

## Supporting Information

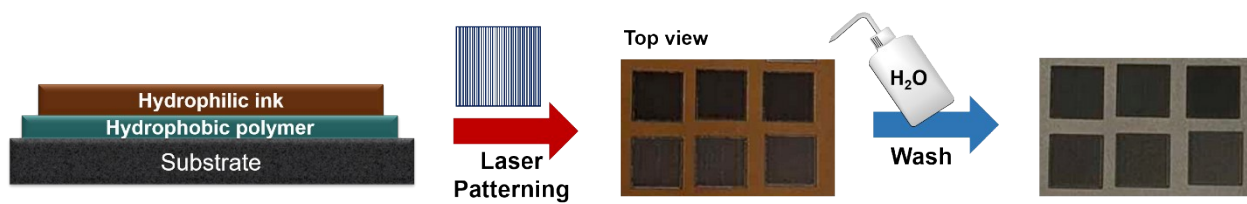
### **Sustainable design of high-performance multifunctional carbon electrodes by one-step laser carbonization for supercapacitors and dopamine sensors**

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**Figure S1.** Schematic images showing the process of LP-C fabrication.

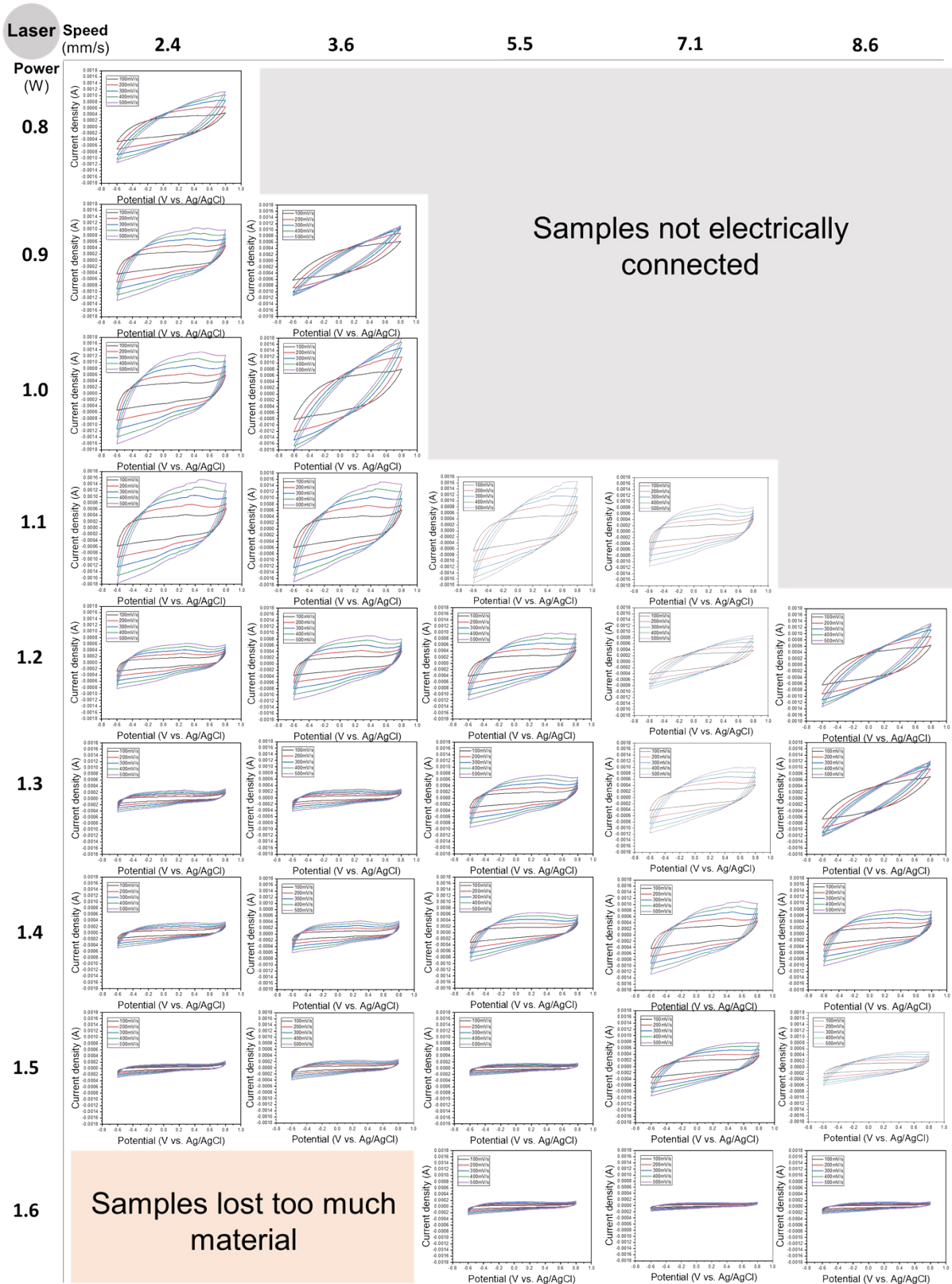
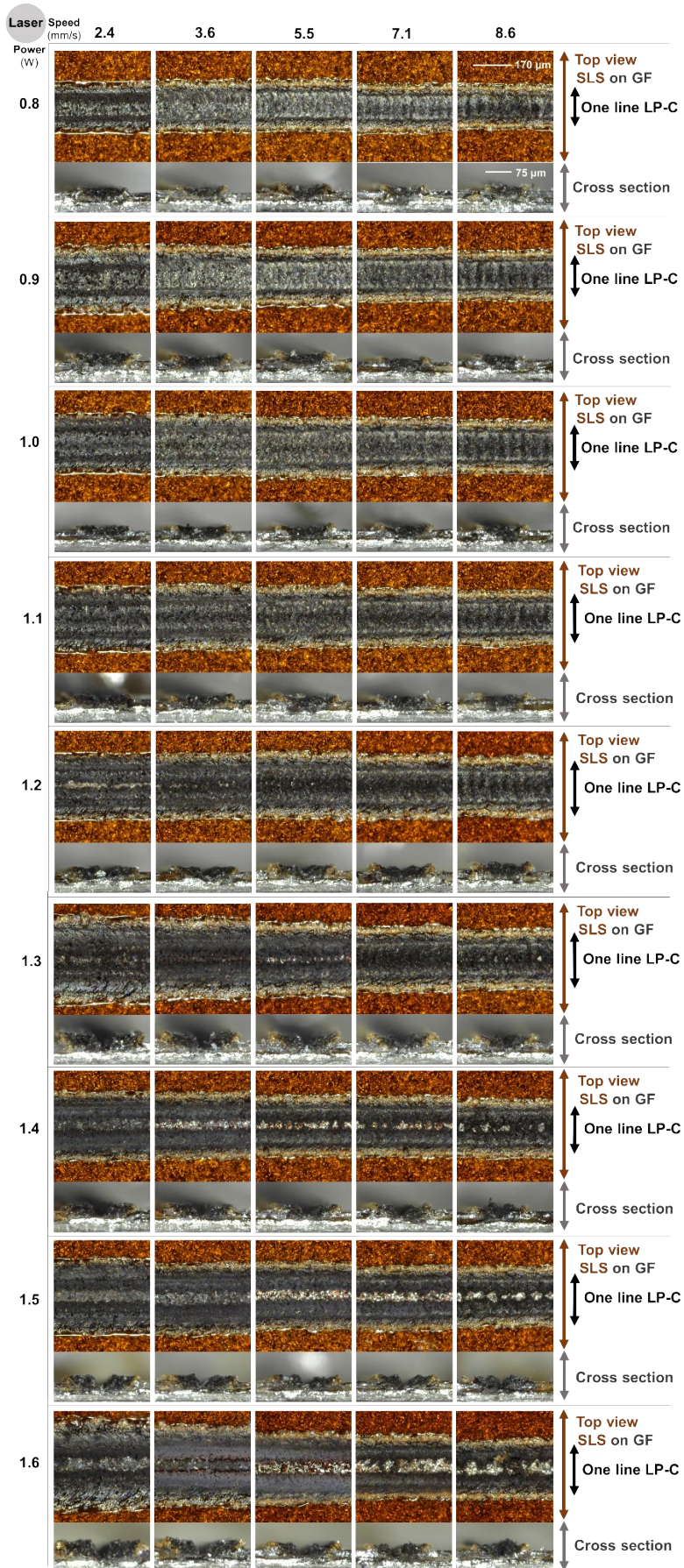
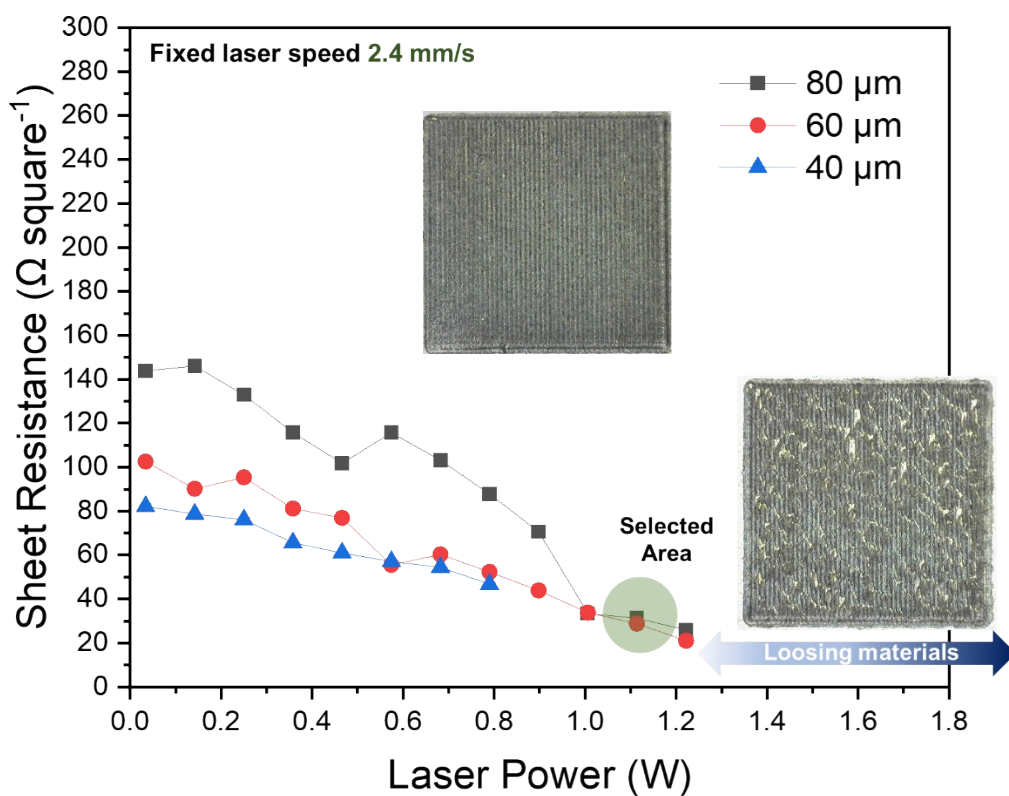


