

Supporting Information

Ultrathin $\text{Ba}_{0.75}\text{Sr}_{0.25}\text{TiO}_3$ nanosheets with highly exposed $\{001\}$ polar facets for high-performance piezocatalytic application

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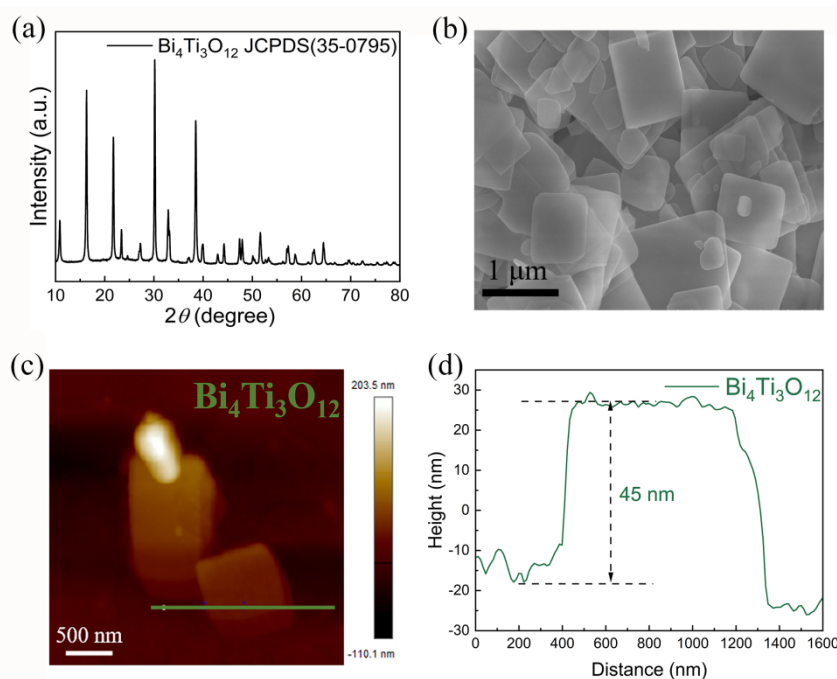


Fig. S1. (a) XRD pattern, (b) SEM image, and (c) AFM image of the $\text{Bi}_4\text{Ti}_3\text{O}_{12}$ nanoplates as synthesized by molten salt method. (d) Height profile of a single $\text{Bi}_4\text{Ti}_3\text{O}_{12}$ nanoplate along the green line in the AFM image.

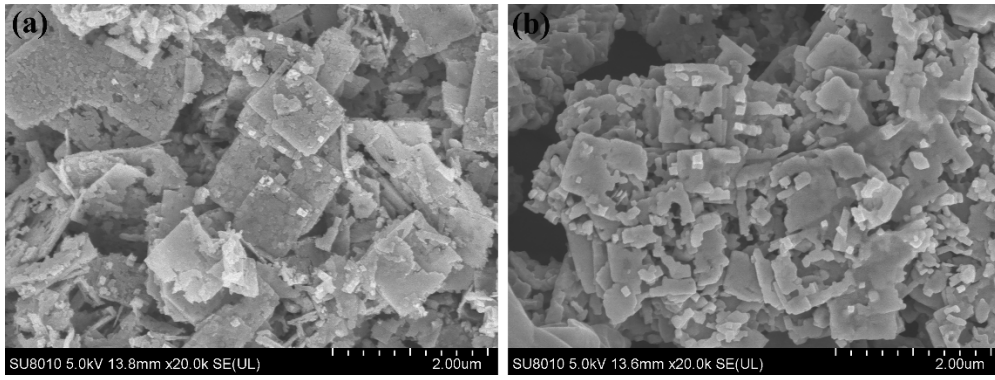


Fig. S2. SEM images of the synthesized (a) $\text{Ba}_{0.85}\text{Sr}_{0.15}\text{TiO}_3$ and (b) $\text{Ba}_{0.7}\text{Sr}_{0.3}\text{TiO}_3$.

