

Supplementary Information

Design of yolk-shell ternary metal sulfide with tunable structure for high-performance electromagnetic wave absorption

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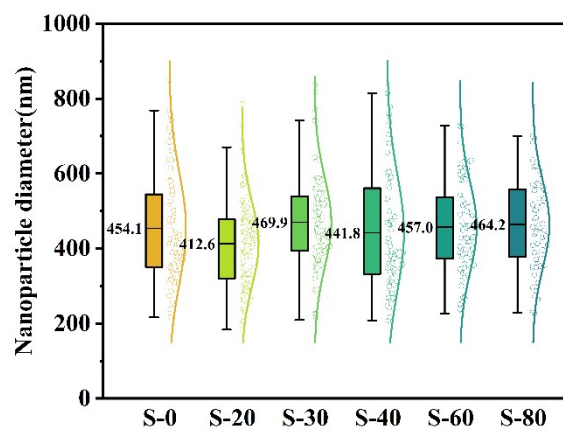


Figure S1. Particle size distribution and average diameter of different samples

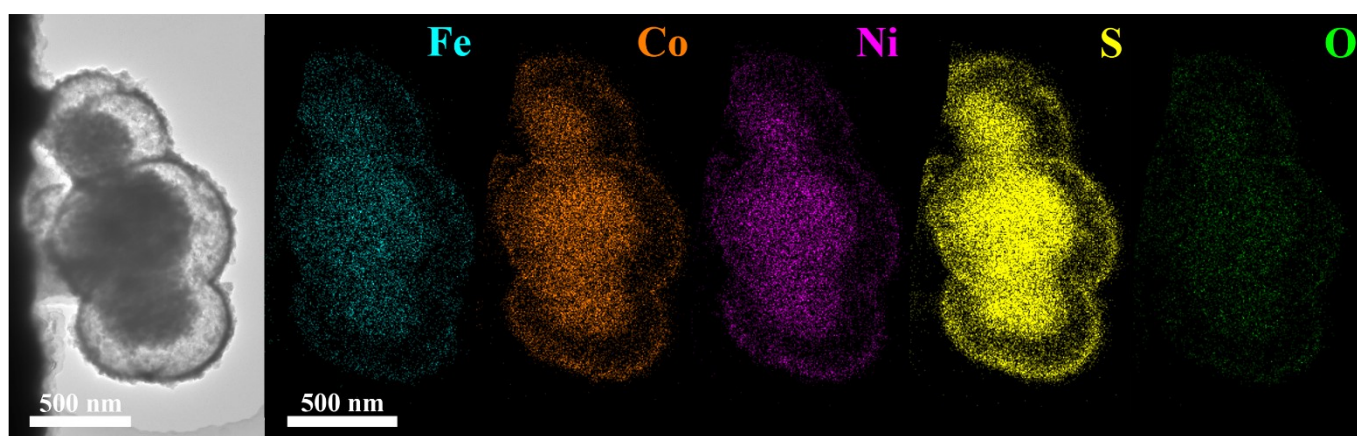


Figure S2. Elemental mapping images of energy dispersive X-ray spectroscopy (EDS) for S-40.

