

Ferroelectrically controlled electromagnetic and transport properties in $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ van der Waals multiferroic heterostructures

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I Structure and electromagnetic properties of the VN_2H_2 monolayer and the Al_2O_3 monolayer

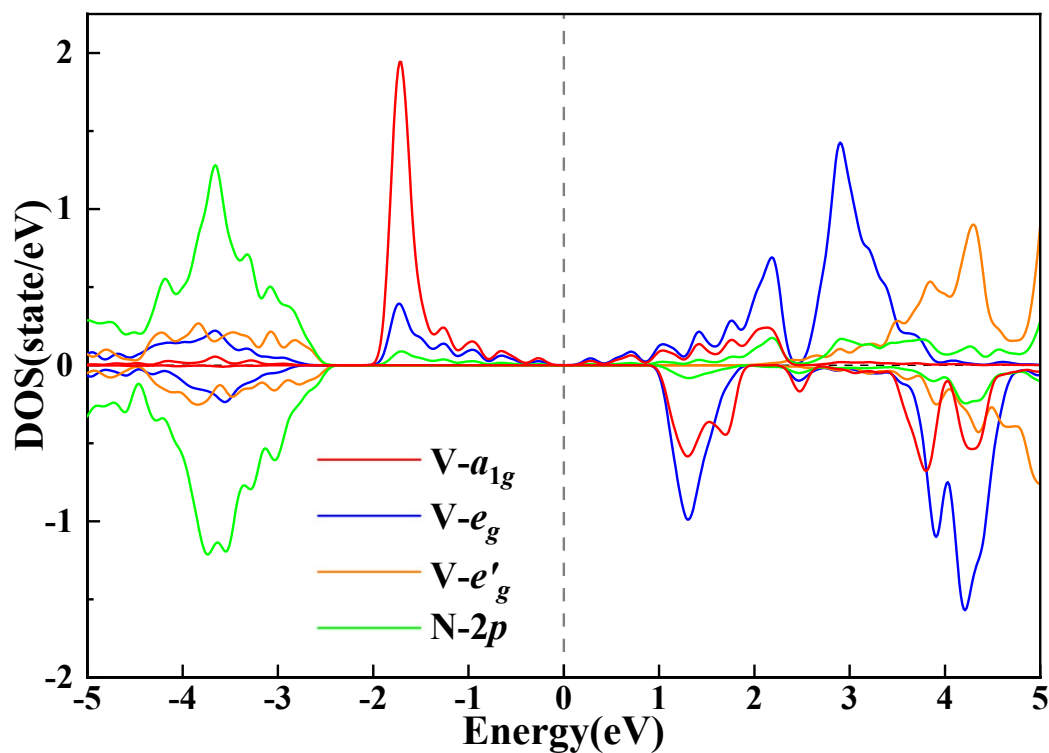


Fig. S1. The orbital projected state density of the VN_2H_2 monolayer. The degenerate orbitals have been merged.

