

## Supporting Information

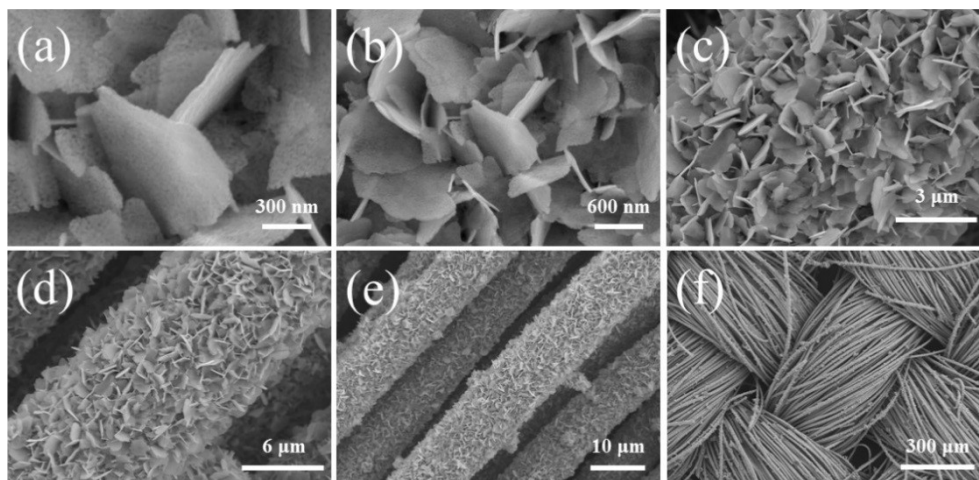
### **Ru doping and interface engineering synergistically boosting the electrocatalytic performance of WP/WP<sub>2</sub> nanosheet array for efficient hydrogen evolution reaction**

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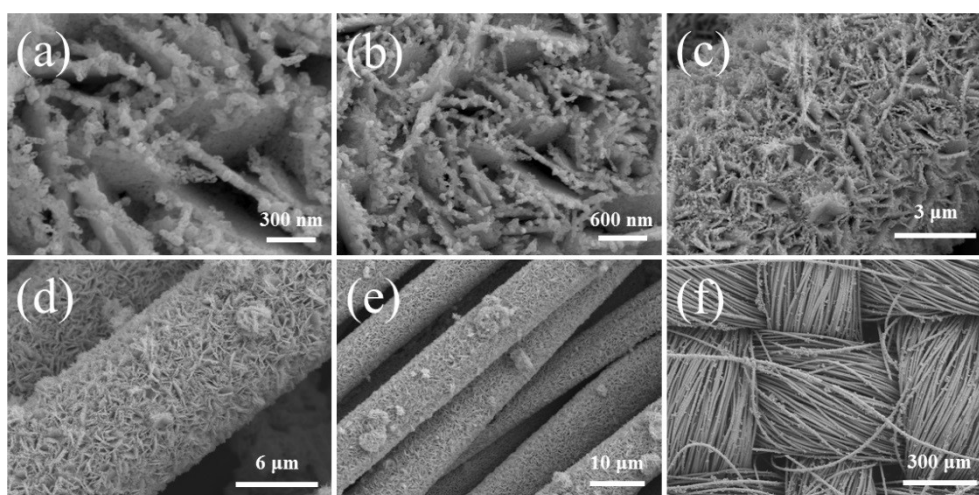
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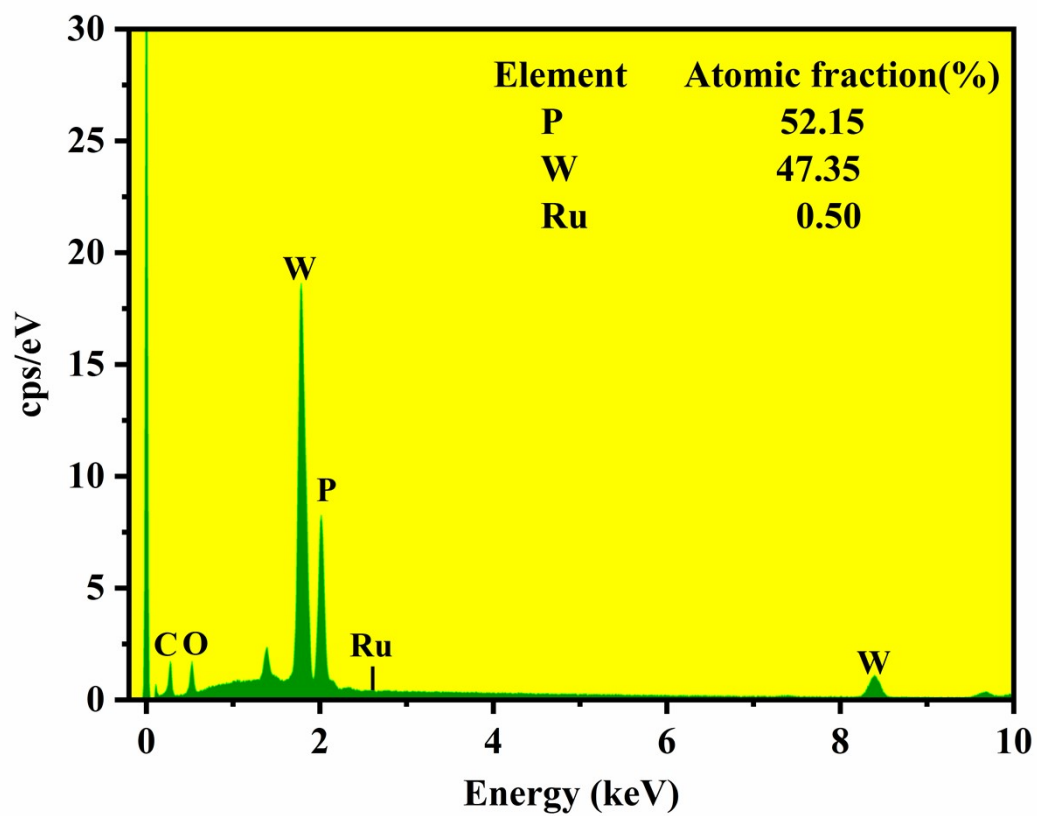
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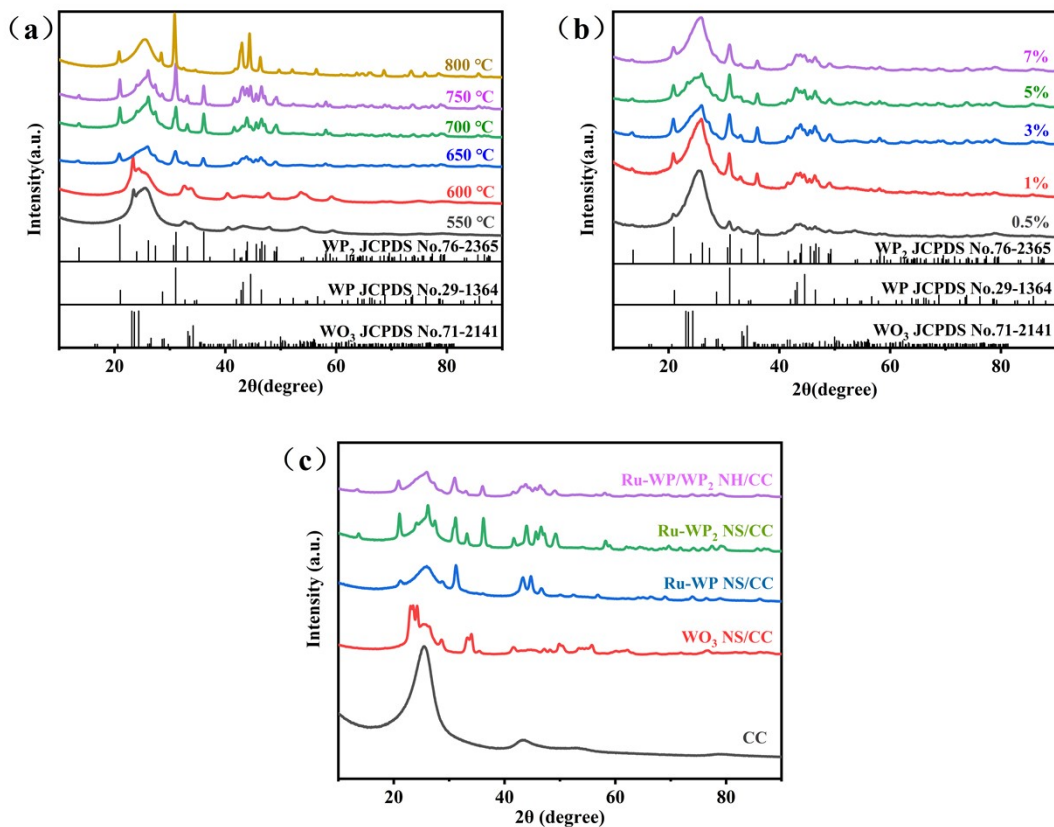
**Fig. S1.** SEM images of Ru-WO<sub>3</sub> NS/CC.