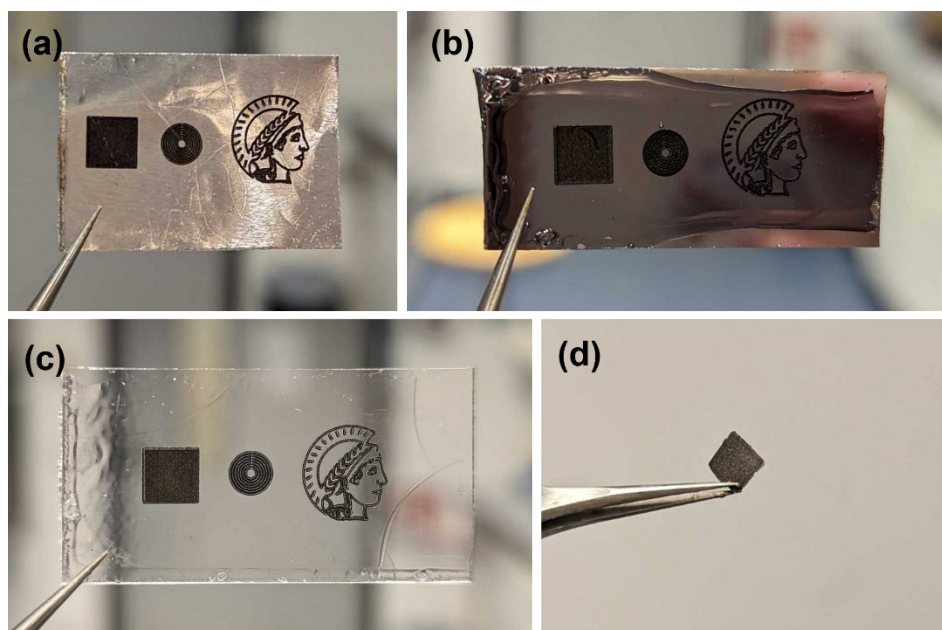
Figure S2. Cyclic voltammetry (CV) curves with varying laser power and speed.