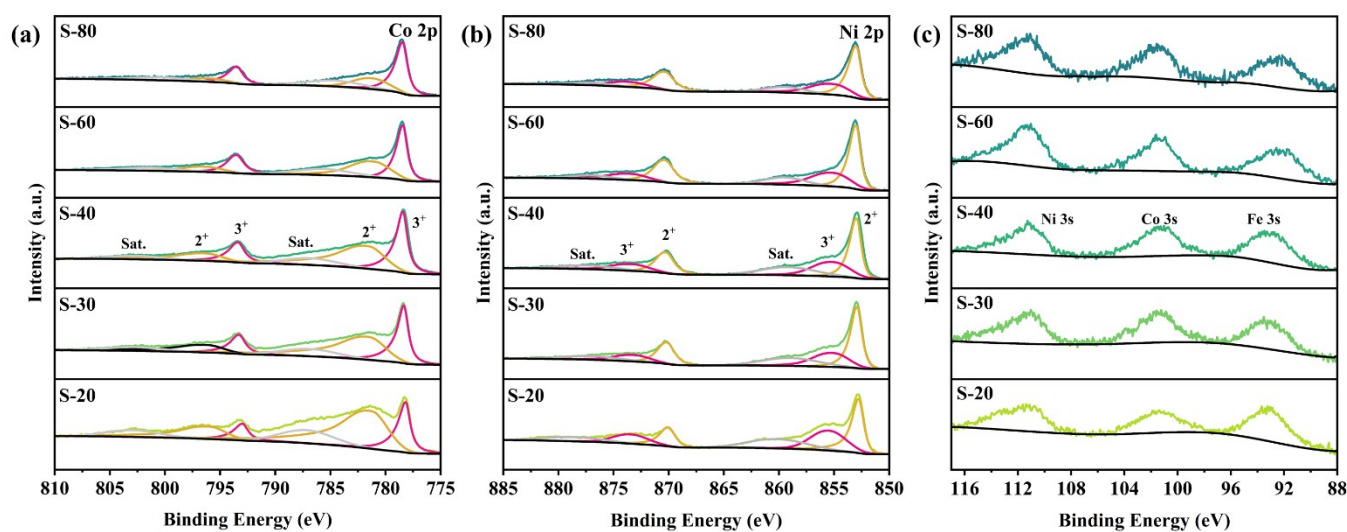


Figure S3. Deconvoluted XPS spectra of (a) Co 2p, (b) Ni 2p, (c) Fe 3s, Co 3s and Ni 3s.