**Fig. S2.** SEM images of Ru-WP/WP<sub>2</sub> NH/CC.

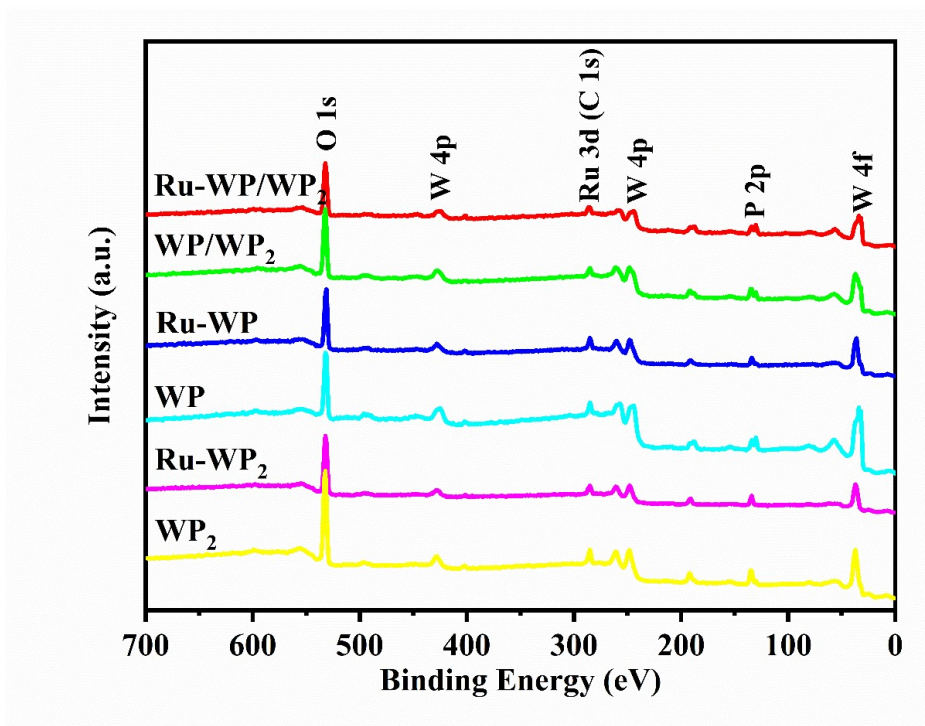


**Fig. S3.** EDS result of Ru-WP/WP<sub>2</sub> NH/CC.



**Fig. S4.** XRD pattern of Ru catalyst. (a) Temperature optimization gradient. (b)

Different Ru doping. (c) Comparison with bare carbon cloth.



