

Supporting information

Self-templated fabrication of P-doped $\text{CoMoO}_4\text{-Co}_3\text{O}_4$ hollow nanocages for the efficient oxygen evolution reaction

Wenzhi Jia ^a, Qian Lu ^a, Tian Tian ^a, Guoxiang Pan ^a, Rui Tan ^c, Bin He ^{a*}, Jiang Liu ^{b*}

^a Huzhou Key Laboratory of Environmental Functional Materials and Pollution Control, Department of Materials Engineering, Huzhou University, Huzhou 313000, China

^b School of Chemistry, South China Normal University, Guangzhou 510006, China

^c Warwick Electrochemical Engineering Group, WMG, Energy Innovation Centre, University of Warwick, Warwick, CV47AL, UK

Corresponding author.

* E-mail address: binhe@zjhu.edu.cn (B. He); liuj0828@m.scnu.edu.cn

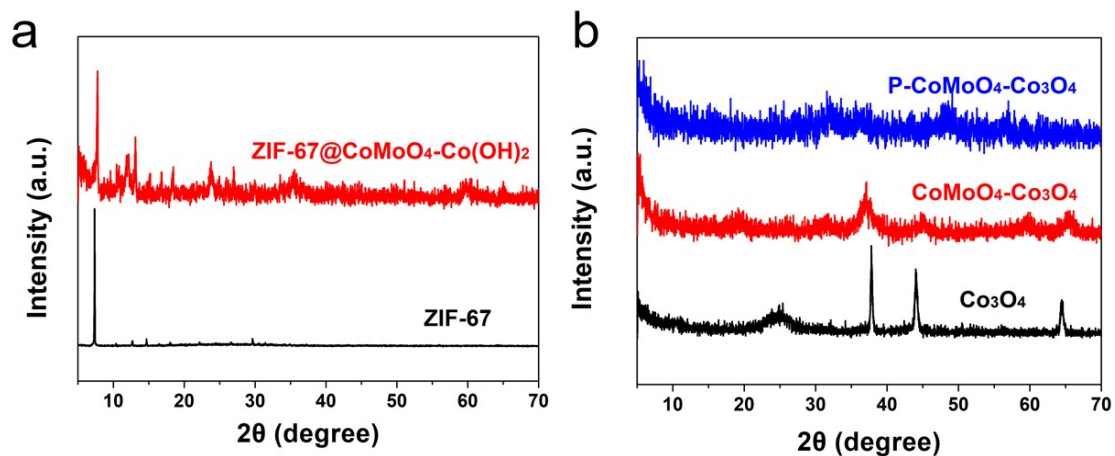


Fig. S1 XRD patterns of prepared samples. a) ZIF-67 and ZIF-67@CoMoO₄-Co(OH)₂, b) Co₃O₄, CoMoO₄-Co₃O₄ and P-CoMoO₄-Co₃O₄.