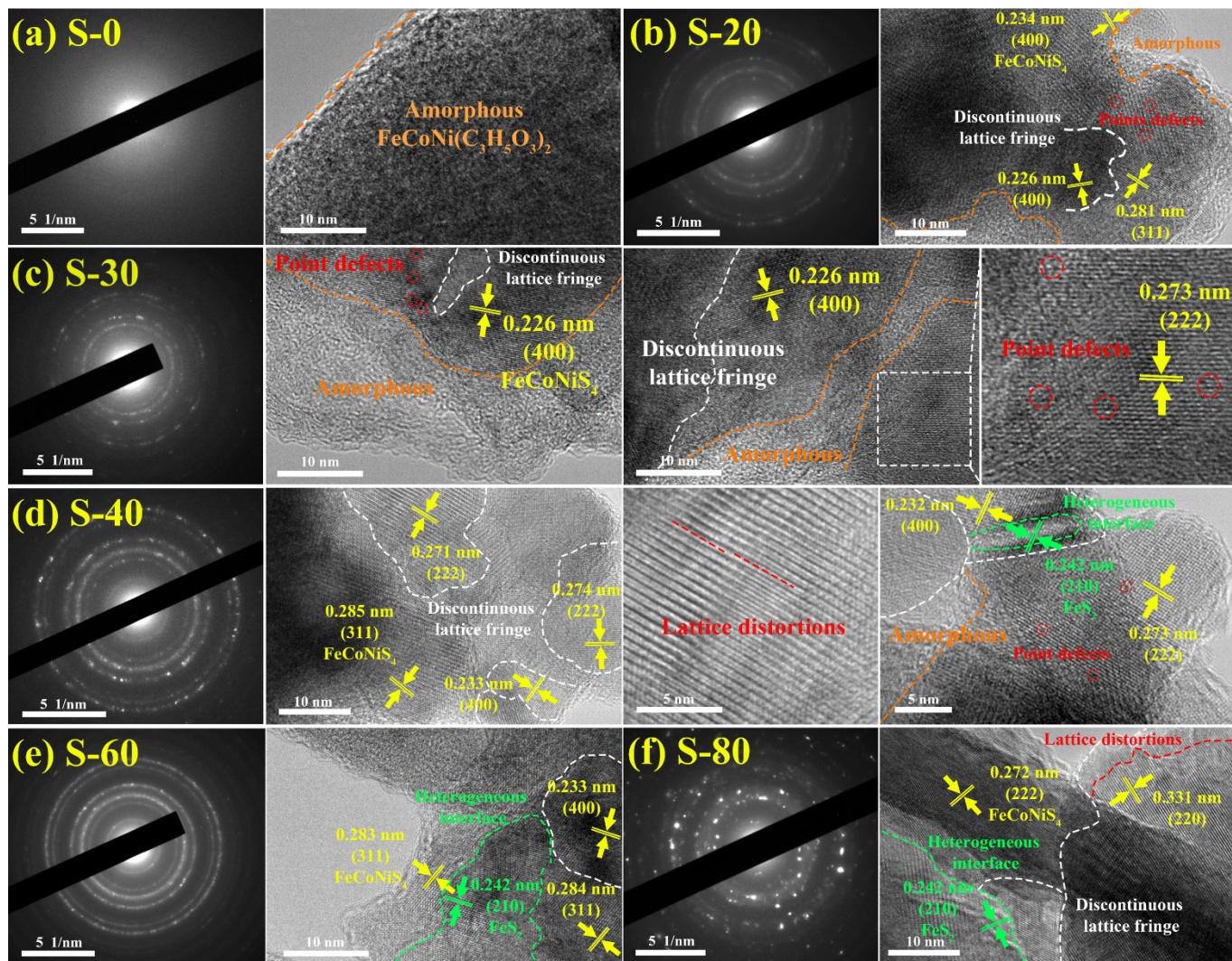


Figure S4. Selected area electron diffraction (SAED) patterns and high-resolution transmission electron microscopy (HR-TEM) images for (a) S-0, (b) S-20, (c) S-30, (d) S-40, (e) S-60 and (f) S-80.

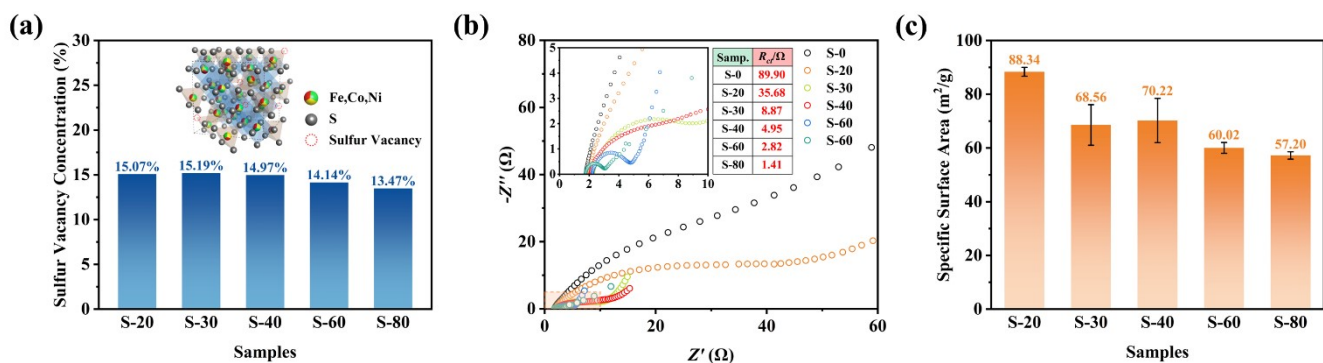


Figure S5. (a) Histogram of the calculated sulfur vacancy concentration; (b) Nyquist plots from the electronic impedance spectra (EIS) test; (c) Brunauer-Emmett-Teller (BET) analysis results obtained by N_2 adsorption-desorption measurement at 77K.

