

Supporting Information for

Designed construction of two new atom-precise three-dimensional and two-dimensional Ag₁₂ cluster-assembled materials

Riki Nakatani,^a Jin Sakai,^a Aishik Saha,^b Ayumu Kondo,^a Rina Tomioka,^a Tokuhisa Kawawaki,^a Saikat Das^a and Yuichi Negishi*^{a,c}*

^aDepartment of Applied Chemistry, Faculty of Science, Tokyo University of Science, Kagurazaka, Shinjuku-ku, Tokyo 162-8601, Japan.

^bDepartment of Metallurgical Engineering and Materials Science, Indian Institute of Technology Bombay, Mumbai 400 076, India.

^cResearch Institute for Science & Technology, Tokyo University of Science, Tokyo 162-8601, Japan.

*Correspondence to: saikatdas@rs.tus.ac.jp (S.D.), negishi@rs.tus.ac.jp (Y.N.)

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Materials and methods

Materials

All starting materials and solvents were obtained from commercial sources and utilized directly as received without further purification, unless otherwise noted. *tert*-butyl mercaptan was obtained from Tokyo Chemical Industry Co., Ltd. Silver trifluoroacetate (CF₃COOAg) was obtained from FUJIFILM Wako Pure Chemical Corporation. Silver nitrate (AgNO₃), methanol, dimethylacetamide (DMAc), dimethylformamide (DMF), acetonitrile, and ethanol were obtained from Kanto Chemical Co., Inc. tris(pyridine-4-ylmethyl)amine (TPMA) and 1,3,5-tris(pyridine-4-ylethynyl)benzene (TPEB) were obtained from ET Co., Ltd.

Methods

Single-crystal X-ray diffraction (SCXRD) was conducted for gathering the data about the single crystal, which was first immersed in cryoprotectant Parabar 10312 (Hampton Research, 34 Journey, Aliso Viejo, CA 92656-3317 USA) followed by mounting on a Dual-Thickness MicroMounts™ (MiTeGen, LLC, Ithaca, NY, USA). A Bruker D8 QUEST diffractometer was utilized for performing SCXRD measurement for **TUS 6** by subjecting the sample to a monochromatic Mo K α radiation ($\lambda = 0.71073 \text{ \AA}$). Regarding **TUS 7**, SCXRD was carried out on a Rigaku XtaLaB Synergy-DW diffractometer equipped with monochromatic Cu K α radiation ($\lambda = 1.5418 \text{ \AA}$). The crystal structures were determined and solved by employing the Apex4 Bruker software¹ and CrysAlis^{Pro} software². Powder X-ray diffraction (PXRD) patterns were procured on a Rigaku MiniFlex 600 X-ray diffractometer with a Cu K α source ($\lambda = 1.5418 \text{ \AA}$) under an applied voltage of 40 kV and current of 15 mA. The range of diffraction angle 2θ was 3-40° over which the intensity peaks were recorded with a step size of 0.01° and scan speed of 3.7°/min. X-ray photoelectron spectroscopy (XPS) spectra were collected on a JPS-9030 electron spectrometer (JEOL, Tokyo, Japan) utilizing a Mg K α radiation (1253.6 eV). All the binding energies were referenced to the neutral C 1s peak at 283.3 eV. An Olympus SZX7 stereo microscope was utilized to record the optical microscope images. Scanning electron microscopy/energy-dispersive X-ray spectroscopy (SEM-EDX) was carried out for high resolution imaging/elemental mapping on a JEOL JSM-IT800SHL field emission scanning electron microscope. The thermal stability of the SCAMs were probed by thermogravimetric analysis (TGA) on a Bruker TG-DTA2010SA thermal analyzer from room temperature to 800

°C at a heating rate of 10 °C min⁻¹ under N₂ atmosphere. Diffuse reflectance spectroscopy (DRS) was accomplished on a JASCO V-670 spectrophotometer. The photoluminescence (PL) spectra were recorded on a SHIMADZU RF-6000 spectrofluorometer.

Synthesis methods

Synthesis of silver *tert*-butylthiolate (AgS'Bu). AgS'Bu was synthesized following previous reports.^{3,4}

Synthesis of TUS 6

[AgS'Bu]_n (20 mg, 0.10 mmol) and CF₃COOAg (22 mg, 0.10 mmol) were mixed together in DMAc (5 mL) under stirring until the solution was transparent. On the other hand, TPMA (15.3 mg, 0.05 mmol) linker was dissolved in another glass vial that contained DMAc (1 mL). This solution was added dropwise to the inorganic solution. After leaving for 1 day in the dark, **TUS 6** (21.9 mg) was obtained in 40.6% yield on the basis of Ag as colourless octahedral-shaped crystals from the bottom of the solution.

Synthesis of TUS 7

To a solution of EtOH/MeCN (v/v=1:1) (5 mL) was added [AgS'Bu]_n (24 mg, 0.12 mmol) and CF₃COOAg (24 mg, 0.12 mmol), and the mixture was stirred vigorously until the solution became transparent. On the other hand, TPEB (11.8 mg, 0.03 mmol) linker was dissolved in another glass vial that contained DMF (5 mL). This solution was added dropwise to the inorganic solution. After leaving for 1 day in the dark, **TUS 7** (35.2 mg) was obtained in 69.6% yield on the basis of TPEB as colourless octahedral-shaped crystals from the bottom of the solution.

Table S1. Crystallographic parameters of **TUS 6**.

Identification code	TUS 6
Empirical formula	$C_{36}H_{45}Ag_6F_9N_4O_6S_3$
CCDC number	2373129
Formula weight	1616.22
Temperature/K	273.15
Crystal system	Orthorhombic
Space group	<i>Pbca</i>
a/Å	21.868(3)
b/Å	22.557(3)
c/Å	22.892(3)
$\alpha/^\circ$	90
$\beta/^\circ$	90
$\gamma/^\circ$	90
Volume/Å ³	11292(3)
Z	8
$\rho_{\text{calc}}/\text{g cm}^{-3}$	1.901
μ/mm^{-1}	2.229
F(000)	6304
Crystal size/mm ³	0.24 × 0.22 × 0.2
Radiation	MoK α ($\lambda = 0.71073$)
2 θ range for data collection/ $^\circ$	2.070 to 24.713 $^\circ$
Index ranges	-25 ≤ h ≤ 25, -25 ≤ k ≤ 26, -24 ≤ l ≤ 26
Reflections collected	76830
Independent reflections	9365 [$R_{\text{int}} = 0.2248$]
Data/restraints/parameters	9365 / 506 / 586
Goodness-of-fit on F ²	0.968
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0530$, $wR_2 = 0.0978$
Final R indexes [all data]	$R_1 = 0.1019$, $wR_2 = 0.1157$
Largest diff. peak/hole / e Å ⁻³	1.565 / -0.843

Table S2. Crystallographic parameters of **TUS 7**.

Identification code	TUS 7
Empirical formula	$C_{15}H_{14}Ag_2F_3NO_2S$
CCDC number	2373236
Formula weight	614.17
Temperature/K	273.15
Crystal system	Trigonal
Space group	<i>R</i> -3
<i>a</i> /Å	24.5604(4)
<i>b</i> /Å	24.5604(4)
<i>c</i> /Å	21.6065(6)
α /°	90
β /°	90
γ /°	120
Volume/Å ³	11287.2(5)
<i>Z</i>	18
$\rho_{\text{calc}}/\text{g cm}^{-3}$	1.626
μ/mm^{-1}	13.672
<i>F</i> (000)	5454
Crystal size/mm ³	0.16 × 0.16 × 0.16
Radiation	MoK α ($\lambda = 0.71073$)
2 θ range for data collection/°	2.915 to 74.530°
Index ranges	-30 ≤ <i>h</i> ≤ 22, -16 ≤ <i>k</i> ≤ 30, -21 ≤ <i>l</i> ≤ 26
Reflections collected	12388
Independent reflections	4922 [<i>R</i> _{int} = 0.0239]
Data/restraints/parameters	4922 / 184 / 220
Goodness-of-fit on <i>F</i> ²	1.070
Final <i>R</i> indexes [<i>I</i> ≥ 2 σ (<i>I</i>)]	<i>R</i> ₁ = 0.0379, <i>wR</i> ₂ = 0.1049
Final <i>R</i> indexes [all data]	<i>R</i> ₁ = 0.0417, <i>wR</i> ₂ = 0.1076
Largest diff. peak/hole / e Å ⁻³	0.822 / -0.951