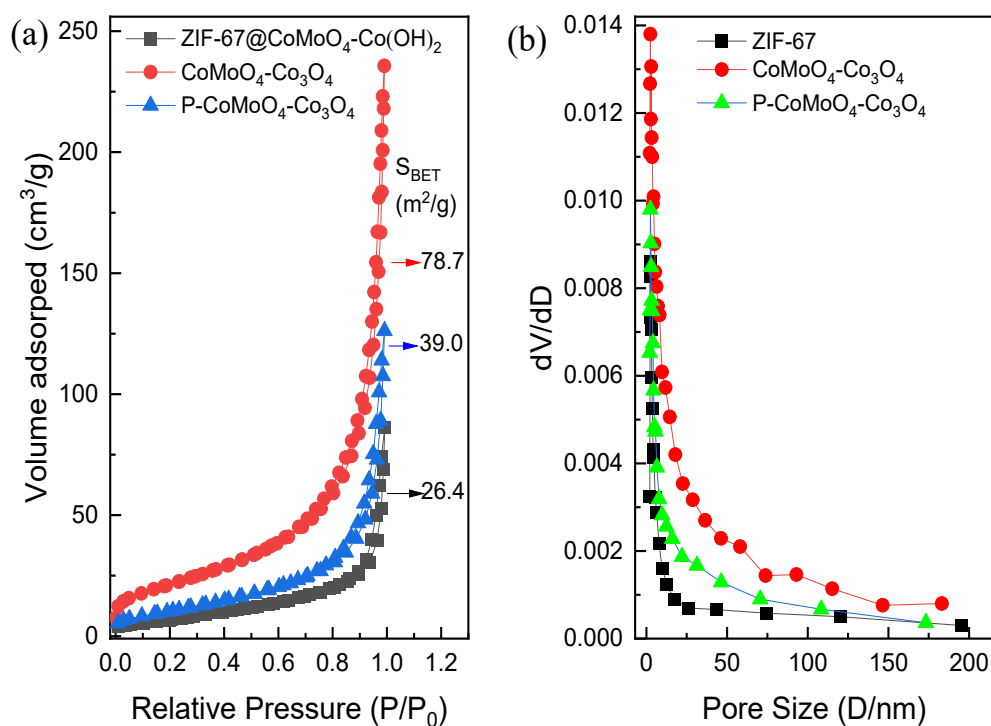


Fig. S2 (a) N₂ adsorption-desorption isotherms, BET areas, (b) pore diameter distribution of ZIF-67@CoMoO₄-Co(OH)₂, CoMoO₄-Co₃O₄ and P-CoMoO₄-Co₃O₄.

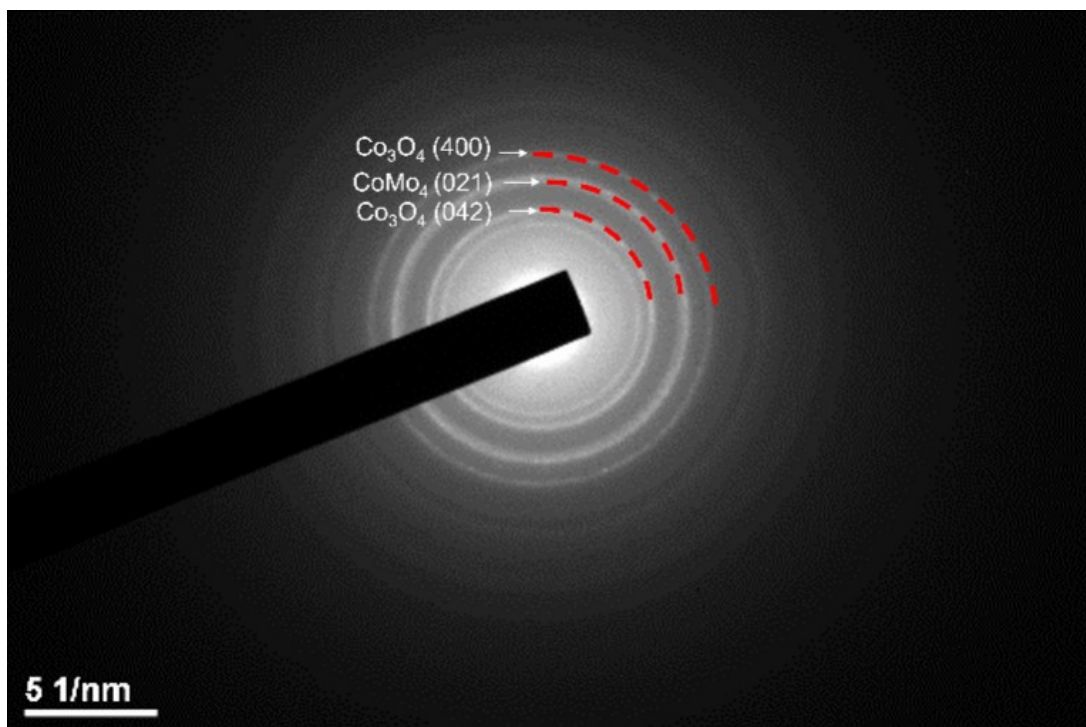


Fig. S3 SAED pattern of P-CoMoO₄-Co₃O₄.

