

Electronic Supplementary Information

Synthesis of Low-Cost Multi-element Pt Base Alloy Nanoparticles as Catalysts for Oxygen Reduction Reaction

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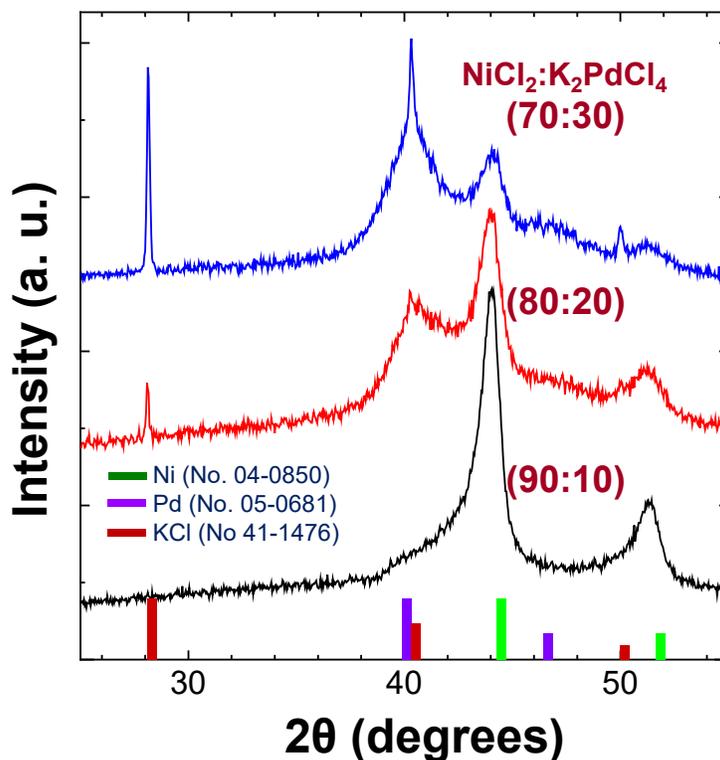


Fig. S1 XRD patterns of Ni-Pd NP samples synthesized using NiCl_2 and K_2PdCl_4 metal precursors at different molar ratios.

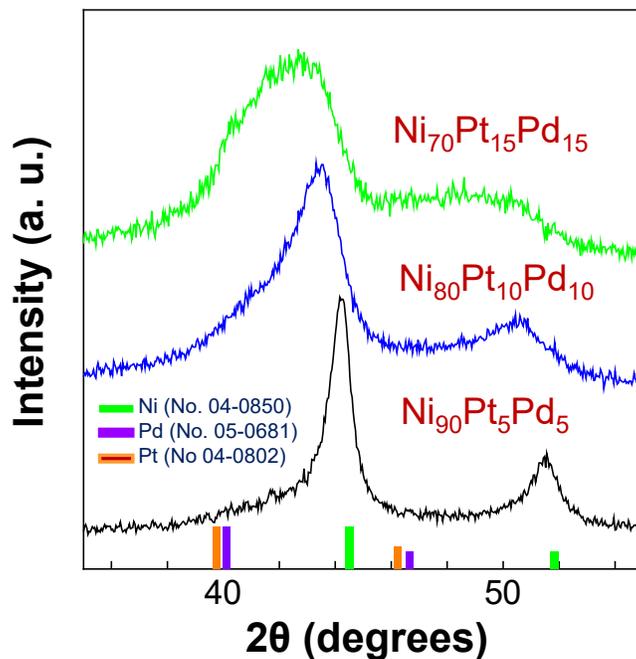


Fig. S2 XRD patterns of $\text{Ni}_{70}\text{Pt}_{15}\text{Pd}_{15}$, $\text{Ni}_{80}\text{Pt}_{10}\text{Pd}_{10}$, and $\text{Ni}_{90}\text{Pt}_5\text{Pd}_5$ NPs samples synthesized in 1-heptanol using $\text{Ni}(\text{OAc})_2$, H_2PtCl_6 and K_2PdCl_4 as metal precursors at different molar ratios.