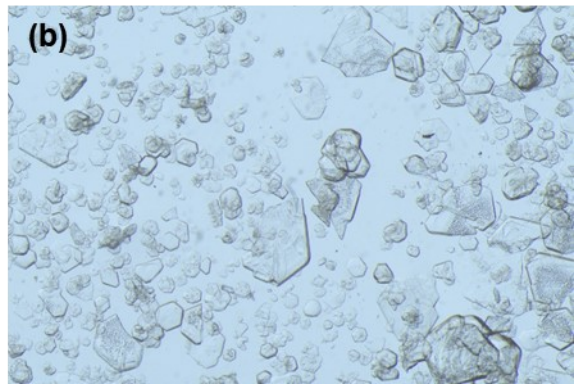
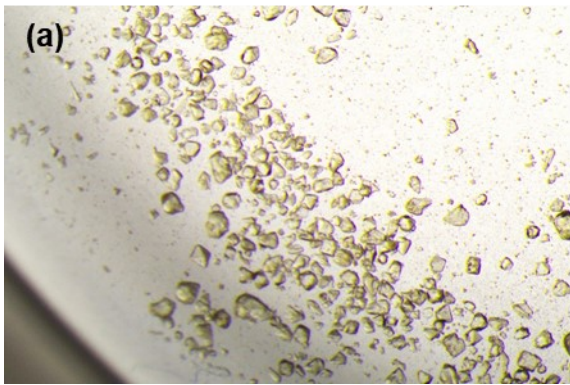


Fig. S1 Optical microscope images of (a) **TUS 6** and (b) **TUS 7**.

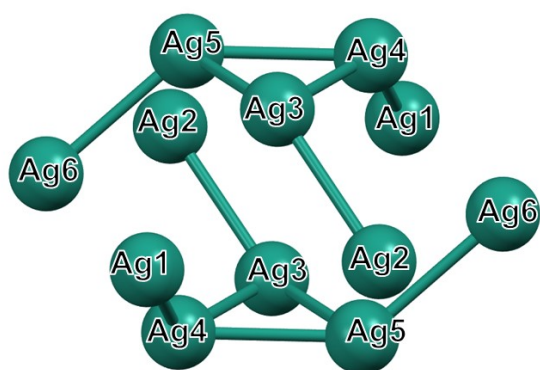


Fig. S2 Ag₁₂ core architecture of TUS 6.

Table S3. Ag–Ag bond lengths for the Ag₁₂ hollow cuboctahedron illustrated in Fig. S2.

Atom1	Atom2	Bond length/Å		Bond length/Å
Ag1	Ag4	3.016	Maximum	3.044
Ag2	Ag3	3.044	Minimum	2.936
Ag3	Ag4	2.961	Average	2.985
Ag3	Ag5	2.936	S.D.	0.042
Ag4	Ag5	2.954		
Ag5	Ag6	3.001		

Table S4. Ag–Ag–Ag bond angles corresponding to Fig. S2.

Atoms	Angle/°
Ag1-Ag4-Ag3	137.8
Ag1-Ag4-Ag5	100.66
Ag2-Ag3-Ag4	99.89
Ag2-Ag3-Ag5	136.93
Ag3-Ag4-Ag5	59.53
Ag3-Ag5-Ag4	60.34
Ag4-Ag3-Ag5	60.13
Ag3-Ag5-Ag6	100.62
Ag4-Ag5-Ag6	137.63

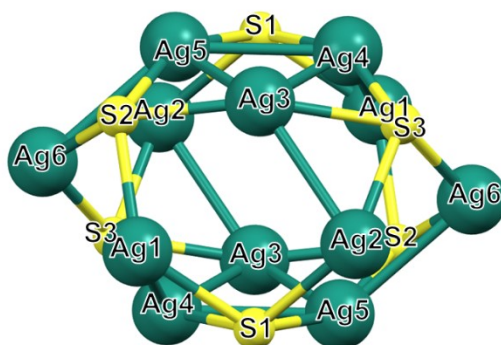


Fig. S3 Attachment of six thiolates on the Ag₁₂ cluster node in TUS 6.

Table S5. Ag–S bond lengths corresponding to Fig. S3.

Atom1	Atom2	Bond length/Å		Bond length/Å
Ag1	S1	2.538	Maximum	2.538
Ag1	S2	2.511	Minimum	2.476
Ag2	S1	2.521	Average	2.504
Ag2	S3	2.537	S.D.	0.023
Ag3	S2	2.480		
Ag3	S3	2.490		
Ag4	S1	2.485		
Ag4	S3	2.476		
Ag5	S1	2.480		
Ag5	S2	2.491		
Ag6	S2	2.522		
Ag6	S3	2.520		

Table S6. Ag–S–Ag and Ag–Ag–S bond angles corresponding to Fig. S3.

Atoms	Angle/°	Atoms	Angle/°
Ag1-S1-Ag2	95.55	Ag2-Ag3-S3	53.46
Ag1-S1-Ag4	73.78	Ag4-Ag3-S3	53.19
Ag2-S1-Ag5	83.96	Ag5-Ag3-S2	53.96
Ag4-S1-Ag5	73.02	Ag2-Ag3-S2	119.11
Ag1-S2-Ag3	84.49	Ag1-Ag4-S1	53.91
Ag3-S2-Ag5	72.40	Ag3-Ag4-S3	53.62
Ag5-S2-Ag6	73.53	Ag1-Ag4-S3	119.71
Ag1-S2-Ag6	94.62	Ag5-Ag4-S1	53.42
Ag2-S3-Ag3	74.51	Ag3-Ag5-S2	53.64
Ag2-S3-Ag6	94.50	Ag6-Ag5-S2	53.72
Ag3-S3-Ag4	73.19	Ag6-Ag5-S1	119.36
Ag4-S3-Ag6	83.31	Ag4-Ag5-S1	53.56
S1-Ag1-Ag4	52.31	Ag5-Ag6-S2	52.75
S1-Ag1-S2	117.56	Ag5-Ag6-S3	89.51
Ag4-Ag1-S2	89.35	S2-Ag6-S3	118.79
Ag3-Ag2-S1	88.67		
Ag3-Ag2-S3	52.03		
S1-Ag2-S3	117.33		

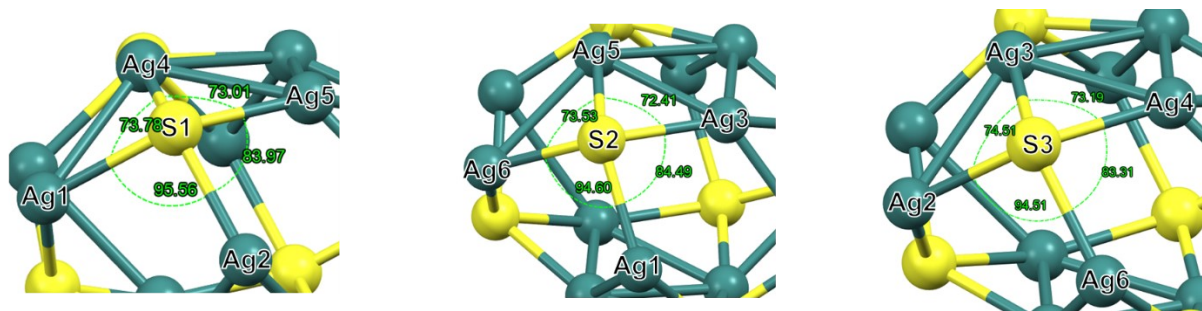


Fig. S4 Connectivities of six S atoms that belong to S'Bu molecules and constructing the cluster nodes with four different Ag atoms in $\mu_4\text{-}\eta^1, \eta^1, \eta^1, \eta^1$ ligation in **TUS 6**.

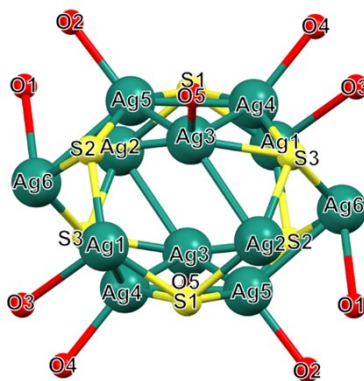


Fig. S5 Attachment of six trifluoroacetates on the Ag_{12}S_6 cluster node in **TUS 6**.