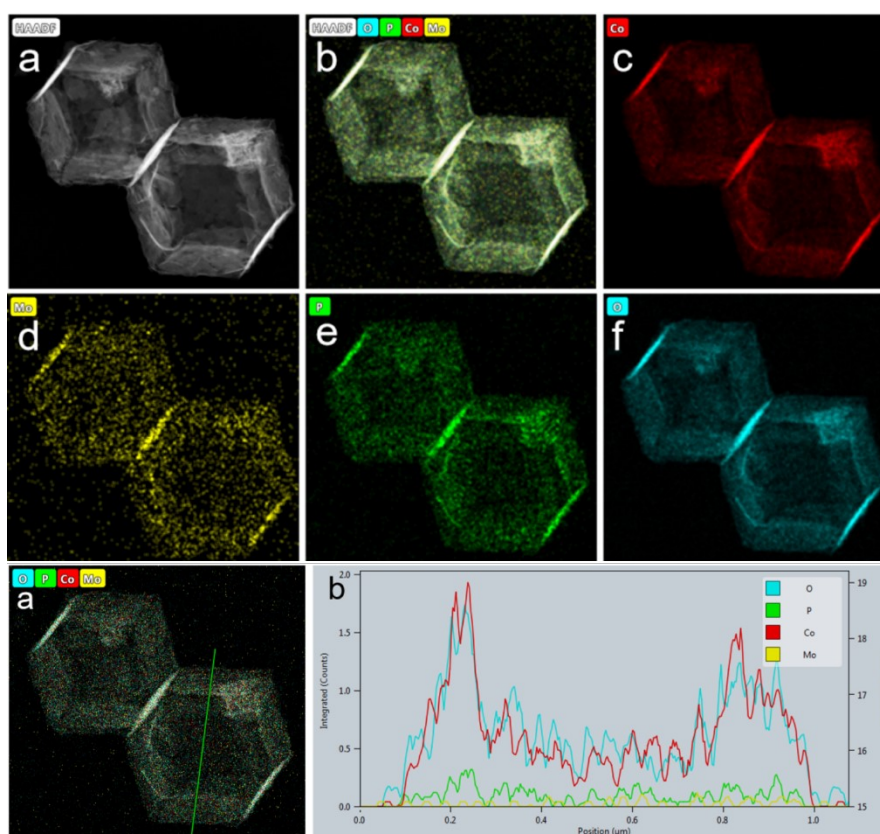


Fig. S4 EDX cross-sectional line scan profiles of the of P-CoMoO₄-Co₃O₄.

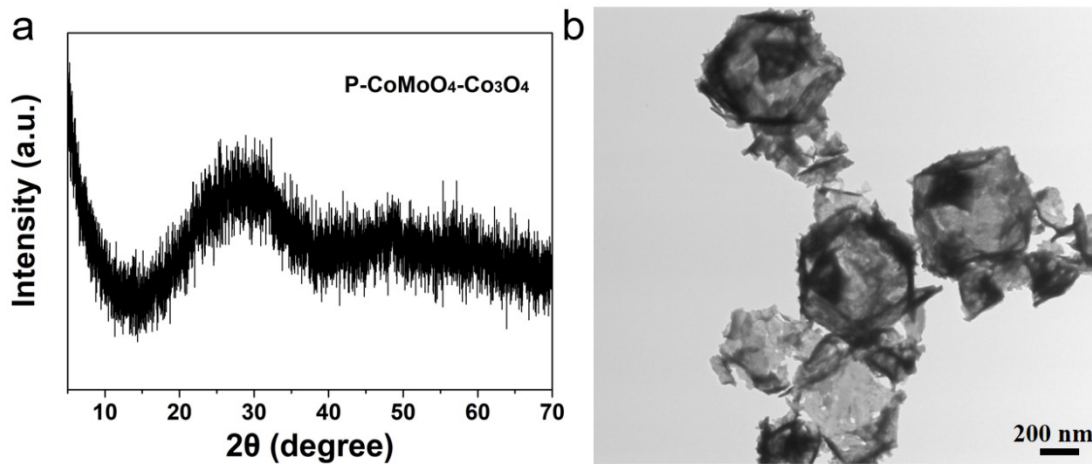


Fig. S5 XRD patterns and TEM images of the used P-Co₃O₄/CoMoO₄ HNC.

