

# **Chromium Cobaltite Based Ternary Composite as Efficient Electrode Material for Hybrid Supercapacitors with Theoretical Investigation**

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## ***Supporting Information (SI)***

### **1. Electrode preparation method**

The electrodes were prepared by blending the synthesized active material, PVDF (binder), and carbon black (conductive element) in a ratio of 8:1:1 to get consistent slurry with the addition of N-Methyl-2-pyrrolidone (NMP). Nickel foam with area of 1 cm<sup>2</sup> was then coated with this slurry and dried at 60 °C overnight. The separator (Whatman paper) wetted by 6M KOH electrolyte was sandwiched between the two electrodes (cathode and anode) and pressed using a hydraulic press to get the desired cell configuration.

### **2. Electrochemical measurements**

The capacitive response of the electrode material corresponding to a voltage window of -1 to +1V was evaluated from CV and GCD in 2-E configuration. The EIS analysis was carried out for 10<sup>5</sup>-0.1 Hz of frequency at open-circuit potential. For asymmetric supercapacitor, the voltage range was 0-1.6 V. The Formulae used for determining various parameters such as specific capacitance, energy density, and power density has been provided in **Table S1**.

**Table S1:** Formulae used for finding various parameters.

Parameter	Formula	Terms used
Bragg's law	$2d \sin \theta = n\lambda$	' $2\theta$ ' is bragg's diffraction angle, ' $d$ ' is inter-planar spacing, ' $\lambda$ ' is the wavelength of X-ray, ' $n$ ' is an integer,.
Interplanar spacing	$d = \frac{1}{\sqrt{\frac{h^2}{a^2} + \frac{k^2}{b^2} + \frac{l^2}{c^2}}}$	( $a, b, c$ ) are lattice parameters of the crystal, ( $hkl$ ) are miller indices of lattice plane.
Crystallite size, $D$ (Schherrer equation)	$D = \frac{K\lambda}{\omega \times \cos \theta}$	' $\omega$ ' is the FWHM, ' $2\theta$ ' is bragg's diffraction angle, $K$ is a constant, ' $\lambda$ ' is the wavelength of X-ray.
Specific capacitance, $C_{sp}$ for single electrode (from CV)	$C_{sp} = \frac{\int IdV}{m \times v \times dV}$	' $I$ ' is the current, ' $m$ ' is the mass of active material, ' $dV$ ' is the potential window, ' $v$ ' is the scan rate.
Specific capacitance, $C_{sp}$ for single electrode (from GCD)	$C_{sp} = \frac{2 \times I \times \Delta t}{m \times dV}$	' $dt$ ' is the discharging time, ' $I$ ' is the current, ' $m$ ' is the mass of active material.
Specific capacitance, $C_{sp}$ for SSC/ASC (from GCD)	$C_{sp} = \frac{I \times \Delta t}{m \times dV}$	' $dt$ ' is the discharging time, ' $I$ ' is the current, ' $m$ ' is the mass of active material.
Energy density, $E_d$	$E_d = \frac{C_{sp} \times (dV)^2}{7.2}$	' $dV$ ' is the voltage window, ' $C_{sp}$ ' is the

(from GCD)		specific capacitance.
Power density, $P_d$	$P_d = \frac{E_d \times 3.6}{\Delta t}$	$\Delta t$ is the discharging time, ' $E_d$ ' is the energy density.
(from GCD)		
Coulombic efficiency, $\eta$	$\eta = \frac{t_d \times 100}{t_c}$	' $t_c$ ' is the charging time and ' $t_d$ ' is the discharging time.
Response time, $\tau$	$\tau = \frac{1}{\nu}$	' $\nu$ ' is the frequency corresponding to phase angle $\theta = 45^\circ$ .
(from EIS)		

### 3. Results and Discussion

**Table S2:** Physico-chemical properties of activated carbon (AC).

Property	Value
Source	Coconut-shell
Appearance	Black powder
Molecular weight	12.01 g mol <sup>-1</sup>
Median size	30-60 $\mu\text{m}$
Ignition temp.	842 $^\circ\text{F}$
Melting point	3550 $^\circ\text{C}$
Resistivity	1375 $\mu\Omega\text{-cm}$