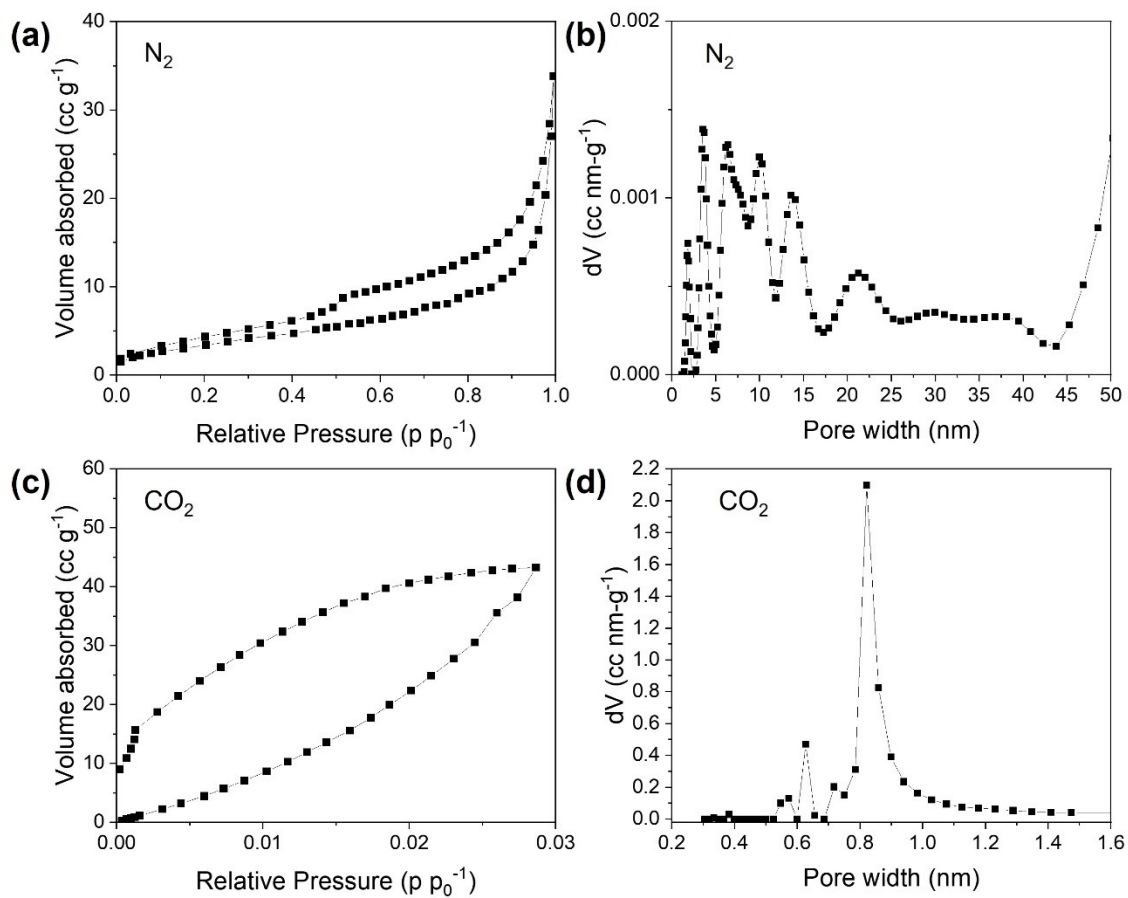
**Figure S3.** Optical microscope images of a laser-carbonized line obtained in top view and cross-section view with varying laser parameters.



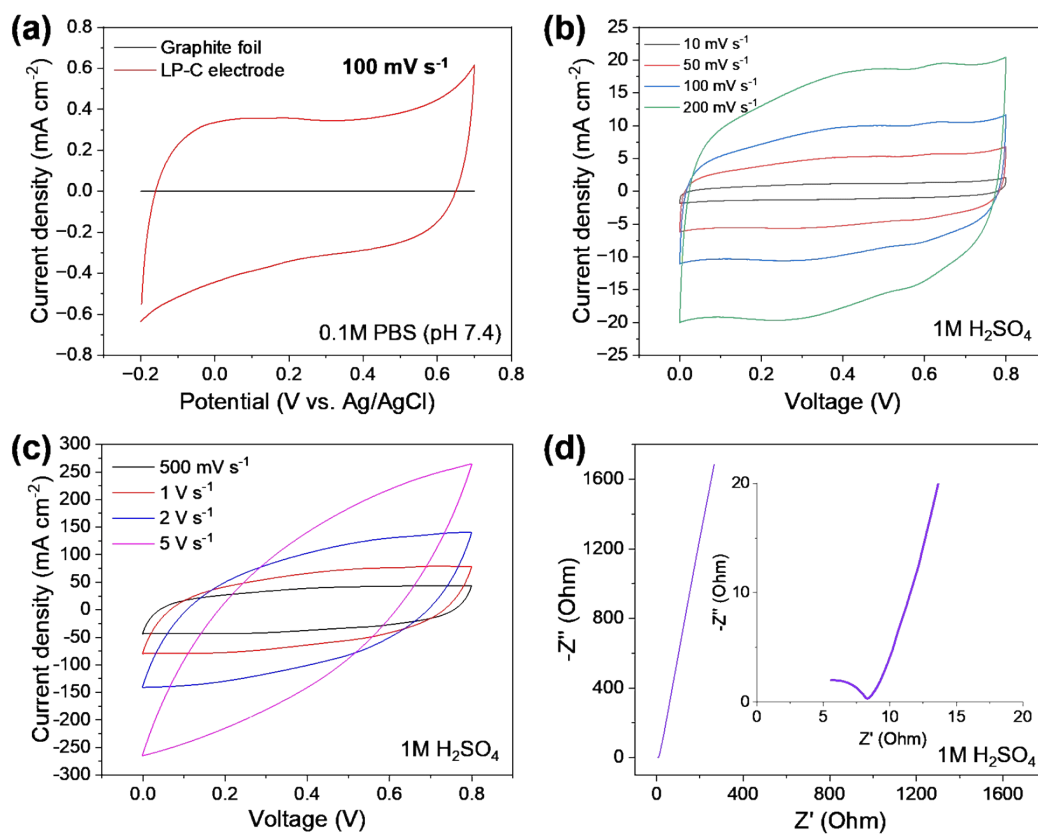
**Figure S4.** Sheet resistance as a function of sodium lignosulfonate precursor film thickness and laser power at the fixed laser speed ( $2.4 \text{ mm s}^{-1}$ ).