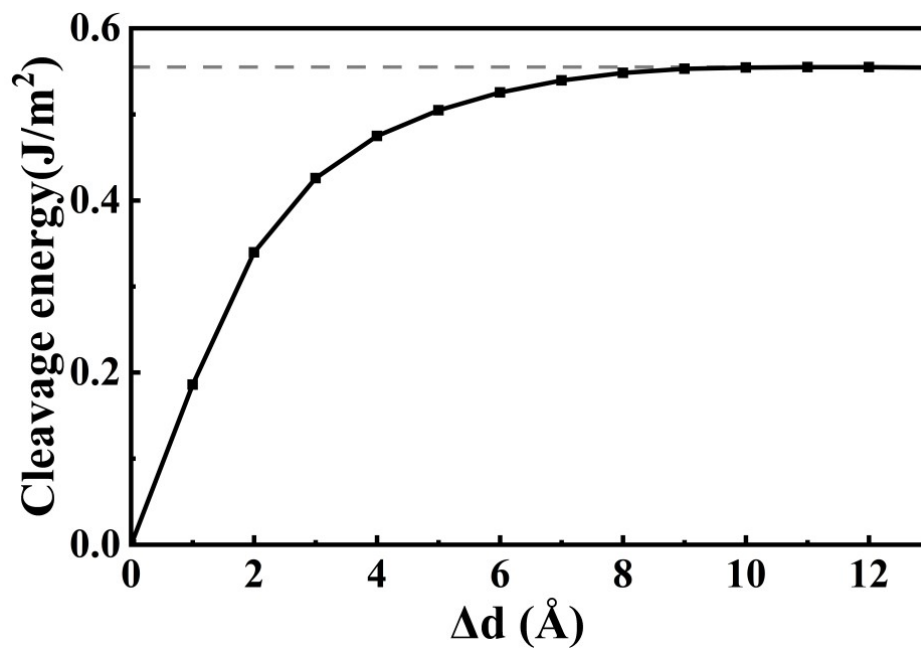


Fig. S2. The cleavage energy of the Al_2O_3 monolayer as a function of separation distance ($\Delta d = d - d_0$), where d_0 and d are the equilibrium and unequilibrium gaps.

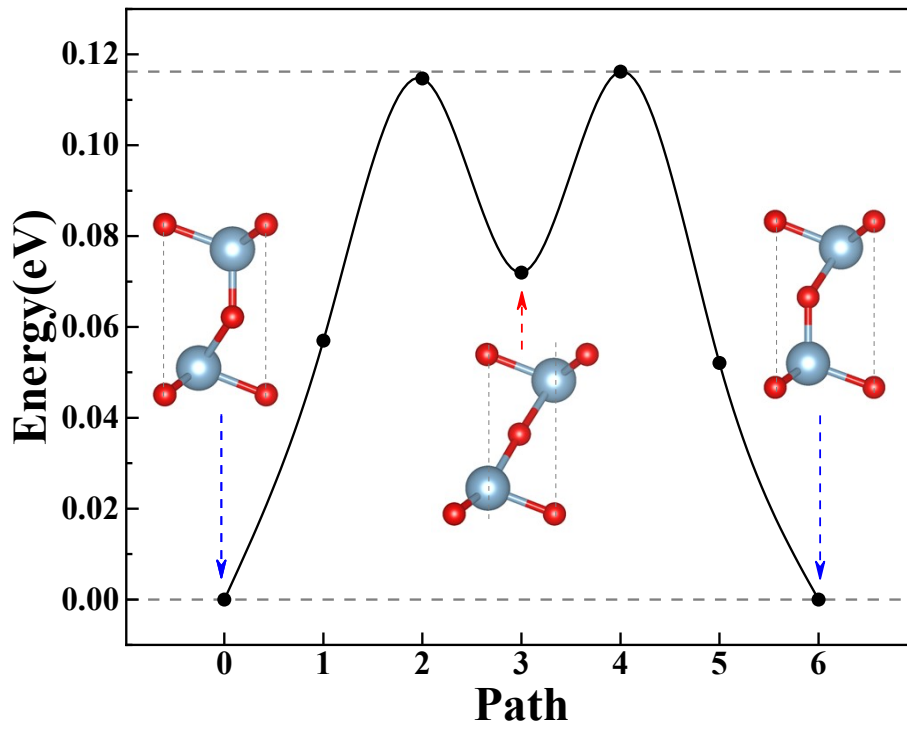


Fig. S3. The FE transition barrier of the Al_2O_3 monolayer.