**Table S3:** XRD analysis for CrCo<sub>2</sub>O<sub>4</sub> (CCO) microspheres.

Angle	Lattice plane	Interplanar	Crystallite
$2\theta$ (degree)	(hkl)	spacing, d ( $\text{\AA}$ )	size (nm)
18.86	(111)	4.69	8.71
31.01	(220)	2.88	15.06

36.42	(311)	2.46	10.49
44.37	(400)	2.03	11.57
54.88	(422)	1.67	8.71
58.61	(511)	1.57	9.66
64.46	(440)	1.44	10.44

**Table S4:** Identification of different functional groups from FTIR.

Wavenumber (cm <sup>-1</sup> )	Functional group	Reference
491.01	Cr-O	1-3
628.93	Co-O	4
745.44	C-H bending	5
1005.87	=C-N deformation (in-plane)	6
1143.70	C-N stretching	6
1156.65	C-O stretching	6
1273.92	C-H in-plane bending	6, 7
1424.69	C-H bending	6
1527.50	N-H bending	6
1575.47	C=C stretching	8
1624.21	C=C stretching	7, 8
2103.95	C $\equiv$ C	6
2358.28	CO <sub>2</sub>	7

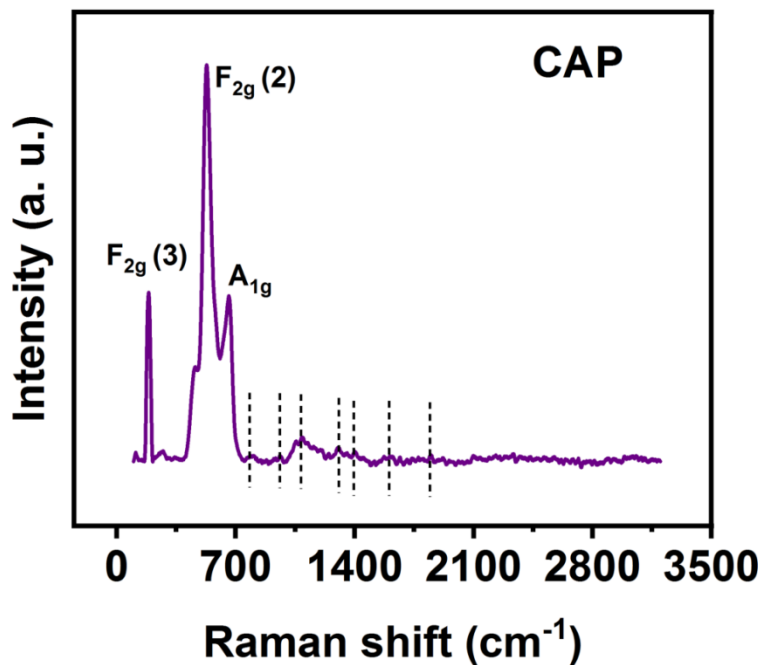


Fig. S1: Raman spectrum of CAP.

Table S5: Raman shift in CAP.

Symmetry	Bond vibration	Raman shift (cm <sup>-1</sup> )	References
$\delta(\text{O-Co-O})$	F <sub>2g</sub> (3)	192.88	9
$\nu(\text{Cr-O})$	F <sub>2g</sub> (2)	527.18	10
$\nu_s(\text{Cr-O})$	A <sub>1g</sub>	663.13	9, 10
PPY ring	Polaron distortion	965.80	11
C-H	In plane bending	1053	12, 13
C-H	In plane bending	1093	4

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C-N	Stretching	1220	14
PPY Ring	Stretching	1347	5
C=C	Asymmetric stretching	1418	14
C=C	Asymmetric stretching	1580	15