**Fig. S5.** XPS spectra of Survey spectrum.

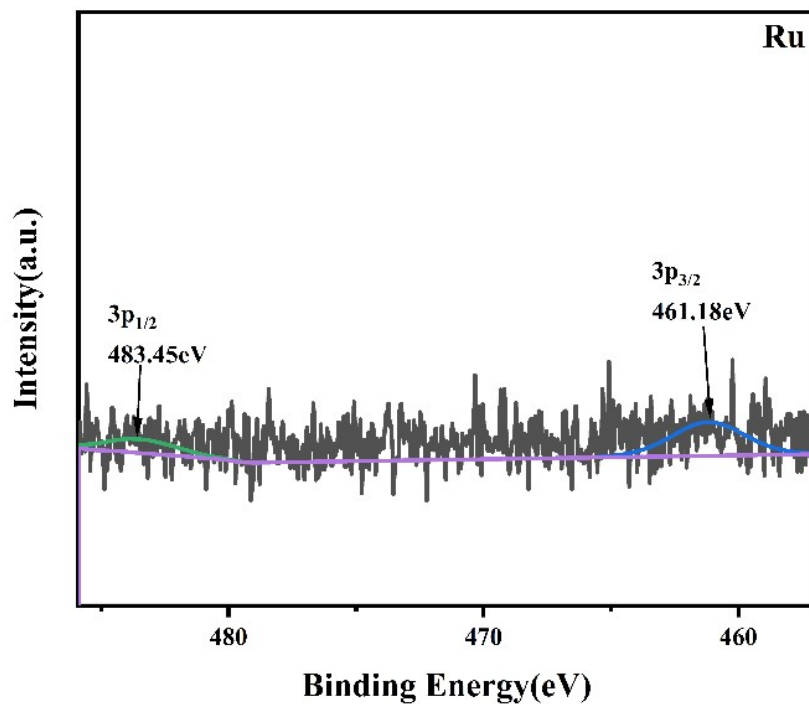


Fig. S6. XPS spectra of Ru 3p

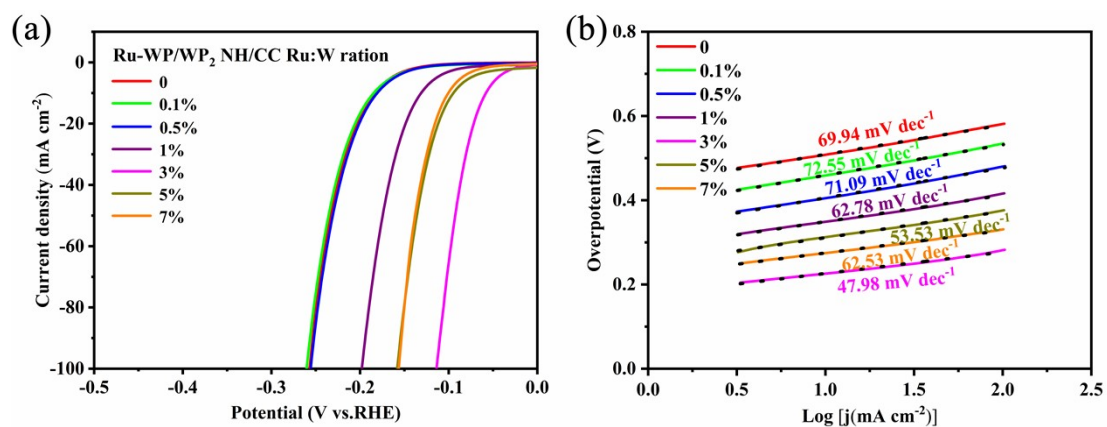
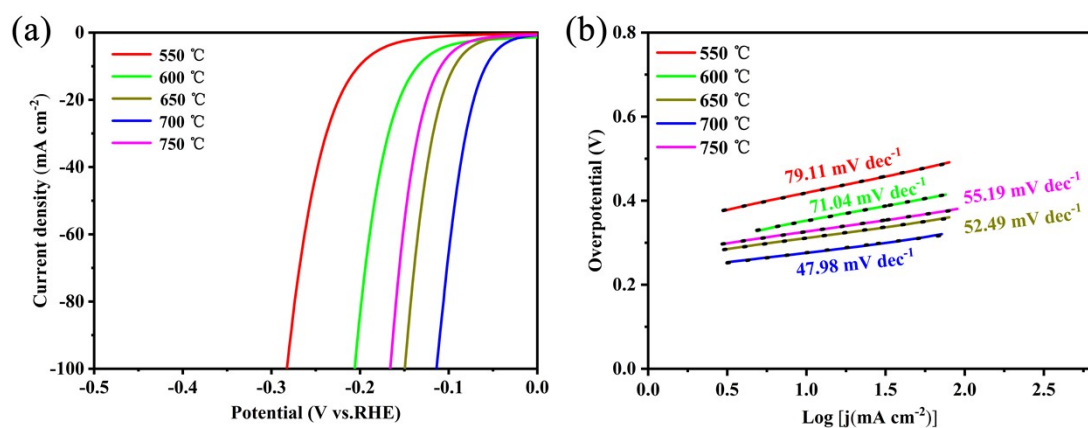
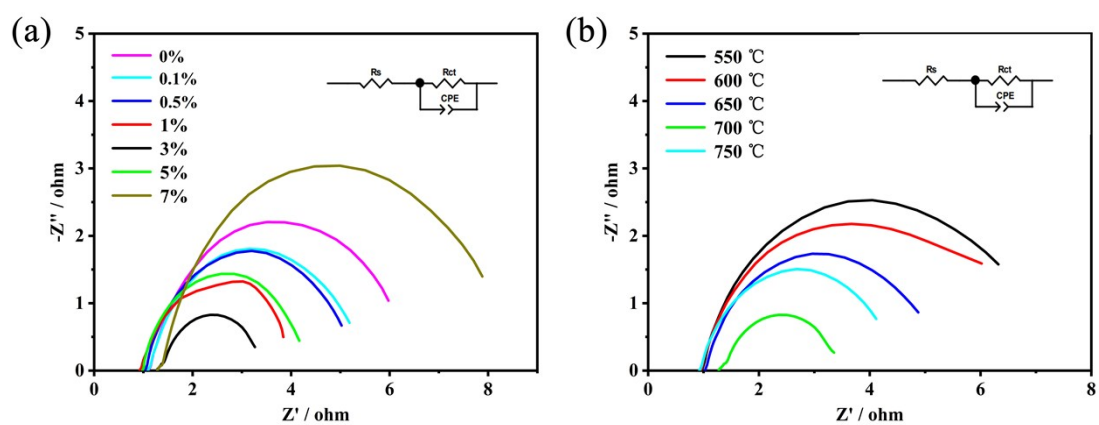


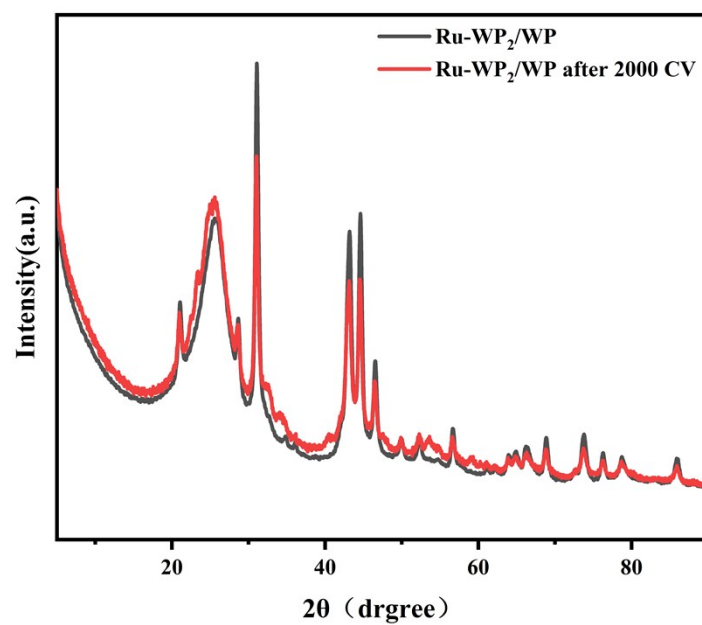
Fig. S7. The Ru: W ratio added to the precursor sample was optimized. (a) Linear sweep voltammetry of the catalyst at a scanning rate of  $5 \text{ mV s}^{-1}$ , and (b) Tafel slope.



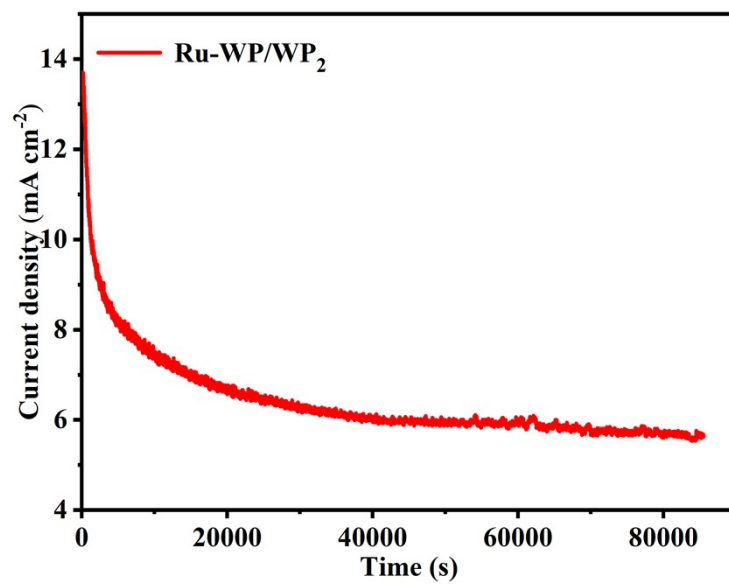
**Fig. S8.** Temperature gradient optimization of catalyst. (a) Linear sweep voltammetry of the catalyst at a scanning rate of  $5 \text{ mV s}^{-1}$  and (b) Tafel slope.



**Fig. S9.** Nyquist curves for (a) different Ru doping concentrations, and (b) different phosphating temperatures.



**Fig. S10.** XRD comparison of Ru-WP<sub>2</sub>/WP before and after 2000 cycles CV



**Fig. S11.** The stability curve of 3% Ru-WP<sub>2</sub>/WP tested at 10 mA cm<sup>-2</sup>.

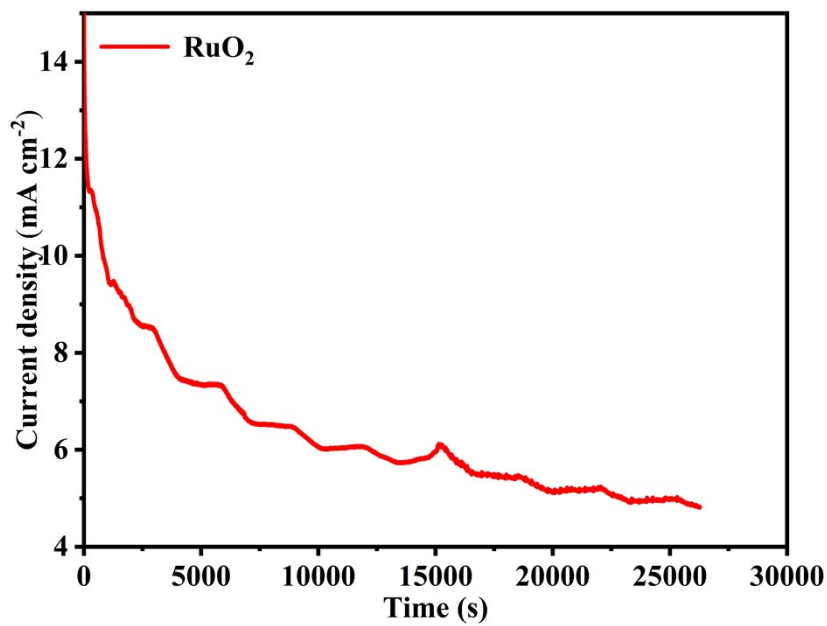


Fig. S12. The stability curve of RuO<sub>2</sub> tested at 10 mA cm<sup>-2</sup>.