Table S7. Ag–O bond lengths corresponding to Fig. S5.

Atom1	Atom2	Bond length/Å		Bond length/Å
Ag1	O3	2.710	Maximum	2.710
Ag3	O5	2.385	Minimum	2.384
Ag4	O4	2.389	Average	2.508
Ag5	O2	2.384	S.D.	0.167
Ag6	O1	2.671		

Table S8. O–Ag–Ag and O–Ag–S bond angles corresponding to Fig. S5.

Atoms	Angle/°	Atoms	Angle/°
O1–Ag6–Ag5	61.56	O4–Ag4–Ag1	95.16
O1–Ag6–S2	100.22	O4–Ag4–Ag3	126.85
O1–Ag6–S3	99.07	O4–Ag4–Ag5	127.13
O2–Ag5–Ag3	127.52	O4–Ag4–S1	102.65
O2–Ag5–Ag4	128.24	O4–Ag4–S3	102.51
O2–Ag5–Ag6	93.81	O5–Ag3–Ag2	96.76
O2–Ag5–S1	102.56	O5–Ag3–Ag4	126.00
O2–Ag5–S2	101.66	O5–Ag3–Ag5	126.14
O3–Ag1–Ag4	61.92	O5–Ag3–S2	102.33
O3–Ag1–S1	98.66	O5–Ag3–S3	102.71
O3–Ag1–S2	102.61		

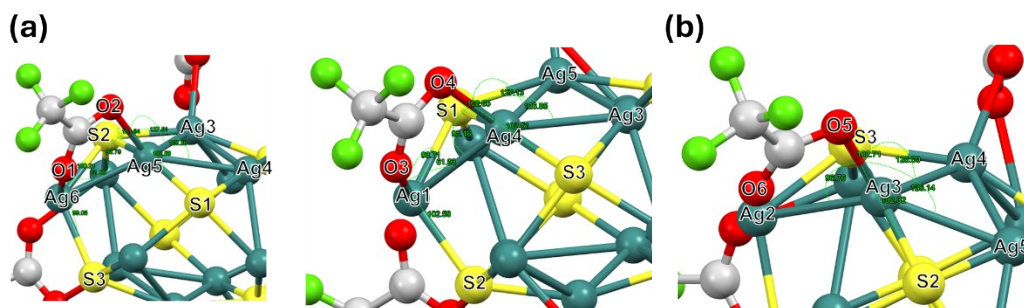


Fig. S6 Connectivities of (a) four CF₃COO⁻ ligands with two different Ag atoms by utilizing two O ($\mu_2\text{-}\eta^1, \eta^1$) and (b) the other two CF₃COO⁻ ligands with one Ag atom by utilizing one O ($\mu_1\text{-}\eta^1$) in TUS 6.

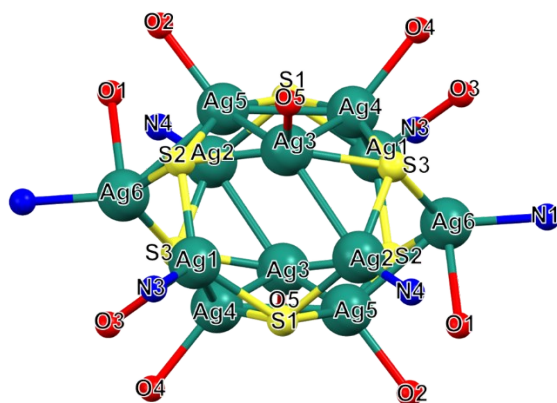


Fig. S7 Attachment of six linker molecules on the Ag₁₂S₆O₁₀ cluster node in TUS 6.

Table S9. Ag–N bond lengths corresponding to Fig. S7.

Atom1	Atom2	Bond length/Å		Bond length/Å
Ag1	N3	2.279	Maximum	2.298
Ag2	N4	2.268	Minimum	2.268
Ag6	N1	2.298	Average	2.282
			S.D.	0.015

Table S10. N–Ag–Ag, N–Ag–O and N–Ag–S bond angles corresponding to Fig. S7.

Atoms	Angle/°
N1–Ag6–Ag5	142.76
N1–Ag6–S2	120.74
N1–Ag6–S3	117.52
N1–Ag6–O1	87.77
N3–Ag1–Ag4	142.45
N3–Ag1–S1	118.27
N3–Ag1–S2	120.51
N3–Ag1–O3	87.87
N4–Ag2–Ag3	139.95
N4–Ag2–S1	120.29
N4–Ag2–S3	120.37

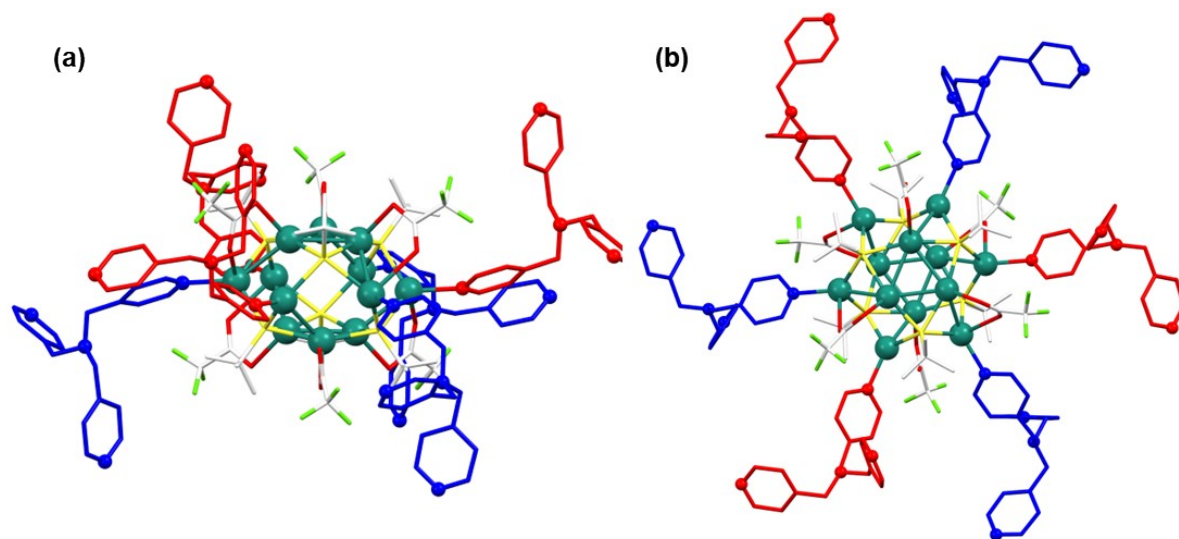


Fig. S8 The connectivities between Ag₁₂ cluster nodes and linkers in **TUS 6** as can be visualized from the (a) side view and (b) top/bottom view.

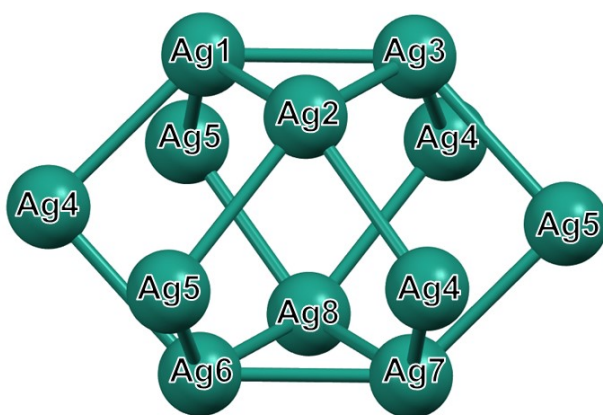


Fig. S9 Ag₁₂ core architecture of TUS 7.

Table S11. Ag–Ag bond lengths for the Ag₁₂ hollow cuboctahedron illustrated in Fig. S9.