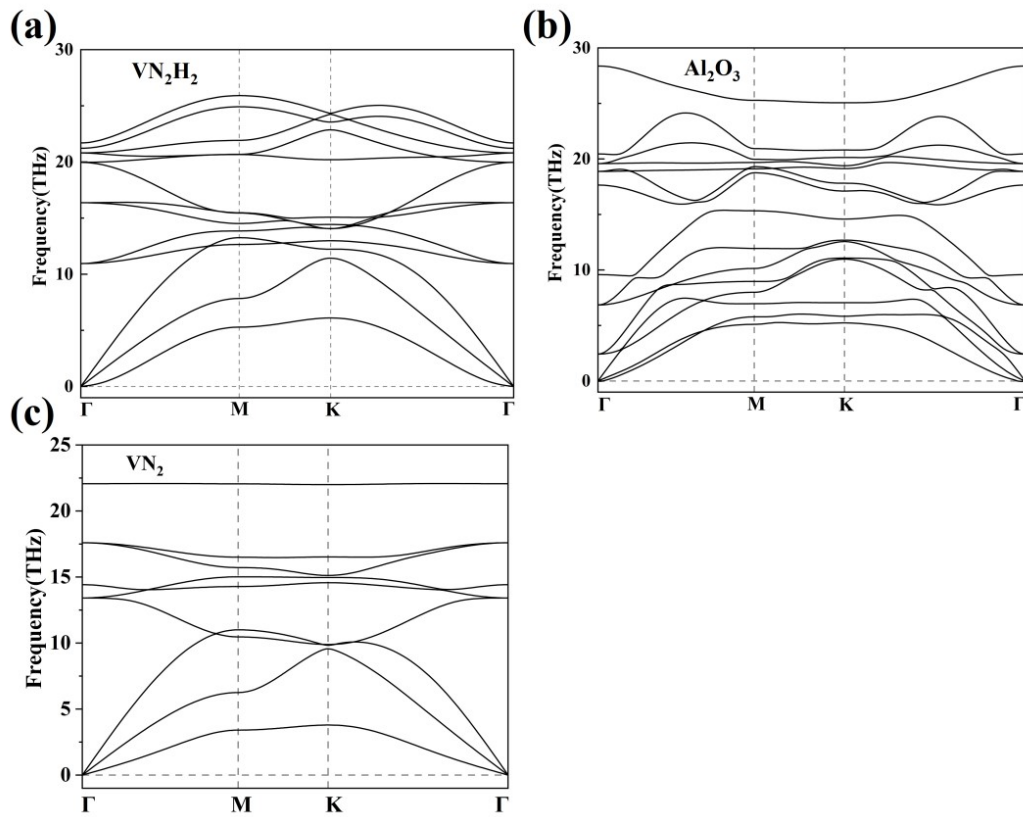


Fig. S4. The phonon spectra of the VN_2H_2 monolayer, the Al_2O_3 monolayer, and the VN_2 monolayer.

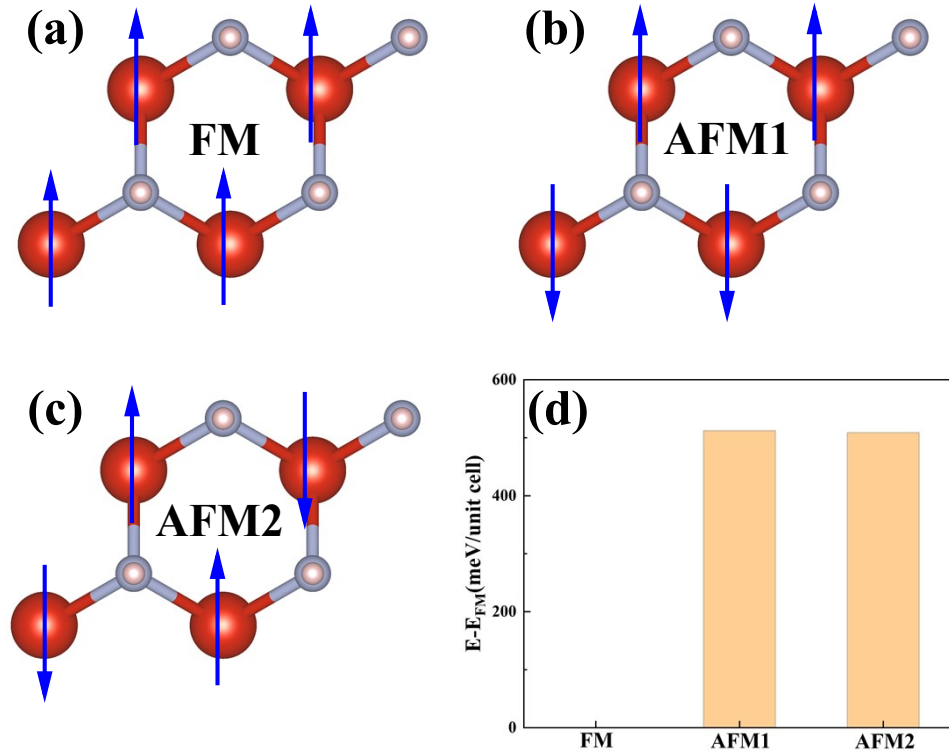


Fig. S5. The different magnetic configurations of the VN₂H₂ monolayer. (a) FM, (b) AFM1, (c) AFM2, and (d) the energy difference relative to FM in 2×2×1 supercell. The direction of the arrow represents the direction of the spin.

