

Supplementary Information for

Photoemission Spectroscopy and Microscopy for Ta@Si₁₆ Superatom and their Assembled-Layers

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Supplementary Note S1. Definition of nominal coverage of Ta@Si₁₆ Superatom (SA).

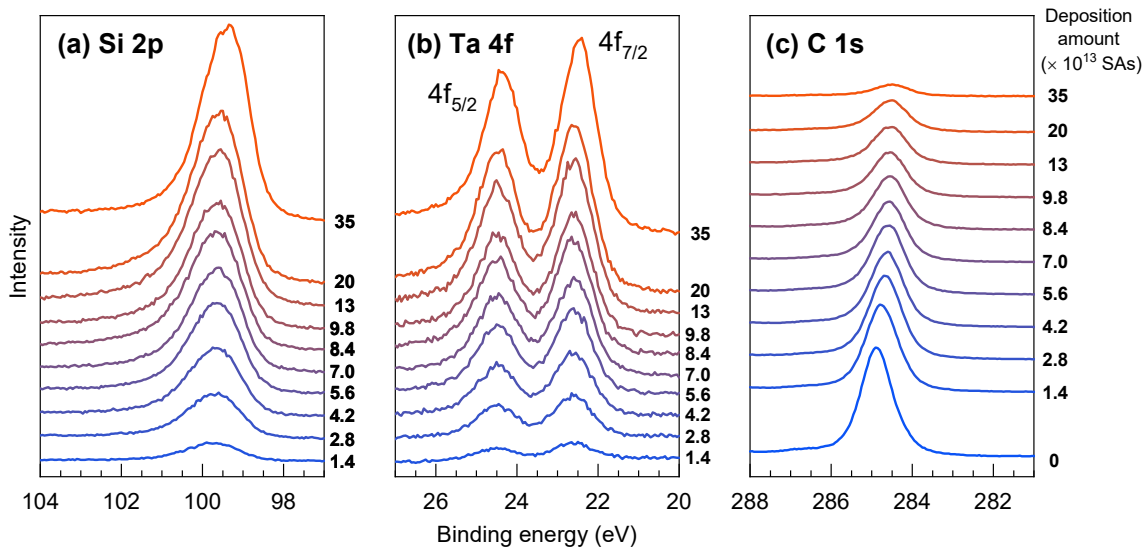
In this note, we estimate the nominal coverage of Ta@Si₁₆ SAs. Assuming the size of SA (0.8~0.9 nm in diameter) and the deposition area (6 mm ϕ , ca. 2.8×10^{13} nm²), the amount of deposition for full coverage (monolayer; ML) corresponds to 5×10^{13} SAs. According to the STM result for Ta@Si₁₆/C₆₀ at very low coverage (much less than 1 ML),^{1,2} individual Ta@Si₁₆ SAs are immobilised on the C₆₀(111) surface. It is considered that deposited SAs spread uniformly over the C₆₀ surface without self-organization (e.g. island formation) up to full coverage (1 ML). This is because the alkaline-like Ta@Si₁₆ SA favorably engages in charge transfer interaction with the n-type organic substrate of C₆₀, which has been discussed in the XPS analysis at a fixed Ta@Si₁₆ SA coverage (0.6 ML).³⁻⁷ After completing the ML, it is considered that the Ta@Si₁₆ SA film grows on the ML surface in a quasi-layer-by-layer growth manner. The above coverage definition and growth manner of the SA film are justified by the step-by-step measurement of XPS spectra.

Supplementary Fig. S1 (a)-(c) shows XPS spectra around the (a) Si 2p, (b) Ta 4f, and (c) C 1s core levels at various deposition amounts (0 – 35×10^{13} clusters, denoted in the right) of Ta@Si₁₆ SAs on a C₆₀ substrate. Both core levels of Ta 4f and Si 2p are clearly resolved by the Ta@Si₁₆ deposition. At the initial deposition (1.4×10^{13} Clusters), the intensity ratio between the Ta 4f and Si 2p core levels is consistent with the atomic composition ratio of 1 : 16 for Ta@Si₁₆ SA.^{3,4} Furthermore, the XPS peaks of Si 2p and Ta 4f are well-reproduced by considering uniform chemical components within the experimental peak broadening, which demonstrates the successful immobilisation of Ta@Si₁₆ on the C₆₀ substrate while preserving its caged structure.