Atom1	Atom2	Bond length/Å		Bond length/Å
Ag1	Ag2	3.020	Maximum	3.228
Ag1	Ag3	3.020	Minimum	3.02
Ag1	Ag4	3.104	Average	3.117
Ag1	Ag5	3.228	S.D.	0.088
Ag2	Ag3	3.020		
Ag2	Ag4	3.104		
Ag2	Ag5	3.228		
Ag3	Ag4	3.104		
Ag3	Ag5	3.228		
Ag4	Ag6	3.228		
Ag4	Ag7	3.228		
Ag4	Ag8	3.228		
Ag5	Ag6	3.104		
Ag5	Ag7	3.104		
Ag5	Ag8	3.104		
Ag6	Ag7	3.020		
Ag6	Ag8	3.020		
Ag7	Ag8	3.020		

Table S12. Ag–Ag–Ag bond angles corresponding to Fig. S9.

Atoms	Angle/°	Atoms	Angle/°
Ag1-Ag2-Ag3	60.00	Ag6-Ag7-Ag8	60.00
Ag1-Ag2-Ag5	94.72	Ag6-Ag7-Ag4	94.72
Ag1-Ag3-Ag2	60.00	Ag6-Ag8-Ag7	60.00
Ag1-Ag3-Ag4	97.71	Ag6-Ag8-Ag5	97.71
Ag2-Ag1-Ag3	60.00	Ag7-Ag6-Ag8	60.00
Ag2-Ag1-Ag4	97.71	Ag7-Ag6-Ag5	97.71
Ag2-Ag3-Ag5	94.72	Ag7-Ag8-Ag4	94.72
Ag3-Ag1-Ag5	94.72	Ag8-Ag6-Ag4	94.72
Ag3-Ag2-Ag4	97.71	Ag8-Ag7-Ag5	97.71

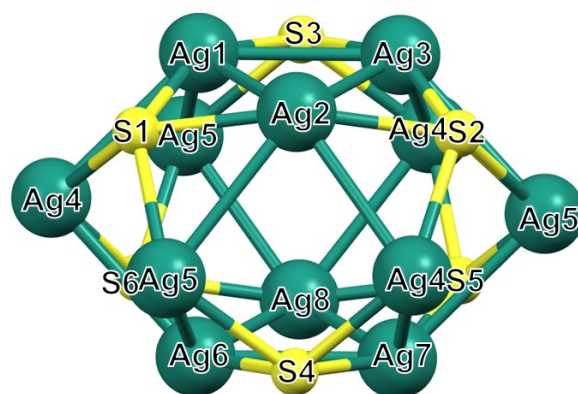


Fig. S10 Attachment of six thiolates on the Ag₁₂ cluster node in TUS 7.

Table S13. Ag–S bond lengths corresponding to Fig. S10.

Atom1	Atom2	Bond length/Å	Atom1	Atom2	Bond length/Å	Bond length/Å	
Ag1	S1	2.483	Ag5	S1	2.509	Maximum	2.544
Ag1	S3	2.471	Ag5	S2	2.509	Minimum	2.471
Ag2	S1	2.471	Ag5	S3	2.509	Average	2.519
Ag2	S2	2.483	Ag5	S4	2.544	S.D.	0.029
Ag3	S2	2.471	Ag5	S5	2.544		
Ag3	S3	2.483	Ag5	S6	2.544		
Ag4	S1	2.544	Ag6	S4	2.544		
Ag4	S2	2.544	Ag6	S6	2.544		
Ag4	S3	2.544	Ag7	S4	2.544		
Ag4	S4	2.509	Ag7	S5	2.544		
Ag4	S5	2.509	Ag8	S5	2.544		
Ag4	S6	2.509	Ag8	S6	2.544		

Table S14. Ag–S–Ag and Ag–Ag–S bond angles corresponding to Fig. S10.

Atoms	Angle/°	Atoms	Angle/°	Atoms	Angle/°	Atoms	Angle/°
Ag1-S1-Ag2	75.12	Ag5-S4-Ag6	76.25	S3-Ag3-Ag1	52.26	S6-Ag8-Ag6	52.26
Ag1-S1-Ag4	76.25	Ag5-S5-Ag7	76.25	S3-Ag3-Ag4	52.76	S6-Ag8-Ag5	52.76
Ag1-S3-Ag3	75.12	Ag5-S6-Ag8	76.25	S3-Ag1-Ag3	52.63	S6-Ag6-Ag8	52.63
Ag1-S3-Ag5	80.84	Ag6-S4-Ag7	75.12	S3-Ag1-Ag5	50.09	S6-Ag6-Ag4	50.09
Ag2-S1-Ag5	80.84	Ag6-S6-Ag8	75.12	S3-Ag4-Ag3	50.99	S6-Ag5-Ag8	50.99
Ag2-S2-Ag3	75.12	Ag7-S5-Ag8	75.12	S3-Ag5-Ag1	49.07	S6-Ag4-Ag6	49.07
Ag2-S2-Ag4	76.25	S1-Ag1-Ag2	52.26	S4-Ag6-Ag7	52.26		
Ag3-S2-Ag5	80.84	S1-Ag1-Ag4	52.76	S4-Ag6-Ag5	52.76		
Ag3-S3-Ag4	76.25	S1-Ag2-Ag1	52.63	S4-Ag7-Ag6	52.63		
Ag4-S1-Ag5	94.31	S1-Ag2-Ag5	50.09	S4-Ag7-Ag4	50.09		
Ag4-S2-Ag5	94.31	S1-Ag4-Ag1	50.99	S4-Ag5-Ag6	50.99		
Ag4-S3-Ag5	94.31	S1-Ag5-Ag2	49.07	S4-Ag4-Ag7	49.07		
Ag4-S4-Ag5	94.31	S2-Ag2-Ag3	52.26	S5-Ag7-Ag8	52.26		
Ag4-S4-Ag7	80.84	S2-Ag2-Ag4	52.76	S5-Ag7-Ag5	52.76		
Ag4-S5-Ag5	94.31	S2-Ag3-Ag2	52.63	S5-Ag8-Ag7	52.63		
Ag4-S5-Ag8	80.84	S2-Ag3-Ag5	50.09	S5-Ag8-Ag4	50.09		
Ag4-S6-Ag5	94.31	S2-Ag4-Ag2	50.99	S5-Ag5-Ag7	50.99		
Ag4-S6-Ag6	80.84	S2-Ag5-Ag3	49.07	S5-Ag4-Ag8	49.07		

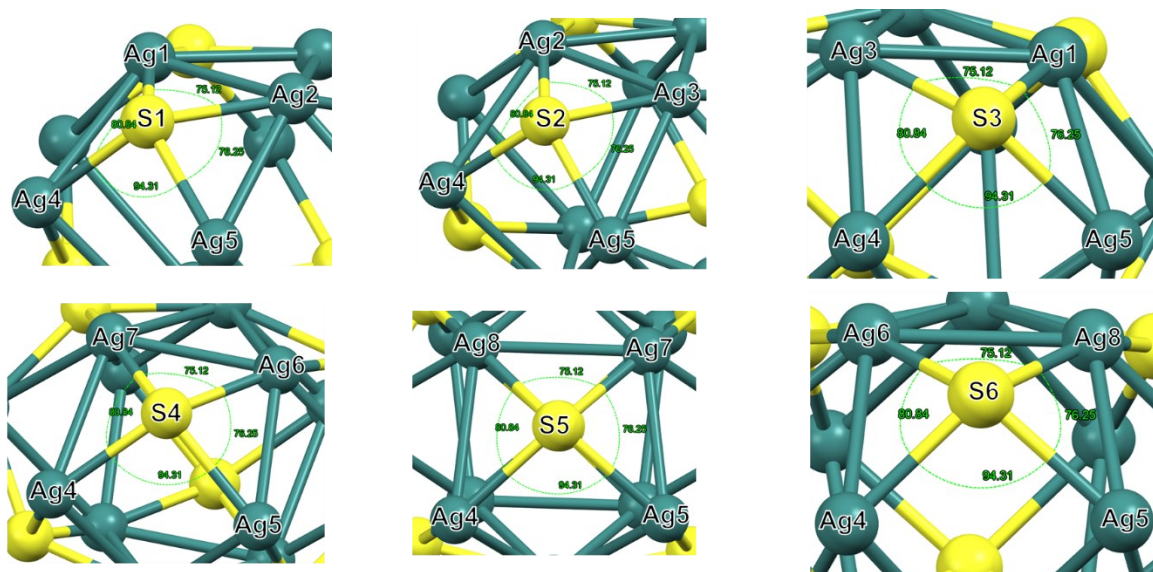


Fig. S11 Connectivities of six S atoms that belong to S'Bu molecules and constructing the cluster nodes with four different Ag atoms in $\mu_4\text{-}\eta^1, \eta^1, \eta^1, \eta^1$ ligation in TUS 7.