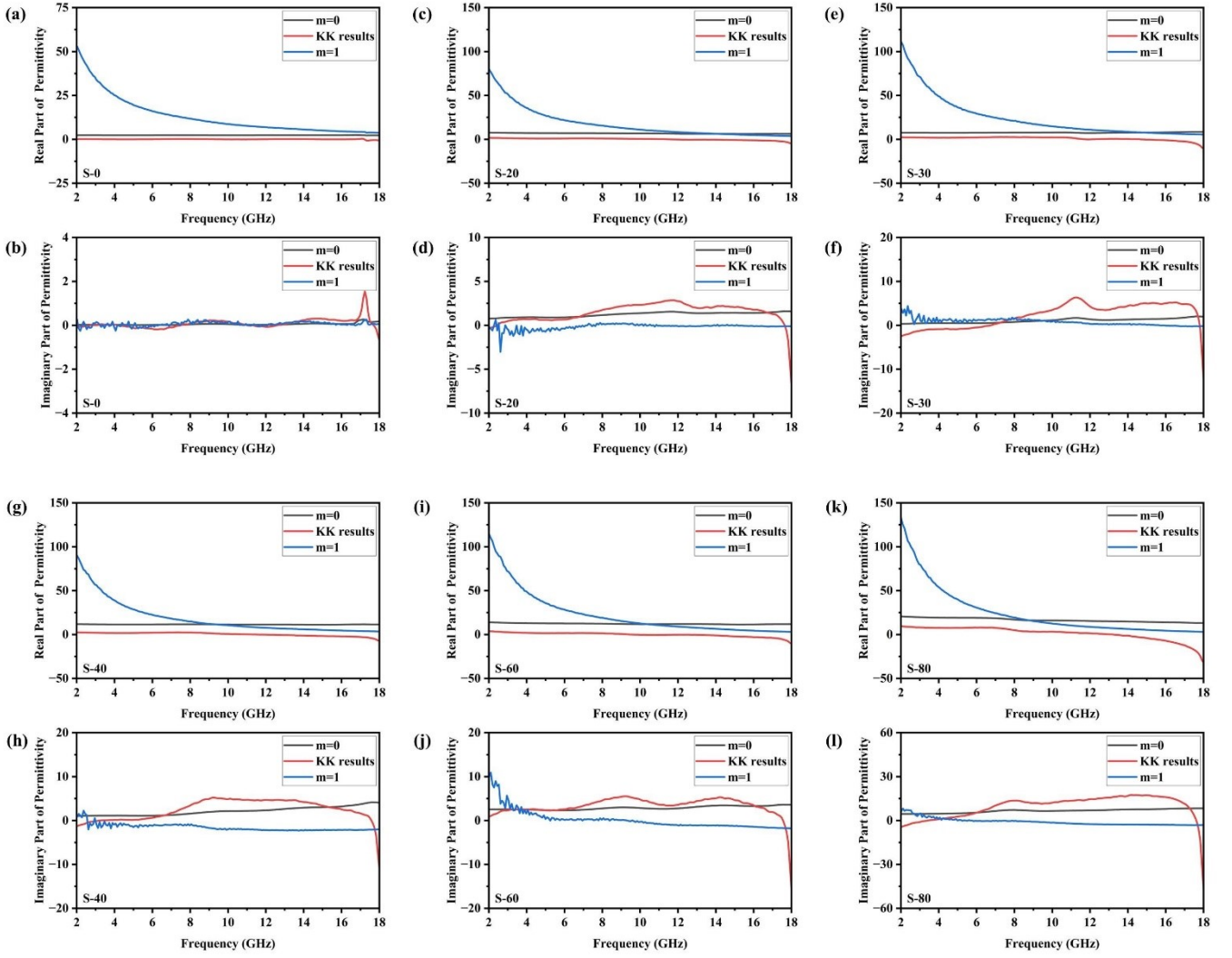


Figure S6. The real and imaginary parts of the permittivity of different m values and Kramers-Kronig (KK) analysis results for (a,b) S-0, (c,d) S-20, (e,f) S-30, (g,h) S-40 (i,j) S-60 and (k,l) S-80.

In the process of calculating permittivity based on scattering parameters, different m values were selected for calculation and Kramers-Kronig (KK) analysis was carried out, the results were shown in **Figure S6**. It can be obviously found that the measure results with $m=0$ agree best with the KK analysis results, which verified the validity of test results. Then we conduct the subsequent absorption performance analysis according to the dielectric parameters with $m=0$.