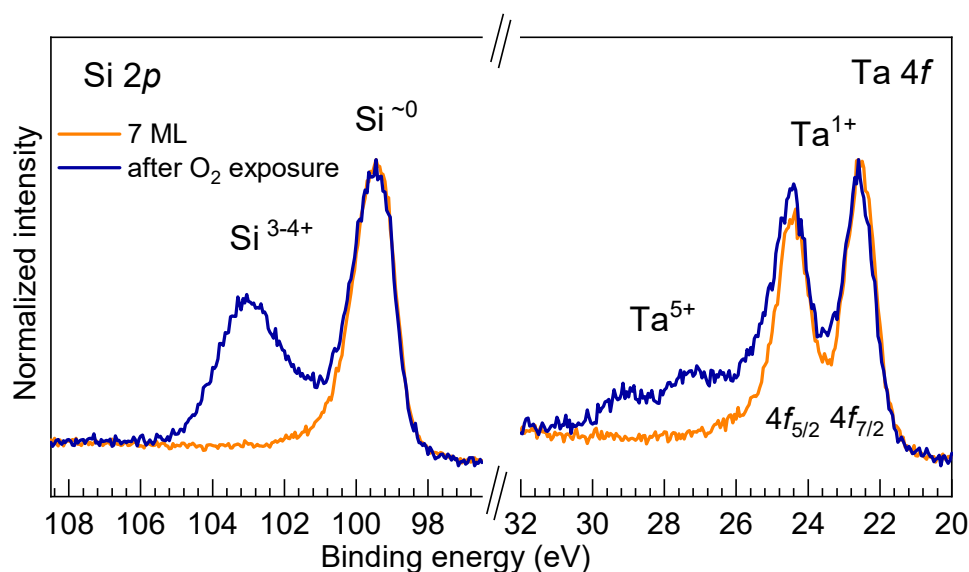
At higher amounts of Ta@Si₁₆ SA deposition of more than 5×10^{13} SAs (i.e., > 1 ML, the intensity ratios, peak widths, and energies remain mostly the same at any deposition amount, suggesting that the SA film formed on the first Ta@Si₁₆ layer, preserving the chemical composition and geometric structure of the SA. The intensities of the Si 2p- and Ta 4f-derived peaks increase with the deposition amount, whereas the XPS intensity of the C 1s peak originating from C₆₀ substrate (Fig. 2 (c)) decreases with SA

deposition at higher coverages, and is mostly quenched at the highest coverage (35×10^{13} SAs) because of the overlayered SA film. The decrease in the C 1s signal intensity with increasing the deposition amount is associated with uniform assembly of SAs on the substrate; otherwise (e.g. 3D growth), the XPS signal from substrates should remain even at higher coverage. From these results, it can be concluded that the Ta@Si₁₆ film grows on an organic substrate in a quasi-layer-by-layer growth manner.

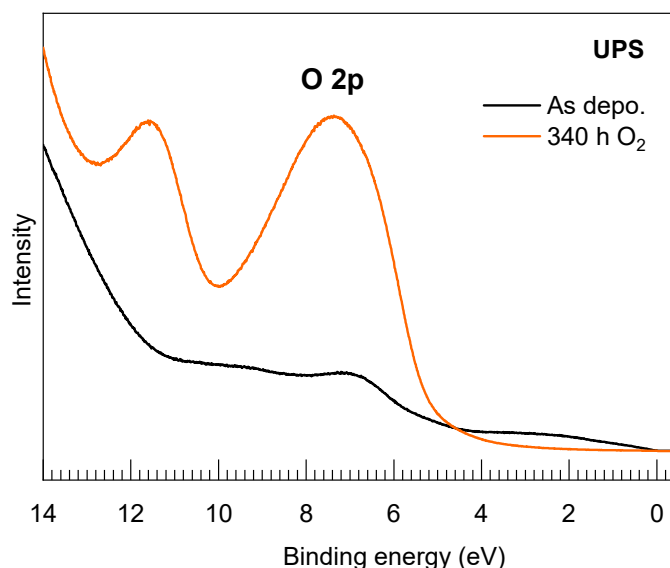
The C 1s peak in Fig. S1(c) shifts slightly towards higher binding energy with increasing Ta@Si₁₆ coverage up to $\sim 5 \times 10^{13}$ SAs (1 ML), which can be explained by the charge transfer interaction between Ta@Si₁₆ SA and C₆₀ molecule.⁴ The saturation of the C 1s peak shift justifies the abovementioned ML definition estimated by the amount of the SA deposition into a deposition area (6 mm ϕ), namely, 1 ML $\approx 5 \times 10^{13}$ SAs, because further deposition on the 1 ML Ta@Si₁₆ film does not affect the charge state of underlying C₆₀.