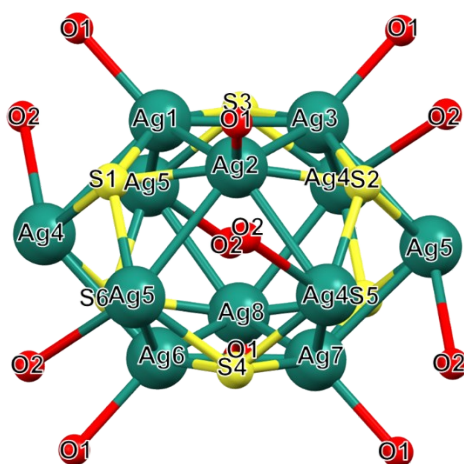


Fig. S12 Attachment of six trifluoroacetates on the Ag_{12}S_6 cluster node in **TUS 7**.

Table S15. Ag–O bond lengths corresponding to Fig. S12.

Atom1	Atom2	Bond length/Å		Bond length/Å
Ag1	O1	2.401	Maximum	2.684
Ag2	O1	2.401	Minimum	2.401
Ag3	O1	2.401	Average	2.472
Ag4	O2	2.684	S.D.	0.131
Ag5	O2	2.684		
Ag6	O1	2.401		
Ag7	O1	2.401		
Ag8	O1	2.401		

Table S16. O–Ag–Ag and O–Ag–S bond angles corresponding to Fig. S12.

Atoms	Angle/°	Atoms	Angle/°	Atoms	Angle/°	Atoms	Angle/°	Atoms	Angle/°
O1-Ag1-Ag2	130.84	O1-Ag3-Ag1	95.16	O2-Ag4-S2	98.50	O2-Ag5-S1	96.56	O1-Ag7-Ag8	130.84
O1-Ag1-Ag3	131.91	O1-Ag3-Ag2	126.85	O2-Ag4-S4	96.56	O2-Ag5-S4	98.50	O1-Ag7-Ag6	131.91
O1-Ag1-Ag4	92.80	O1-Ag3-Ag4	127.13	O2-Ag4-Ag3	61.28	O2-Ag5-Ag3	139.38	O1-Ag7-Ag4	96.12
O1-Ag1-Ag5	96.12	O1-Ag3-Ag5	102.65	O2-Ag4-Ag8	139.38	O2-Ag5-Ag7	61.28	O1-Ag7-Ag5	92.80
O1-Ag1-S1	102.33	O1-Ag3-S3	102.51	O2-Ag4-S3	98.50	O2-Ag5-S2	96.56	O1-Ag7-S5	102.33
O1-Ag1-S3	103.68	O1-Ag3-S2	96.76	O2-Ag4-S5	96.56	O2-Ag5-S5	98.50	O1-Ag7-S4	103.68
O1-Ag2-Ag3	130.84	O2-Ag4-Ag1	61.28	O2-Ag5-Ag1	139.38	O1-Ag6-Ag7	130.84	O1-Ag8-Ag6	130.84
O1-Ag2-Ag1	131.91	O2-Ag4-Ag6	139.38	O2-Ag5-Ag6	61.28	O1-Ag6-Ag8	131.91	O1-Ag8-Ag7	131.91
O1-Ag2-Ag4	92.80	O2-Ag4-S1	98.50	O2-Ag5-S3	96.56	O1-Ag6-Ag4	96.12	O1-Ag8-Ag4	96.12
O1-Ag2-Ag5	96.12	O2-Ag4-S6	96.56	O2-Ag5-S6	98.50	O1-Ag6-Ag5	92.80	O1-Ag8-Ag5	92.80
O1-Ag2-S2	102.33	O2-Ag4-Ag2	61.28	O2-Ag5-Ag2	139.38	O1-Ag6-S4	102.33	O1-Ag8-S6	102.33
O1-Ag2-S1	103.68	O2-Ag4-Ag7	139.38	O2-Ag5-Ag6	61.28	O1-Ag6-S6	103.68	O1-Ag8-S5	103.68

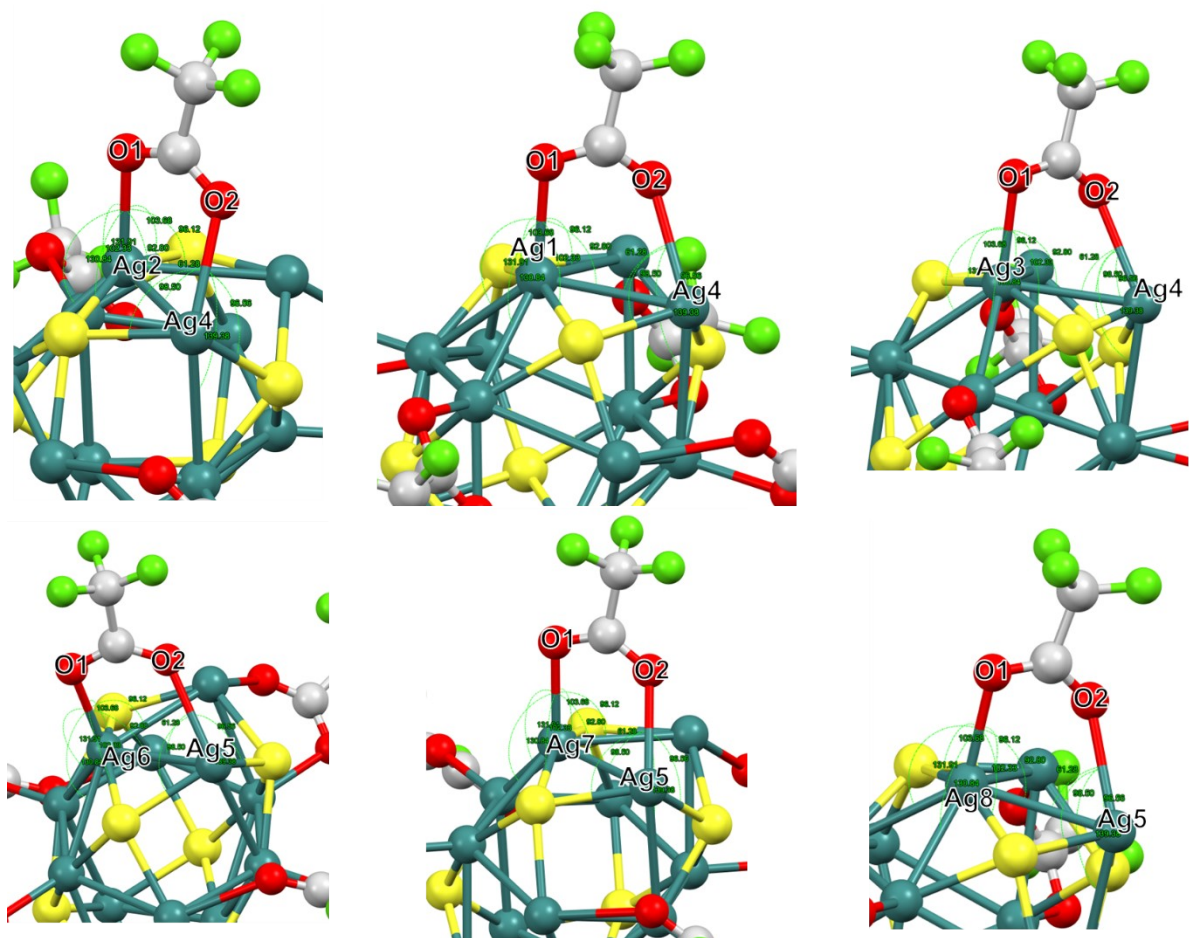


Fig. S13 Connectivities of CF_3COO^- ligands with two different Ag atoms by utilizing two O ($\mu_2\text{-}\eta^1, \eta^1$) in TUS 7.