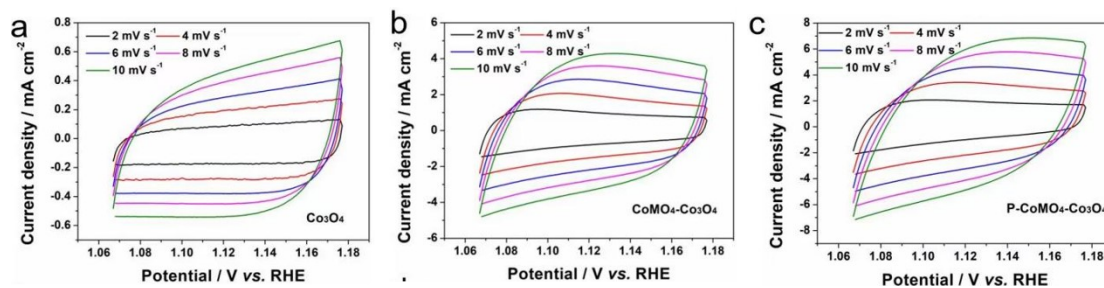


Fig. S6 Cyclic voltammetry of a) Co₃O₄, b) CoMoO₄-Co₃O₄ and c) P-CoMoO₄-Co₃O₄ and d) Values of the electrochemical double layer capacitance (Cdl).

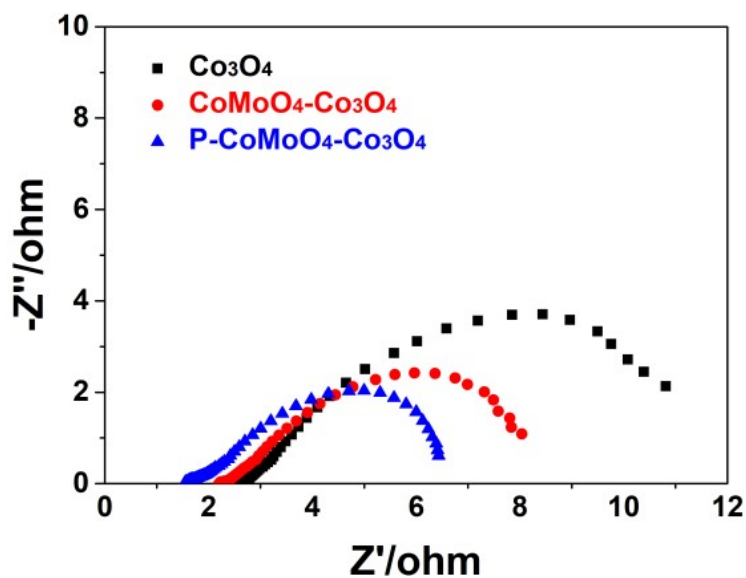


Fig. S7 EIS Nyquist diagram of Co₃O₄, CoMoO₄-Co₃O₄ and P-CoMoO₄-Co₃O₄.

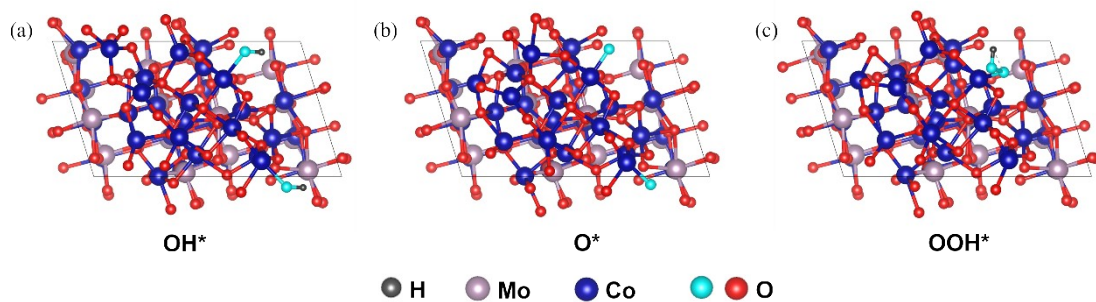


Fig. S8 The top view and side view of the (a) OH*, (b) O* and (c) OOH* structures on CoMoO₄-Co₃O₄ in OER.