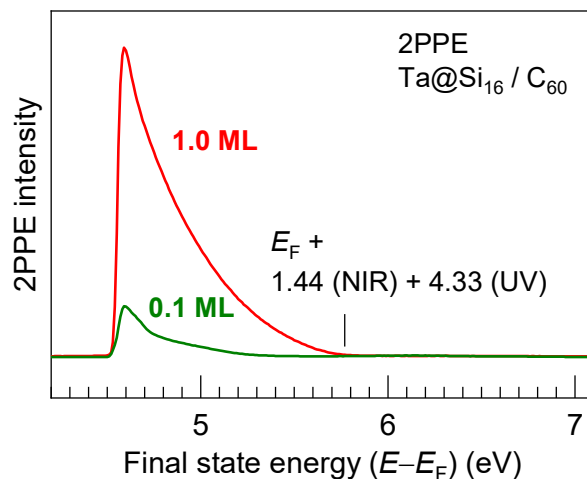
Supplementary Fig. S1. Coverage dependence of the XPS spectra of Ta@Si₁₆ SAs on C₆₀ substrate. The XPS spectra are near the (a) Si 2p, (b) Ta 4f, and (c) C 1s core levels. The deposition amounts, in units of 10^{13} SAs, are presented on the right side of each figure. Both Ta 4f and Si 2p peaks increase with SA coverage, while their widths can be reproduced by single chemical components.^{3,4} The C 1s signal is attenuated by SA deposition, suggesting the formation of a uniform Ta@Si₁₆ SA film on the C₆₀ substrate.



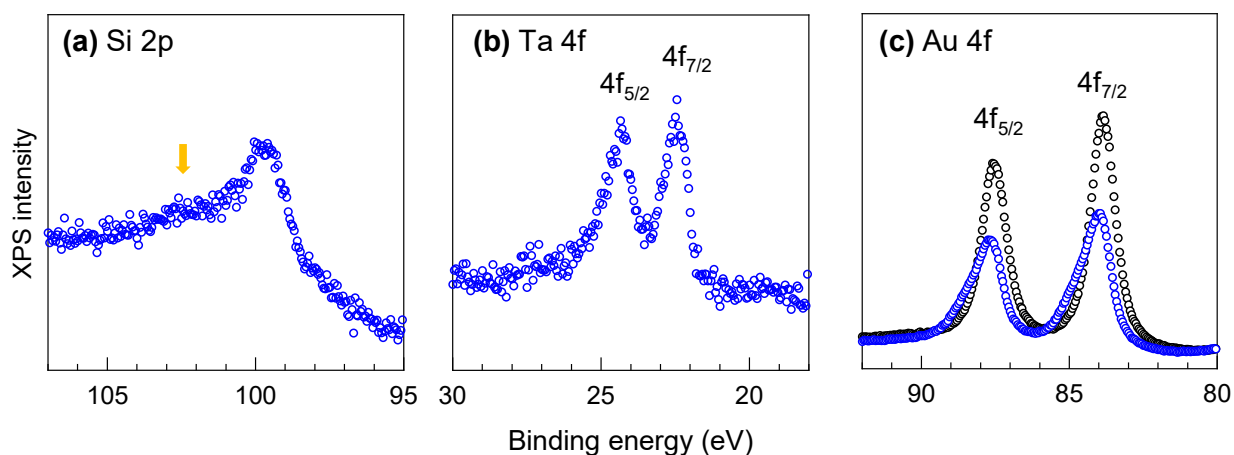
Supplementary Fig. S2. XPS spectra for Ta@Si₁₆ thick film (7 ML) before and after exposing oxygen (1atm for 40 hours). The Si 2p derived peak shows Si³⁻⁴⁺-derived chemical components owing to the oxidation of the Ta@Si₁₆. Ta atom is also oxidized in part, while it is rather robust toward oxidative reaction because of the metal encapsulation.