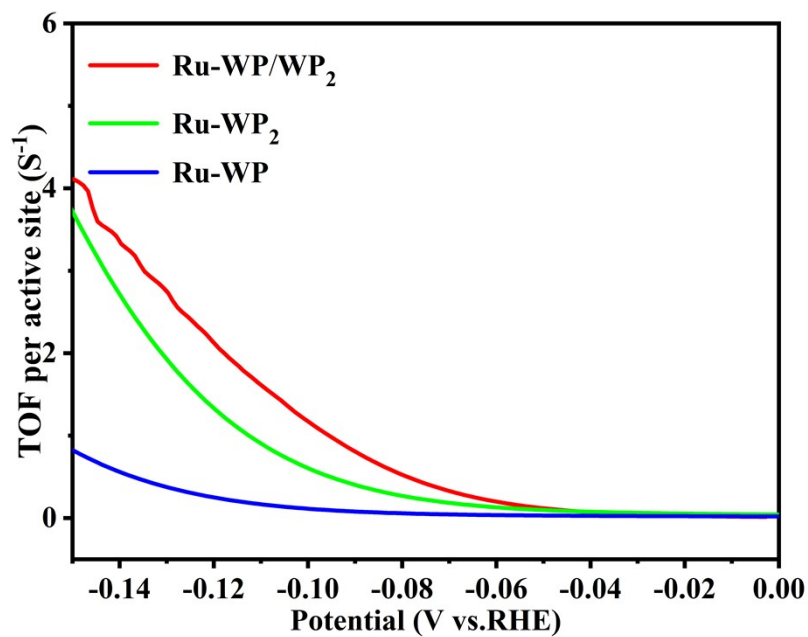
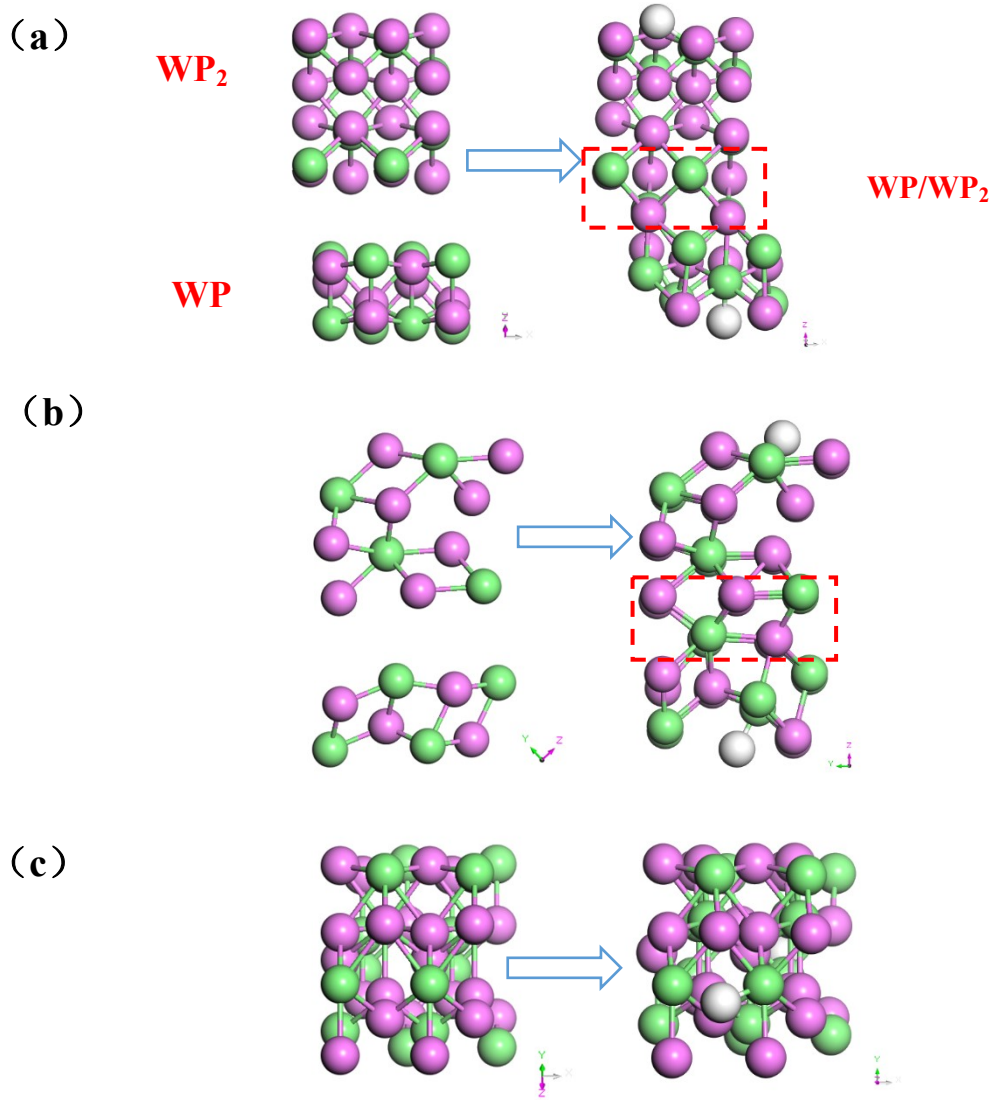
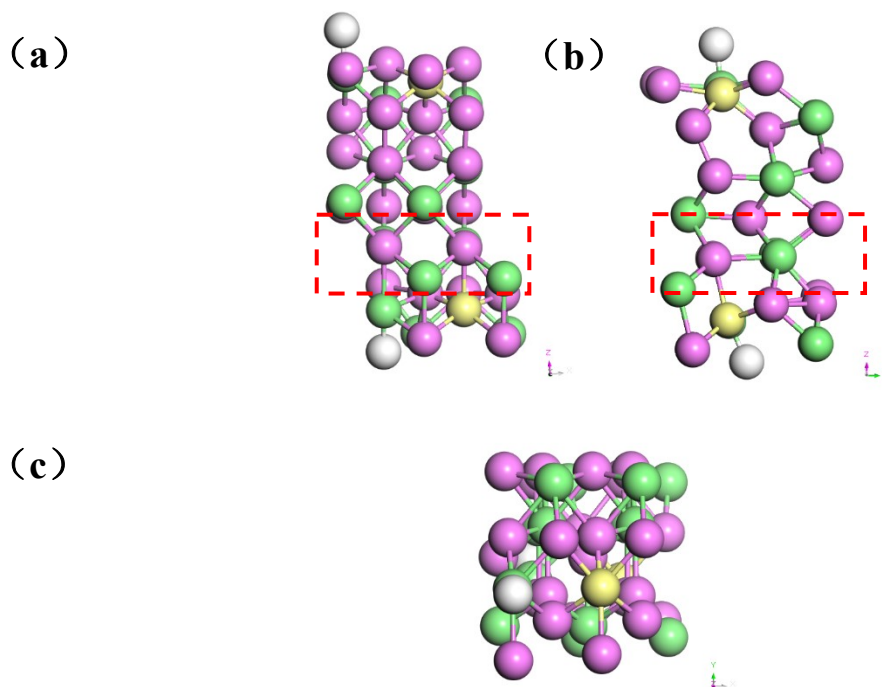


Fig. S13. TOF curves of Ru-WP<sub>2</sub>/WP, Ru-WP<sub>2</sub> and Ru-WP.