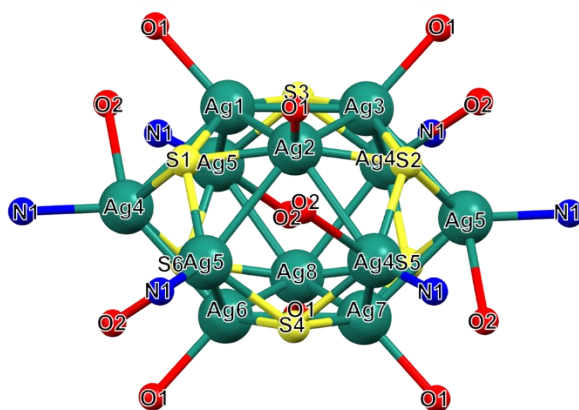


Fig. S14 Attachment of six linker molecules on the $\text{Ag}_{12}\text{S}_6\text{O}_{12}$ cluster node in **TUS 7**.

Table S17. Ag–N bond lengths corresponding to Fig. S14.

Atom1	Atom2	Bond length/Å		Bond length/Å
Ag4	N1	2.275	Maximum	2.275
Ag5	N1	2.275	Minimum	2.275
			Average	2.275
			S.D.	0.000

Table S18. N–Ag–Ag, N–Ag–O and N–Ag–S bond angles corresponding to Fig. S14.

Atoms	Angle/°	Atoms	Angle/°	Atoms	Angle/°
N1–Ag4–Ag1	137.87	N1–Ag5–Ag1	129.22	N1–Ag4–O2	85.22
N1–Ag4–Ag2	137.87	N1–Ag5–Ag2	129.22	N1–Ag5–O2	85.22
N1–Ag4–Ag3	137.87	N1–Ag5–Ag3	129.22		
N1–Ag4–Ag6	129.22	N1–Ag5–Ag6	137.87		
N1–Ag4–Ag7	129.22	N1–Ag5–Ag7	137.87		
N1–Ag4–Ag8	129.22	N1–Ag5–Ag8	137.87		
N1–Ag4–S1	117.65	N1–Ag5–S1	123.04		
N1–Ag4–S2	117.65	N1–Ag5–S2	123.04		
N1–Ag4–S3	117.65	N1–Ag5–S3	123.04		
N1–Ag4–S4	123.04	N1–Ag5–S4	117.65		
N1–Ag4–S5	123.04	N1–Ag5–S5	117.65		
N1–Ag4–S6	123.04	N1–Ag5–S6	117.65		

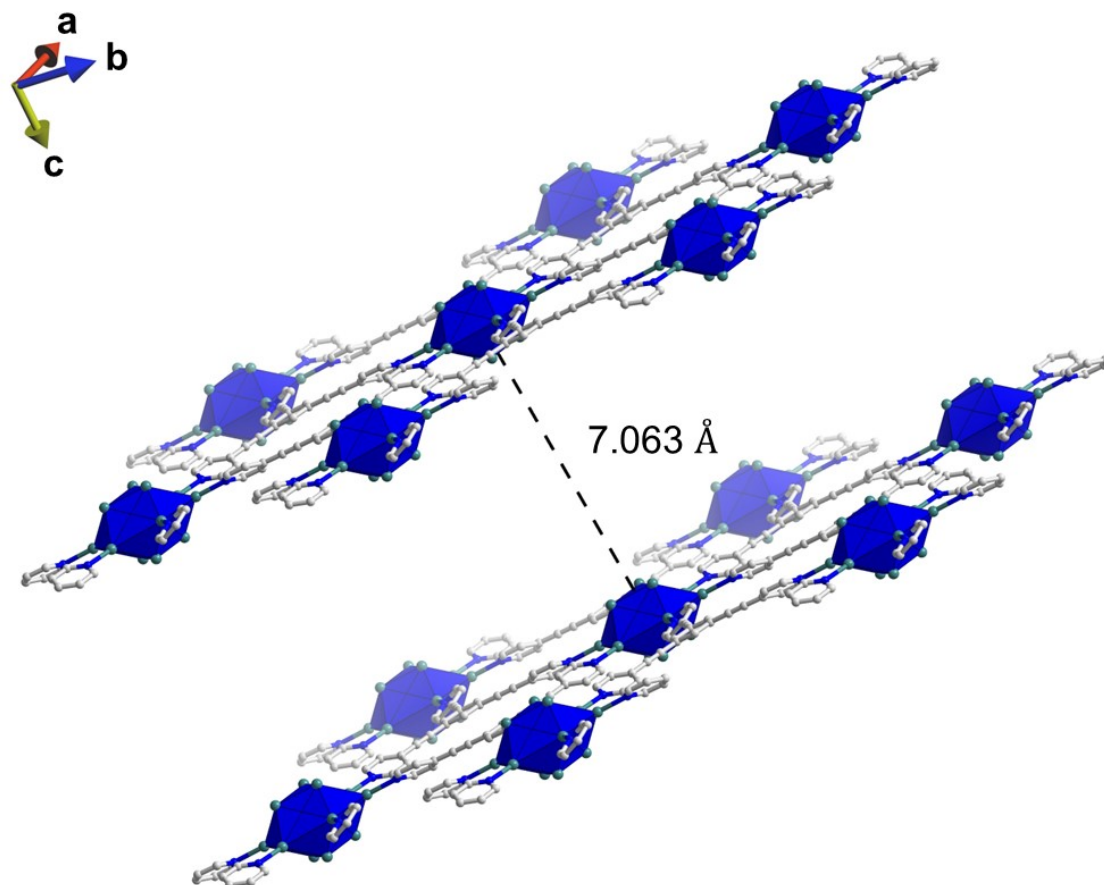
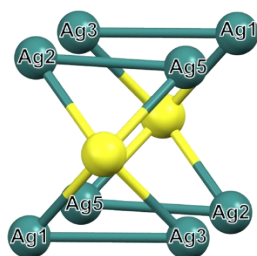
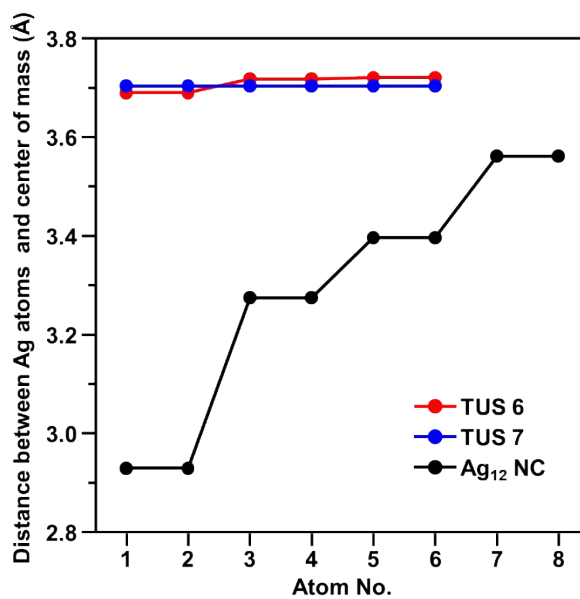
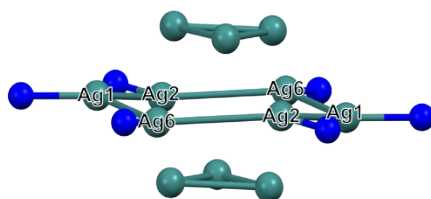


Fig. S15 The layer distance between the structure of TUS 7.

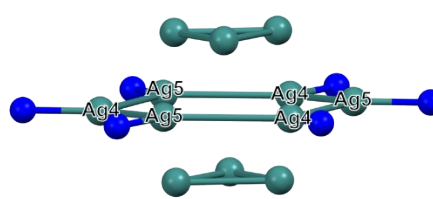
Atom No.	Ag ₈ cuboid core	Ag ₆ middle layer	
	Ag ₁₂ NC	TUS 6	TUS 7
1	Ag5	Ag6	Ag4
2	Ag5	Ag6	Ag4
3	Ag6	Ag1	Ag4
4	Ag6	Ag1	Ag5
5	Ag4	Ag2	Ag5
6	Ag4	Ag2	Ag5
7	Ag1		
8	Ag1		



Ag₁₂ NC



TUS 6



TUS 7

Fig. S16 Ag₁₂ nanocluster symmetry in relation to Ag₈ cuboid core versus SCAMs symmetry in relation to Ag₆ middle layer.

