Supplementary Information

Computational exploration of chemically ordered and disordered M-sites in M'₂M''B₂ and M'₄M''B₃ compounds

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SQS convergence

The disordered structures used herein modeled by SQS structures to simulate solid solution on the Msites were derived through a convergence of the energy when increasing the cell size. Below is the convergence for the P4/mbm (M'₂M''B₂ prototype) and I4/mcm (M'₄M''B₃ prototype) structures where the energy is displayed as ΔE_0 which is the energy with respect to the energy of the largest cell size. The convergence was performed for the M' = Cr and M'' = W to replicate the scenario in Ref. [1].



Figure S1. Energy convergence with respect to the energy of the cell with the greatest size for the SQS structures with disordered M-sites versus the number of atoms of the generated supercell.

The energy difference with respect to the cell of the greatest number, ΔE_0 , is seen to vary and remain constant for both considered symmetries as the cells are increased. The *P4/mbm* symmetry yielded an energy difference of 0.4 meV/atom when occupied by 90 atoms which was the considered cell for further analysis. Similarly, the cell with 80 atoms yielded the same energy as the cell with 320 atoms for the I4/mbm symmetry.

Complete phase stability for M'₂M''B₂

Complete heatmap for the M'₂M''B₂ composition with M' and M'' being occupied by elements ranging from period 3 to 6 arranged after their atomic number. The colors range from red (metastable) to blue (stable) whereas grey illustrates phases with $\Delta H_{cp} > 150$ meV/atom. The symbols of the elements illustrate the arrangement of the M-sublattice with a filled representing ordered and unfilled disordered. Experimentally known phases are distinguished by a black edge and phases deemed uncertain with a grey square.



Figure S2. Complete phase stability heatmap at 2000 K where stability is depicted as the color ranging from red (metastable) to blue (stable). Phases with $\Delta H_{cp} > 150 \text{ meV/atom}$ are colored grey and denoted unstable.

Μ'	M''	ΔH_{cp} or ΔG_{cp} (meV/atom)	M' and M" arrangement	Set of most competing phases
Та	V	-56.8318	Ordered	$Ta_3B_2 + V_3B_2$
Та	Cr	-53.7503	Ordered	Cr + TaB
Та	Мо	-52.6921	Ordered	Mo + TaB
Та	Tc	-44.9297	Ordered	$Tc_7B_3 + TaTc + TaB$
V	V	-44.5126	Ordered	VB + V
Та	Re	-43.1232	Ordered	Re + TaB
Ti	Re	-40.9885	Ordered	$ReTi + TiB_2$
Та	Ti	-40.6692	Disordered	$Ta_3TiB_4 + Ta_3B_2 + Ti$
Nb	Tc	-40.4089	Ordered	Tc + NbB
Nb	Ti	-37.9343	Disordered	$NbTiB_2 + Nb$
Мо	Fe	-37.2176	Ordered	$FeMo_4B_3 + Fe_2B + MoB$
Ti	Mn	-34.807	Ordered	$Ti_3B_4 + Mn_2Ti$
W	Mn	-34.7966	Ordered	$Mn_3WB_4 + MnW_4B_3 + Mn_2B$
Та	Mn	-33.8266	Ordered	$Mn_2Ta + Mn_2B + TaB$
Та	Nb	-33.0403	Disordered	$Ta_3B_2 + Nb_2TaB_2 \\$
Nb	Та	-32.8848	Disordered	$Nb_3B_2 + NbTa_2B_2$
Мо	Mn	-31.1031	Ordered	$MnMo_4B_3 + MoB + Mn_2B$
Nb	V	-30.1405	Ordered	$Nb_3B_2+V_3B_2\\$
Ti	W	-26.2655	Disordered	TiB + W
V	Mo	-25.7328	Disordered	Mo + VB
Tc	Al	-25.0545	Ordered	$Al_4Tc + AlTc_2 + TcB_2$
Nb	Mo	-24.8286	Ordered	Mo + NbB
V	Cr	-23.3763	Disordered	$Cr_3V_9B_8 + Cr_2VB_2$
Zr	Tc	-23.3644	Ordered	$Zr + Tc_2Zr + ZrB_2$
Та	Та	-20.4912	Ordered	Ta + TaB
Та	W	-18.534	Ordered	TaB + W
V	Mn	-16.9839	Ordered	$Mn_2B + MnV + VB$
Nb	Re	-16.9464	Ordered	Re + NbB
Nb	Nb	-14.7118	Ordered	Nb + NbB
V	W	-14.3244	Disordered	VB + W
Ti	Tc	-14.0164	Ordered	$TcTi + TiB_2$
Ti	Mo	-13.2575	Disordered	$Ti_{3}B_{4}+TiB+Mo_{3}Ti$
Mo	V	-12.8257	Disordered	$MoV_2B_2 + Mo_4VB_3 + MoB$
W	V	-11.4814	Disordered	$VW_4B_3 + VB$
W	Mo	-11.3594	Disordered	$MoW_4B_3 + MoB$
Nb	Cr	-11.1174	Ordered	Cr + NbB
Mo	Cr	-8.22328	Ordered	$CrMo_4B_3 + CrB$
Mo	W	-5.43422	Disordered	MoB + W
Cr	V	-5.29457	Disordered	$Cr_4VB_3 + VB$
Hf	Re	-3.69273	Ordered	$Hf_9Re_4B+Hf_{21}Re_{25}+HfB_2$
W	Cr	-2.91818	Ordered	$CrW_4B_3 + CrB$
Cr	Mn	-2.78801	Disordered	$Cr_4MnB_3 + CrB + Mn_2B$
V	Tc	-2.15403	Disordered	$Tc_7B_3 + TcV + VB$
Mo	Та	-1.64373	Disordered	$MoTa_2B_2 + Mo + MoB$
Mo	Ti	-1.04939	Disordered	$MoTiB_2 + Mo$