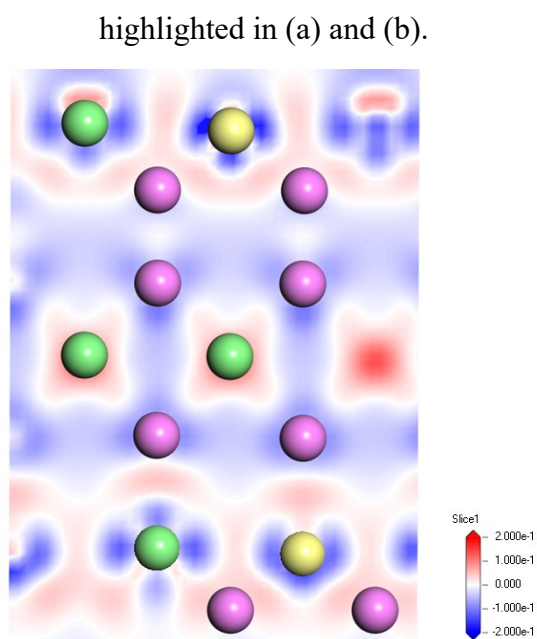




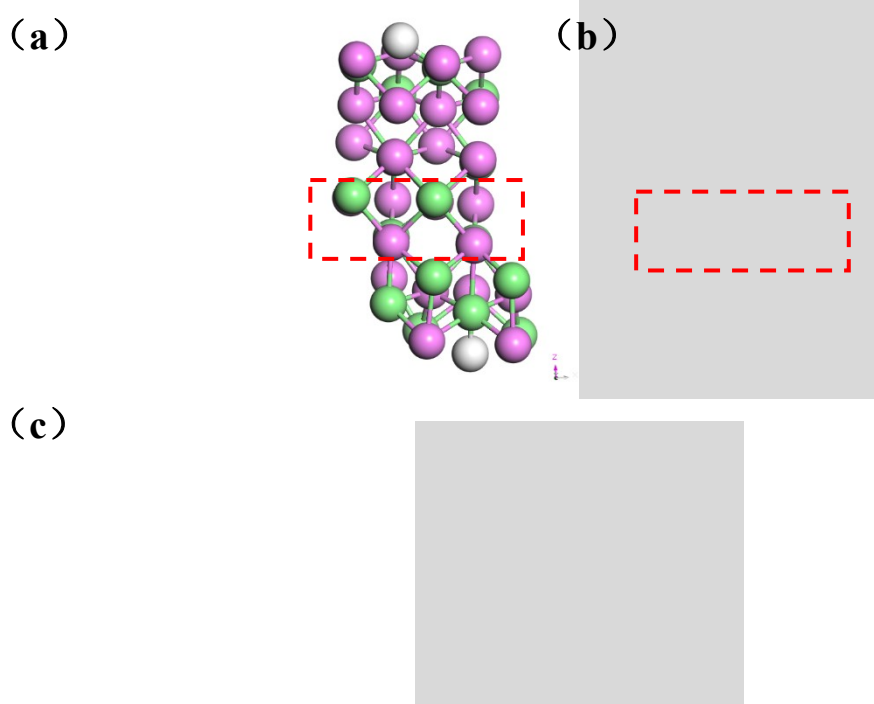
**Fig. S14.** WP/WP<sub>2</sub> optimization front and rear models, the left is optimized before the left is optimized. (a) Front view, (b) Side view, and (c) Top view. White, Green and Pink balls represent H, W and P atoms, respectively. The heterojunction interface and interface atoms were highlighted in (a) and (b).



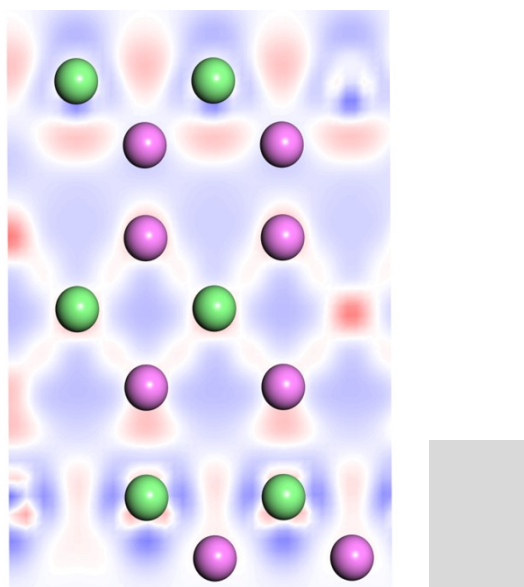
**Fig. S15.** Ru-WP/WP<sub>2</sub> model and after structural optimization. (a) Front view, (b) Side view, and (c) Top view. Yellow, White, Green and Pink balls represents Ru, H, W and P atoms, respectively. The heterojunction interface and interface atoms were



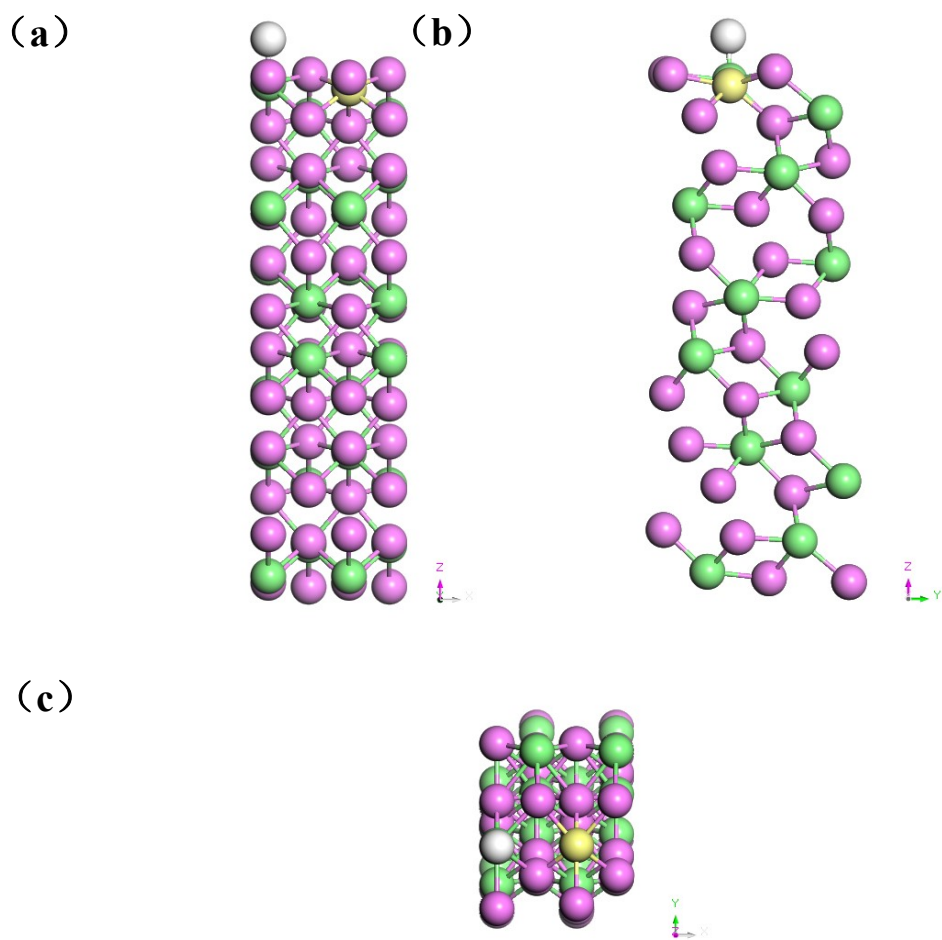
**Fig. S16.** Charge density difference diagram of Ru-WP/WP<sub>2</sub>.