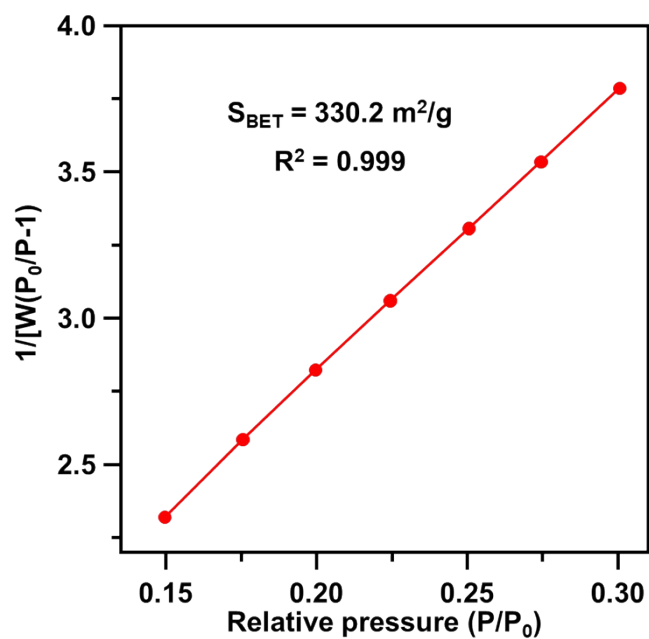


Fig. S17 BET plot for TUS 6 calculated from the N₂ adsorption isotherms at 77 K.

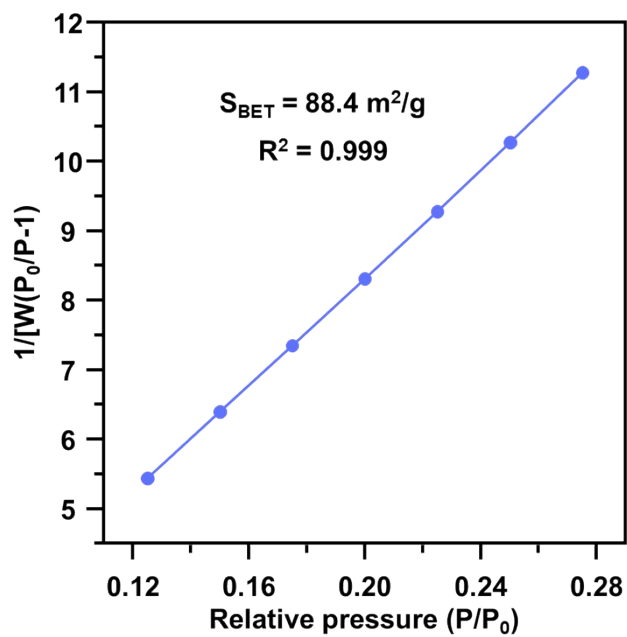


Fig. S18 BET plot for TUS 7 calculated from the N₂ adsorption isotherms at 77 K.

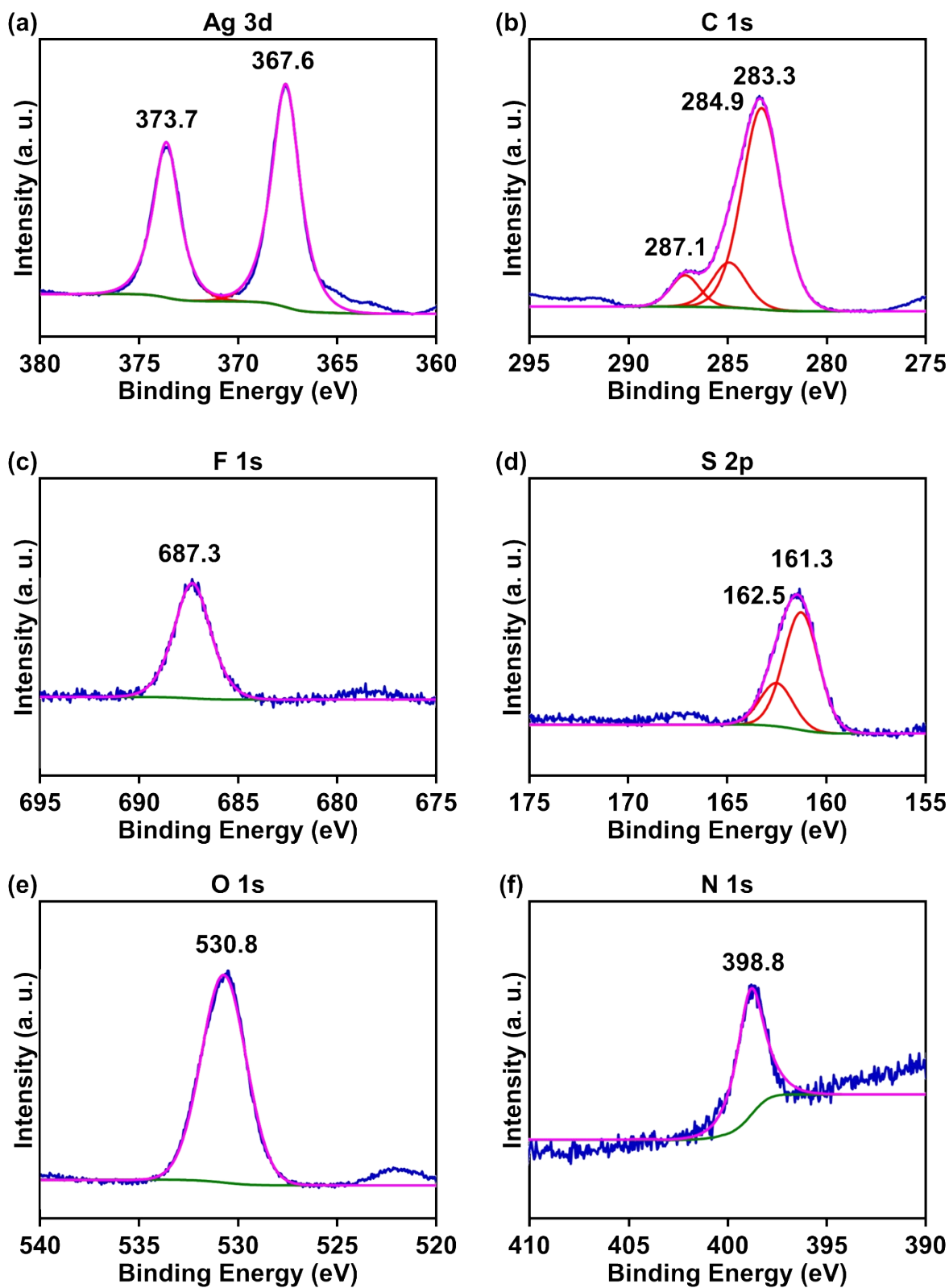


Fig. S19 High resolution binding energy plot of each element obtained from the XPS measurement of TUS 6.