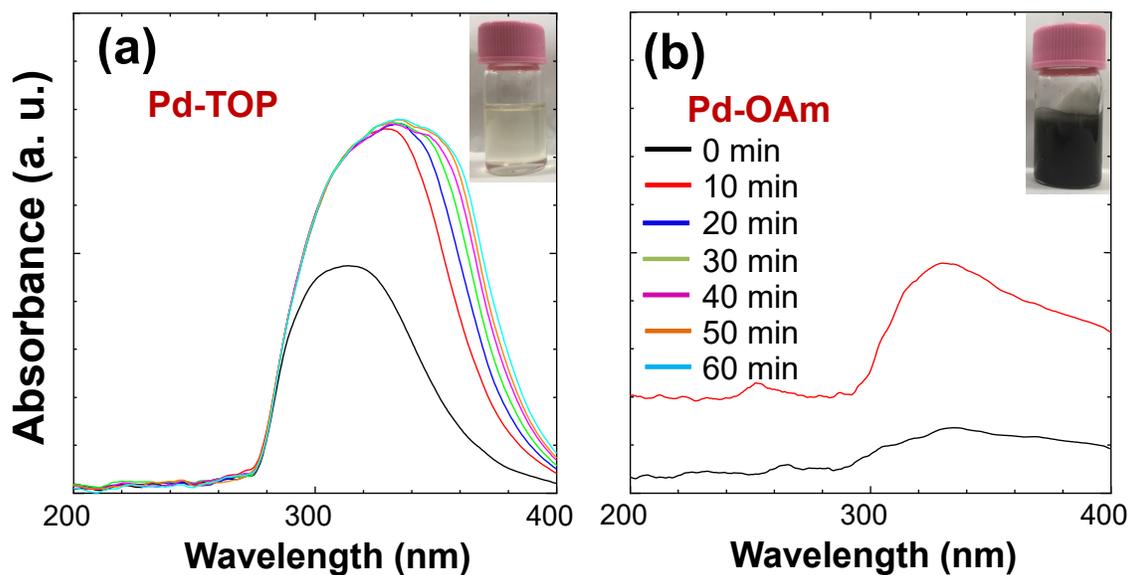
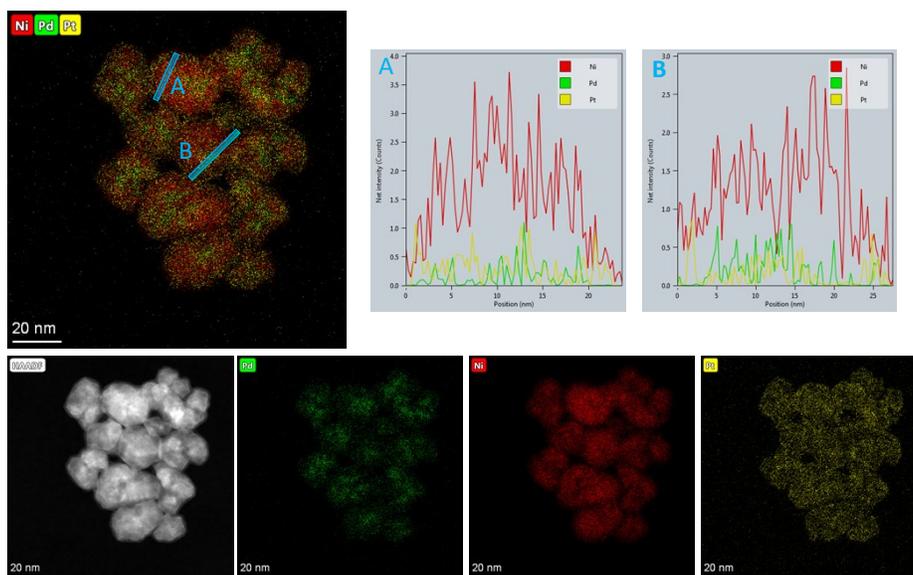


Fig. S3 In situ UV-vis spectroscopic measurements during synthesis of Pd NPs in 1-heptanol at $173\text{ }^\circ\text{C}$ using (a) TOP and (b) OAm as surfactants.