Table S1. The calculated lattice parameters of FM and AFM states of the VN₂H₂ monolayer, including lattice constant (*a*), the static energy (*E*₀), the length of V-N bond (*L*_{V-N}), the length of N-H bond (*L*_{N-H}), the angle of V-N-V bond (*θ*_{V-N-V}), the total magnetic moment (*mag*).

	<i>c</i> (Å)	<i>E</i> ₀ (eV)	<i>L</i> _{V-N} (Å)	<i>L</i> _{N-H} (Å)	<i>θ</i> _{V-N-V} (°)	<i>mag</i> (μB)
FM	2.88	-34.15	2.01	1.02	91.33	1.00
AFM1	2.88	-34.02	2.00	1.02	91.51	0.00
AFM2	2.90	-34.02	2.01	1.02	91.51	0.00

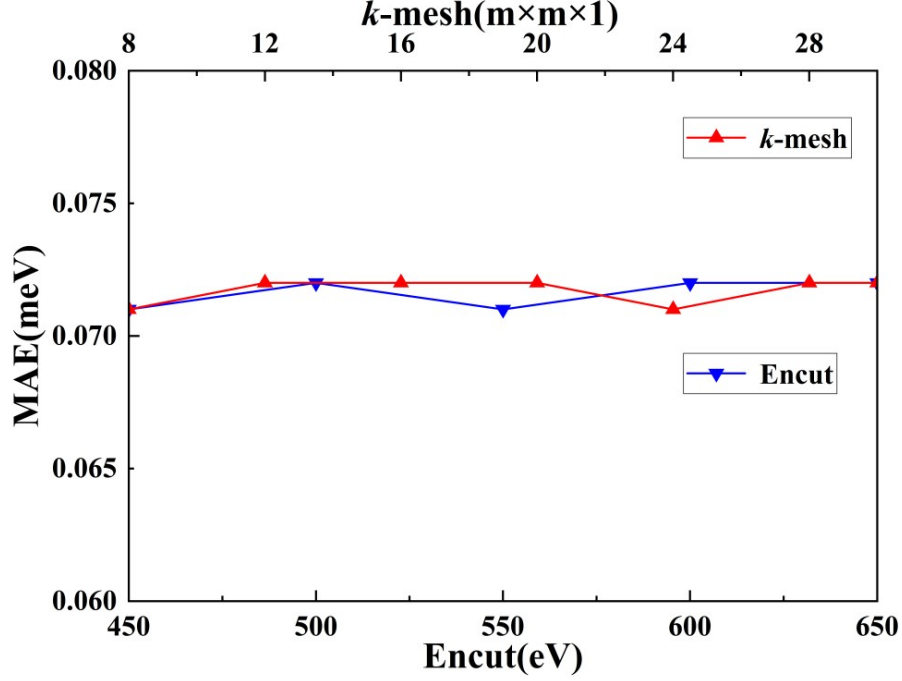


Fig. S6. The magnetic anisotropy energy of the VN₂H₂ monolayer with different cutoff energy and kpoint.

II The second-order perturbation theory analysis of the magnetic anisotropy energy

The energy expression considering SOC effect is written as

$$E^{SL} = \sum_o \sum_u \frac{|\langle o | H^{SL} | u \rangle|^2}{E_o - E_u}$$

where the perturbative Hamiltonian is $H^{SL} = \lambda \vec{\sigma} \cdot \vec{L}$, and according to the expression of MAE, MAE can be represented as

$$MAE = E_{(z)} - E_{(x)} = \pm \lambda^2 \sum_{o, u} \frac{|\langle o | L_z | u \rangle|^2 - |\langle o | L_x | u \rangle|^2}{E_o - E_u}$$

From the orbital-resolved MAE shown in **Fig. 2** (b), the MAE of monolayer VN₂H₂ is mainly contributed by the SOC interaction of the d_{xy} orbitals and the $d_{x^2-y^2}$ orbitals of the vanadium atom. The differences of matrix elements are given in **Table S2**, and the energies of occupied and unoccupied states can be analyzed from the PDOS shown in **Fig. S7**. Since only the spin-up channel contributes to the occupied state, the expression for MAE can be written as

$$(d_{xy}, d_{x^2-y^2}) \Delta E = \lambda^2 \sum_{E_{o^+} - E_{u^+}} \frac{4}{E_{o^+} - E_{u^+}} + \frac{-4}{E_{o^+} - E_{u^-}} = \lambda^2 \left(\frac{4}{-m} + \frac{-4}{-n} \right)$$

as shown in **Fig. S7**, $m > n$, the calculated MAE is positive, indicating that monolayer VN₂H₂ has

an in-plane easy magnetization axis, which is consistent with the results of DFT.