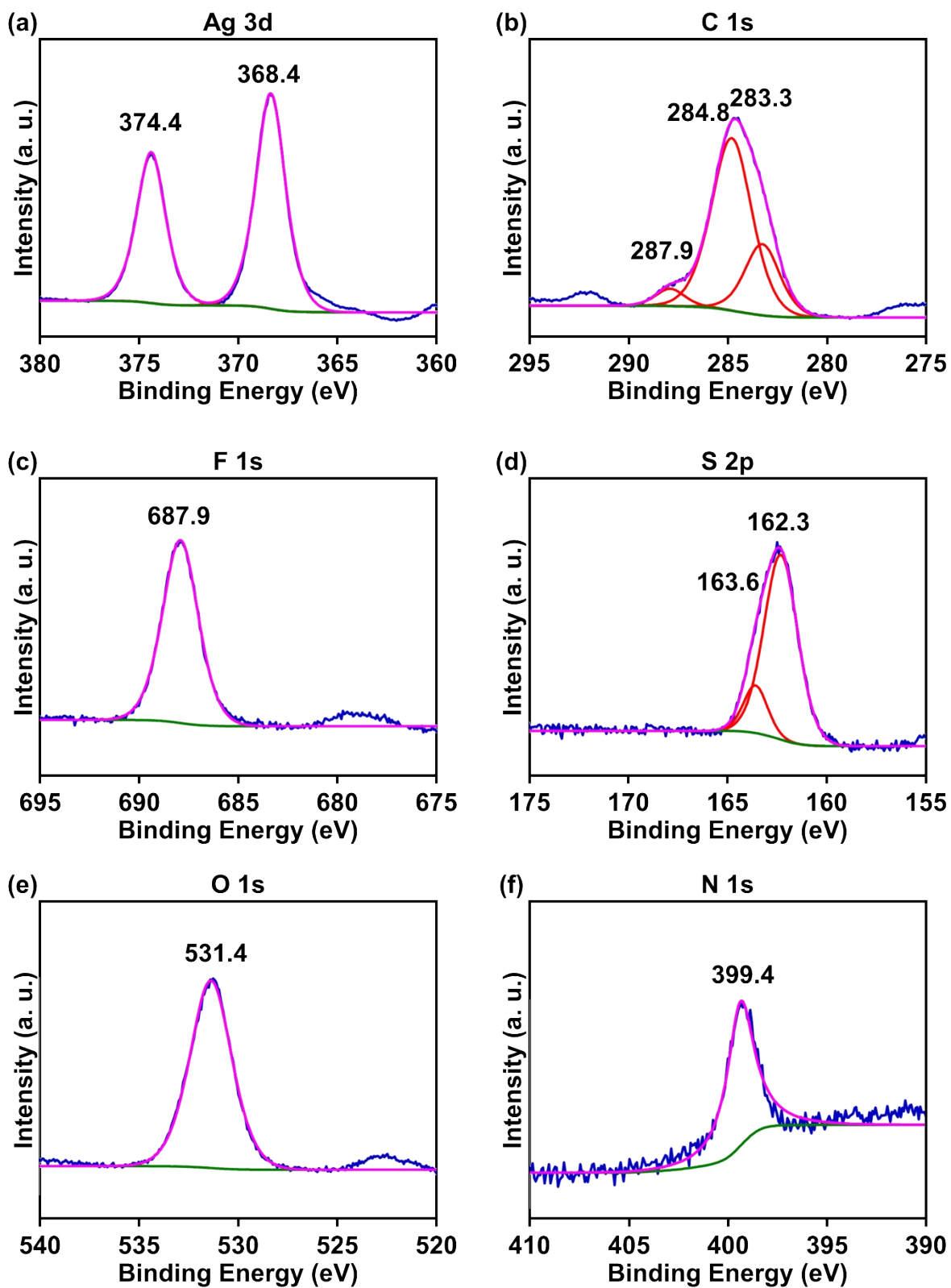


Fig. S20 High resolution binding energy plot of each element obtained from the XPS measurement of TUS 7.

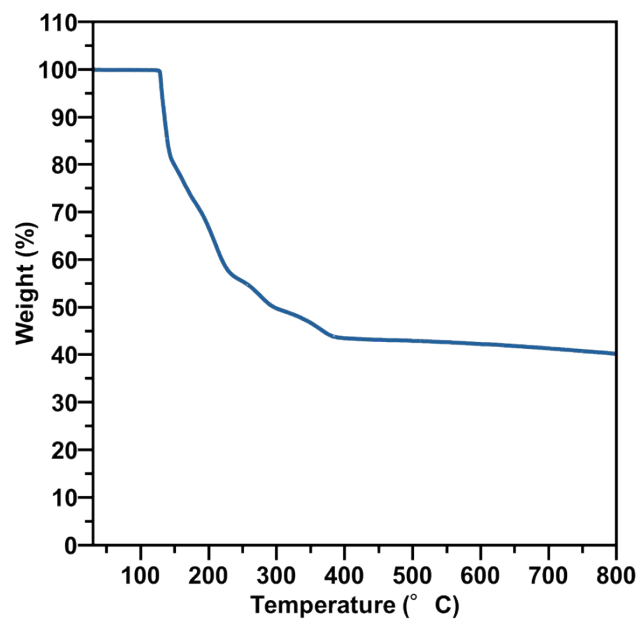


Fig. S21 TGA curve of TUS 6 under N₂ atmosphere.

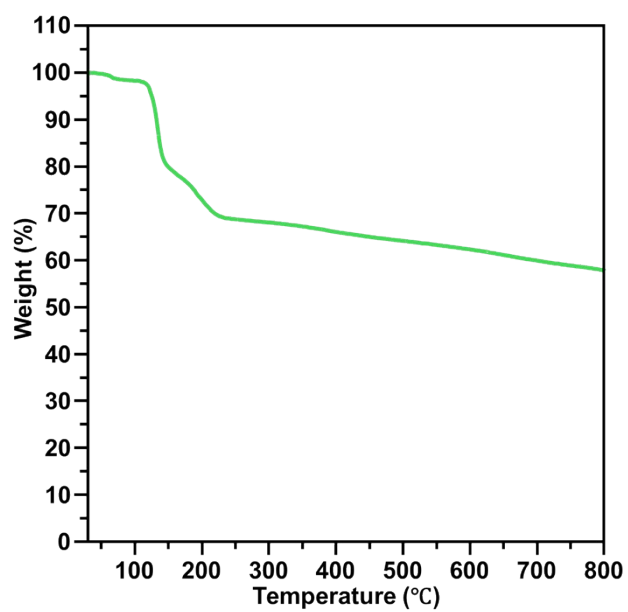


Fig. S22 TGA curve of TUS 7 under N₂ atmosphere.

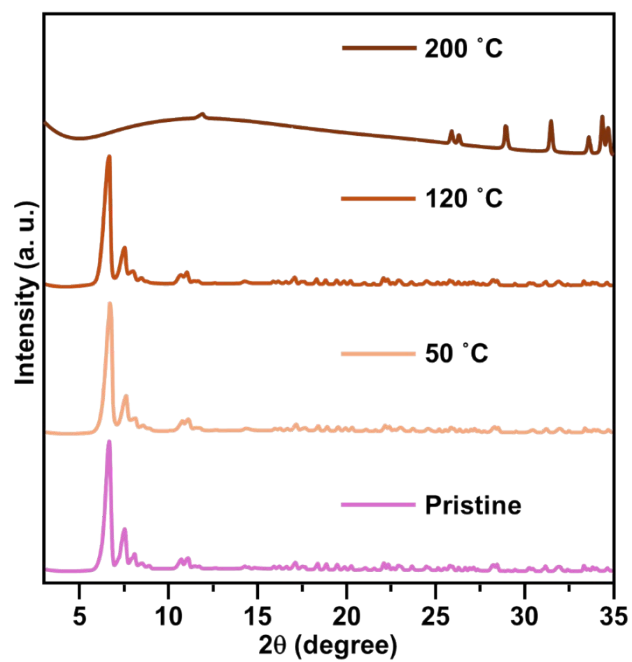


Fig. S23 PXR D profiles of TUS 6 after heating at different temperatures.

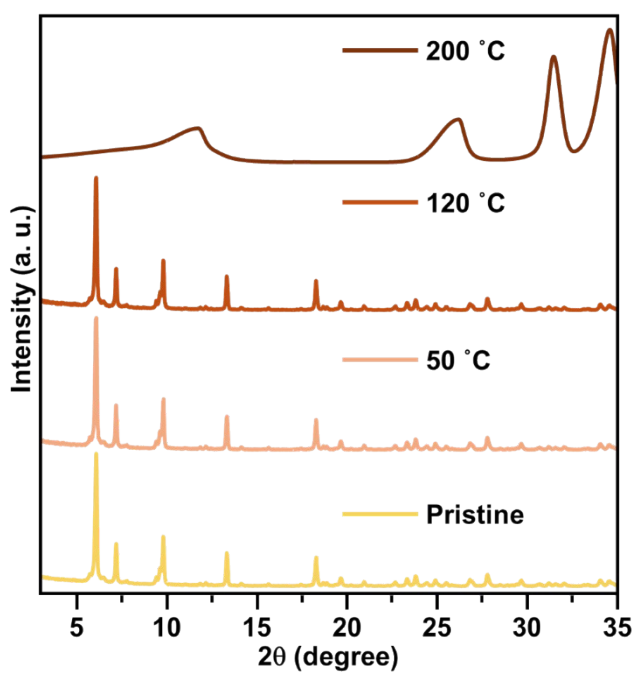


Fig. S24 PXR D profiles of TUS 7 after heating at different temperatures.

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