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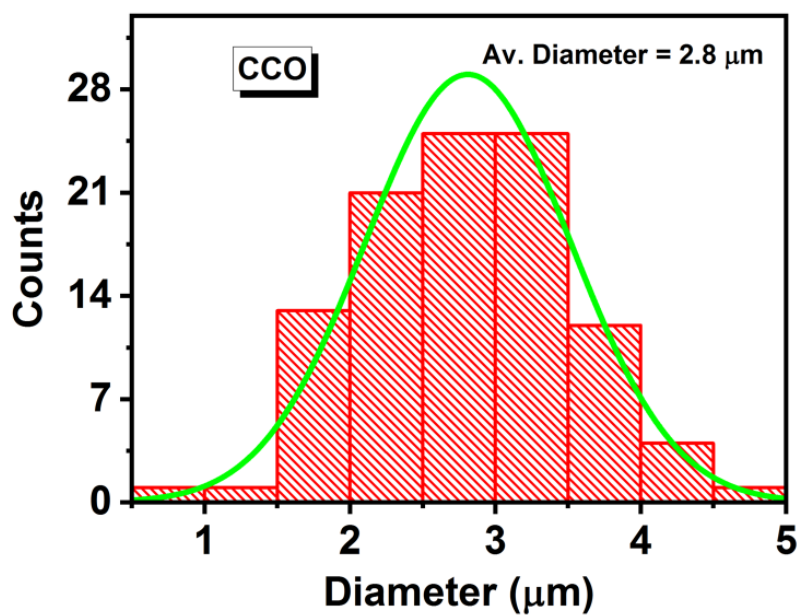
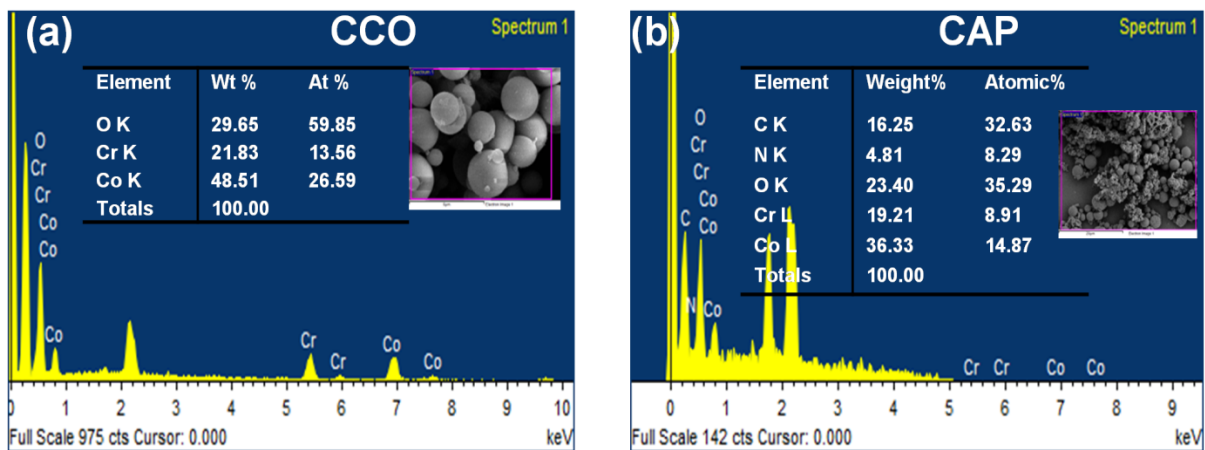
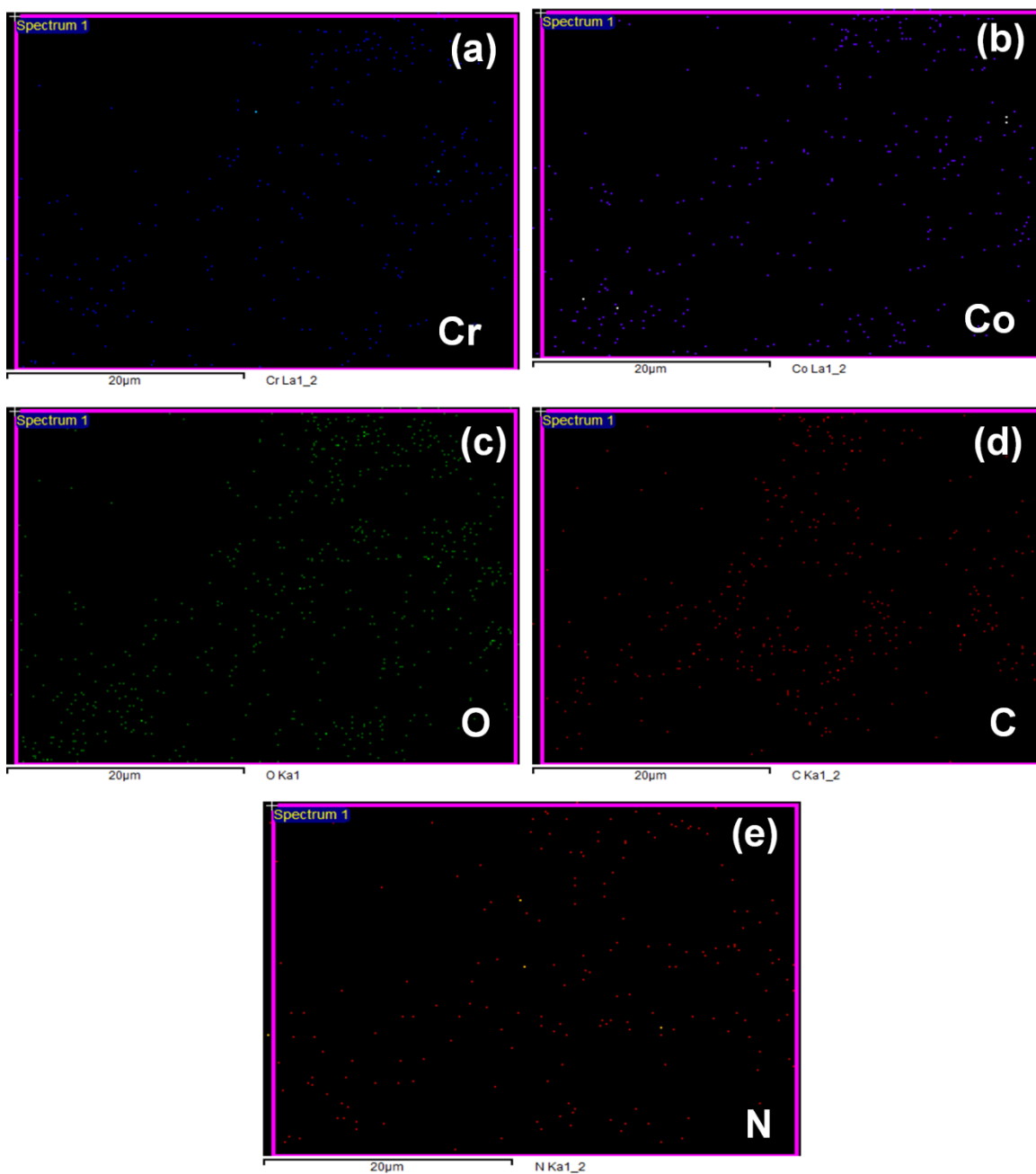