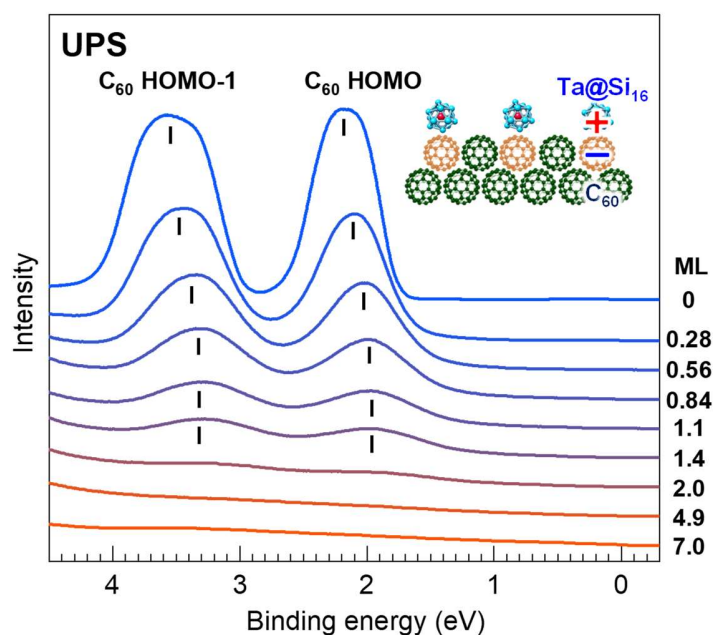
Supplementary Fig. S3. UPS spectra (raw data) for Ta@Si₁₆ thick film (7 ML) before and after exposing oxygen (1atm for 40 hours). After O₂ exposures, strong O 2p derived feature is observed at the binding energy of 7 eV, showing the oxidative reaction of Ta@Si₁₆ SAs. UPS is more sensitive to the surface oxidations rather than that for the XPS (Fig. S2) because of the shorter mean-free-path of photoelectrons.



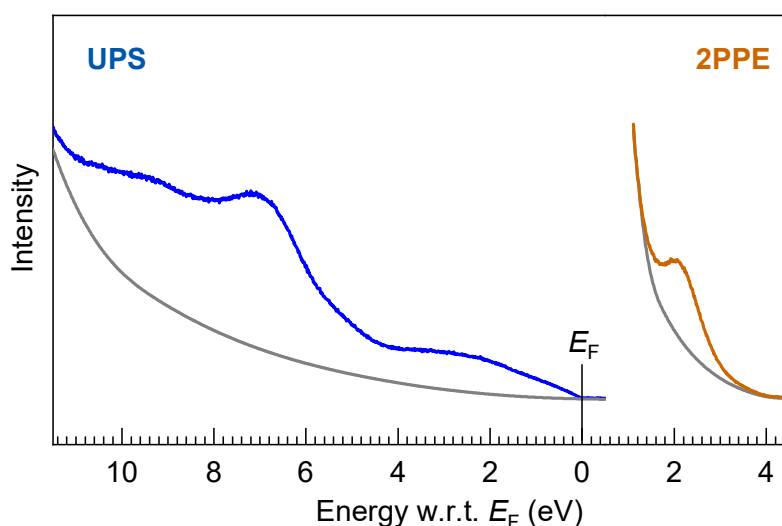
Supplementary Fig. S4. Two-color 2PPE spectra for Ta@Si₁₆ SAs on C₆₀. The photon energies of the NIR and UV lasers are 1.44 eV and 4.33 eV, respectively. The 2PPE intensity below the Fermi edge (1.44 eV + 4.33 eV in the final state energy) is enhanced by the SA deposition. It is important to note that no measurable photoemission is detected in the absence of SA deposition.