Table S2. Matrix elements differences between two magnetization directions

$$|\langle o^+ | L_z | u^+ \rangle|^2 - |\langle o^+ | L_x | u^+ \rangle|^2 \text{ and } |\langle o^+ | L_z | u^- \rangle|^2 - |\langle o^+ | L_x | u^- \rangle|^2$$

	u^+					u^-				
o^+	d_{xy}	d_{yz}	d_{z^2}	d_{xz}	$d_{x^2-y^2}$	d_{xy}	d_{yz}	d_{z^2}	d_{xz}	$d_{x^2-y^2}$
d_{xy}	0	0	0	-1	4	0	0	0	1	-4
d_{yz}	0	0	-3	1	-1	0	0	3	-1	1
d_{z^2}	0	-3	0	0	0	0	3	0	0	0
d_{xz}	-1	1	0	0	0	1	-1	0	0	0
$d_{x^2-y^2}$	4	-1	0	0	0	-4	1	0	0	0

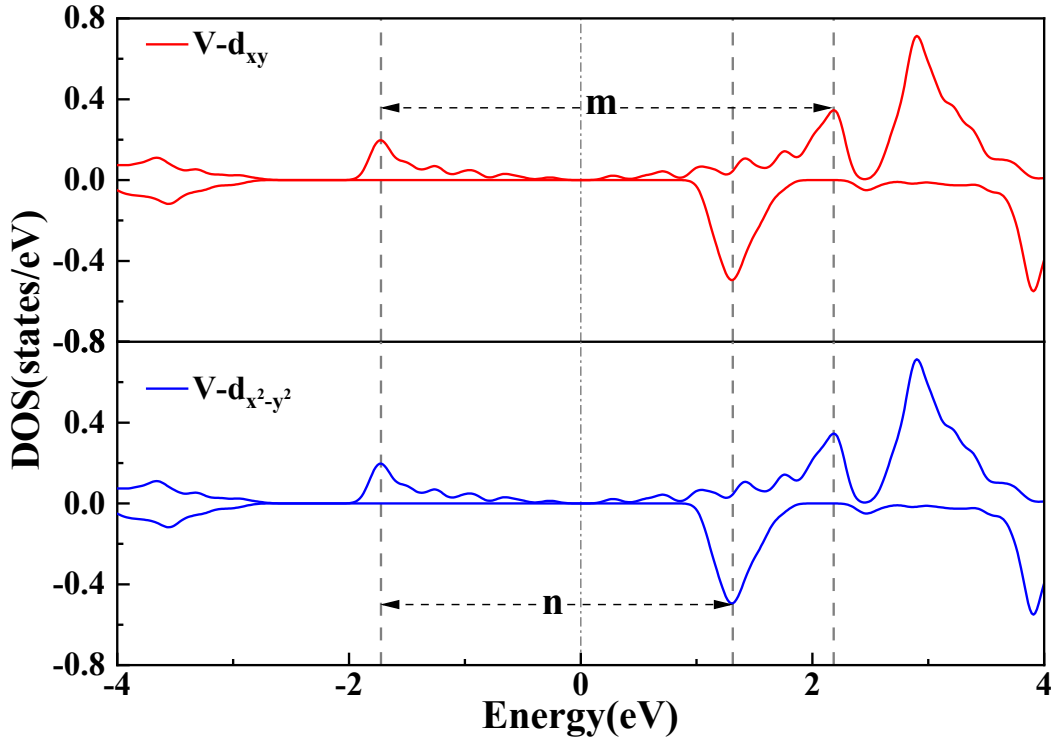


Fig. S7. The projected densities of state of d_{xy} orbitals and $d_{x^2-y^2}$ orbitals of the VN_2H_2 monolayer.