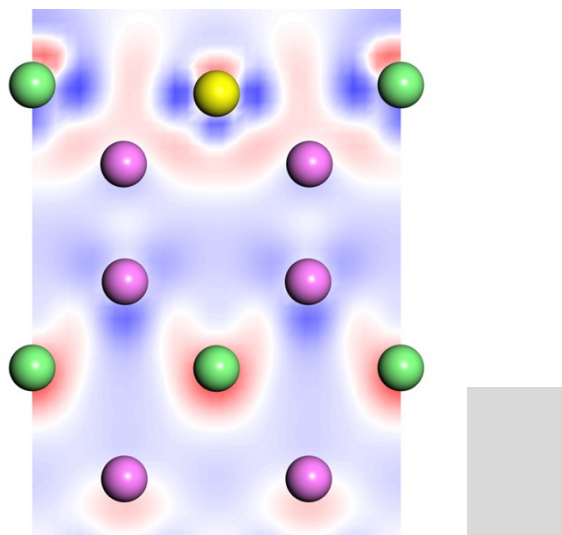
**Fig. S17.** WP/WP<sub>2</sub> model at after structural optimization. (a) Front view, (b) Side view, (c) Top view. White, Green and Pink balls represent H, W, and P atoms, respectively. The heterojunction interface and interface atoms were highlighted in (a) and (b).



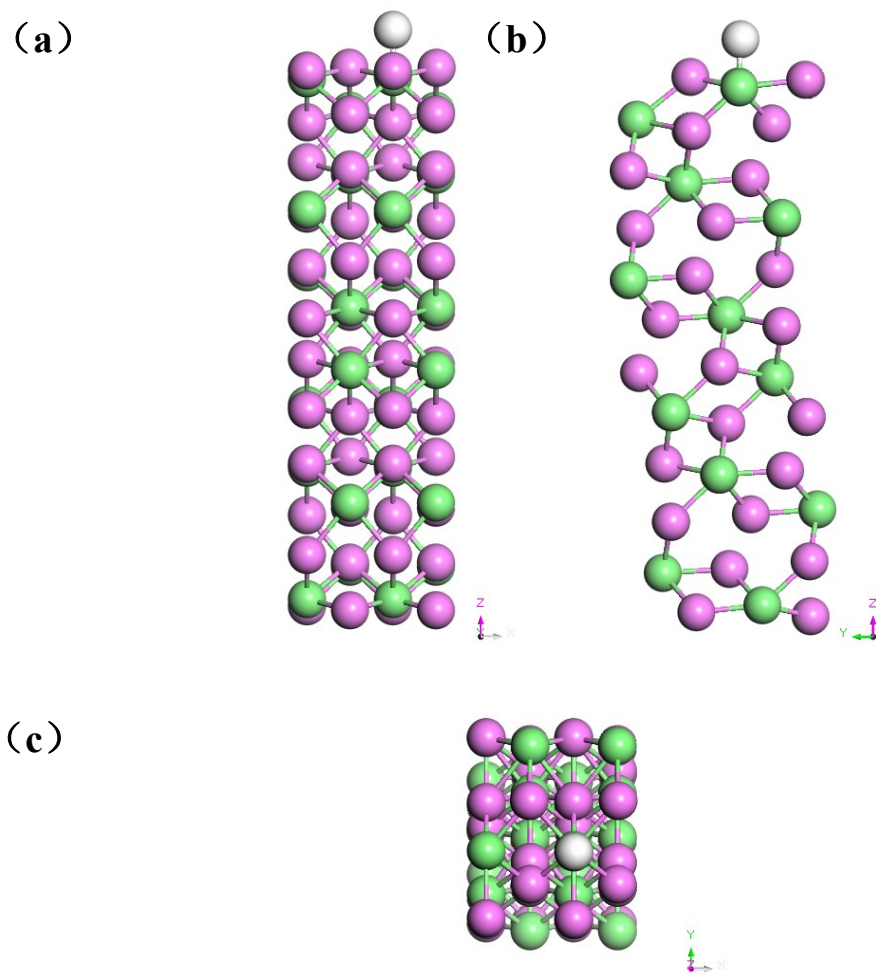
**Fig. S18.** Charge density difference diagram of WP/WP<sub>2</sub>.



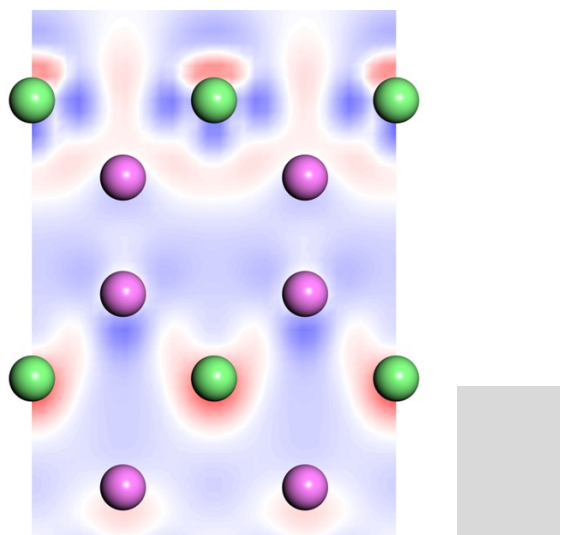
**Fig. S19.** Optimized atomic model for Ru-WP<sub>2</sub> (a) Front view, (b) Side view, and (c) Top view. Yellow, White, Green, and Pink balls represent Ru, H, W and P atoms, respectively.