Fig. S2: Size distribution of CCO.



**Fig. S3:** EDS of (a) CCO and (b) CAP.



**Fig. S4:** Elemental mapping of CAP.



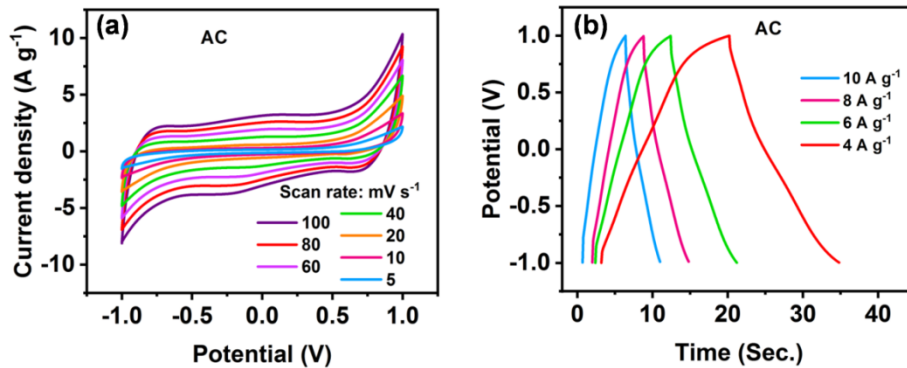


Fig. S5: (a) CV and (b) GCD curves of AC.