Supplementary Fig. S5. XPS spectra for Ta@Si₁₆ (0.6 ML) deposited on an Au(111) substrate. (a) Si 2p, (b) Ta 4f, and (c) Au 4f core levels. Si 2p derived peak shows multiple chemical components (vertical arrow), suggesting that the deposited Ta@Si₁₆ SA is chemically affected by the interaction with Au (111) substrate. The Au 4f peaks also differ from that of a clean Au substrate, showing the chemical modification of Au side by the deposition of Ta@Si₁₆ SAs.



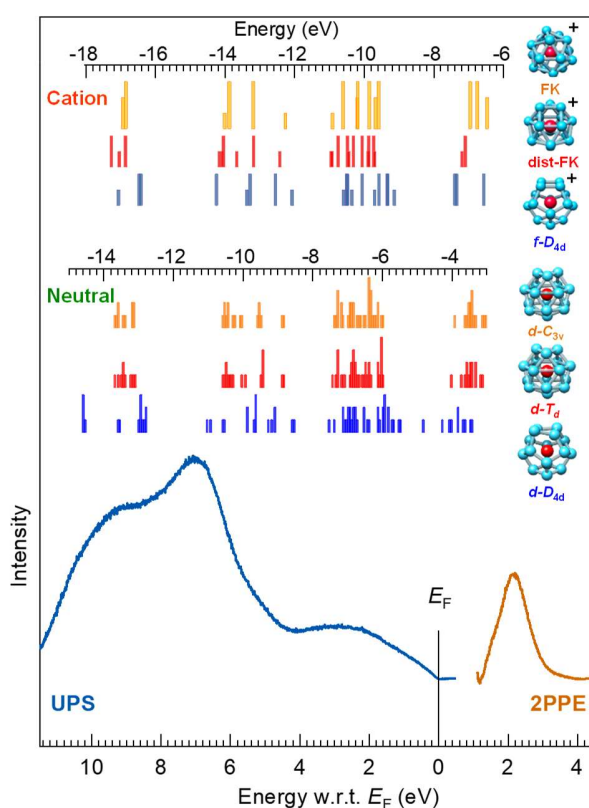
Supplementary Fig. S6. Coverage-dependent UPS spectra ($h\nu = 21.22$ eV) of Ta@Si₁₆ SA on C₆₀. The C₆₀-derived HOMO and HOMO-1 peaks are attenuated by SA deposition up to 7 ML, with a slight upshift of 0.2 eV, suggesting that a charge transfer interaction occurs between SA and C₆₀ at the interface. The C₆₀-derived features completely disappear in multilayered SA films thicker than 2 ML, indicating that fully covered SA films are formed. The inset shows a schematic illustration of the charge transfer complexation of Ta@Si₁₆ SAs with C₆₀ molecules.



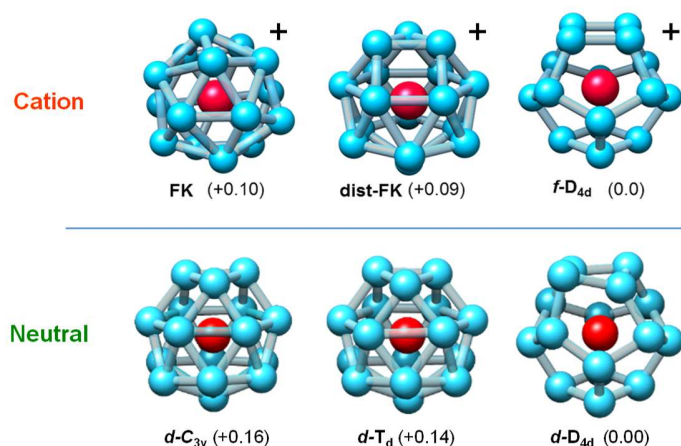
Supplementary Fig. S7. Raw UPS/2PPE spectral data and background subtraction. Appropriate backgrounds (grey) with spline curves are subtracted from raw data of UPS and 2PPE spectra (colored spectra).

Note S2. Computational methods.