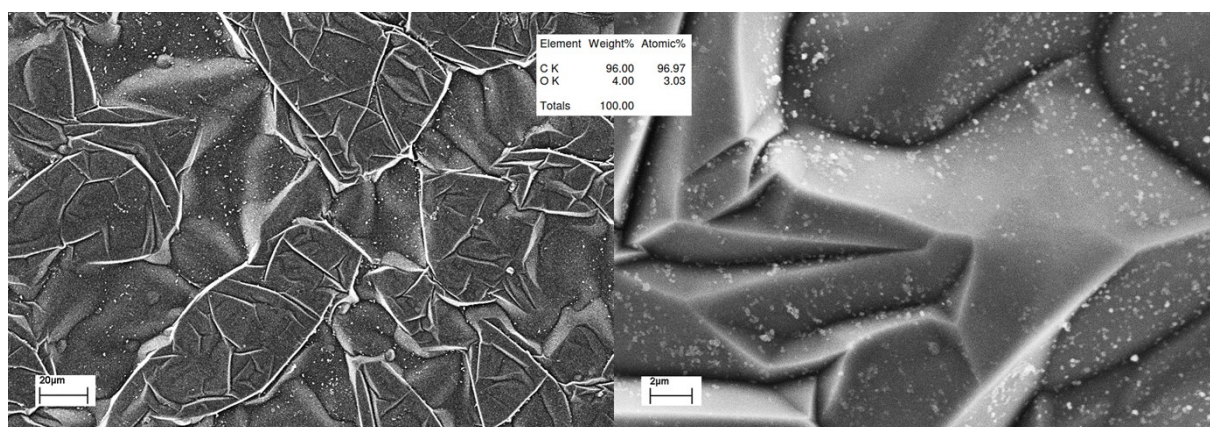
**Figure S5.** LP-C on (a) Al foil, (b) Si wafer, and (c) glass substrates with different laser patterns; (d) free-standing LP-C.



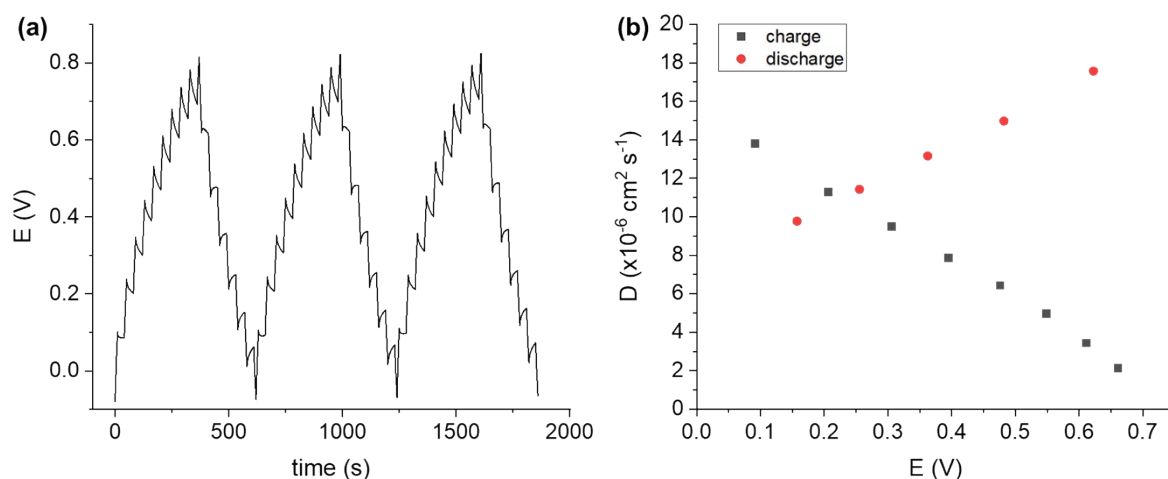
**Figure S6.** (a) N<sub>2</sub> adsorption–desorption isotherms and (b) pore size distribution of LP-C powder (N<sub>2</sub>); (c) CO<sub>2</sub> adsorption–desorption isotherms and (d) pore size distribution of LP-C powder (CO<sub>2</sub>).



