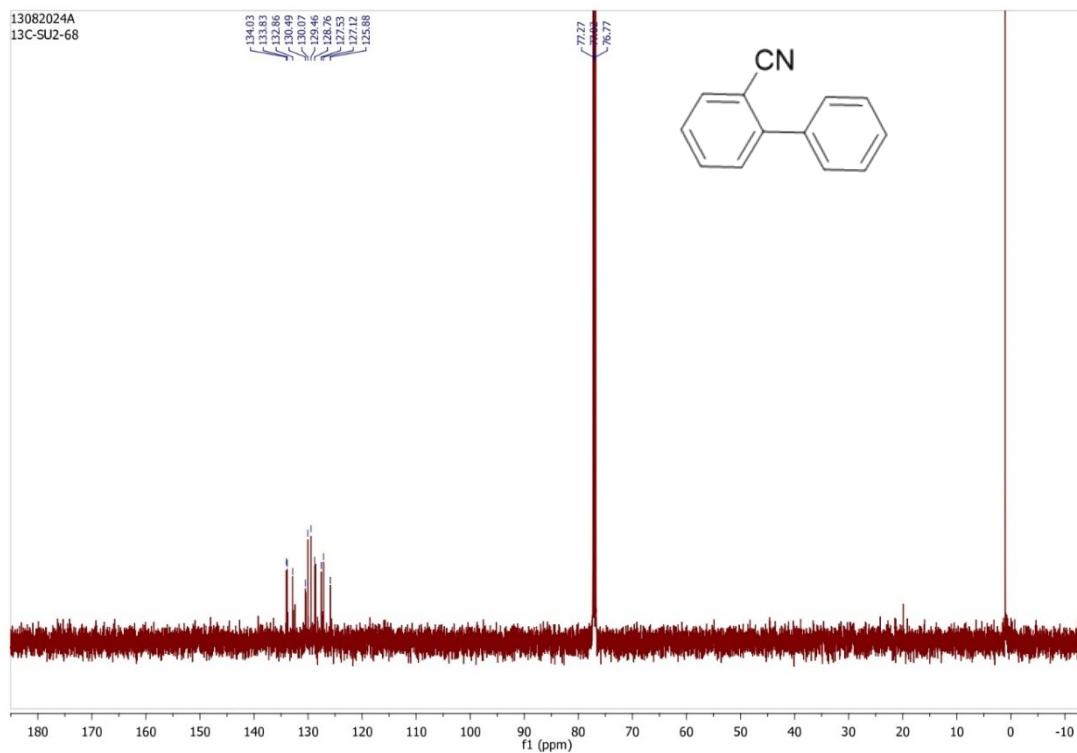
¹³C NMR Spectrum of 2,3'-bithiophene (17)



¹H NMR Spectrum of 1,1'-biphenyl-2-carbonitrile (11)



¹³C NMR Spectrum of 1,1'-biphenyl-2-carbonitrile (11)



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