The electronic structures of optimized Ta@Si₁₆ neutrals and cations were calculated using the Gaussian 09 (cations)⁸ and Gaussian 16 (neutrals)⁹ programs, at several different levels of the density functional theory. The Ahlrichs' basis sets def-TZVP (for cations)¹⁰ or def2-TZVP (for neutrals) were utilized for Si and Ta. For Ta, a 60-electron relativistic effective core potential was also employed. The hybrid exchange-correlation functional of B3LYP¹¹ and pure functional of PBEPBE,¹² and hybrid PBE0 functional¹³ and Slater-type TZ2P basis-set¹⁴ were utilized. These calculations were carried out for the optimized structures of cationic M@Si₁₆ isomers at the PBEPBE/def-TZVP level (Gaussian 09). The details of the coordinates are described elsewhere.¹⁵



Supplementary Fig. S8. UPS, 2PPE, and calculated results. (bottom) UPS and 2PPE spectra of a thick Ta@Si₁₆ SA film (7 ML). The energy onset at E_F in the UPS spectrum indicates Ohmic electronic conduction of the Ta@Si₁₆ SA film. The broad valence and conduction band features at $E_F - 2.6$ eV, -7.0 eV, and -9.6 eV in UPS and at $E_F + 2.1$ eV in 2PPE are observed. (upper) Calculated energy levels of free Ta@Si₁₆ SAs in a cationic state (top three) and in a neutral state (lower three) for optimized isomeric structures, where the total energies in each charge state vary minimally among the three structures (see also Fig. S9): Frank-Kasper (FK), distorted FK, and f -D_{4d} for cations, and d -C_{3v}, d -T_d, and f -D_{4d} for neutrals, as shown in the right part of the figure. The energy levels align moderately with the experimental observations (lower spectra), whereas the energies of unoccupied levels are underestimated in the present calculations.



Supplementary Fig. S9. Optimized structures of cationic and neutral Ta@Si₁₆. The symmetries and energies relative to the most stable isomer (in eV) are indicated at the bottom of each structure.

Supplementary Table S1. Eigenvalues of calculated results (eV) for neutral Ta@Si₁₆ and their degeneracies. The values in blue and red correspond to HOMO and LUMO levels, respectively.

FK	Degen.	Dist.-FK	Degen.	$f\text{-D}_{4d}$	Degen.
-20.70	1	-20.71	1	-19.80	1
-18.98	3	-19.05	2	-18.67	1
-16.97	2	-18.80	1	-18.09	2
-16.89	3	-17.14	2	-17.02	1
-14.05	1	-16.91	1	-16.42	2
-13.92	3	-16.73	2	-16.38	2
-13.23	3	-14.03	1	-14.20	2
-12.31	1	-13.96	1	-13.33	1
-10.95	1	-13.91	2	-13.23	2
-10.65	3	-13.53	1	-12.51	2
-10.24	2	-13.05	2	-12.03	1
-10.23	3	-12.30	1	-10.54	1
-9.90	3	-10.82	1	-10.47	1
-9.72	2	-10.80	1	-10.46	2
-9.63	3	-10.62	2	-10.44	2
-6.99	3	-10.35	2	-10.31	1
-6.78	3	-10.32	1	-10.01	2
-6.51	2	-10.18	2	-9.65	1
-5.70	3	-9.92	2	-9.53	2
-5.70	1	-9.76	1	-9.29	2
-5.42	3	-9.74	2	-9.28	2
-4.05	3	-9.59	2	-9.09	1
-3.96	2	-9.57	1	-7.35	2
-3.94	1	-7.06	1	-7.30	2
-3.63	3	-6.96	2	-6.52	2
		-6.94	1	-5.66	2
		-6.89	2	-5.36	1

-6.50	2	-5.28	2
-5.81	2	-4.98	1
-5.75	1	-4.95	2
-5.46	2	-4.49	1
-5.39	1	-4.48	2
-5.13	1	-4.11	1
-4.29	2	-3.82	1
-4.07	1	-3.60	1
-3.98	1	-3.46	2
-3.86	2	-2.80	2
-3.56	1	-2.65	2
-3.45	2		

Supplementary Table S2. Eigenvalues of calculated results (eV) for neutral Ta@Si₁₆ and their spin orbitals. The values in blue and red correspond to HOMO and LUMO levels, respectively.