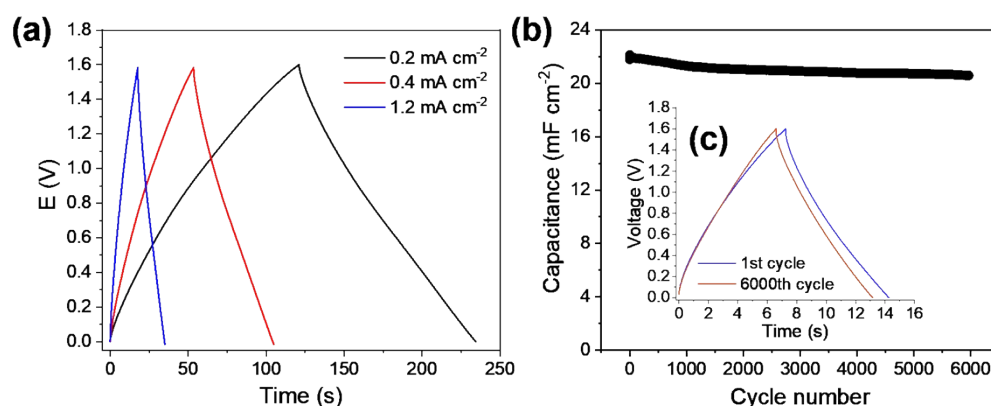
**Figure S7.** Cyclic voltammetry (CV) curves of LP-C measured with a 3-electrode system (a) in 0.1M PBS (pH 7.4) at 100 mV s<sup>-1</sup>, (b, c) in 1M H<sub>2</sub>SO<sub>4</sub> as a function of voltage scan rate and (d) Nyquist plot.



**Figure S8.** Scanning electron microscope (SEM) images of bare graphite foil.



**Figure S9.** (a) Charge and discharge curves of the galvanostatic intermittent titration technique (GITT) test and (b) diffusion coefficient ( $D$ ) values as a function of potential tested by cycling a symmetrical cell at  $50 \text{ mA g}^{-1}$  for 10 sec followed by an open circuit relaxation period of 30 s.

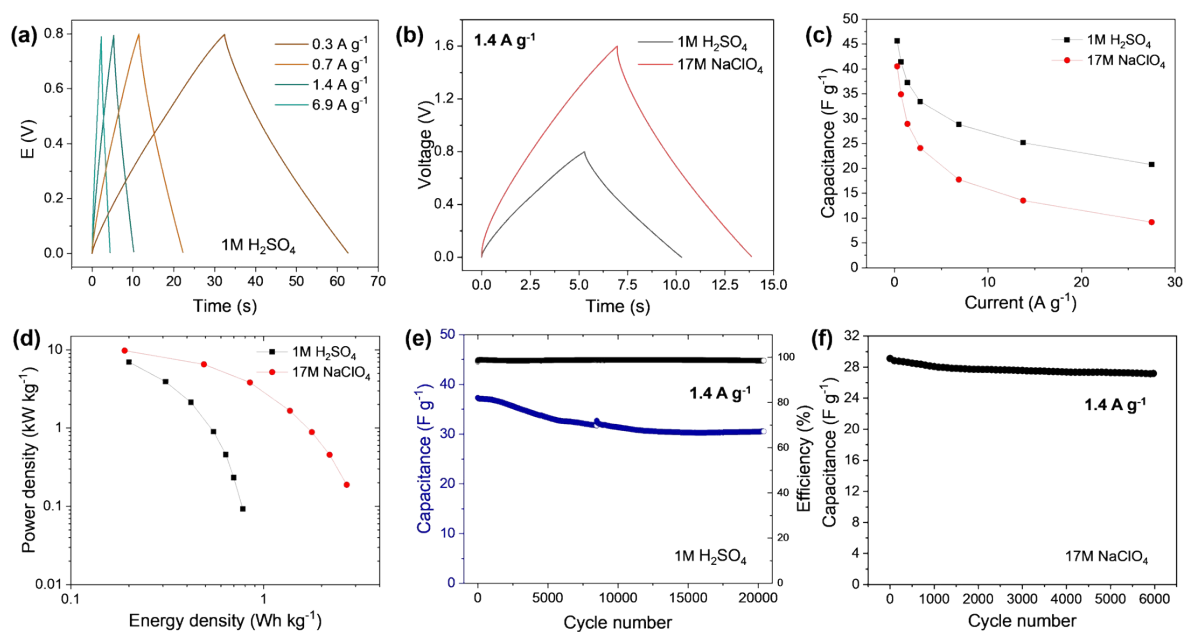


**Figure S10.** Galvanostatic charge-discharge (GCD) profiles in  $17\text{M NaClO}_4$  (a) with different current densities, (b) GCD cyclability test showing the retention of specific areal capacitance at  $2.2 \text{ mA cm}^{-2}$  and (c) GCD profiles of the 1<sup>st</sup> and 6000<sup>th</sup> cycles.