III Different stacking structural and electronic properties of the $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ vdW

heterostructures

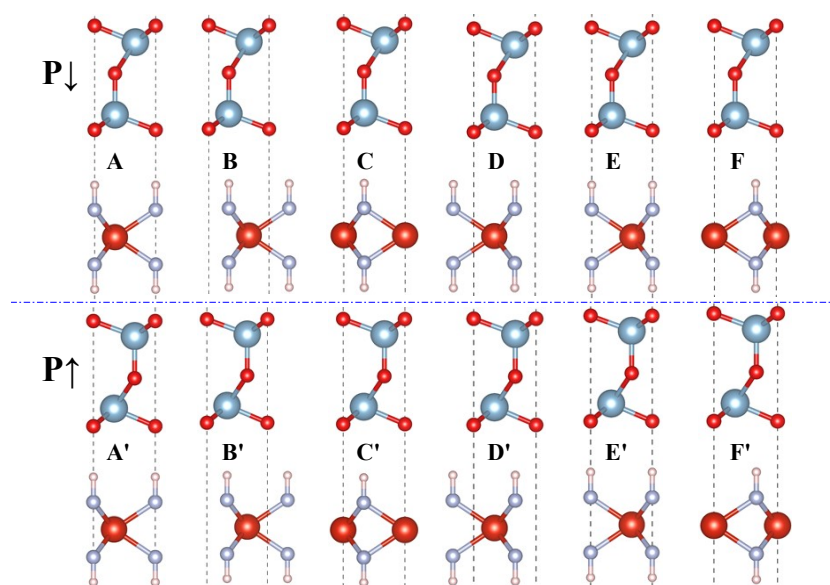


Fig. S8. The stack structures of $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ vdW heterostructures, including polarization up state and polarization down state. In the same polarization state, (A-C) is the Al_2O_3 layer fixed, the VN_2H_2 layer translated along the high symmetric point, (D-F) is the Al_2O_3 layer fixed, the VN_2H_2 layer first flipped 180° around the z axis, and then translated along the high symmetric point.

Table S3. Calculated parameters of $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ vdW heterostructures with different stacking configurations, including lattice constant (a), energy difference with respect to A' configuration (ΔE), and layer spacing (d). It should be noted that all stacking configurations are FM ground state, and the energy is based on the FM state energy. Although the energy of the D' configuration is very close to that of the A' configuration, calculations with sufficient precision still show that the energy of the D' configuration is greater than that of the A' configuration, so the A' configuration is used as the ground state.

P↓	A	B	C	D	E	F
a (Å)	2.91	2.91	2.91	2.91	2.91	2.91
ΔE (meV)	102.92	119.78	119.37	102.58	102.71	102.15
d (Å)	2.28	2.28	2.28	2.16	2.28	2.16
P↑	A'	B'	C'	D'	E'	F'
a (Å)	2.91	2.91	2.91	2.91	2.91	2.91
ΔE (meV)	0	48.75	48.39	24.51	0.087	24.27

d (Å)	2.08	2.19	2.19	2.04	2.07	2.04
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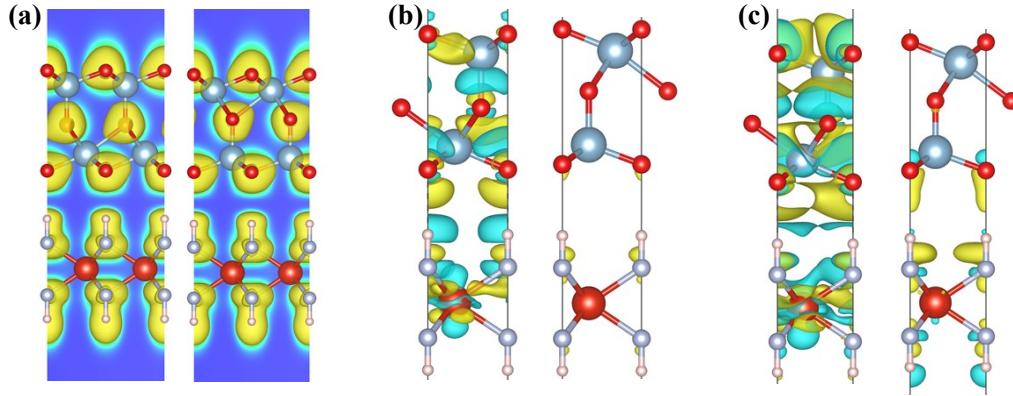