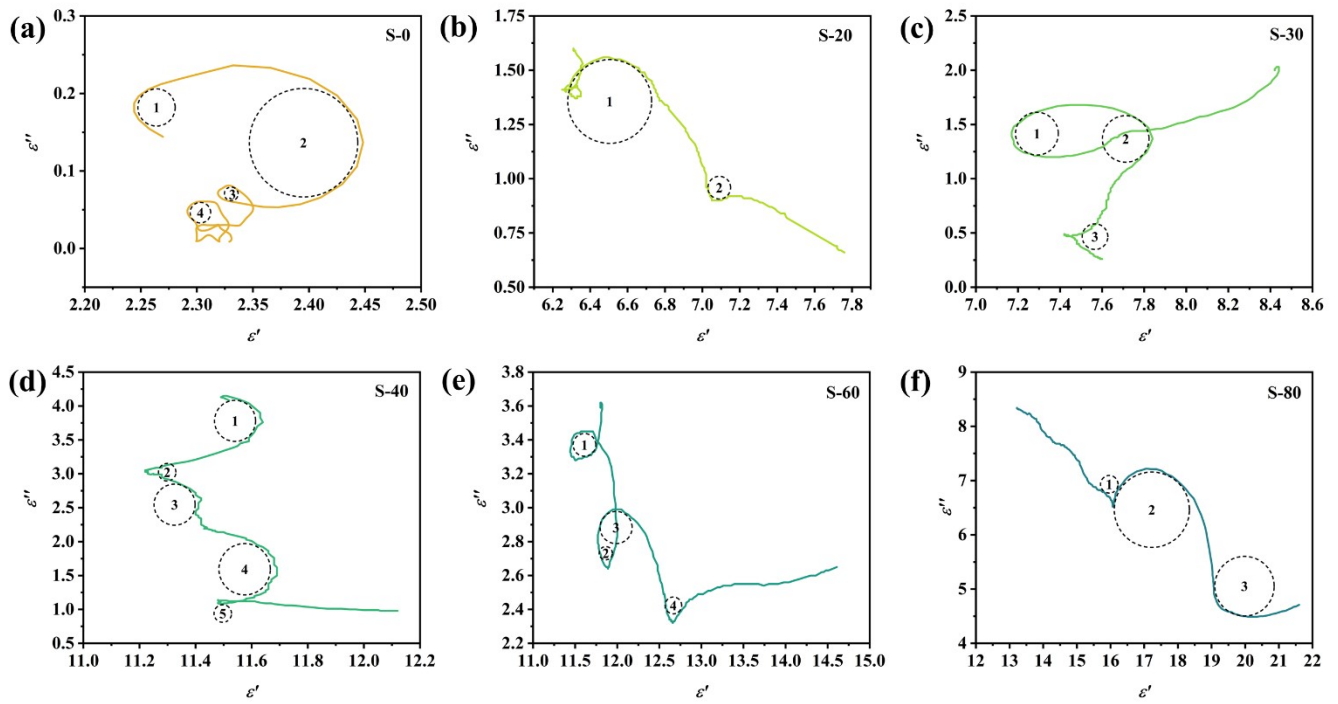


Figure S7. Cole-Cole semicircles of (a) S-0, (b) S-20, (c) S-30, (d) S-40, (e) S-60 and (f) S-80.

Table S1. Comparison of standard X-ray diffraction patterns of different spinel metal sulfides

crystallographic planes	This work $2\theta/^\circ$	Ni ₃ S ₄ (PDF#01-076-1813) $2\theta/^\circ$	Co ₃ S ₄ (PDF#01-072-8728) $2\theta/^\circ$	CoNi ₂ S ₄ (PDF#01-073-1297) $2\theta/^\circ$	NiCo ₂ S ₄ (PDF#01-073-1704) $2\theta/^\circ$	FeNi ₂ S ₄ (PDF#01-073-8669) $2\theta/^\circ$
(220)	26.8	26.639	26.786	26.734	26.841	26.619
(311)	31.6	31.346	31.520	31.459	31.585	31.322
(400)	38.3	38.029	38.243	38.168	38.323	38.000
(422)	47.4	47.036	47.306	47.210	47.407	46.999
(511)	50.4	50.079	50.369	50.266	50.477	50.039
(440)	55.3	54.873	55.196	55.081	55.316	54.829