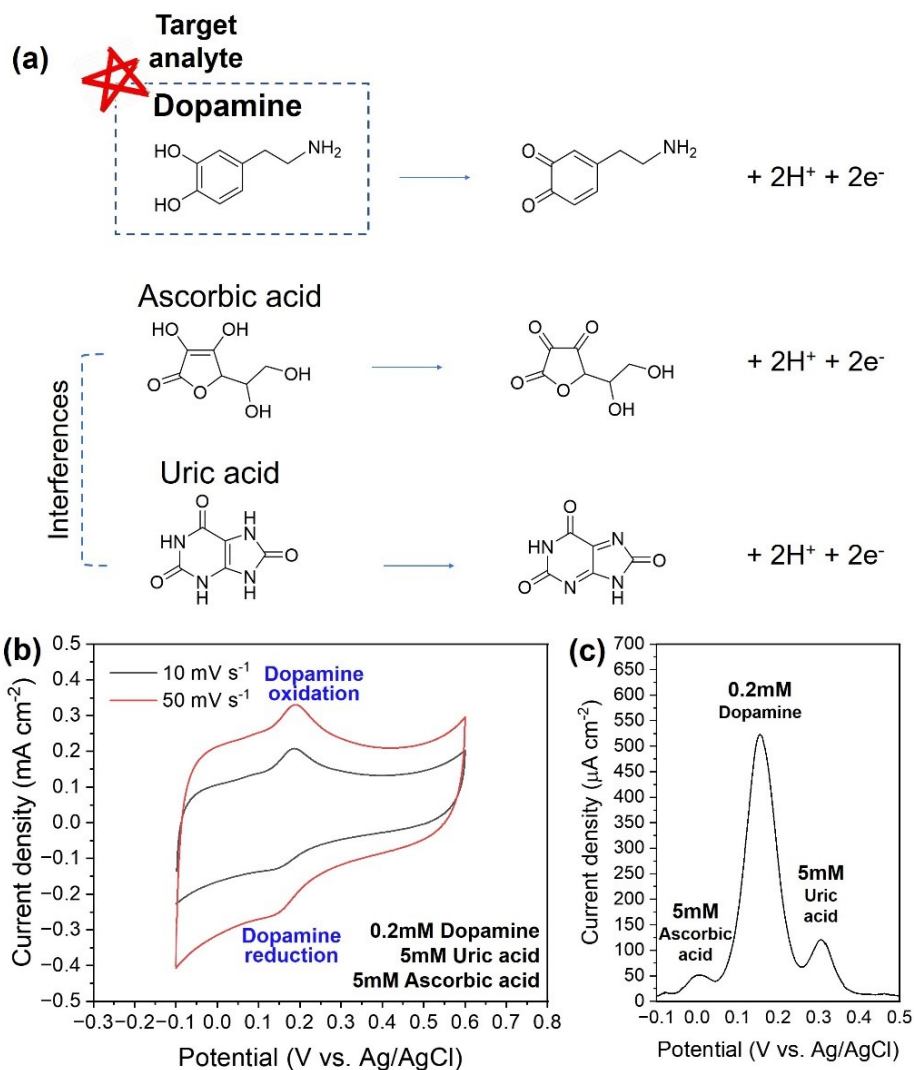
Ref	Precursor	Laser $\lambda$ ( $\mu\text{m}$ )	BET ( $\text{m}^2 \text{g}^{-1}$ )	Current collector	Electrolyte	Capacitance ( $\text{mF cm}^{-2}$ )	Energy Density ( $\mu\text{Wh cm}^{-2}$ )	Power Density ( $\text{mW cm}^{-2}$ )	Retention / cycles / Current density ( $\text{mA cm}^{-2}$ )	Current density range ( $\text{mA cm}^{-2}$ )
This work	Sodium lignosulfonate	10.6	13 ( $\text{N}_2$ ), 504 ( $\text{CO}_2$ )	Graphite foil	$\text{H}_2\text{SO}_4$	38.8	1.3 (1.7)	11.4 (14.9)	81% / 20k / 2.2	0.44 - 44
					$17\text{M NaClO}_4$	32.8	4.3 (5.8)	16 (21.7)	93% / 6k / 2.2	
35	Kraft Lignin/PVA	10.6	338.3 ( $\text{N}_2$ )	Copper strip, Silver paint	$\text{H}_2\text{SO}_4/\text{PVA}$	17	2.6	2.2	99.2% / 12k / 2	0.05 - 2
36	Lignin/PEO	10.6	-	Copper tape, Silver paint	$\text{H}_2\text{SO}_4/\text{PVA}$	25.4	3.5	0.8	93% / 4.5k / 0.2	0.1 - 1
37	Kraft Lignin/PEO	10.6	-	Copper tape, Silver paint	$\text{H}_2\text{SO}_4/\text{PVA}$	2.5	0.3	0.03	91% / 10k / 0.02	0.01 - 0.02
38	Agglomerate d Cork	1.06	-	-	$\text{H}_2\text{SO}_4/\text{PVA}$	1.4	0.1	0.08	106% / 10k / 0.05	0.05 - 0.1
39	Nageia fleuryi leaves	0.346	-	Copper foil, Silver paste	$\text{H}_2\text{SO}_4/\text{PVA}$	35.3	1.2	0.3	99% / 50k / -	0.005