Fig. S9. (a) The electron localization function (ELF) of the $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ (\uparrow) state and $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ (\downarrow) state. The yellow region is the electron concentration region, and the blue region has no electron distribution. The charge density difference (isoval: (b) $0.0007\text{e}\cdot\text{\AA}^{-3}$, (c) $0.0002\text{e}\cdot\text{\AA}^{-3}$) of the $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ vdW heterostructure. The yellow and blue distributions correspond to the charge accumulation and depletion.

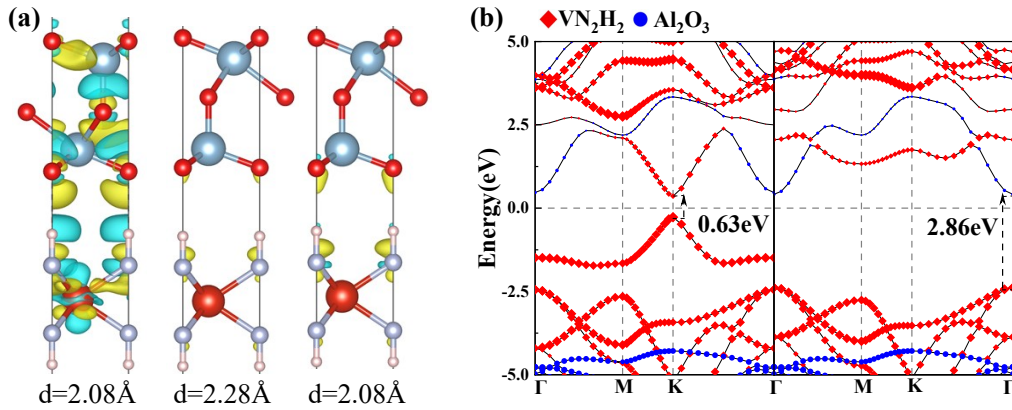


Fig. S10. (a) The charge density difference of (isoval: $0.0007\text{e}\cdot\text{\AA}^{-3}$) of the $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ vdW heterostructure with different layer spacing. (b) The band structure of $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ (\downarrow) with layer spacing of 2.08\AA .