Table S2. Comparison of Raman vibration modes in this work with references

Ref.	A_g/cm^{-1}	E_g/cm^{-1}	$T_{2g,1}/cm^{-1}$	$T_{2g,2}/cm^{-1}$
This work	376-383	249	346	300
1 (CoNi ₂ S ₄)	386	230	351	299
2 (FeS ₂)	372	337	-	-

Table S3. Peak areas (CPS.eV) and atomic% of Fe 3s (92.8 eV), Co 3s (101.1 eV) and Ni 3s (111.4eV) in XPS spectra of different samples

Samples	Fe 3s (CPS.eV)	Co 3s (CPS.eV)	Ni 3s (CPS.eV)	Fe 3s ratio (%)	Co 3s ratio (%)	Ni 3s ratio (%)
Sensitivity factors	0.548	0.606	0.671			
S-20	8498.88	7623.10	9625.94	36.55	29.64	33.81
S-30	9187.05	9805.61	11059.36	33.92	32.74	33.35
S-40	9211.39	9559.52	10473.28	34.88	32.73	32.39
S-60	8881.00	9494.73	11636.28	32.93	31.84	35.24
S-80	6204.80	7005.11	8101.82	32.39	33.07	34.54

Table S4. Peak areas (CPS.eV) of S 2p, O 1s O-M (530.1 eV) in XPS spectra of different samples

Samples	S 2p (CPS.eV)	O 1s O-M (CPS.eV)	S 2p after correction (CPS.eV)	O1s O-M after correction (CPS.eV)	FeCoNiS ₄ ratio (%)	FeCoNi (C ₃ H ₅ O ₃) ₂ ratio (%)
Sensitivity factors	1.881	2.881				
S-20	68806.70	100401.54	36579.85	34849.55	61.16	38.84
S-30	115706.32	75597.17	61513.20	26239.91	77.86	22.14
S-40	130895.62	67279.39	69588.31	23352.79	81.72	18.28
S-60	179418.38	13701.87	95384.57	4755.94	96.78	3.22
S-80	143573.18	3361.79	76328.11	1166.88	98.99	1.01

Table S5. Binding energies (eV) of S 2p and the calculated sulfur vacancy concentration for FeCoNiS_xG_y

Samples		S-20	S-30	S-40	S-60	S-80	
Area of S 2p (CPS.eV)	S ²⁻ 2p _{3/2}	161.5 (eV)	20304.4	34410.05	38360.23	49667.34	37859.55
	S ₂ ²⁻ 2p _{3/2}	162.5(eV)	11205.47	20597.57	23897.33	40744.5	32371.62
	S ²⁻ 2p _{1/2}	162.7 (eV)	10371.71	17577.03	19594.83	25370.63	19339.08
	S ₂ ²⁻ 2p _{1/2}	163.7 (eV)	5723.88	10521.47	12207.02	20812.74	16535.78
	S-S 2p _{3/2}	164.3 (eV)	5081.82	8936.93	9829.51	16266.93	13991.69
	S-S 2p _{1/2}	165.5 (eV)	2595.85	4565.08	5021.02	8309.33	7147.11
	SO ₃ ²⁻ 2p _{3/2}	167.1 (eV)	3500.93	5167.06	6137.81	5564.23	4479.48
	SO ₃ ²⁻ 2p _{1/2}	168.3 (eV)	1788.31	2639.39	3135.26	2842.27	2288.17
	SO ₄ ²⁻ 2p _{3/2}	168.9 (eV)	5450.27	7473.96	8414.43	6513.33	6328.19
	SO ₄ ²⁻ 2p _{1/2}	170.1 (eV)	2784.06	3817.78	4298.18	3327.08	3232.51
	S ²⁻ 2p _{1/2} /S _{total} (%)		15.07	15.19	14.97	14.14	13.47

Supplementary References

- 1 D. Larcher, G. Sudant, R. Patrice and J. M. Tarascon, *Chem. Mat.*, 2003, **15**, 3543-3551.
- 2 X. Liu, M. Gong, S. Deng, T. Zhao, J. Zhang and D. Wang, *J. Mater. Chem. A*. 2020, **8**, 113-1149.