**Table S1.** Comparison of the performance and material properties of laser-carbonized supercapacitors using biomass-derived precursors.

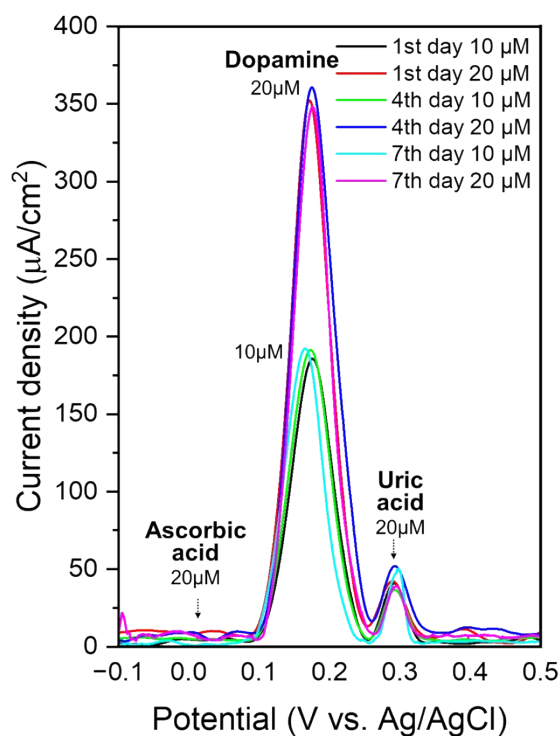




**Figure S11.** Gravimetric supercapacitor performance of LP-C electrodes using a symmetrical cell: (a) Schematic image of the EDLC charge-discharge process; (b) Galvanostatic charge and discharge (GCD) profiles in 1M H<sub>2</sub>SO<sub>4</sub> at different current densities and (c) at a current density of 1.4 A g<sup>-1</sup> for different electrolytes; (d) Specific gravimetric capacitance as a function of current density; (e) Ragone plot; (f) Long-term cyclability at a current density of 1.4 A g<sup>-1</sup> showing the retention of specific areal capacitance and coulombic efficiency.



**Figure S12.** (a) Oxidation process of dopamine (DA), ascorbic acid (AA) and uric acid (UA), (b) cyclic voltammetry (CV) and (c) differential pulse voltammetry (DPV) curves of DA, AA and UA ternary mixtures



**Figure S13.** (a) Differential pulse voltammetry (DPV) detection curves of ternary mixtures (dopamine (10 and 20  $\mu\text{M}$ ), uric acid (20  $\mu\text{M}$ ) and ascorbic acid (20  $\mu\text{M}$ )) on days 1, 4 and 7.

Ref	Precursor	Laser $\lambda$	Current collector	Electrolyte	LOD ( $\mu\text{M}$ ) Measured / Extrapolated	Sensitivity ( $\mu\text{A } \mu\text{M}^{-1} \text{cm}^{-2}$ )	Linear range	$R^2$	Detection technique
This work	Sodium lignosulfonate	10.6 $\mu\text{m}$	Graphite foil	0.1M PBS (pH 7.4)	0.1 / 0.5	13.38	0.1 – 20 $\mu\text{M}$	0.996	CV, DPV
49	Kraft lignin/Cellulose nanofibers	10.6 $\mu\text{m}$	Glassy carbon	PBS	5 / 3.4	4.39	5 – 40 $\mu\text{M}$	0.995	CV, DPV
50	Alkaline lignin/Ag NPs/PVA	800 nm	-	0.1M PBS (pH 7.4)	1 / 0.098	0.98	1 – 45 $\mu\text{M}$	0.99	CV, DPV

**Table S2.** Comparison of the performance and material properties of laser-carbonized electrochemical dopamine sensors using biomass-derived precursors.