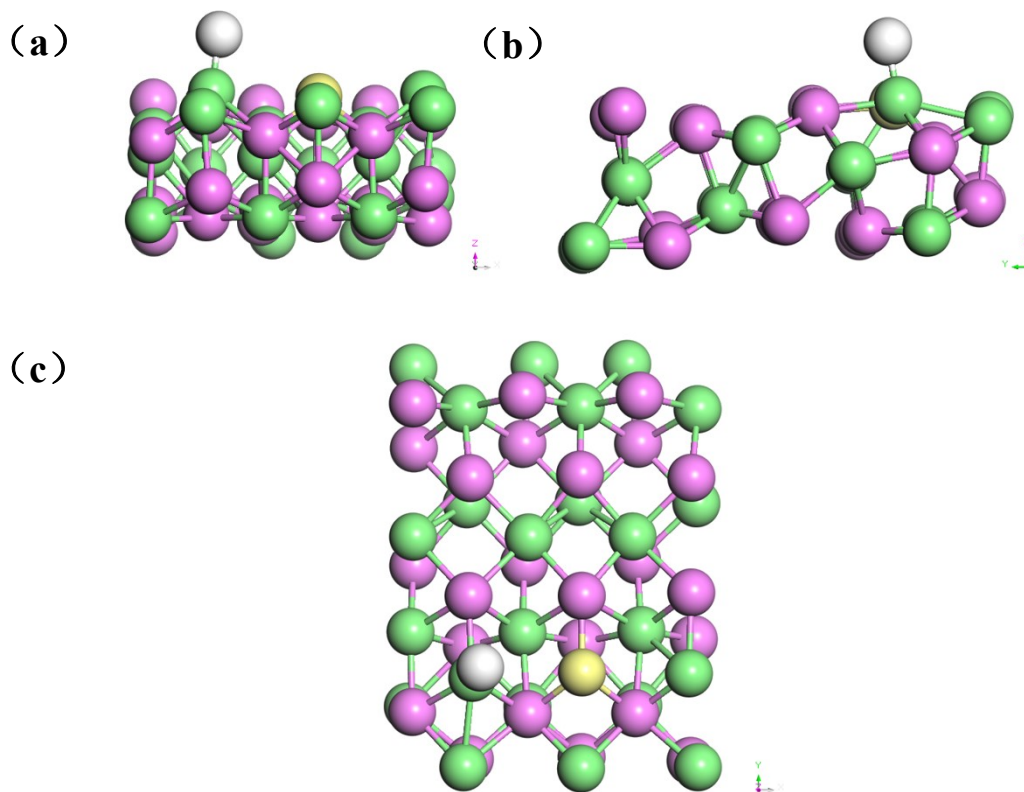
**Fig. S20.** Charge density difference diagram of Ru-WP<sub>2</sub>.



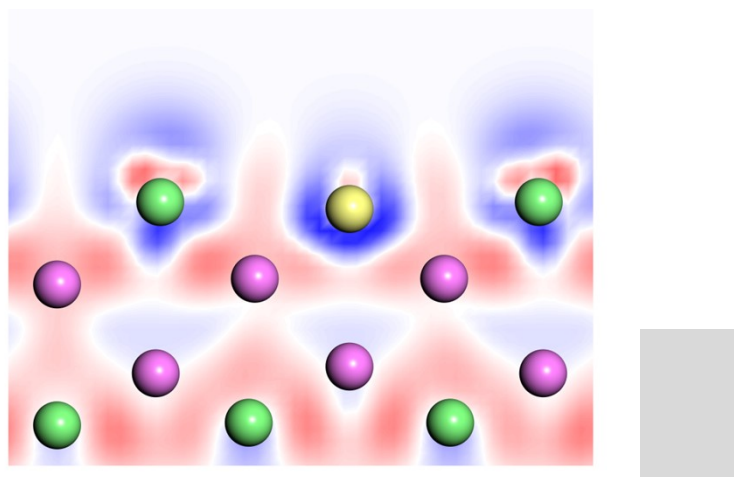
**Fig. S21.**  $WP_2$  model and after structural optimization. (a) Front view, (b) Side view, and (c) Top view. White, Green, and Pink balls represent H, W, and P atoms, respectively.



**Fig. S22.** Charge density difference diagram of WP<sub>2</sub>.

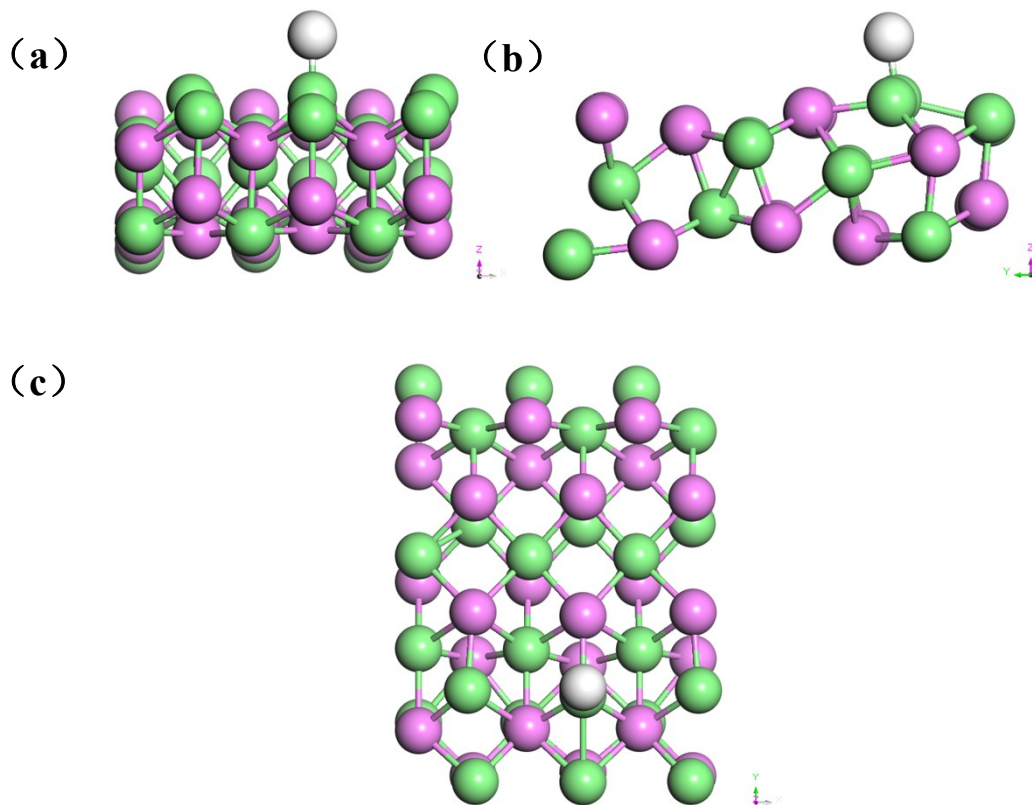


**Fig. S23.** Ru-WP model and after structural optimization. (a) Front view, (b) Side view, and (c) Top view. Yellow, White, Green, and Pink balls represent Ru, H, W, and P atoms, respectively.