Table S1. Gibb's free energy, M' and M'' arrangement and the set of most of the competing phases for the predicted stable $M'_2M''B_2$ phases with symmetry *P4/mbm* are displayed in Figure 2 at 2000 K.

Figure S3 to S27 shows phonon dispersion spectrum for stable $M'_2M''B_2$ phases with symmetry *P4/mbm* and with M' and M'' being ordered. Note that Tc-based phases have intentionally been left out since Tc is very unstable and can only be made artificially. The absence of imaginary frequencies indicates that the ordered MAB phase is dynamically stable. One $M'_2M''B_2$ phases is found dynamically unstable, i.e., Ta₂MnAlB₂, whereas the other 25 are dynamically stable.





Figure S4. Phonon dispersion spectra and phonon DOS for Ta₂CrB₂.







Figure S13. Phonon dispersion spectra and phonon DOS for Ta₂MnB₂. Presence of imaginary frequencies indicate dynamical unstable structure.







Figure S15. Phonon dispersion spectra and phonon DOS for Mo₂MnB₂.





S9











Complete phase stability for M'4M''B₃

Complete heatmap for the M'₄M''B₃ composition with M' and M'' being occupied by elements ranging from period 3 to 6 arranged after their atomic number. The colors range from red (metastable) to blue (stable) whereas grey illustrates phases with $\Delta H_{cp} > 150 \text{ meV/atom}$. The symbols of the elements illustrate the arrangement of the M-sublattice with a filled representing ordered and unfilled disordered. Experimentally known phases are distinguished by a black edge.



Figure S28. Complete phase stability heatmap at 2000 K where stability is depicted as the color ranging from red (metastable) to blue (stable). Phases with $\Delta H_{cp} > 150 \text{ meV/atom}$ are colored grey and denoted unstable.

Μ'	M''	$\Delta H_{\rm cp} \text{ or } \Delta G_{\rm cp} \text{ (meV/atom)}$	M' and M" arrangement	set of most competing phases
Cr	V	-42.2856	Disordered	$Cr_2B + VB$
W	Mn	-39.2617	Ordered	$W_2B + MnW_2B_2 \\$
Fe	Mn	-34.1583	Disordered	$Fe_2B + FeB + Mn_2B$
Mo	V	-32.0671	Ordered	Mo + MoB + VB
W	Cr	-31.1476	Ordered	$W_2B + CrB$
W	V	-29.7659	Ordered	$W_2B + VB$
Mo	Mn	-26.3861	Ordered	$Mo + MoB + Mn_2B$
Мо	Fe	-10.1717	Ordered	Fe + Mo + MoB
Mo	Cr	-10.0398	Ordered	$CrMo_2B_2 + Mo + MoB$
Tc	Fe	-8.93906	Ordered	$Tc_7B_3 + FeB + TcB_2$
Cr	Tc	-6.71052	Disordered	$Cr_5B_3 + Tc_7B_3 + CrB$
Мо	W	-5.75225	Disordered	Mo + MoB + W
Cr	Cr	-5.1054	Ordered	$Cr_2B + CrB$
Tc	Co	-1.02338	Ordered	$Tc_7B_3 + TcB_2 + CoB$
Cr	W	-0.25586	Disordered	$Cr_5B_3 + CrB + W$
Cr	Mo	-0.24018	Disordered	$Cr_5B_3 + Cr_2B + CrMo_2B_2$

Table S2. Gibb's free energy, M' and M'' arrangement and the set of most competing phases for the predicted stable $M'_4M''B_3$ phases with symmetry I4/mcm are displayed in Figure 3 at 2000 K.