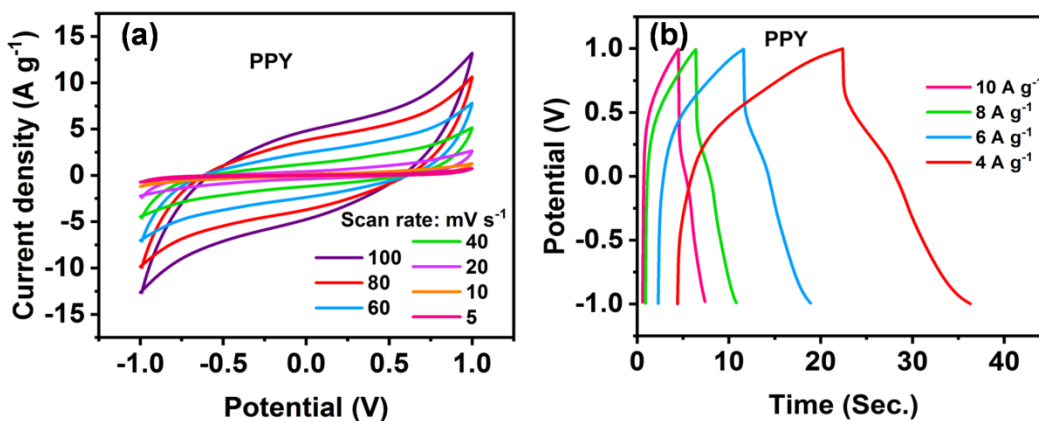


Fig. S6: (a) CV and (b) GCD curves of PPY.

Table S6:  $R_s$  and  $R_{ct}$  of CCO, CAC, and CAP.

Sample	CCO	CAC	CAP
$R_s$ ( $\Omega$ )	3.69	2.67	2.14
$R_{ct}$ ( $\Omega$ )	0.37	0.34	0.36

Table S7: Supercapacitive performance comparison of the prepared samples with literature.