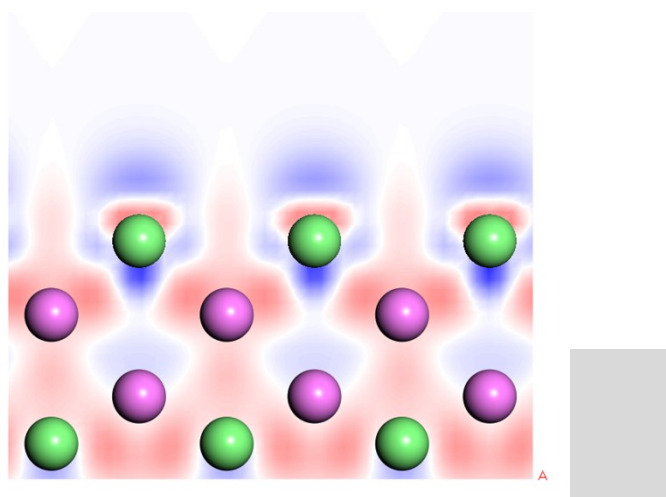


**Fig. S24.** Charge density difference diagram of Ru-WP.

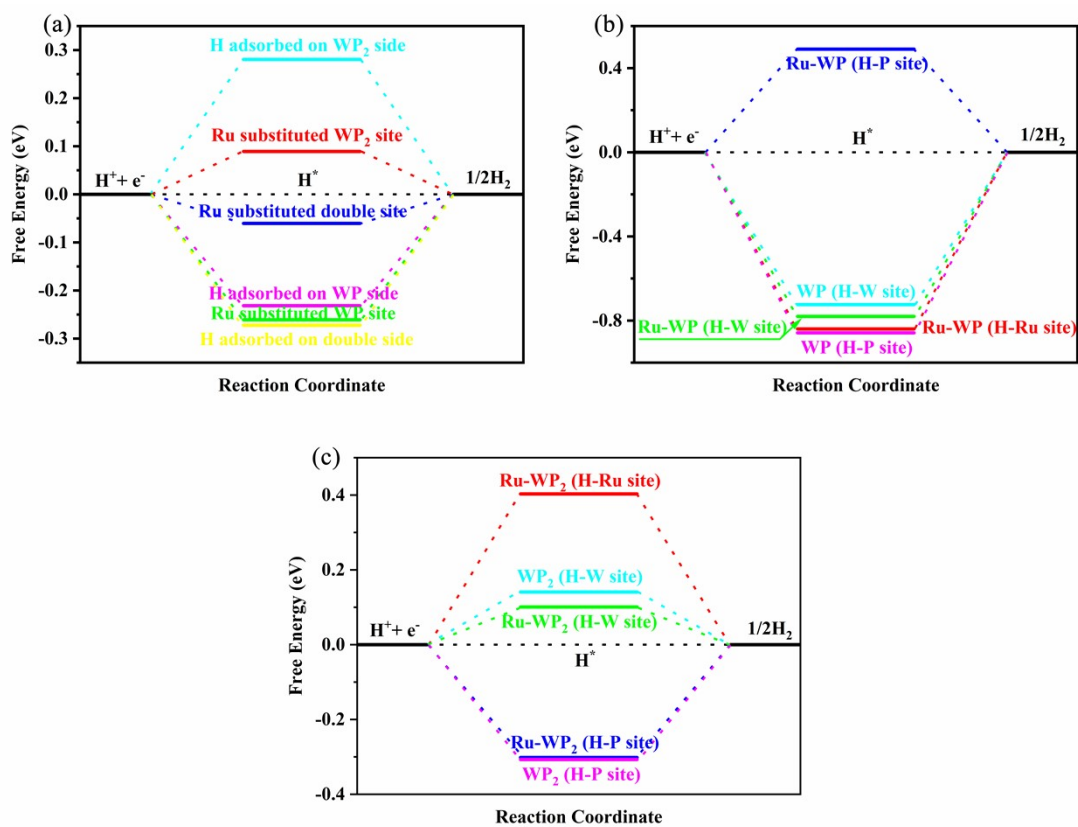




**Fig. S25.** WP model and after structural optimization. (a) Front view, (b) Side view, (c) Top view. White is an H atom, green is W atom and pink is P atom.



**Fig. S26.** Charge density difference diagram of WP.



**Fig. S27.** Gibbs free energy volcano diagram of hydrogen adsorption point of catalyst.

(a) Ru-WP/WP<sub>2</sub> NH/CC. (b) Ru-WP NS/CC, and (c) Ru-WP<sub>2</sub> NS/CC.

**Table S1.** Acidic HER performance of catalysts prepared with different Ru: W ratio.

Electrocatalyst <sup>a</sup>	Overpotential (mV vs. RHE)	Tafel slope (mV dec <sup>-1</sup> )	R <sub>ct</sub> (Ω)	The atomic ratio of Ru to W <sup>b</sup>
0	183.8	69.64	5.26	0
0.1%	184.7	72.55	4.27	0.22%
0.5%	180.7	71.09	4.17	0.61%
1%	131.1	62.78	3.04	2.6%
3%	58.0	47.98	2.20	4.2%
5%	94.7	62.53	3.19	5.61%
7%	100.4	53.53	7.10	6.1%

<sup>a</sup>The Ru: W ratio% joined when the catalyst was prepared.

<sup>b</sup>The results of ICP data analysis

**Table S2.** Acidic HER activities of Ru-WP/WP<sub>2</sub> NH/CC catalysts prepared at different phosphating temperatures. (Temperature effect on optimized Ru-WP/WP<sub>2</sub> NH/CC)

Electrocatalyst <sup>a</sup>	Overpotential (mV vs. RHE)	Tafel slope(mV dec <sup>-1</sup> )	R <sub>ct</sub> (Ω)
550	201.2	71.11	6.21
600	134.9	71.04	5.52
650	94.1	52.49	4.19
700	58.0	47.98	2.20
750	108.8	55.19	3.45

<sup>a</sup> The catalysts of prepared at different phosphating temperatures.

**Table S3.** Summarized HER performance of some previously reported catalysts with present work.

Electrocatalyst	Overpotential (mV vs. RHE)	Tafel slope (mV dec <sup>-1</sup> )	Reference
Ru-WP/WP <sub>2</sub>	58	47.98	This work
ES-WC/W <sub>2</sub> C	75	59	[1]
NiP <sub>2</sub> /CoP <sub>2</sub>	71	48.5	[2]
Mo <sub>2</sub> C@MoS <sub>2</sub>	67	67	[3]
CoP <sub>3</sub> /CoMoP/NF	125	61.1	[4]
MoS <sub>2</sub> -MoP	102	58	[5]
CoP/Co-MOF	52	49	[6]
Co-Mo <sub>5</sub> N <sub>6</sub>	19	29	[7]
CoP/NiCoP	133	88	[8]
Se-MnS/NiS	56	55	[9]
Ru-NiCoP/NF	44	45.4	[10]
Ru-CuO/MoS <sub>2</sub>	198	113	[11]
Ru-Zn <sub>3</sub> V <sub>3</sub> O <sub>8</sub>	70	50.6	[12]
Ru-FeCoP	45	32.1	[13]
Ru-VN	134	35	[14]
Ru-NMCNs-T	28	35.2	[15]
Pd-Ru@NG	42	73	[16]
Ru@B-Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub>	62.9	100	[17]

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Ru/CoO	55	72	[18]
NiFe-P	44	80	[19]
WP <sub>2</sub> NW/NF	130	94	[20]
Ni-WP <sub>2</sub> NS/CC	110	65	[21]
CoWP-CA/KB	111	58	[22]
WP-Mesop	104	58	[23]
Co/WP <sub>2</sub> NWs	72	52	[24]
W-Ni <sub>2</sub> P NS/CC	71	67.4	[25]
P-WP <sub>2</sub> NSs/GP	155	66	[26]
C-WP/W	109	79.8	[27]

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**Table S4. Gibbs free energy of adsorbed hydrogen at each point of the catalyst.**

Electrocatalyst	Adsorption site of H	$\Delta G_{H^*}$ (eV)
Pt	Pt	-0.09
Ru-WP	Ru	-0.84
	W	-0.71
	P	0.49
WP	W	-0.72
	P	-0.86
Ru-WP <sub>2</sub>	Ru	0.40
	W	-0.10
	P	-0.3
WP <sub>2</sub>	Ru	0.14
	W	-0.31

**Table S5.** Gibbs free energy of adsorbed hydrogen with different Ru substitution sites in Ru-WP/WP<sub>2</sub> heterojunction. When Ru is doped, H is adsorbed on the W atom next to Ru.

Electrocatalyst	Ru Substitution Sites		$\Delta G_{H^*}$ (eV)	
Ru-WP/WP <sub>2</sub>	Ru substituted WP <sub>2</sub> side	H adsorbed on WP <sub>2</sub> side	0.089	
		H adsorbed on WP side	-0.241	
	Ru substituted WP side	H adsorbed on WP <sub>2</sub> side	0.175	
		H adsorbed on WP side	-0.261	
	Both WP and WP <sub>2</sub> are replaced by Ru	H adsorbed on WP <sub>2</sub> side	0.119	
		H adsorbed on WP side	-0.235	
		H adsorption on both sides of WP and WP <sub>2</sub> .	-0.06	
	WP/WP <sub>2</sub>	H adsorbed on WP <sub>2</sub> side		0.281
		H adsorbed on WP side		-0.232
H adsorption on both sides of WP and WP <sub>2</sub> .		-0.272		



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