Figure S29 to S35 shows phonon dispersion spectrum for stable $M'_4M''B_3$ phases with symmetry *I4/mcm* and with M' and M'' being ordered. We have intentionally left out Tc-based phases since Tc is very unstable and can only be made artificially. The absence of imaginary frequencies indicates that the ordered MAB phase is dynamically stable. Only 3 out of 7 $M'_4M''B_3$ phases considered fulfills this. Note that only non-magnetic and ferromagnetic spin configurations have been considered while a more indepth analysis of phonon dispersion for various anti-ferromagnetic spin configuration may lead to dynamically stable structures.





Figure S32. Phonon dispersion spectra for W₄CrB₃. Presence of imaginary frequencies indicate dynamical unstable structure.



Figure S33. Phonon dispersion spectra for W₄MnB₃. Presence of imaginary frequencies indicate dynamical unstable structure.



Figure S34. Phonon dispersion spectra for Mo₄MnB₃. Presence of imaginary frequencies indicate dynamical unstable structure.



Figure S35. Phonon dispersion spectra for Mo₄CrB₃. Presence of imaginary frequencies indicate dynamical unstable structure.

Stability at finite temperatures

All stability calculations have been performed by calculating the formation enthalpy of an arbitrary M'-M"-B phase, denoted $M'_{x}M''_{y}B_{z}$, by comparing the energy of an arbitrary phase with the set of most competing phases at 0 K. However, the enthalpy of a material decreases with temperature due to, e.g., phonon vibrations. It has previously been demonstrated that due to a mutual cancellation of the temperature dependent energy terms, the stability at finite temperatures does not differ significantly from the calculated 0 K formation enthalpies [2,3].

The contribution from phonon vibrations to the free energy were calculated using the small displacement supercell method implemented in Phonopy [4] by applying finite displacements of 0.01 Å. The supercell for each phase was sufficiently large to converge the phonon free energy to 1 meV/atom, which was also the convergence criterion for the density of Brillouin zone mesh points for sampling the phonons. Figure S36a shows the individual vibrational phonon free energy for Ta_2TiB_2 and Ti_2ReB_2 and their set of most competing phases. Note that the vibrational free energy decreases significantly with temperature.

Next we look at the formation Gibbs free energy which is the difference in Gibbs free energy of $M'_xM''_yB_z$ with respect to the Gibbs free energy of its competing phases. Figure S36b shows the formation Gibbs free energy as function of temperature for four different $M'_2M''B_2$ phases of relevance for this work. In comparison, corresponding 0 K formation enthalpies are included as reference. Only minor changes are observed with temperature, thus demonstrating that the contribution from phonon vibrations at finite temperature mostly cancel one another for chemically similar material and further supports the approximation of only using 0 K energy terms when calculating the formation enthalpy.



Figure S36. (a) Individual vibrational phonon free energy and (b) formation Gibbs free energy as function of temperature.

Simulated XRD patterns



Figure S37. (a) Measured XRD pattern compared to simulated pattern of (b) ordered Ta_2MoB_2 , (c) disordered ($Ta_{0.66}Mo_{0.33}$)₃B₂, and (d) TaB.

SEM images



Figure S38. Representative SEM images. Orange crosses represent particles analyzed with EDX.

EDX measurements

Table S3: EDX Measurements of Ta and Mo (at%) and their calculated ratio for different particles of the Ta_2MoB_2 phase (TaB₂ particles are not shown).

Mo at%	Ta at%	Ta/Mo ratio
25.33	74.67	2.9
27.17	72.83	2.7
35.16	64.84	1.8
36.83	63.17	1.7
38.05	61.95	1.6
38.14	61.86	1.6
40.83	59.17	1.4
43.33	56.67	1.3
43.97	56.03	1.3
45.43	54.57	1.2
46.75	53.25	1.1
50.37	49.63	1.0
54.96	45.04	0.8

Supplementary References

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