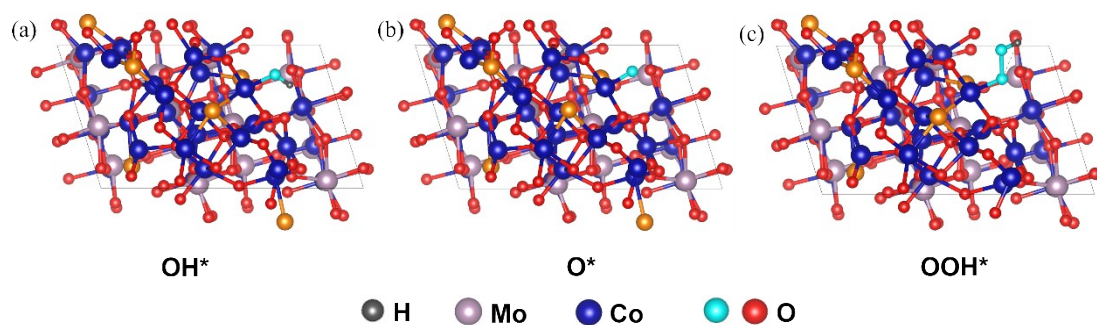


Fig. S9 The top view and side view of the (a) OH*, (b) O* and (c) OOH* structures on P-CoMoO₄-Co₃O₄ in OER.

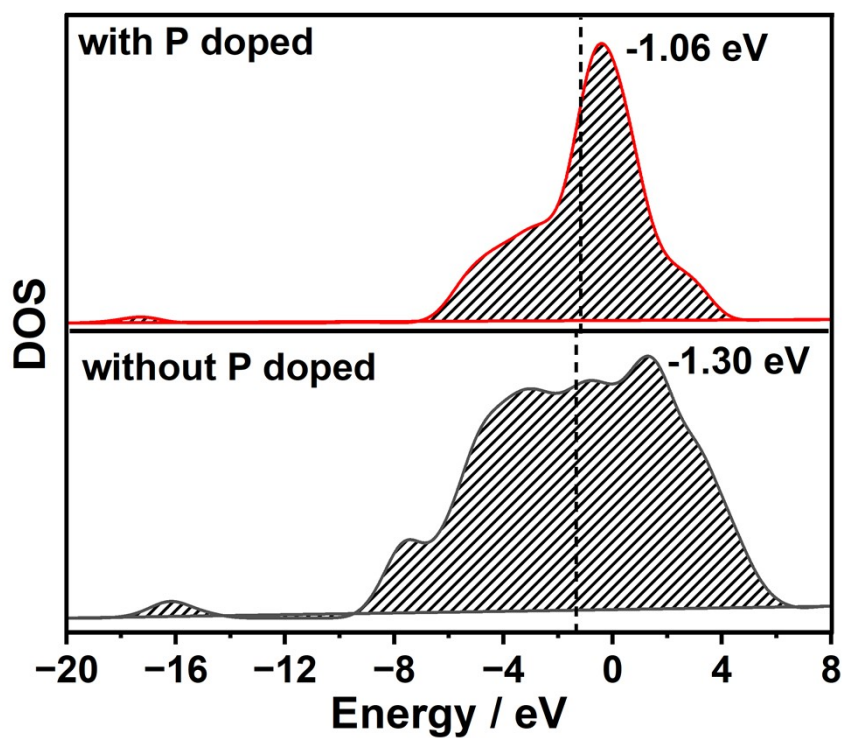


Fig. S10. Calculated density of states (DOS) of Co-3d in P-CoMoO₄-Co₃O₄ and CoMoO₄-Co₃O₄.

Table S1 OER performances comparison of recently reported representative electrocatalysts in alkaline medium (1.0 M KOH, at 10 mA cm⁻²).

Catalyst	Overpotential (mV)	Reference
P-CoMoO ₄ -Co ₃ O ₄	279	This work
Co/CoP	340	[1] Adv. Energy Mater. 7 (2017) 1602355.
Ni ₃ ZnC _{0.7} /NCNT-700	380	[2] Carbon 148 (2019) 496–503.
CoNiP/CoNi	300	[3] ChemSusChem 14(2021) 1921–1935.
N doped NiS/NiS ₂	270	[4] Chem. Eng. J. 397 (2020) 125507.
MnCo@NiS	286	[5] J. Power Sources 489 (2021) 229525.
Mn ₁₃ -NS/NF(Ni foam)	357	[6] ACS Appl. Nano Mater. 5 (2022) 326–330.
NiOOH/Ni ₃ S ₂ /NF	255	[7] Mater. Today Energy (2022) 101008.
Ni ₂ P-Ni ₃ S ₂ HNAs/NF	210	[8] ACS Catal. 8 (2018) 8107–8114.
Ni(Co)(Fe)OxHy	278	[9] Nano Res. 12 (2019) 2288–2295.
NiS/Ni interface	301	[10] J. Mater. Sci. Technol. 42 (2020) 10–16.