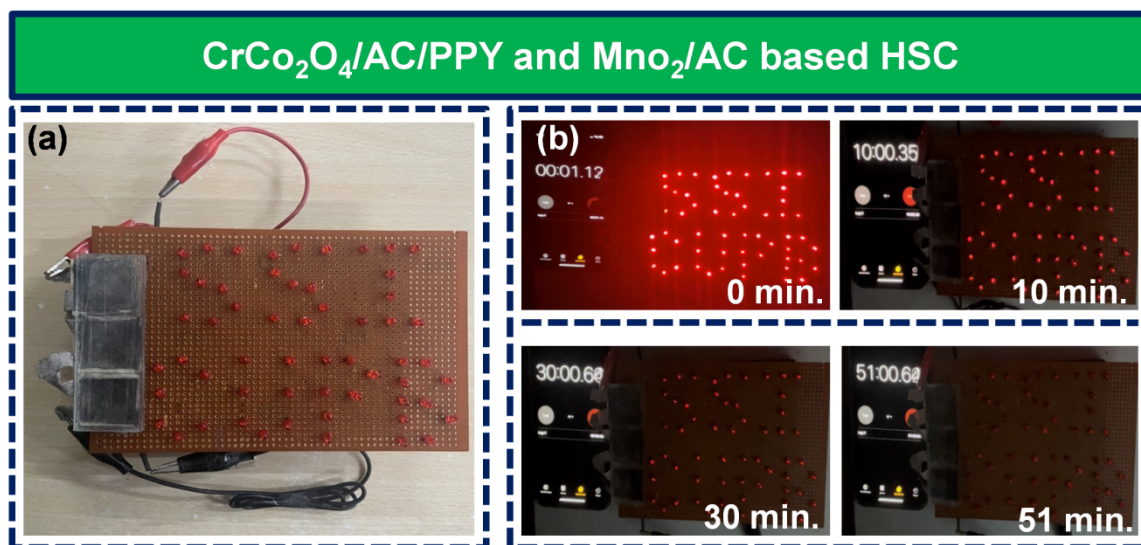
Material	$C_{sp}$	Voltage	ASC/SSC	$E_d$	Pd	Stability, Reference
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	(Fg <sup>-1</sup> )	window (V)		(Wh kg <sup>-1</sup> )	(kW kg <sup>-1</sup> )	% (cycles)	
MnCo <sub>2</sub> O <sub>4</sub> - graphite@PPY	2364	0-1.6	ASC	25.7	16.1	85.5 (10,000)	16
NiCo <sub>2</sub> O <sub>4</sub> /NF@PPY	1717 C g <sup>-1</sup>	0-1.6	ASC	68.9	1.77	89.2 (10,000)	17
MgCo <sub>2</sub> O <sub>4</sub> /PPY	988 (1 A g <sup>-1</sup> )	0-1.6	ASC	40.0	1.54	84.0 (10,000)	18
NiCo <sub>2</sub> O <sub>4</sub> /CNF@PPY	910 (1 A g <sup>-1</sup> )	0-1.5	ASC	40.8	0.73	88.0 (10,000)	19
NiCo <sub>2</sub> O <sub>4</sub> /Co <sub>3</sub> S <sub>4</sub> /MnS @PPY	2557 (1 A g <sup>-1</sup> )	0-1.6	ASC	81.1	0.80	83.6 (20,000)	20
MnNi <sub>2</sub> O <sub>4</sub> /PPY	304 (1 A g <sup>-1</sup> )	0-1.6	ASC	35.9	0.80	--	21
CC@NiCo <sub>2</sub> O <sub>4</sub> @PPY	1687 (1 A g <sup>-1</sup> )	0-1.5	ASC	46.5	0.72	80.0 (10,000)	22
NiCo <sub>2</sub> O <sub>4</sub> @PANI	561 (10 mV s <sup>-1</sup> )	0-1.2	ASC	6.4	0.28	86.2 (3,000)	23
CoFe <sub>2</sub> O <sub>4</sub> /PANI/GO	346.9 (1 A g <sup>-1</sup> )	0-1.2	SSC	69.3	5.98	79.0 (5,000)	24

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NiMoO <sub>4</sub> /rGO/PANI	1150 C g <sup>-1</sup> (1 A g <sup>-1</sup> )	0-1.7	ASC	82.43	0.85	94.5 (10,000)	25
CuCo <sub>2</sub> O <sub>4</sub> /GO@PANI	312.7 (1 A g <sup>-1</sup> )	0-1.2	SSC	62.5	5.99	84.2 (5,000)	26
Fe-MnCo <sub>2</sub> O <sub>4</sub> @PPY	422.4 (2 mA cm <sup>-2</sup> )	0-1	SSC	519.9 mWh cm <sup>-2</sup>	--	94.7 (7,000)	27
CoCr <sub>2</sub> O <sub>4</sub> /Ti <sub>3</sub> C <sub>2</sub>	417	-0.2-0.5	3-E	20.89	0.60	--	28
NiCo <sub>2</sub> O <sub>4</sub> /CF@PANI	369 mAh g <sup>-1</sup>	0-1.5	ASC	60.60	2.32	--	29
CoCr <sub>2</sub> O <sub>4</sub> /Co-MOF	596.8 C g <sup>-1</sup> (1 A g <sup>-1</sup> )	0-1.6	ASC	34.36	0.20	96.2 (5,000)	30
CrCo <sub>2</sub> O <sub>4</sub> /AC/PPY	991.25 (5 mV s <sup>-1</sup> ), 879.37 (4 A g <sup>-1</sup> )	0-1.6	HSC	97.77	1.6	76.75 (10,000)	This work

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**Fig. S7:** (a) LED-setup for device testing; and (b) Digital images of glowing LED-panel at different times.

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