(a) Pt-rich shell NiPtPd NPs



(b) Pd-rich shell NiPtPd NPs

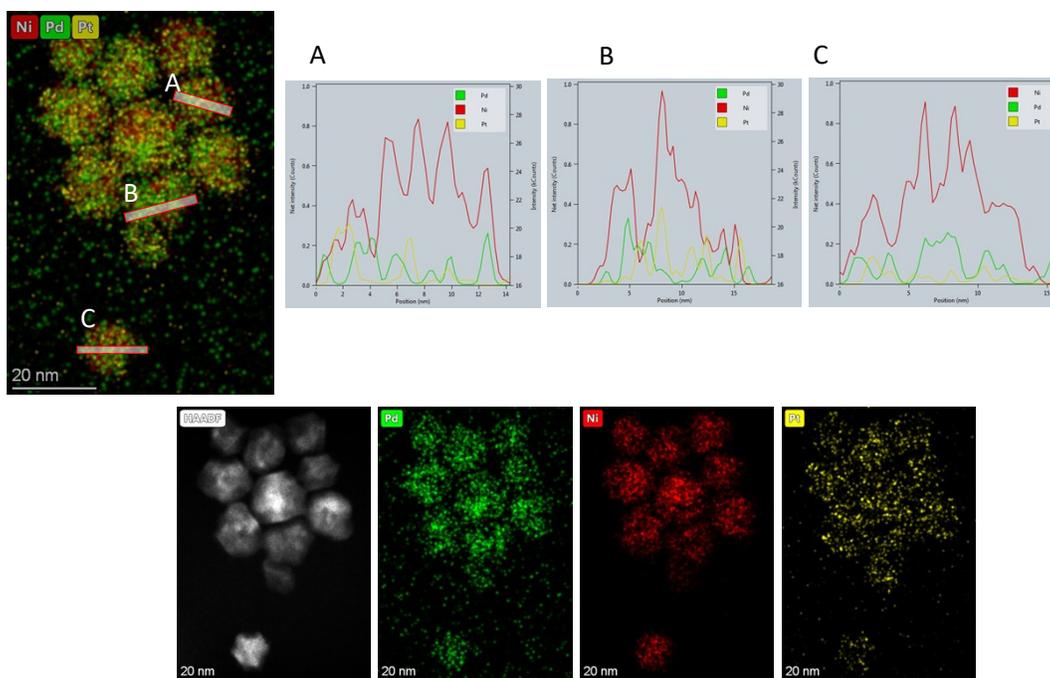
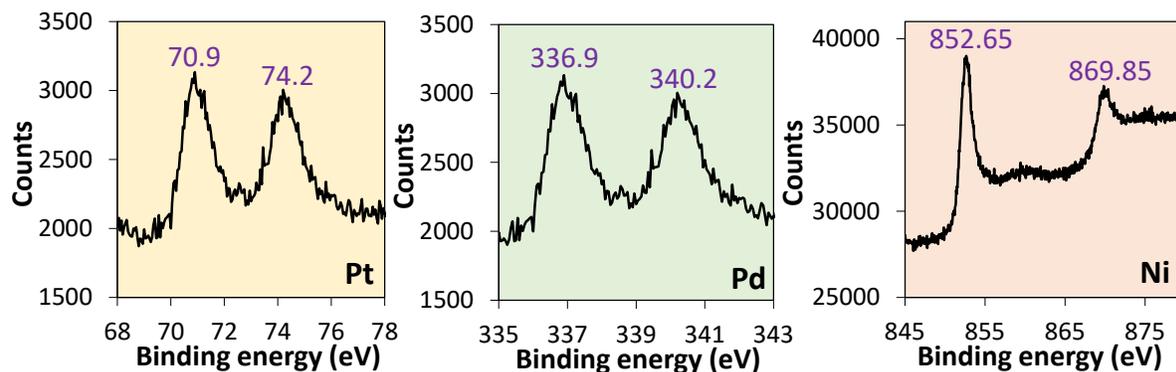


Fig. S4 Elemental and line mapping of synthesized samples with controlled OAm/Pt and TOP/Pd molar ratio of (a) 1.4 and 1 and (b) 1.4 and 3, respectively.

(A) Pt-rich shell



(B) Pd-rich shell

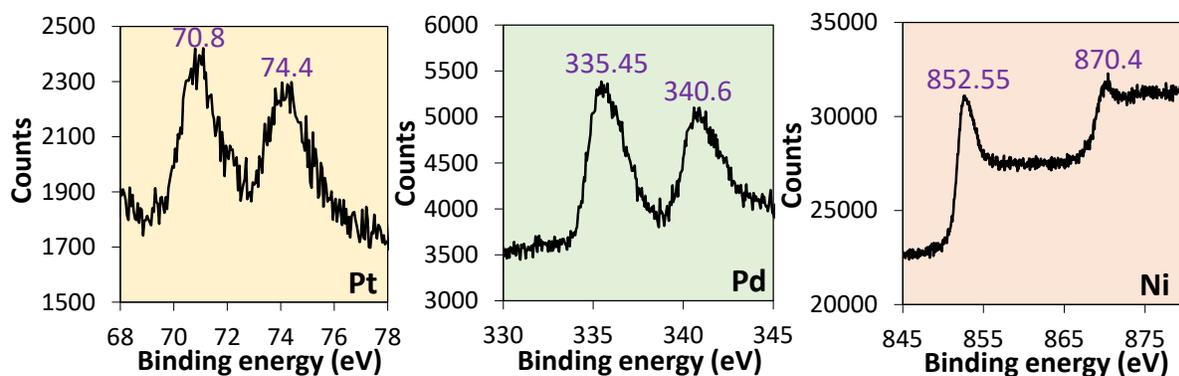


Fig. S5 XPS spectra in the regions of Pt 4f, Pd 3d and Ni 2p for (A) Pt-rich and (B) Pd-rich shell of NiPtPd nanoparticles.