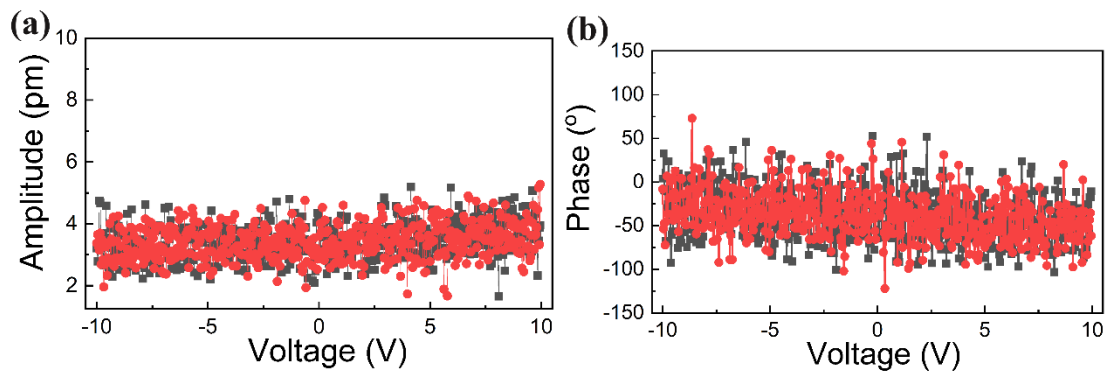


Fig. S3. In-plane (a) amplitude butterfly loop and (b) phase hysteresis loop of the prepared $\text{Ba}_{0.75}\text{Sr}_{0.25}\text{TiO}_3$.

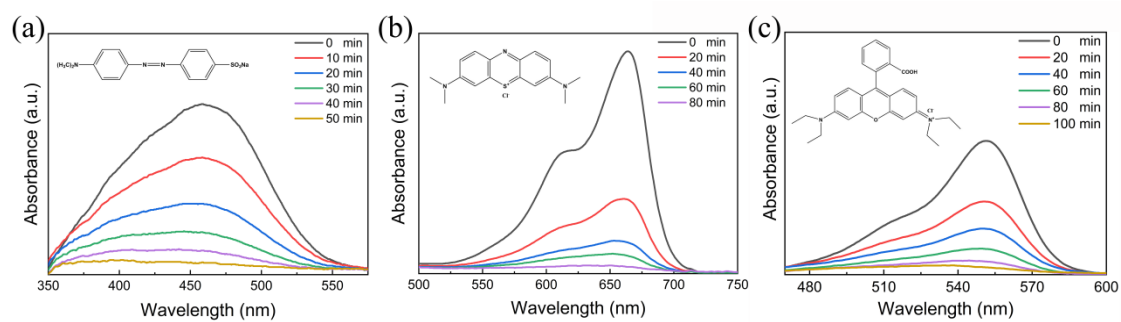


Fig. S4. Evolution of UV-vis absorption spectra of (a) MO, (b) MB and (c) RhB solutions during the piezocatalytic degradation using $\text{Ba}_{0.75}\text{Sr}_{0.25}\text{TiO}_3$ nanosheets as piezocatalyst.

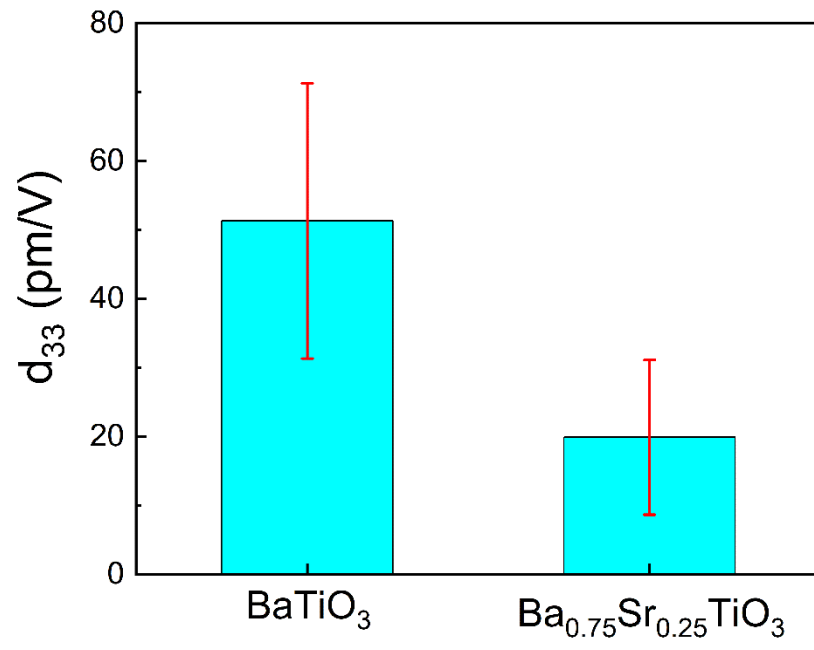


Fig. S5. d_{33} of the synthesized BaTiO_3 and $\text{Ba}_{0.75}\text{Sr}_{0.25}\text{TiO}_3$.