IV Stability and band structures of electrostatic doping

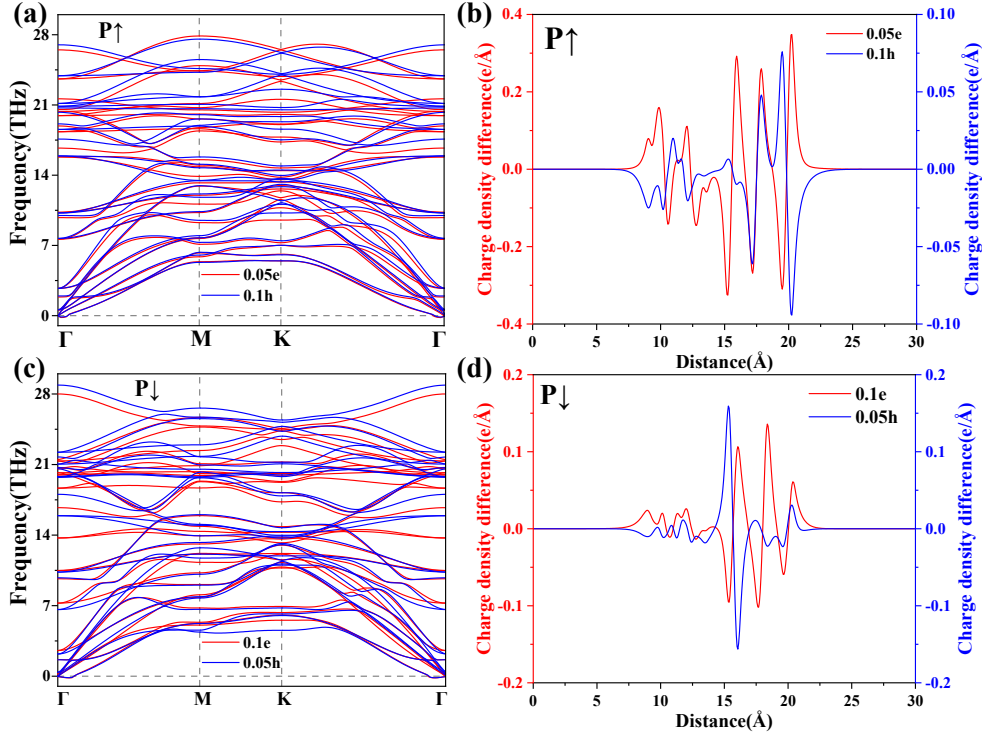


Fig. S11. (a) The phonon spectrum of the $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ (\uparrow) state under 0.05 e and 0.1 h per unit cell doping. (b) Charge density difference between the doped and undoped $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ (\uparrow) state under 0.05 e and 0.1 h per unit cell doping. (c) The phonon spectrum of the $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ (\downarrow) state under 0.1 e and 0.05 h per unit cell doping. (d) Charge density difference between the doped and undoped $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ (\downarrow) state under 0.1 e and 0.05 h per unit cell doping.

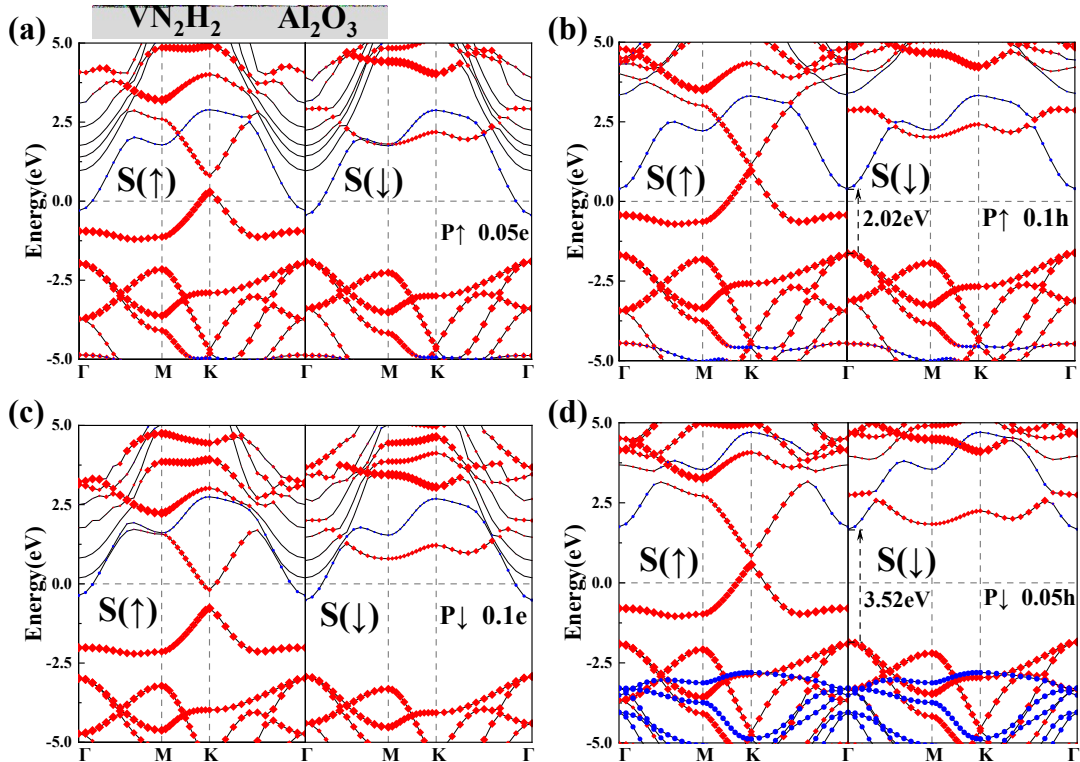


Fig. S12. The calculated band structures of the $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ (\uparrow) state under (a) $0.05 e$ and (b) $0.1 h$ per unit cell doping. The calculated band structures of the $\text{VN}_2\text{H}_2/\text{Al}_2\text{O}_3$ (\downarrow) state under (c) $0.1 e$ and (d) $0.05 h$ per unit cell doping.