<i>d-D_{4d}</i>	Spin	<i>d-T_d</i>	Spin	<i>d-C_{3v}</i>	Spin
-14.58	α	-13.66	α	-13.65	α
-14.57	α	-13.60	β	-13.61	α
-14.56	β	-13.46	α	-13.60	β
-14.54	β	-13.44	α	-13.57	β
-13.59	α	-13.41	β	-13.42	α
-13.55	β	-13.39	β	-13.37	β
-12.99	α	-13.24	α	-13.18	α
-12.94	α	-13.18	β	-13.15	α
-12.93	β	-13.11	α	-13.13	β
-12.92	β	-13.09	β	-13.12	β
-12.85	α	-10.59	α	-10.56	α
-12.81	β	-10.53	α	-10.51	α
-12.79	α	-10.49	β	-10.50	β
-12.76	β	-10.46	β	-10.47	α
-11.04	α	-10.42	α	-10.42	β
-10.94	β	-10.37	β	-10.40	β
-10.56	α	-10.32	α	-10.27	α
-10.52	β	-10.27	β	-10.23	β
-9.88	α	-10.01	α	-10.06	α
-9.86	β	-9.94	β	-10.00	β
-9.70	α	-9.49	α	-9.56	α
-9.65	β	-9.44	β	-9.54	α
-9.64	α	-9.44	α	-9.53	β
-9.62	β	-9.41	β	-9.49	β
-9.26	α	-8.88	α	-8.87	α
-9.16	β	-8.83	β	-8.82	β
-9.08	α	-7.41	α	-7.37	α
-9.07	β	-7.33	α	-7.32	α
-8.55	α	-7.27	β	-7.29	β
-8.51	β	-7.26	β	-7.26	β
-7.55	α	-7.25	α	-7.26	α
-7.36	β	-7.18	β	-7.17	β
-7.14	α	-7.14	α	-7.17	α
-7.11	α	-7.10	β	-7.14	β
-7.09	β	-6.94	α	-6.97	α
-7.05	β	-6.88	β	-6.92	α
-6.99	α	-6.87	α	-6.92	α
-6.97	β	-6.84	α	-6.87	β
-6.94	α	-6.83	α	-6.87	β
-6.91	β	-6.82	β	-6.83	β
-6.86	α	-6.78	β	-6.80	α
-6.83	β	-6.77	β	-6.72	β
-6.81	α	-6.70	α	-6.66	α
-6.78	β	-6.67	β	-6.64	β
-6.78	α	-6.58	α	-6.54	α

-6.70	β	-6.51	β	-6.48	β
-6.55	α	-6.49	α	-6.40	α
-6.53	β	-6.45	β	-6.40	α
-6.45	α	-6.42	α	-6.36	α
-6.37	β	-6.36	α	-6.36	β
-6.13	α	-6.35	β	-6.32	β
-6.11	β	-6.32	β	-6.32	α
-6.06	α	-6.13	α	-6.31	β
-5.98	α	-6.10	β	-6.28	β
-5.97	β	-6.09	α	-6.20	α
-5.93	α	-6.05	α	-6.17	β
-5.92	β	-6.04	β	-6.15	α
-5.91	β	-6.02	β	-6.11	β
-5.84	α	-6.01	α	-6.02	α
-5.81	β	-5.97	β	-5.97	β
-5.72	α	-4.01	α	-3.93	α
-5.68	β	-3.71	β	-3.63	β
-5.52	α	-3.60	α	-3.55	α
-5.48	β	-3.58	α	-3.51	α
-4.82	α	-3.57	β	-3.51	β
-4.29	β	-3.55	β	-3.48	β
-4.08	α	-3.50	α	-3.46	α
-4.04	β	-3.46	α	-3.44	α
-3.83	α	-3.45	β	-3.42	β
-3.81	β	-3.40	β	-3.41	β
-3.67	α	-3.35	α	-3.36	α
-3.61	β				
-3.48	α				
-3.43	β				
-2.91	α				

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