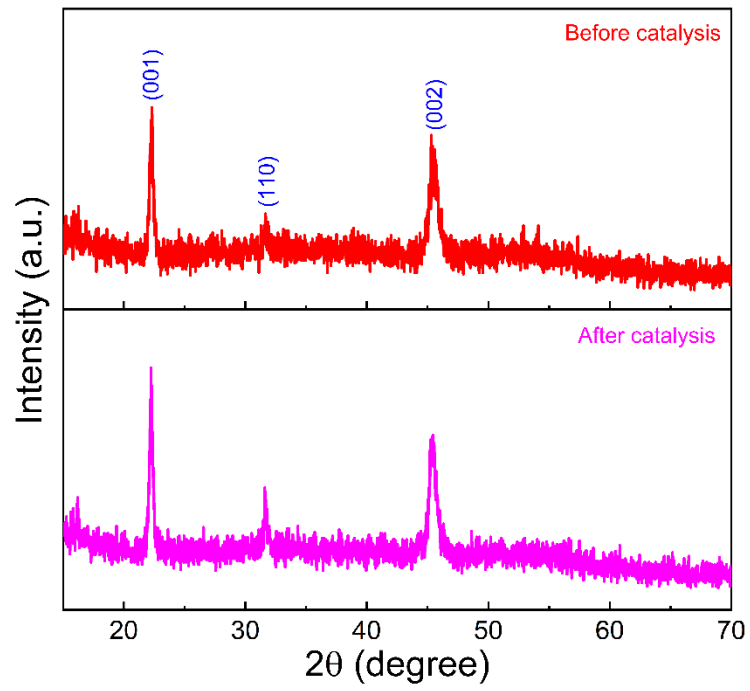


Fig. S6. XRD patterns of the $\text{Ba}_{0.75}\text{Sr}_{0.25}\text{TiO}_3$ before and after cyclic test.

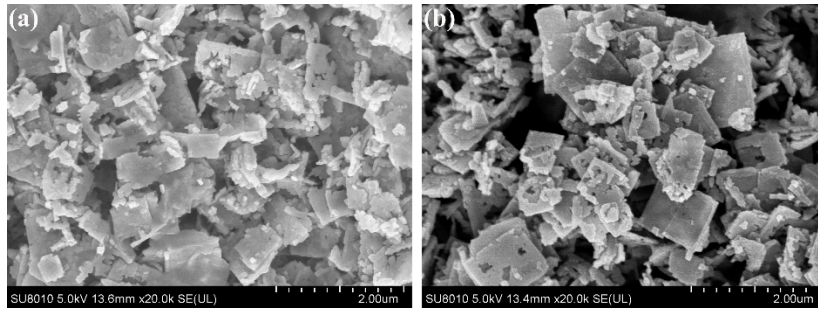


Fig. S7. SEM images of the (a) pristine and (b) recycled $\text{Ba}_{0.75}\text{Sr}_{0.25}\text{TiO}_3$.

DFT Calculation

The spin-polarized density functional theory (DFT) calculations were conducted by using the projected augmented wave (PAW) pseudopotentials [1], as implemented in the Vienna ab-initio Simulation Package (VASP) [2,3]. The exchange correlation function of Perdew-Burke-Ernzerhof (PBE) form was employed [4], with a plane wave basis cut-off energy set to 450 eV. The convergence criteria were set to be 10^{-4} in energy and 0.02 eV/Å in force. The BaTiO₃ and Ba_{0.75}Sr_{0.25}TiO₃ with the BaO-terminated (001) surface was modelled by a seven-layer slab, where the atoms in the bottom three layers were fixed at the theoretical bulk lattice positions. The Brillouin zone was sampled by a Monkhorst–Pack k-point mesh of 4×4×1 grid and a vacuum layer of 15 Å was employed to avoid interactions of neighboring images. The van der Waals (vdW) interactions between BaTiO₃ and reactants were treated using the Grimme’s D3-type of the semiempirical method [5].

Adsorption energies (E_{ads}) were calculated by using the following equation.

$$E_{\text{ads}} = E_{\text{total}} - E_{\text{substrate}} - E_{\text{reactant}}$$

where E_{total} is the total energy of adsorbed systems, $E_{\text{substrate}}$ and E_{reactant} are the energies of the substrate BaTiO₃/Ba_{0.75}Sr_{0.25}TiO₃ and reactants, respectively. According to the definition, a negative E_{ads} indicates an energetically stable configuration.

References

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3. G. Kresse, J. Hafner, Phys. Rev. B 49 (1994) 14251-14269.
4. J. P. Perdew, K. Burke, M. Ernzerhof, Phys. Rev. Lett. 77 (1996) 3865-3868.
5. S. Grimme, J. Antony, S. Ehrlich, H. Krieg, J. Chem. Phys. 132 (2010) 154104.