Effectively enhancing ion diffusion in superconcentrated ionic liquid electrolytes by co-solvent additives

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SUPPLEMENTARY INFORMATION

Supplementary Information

 $E = \frac{abs(Experiment - Model)}{Experiment} 100\%$ for predicted and experimental

diffusivities (%) calculated at 80 °C.

	Li	FSI	N1113	Co-solvent
100IL	75	36	62	
20EC	42	9	14	8
20DME	36	15	42	23

Table S- 2: Donnor number and dielectric constant of EC and DME.

	Donnor number	Dielectric constant	
EC	16.3 ¹	89.8 ²	
DME	20.01	7.2 ²	

Table S- 3: Number of ions/molecules used in the simulations.

System	Li	FSI	N1113	EC	DME
100IL	320	640	320		
20EC	320	603	283	227	
20DME	320	603	283		222



Figure S-1: (a to d) Chemical structures of FSI, N1113, DME and EC.



Figure S- 2: (a) Li-EC and (b) Li-DME binding energies calculated using density functional theory at a B3LYP/6-31G(d,p) level of theory with Gaussian 09 software based on the equation Eb = E(Lisolvent)-E(Li)-E(solvent). Literature reports Eb's of -2.24 eV and -2.84 eV for Li-EC and Li-DME, respectively³.



Figure S- 3: (a) Fraction of EC that is not coordinated with Li at 5, 10 and 20 wt% EC concentrations.



Figure S- 4: (a and b) MSD functions for Li in full FSI shell (solid lines) and hybrid shells (dashed lines). The left and right plots correspond to the 20 wt% EC and the 20 wt% DME systems, respectively.



Figure S- 5: (a) Fraction of Li along with their solvation shell that are detached from the Li-FSI network for the 20 wt% EC and the 20 wt% DME systems. (b and c) Simulation snapshots that highlight Li that are detached from the Li-FSI network colored in red for 20EC and 20DME systems, respectively.



Figure S- 6: (a to c) Li-Li RDF's along with their integration curves, colored in red, for the 100IL, 20EC and 20DME systems, respectively. (d) Calculated average distance between Li and coordination number.

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