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Trivalent Dialkylaminopyridine-Catalyzed Site-Selective Mono-O-acylation of Partially-

Protected Pyranosides

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1. Reaction optimization:

The screening of the reactions was performed under kinetically controlled condition.¹ As given in the tables below, a catalyst loading of 1.5 mol% was optimized for all the acylations. With the increase in steric bulkiness of the anhydrides, reactions were conducted at different durations, in order to optimize the yield of the major product, with 1.5 mol% of the catalyst **1**.

Ph 0 HO 6	нооме	Catalyst 1 Ac ₂ O (1.1 mol. equiv.) CH ₃ CN, 0 °C - rt, time	Ph O HO HO 7	O Ph C + Ac O OMe	HO OM	Ph 0 + Ac0- e	Aco OMe
	Entry	Catalyst (mol%)	Time (h)	7 (%)	8 (%)	9 (%)	
	1	10	2	41	3	56	
	2	5	2	42	6	52	
	3	3.3	2	48	8	34	
	4	1.1	3	53	8	21	
	5	1.5	3	65	10	25	

Table S1. Acetylation of 6 under varied reaction conditions.^a

^aReactions were conducted in 30 mg (0.106 mmol). Product ratios determined by HPLC analysis.

Table S2. Acetylation of 40 under varied reaction condition.^a

TrC HO- HC	HO OMe	catalys Ac ₂ O (1.1 CHCl ₃ , 0 °C-	t 1 T equiv) Hi ────────────────────────────────────		+ AcO HO HO 42 OMe + Other mixtu	regioisomeric res of acetate
Entry	Catalyst	Time	41 (%)	42 (%)	Multiple acylated	Conversion
	(mol%)	(min)			products	
1	10	30	32	ND	19% + 25% + 14% + 8%	Quantitative
2	5	30	41	ND	14% + 23% + 13% + 9%	Quantitative
3	3.3	30	54	9	4% + 8%	86

4	1.1	30	58	6	5% + 1% + 4%	77
5	1.5	30	64	9	6% + 2%	86
6	1.1	45	65	11	6% + 3%	85
7	1.5	45	74	9	7% + 3%	94

2. 4-N-Methylamino pyridine (3)-mediated acylation:



The product ratios determined by HPLC analysis. 4-*N*-methylamino pyridine **3** was not an active catalyst for site-selective acetylation of **6**.

3. Calculation of the regioisomeric ratio:

An aliquot of the reaction mixture was quenched with a few drops of MeOH. After 30 min, the solvent was evaporated, diluted in CH_2Cl_2 (1 mL) and analyzed by HPLC (analytical silica column, photodiode array detector and using hexane/EtOAc binary elution).

Table S3. Site-selectivity for the acetylation of 6 at different time intervals.^a

Ph⁄	70-	No.	Catalyst 1 ([1.5 mol%) → P	h 697	O Ph OT	5070	
	Η̈́Ο-	6 HO OMe	Ac₂O (1.1 n CH₃CN, 0 ^o	nol. equiv.) C - rt, time	HO_Ac	OMe	внооме	
		Time (m	nin) '	7 (%)	8 (%)	Ratio of 7/8		
		15		13	2	6.5		
		30		25	4	6.2		
		45		34	5	6.8		

-				
	60	41	6	6.8
	90	49	7	7.0
	120	57	8	7.1
	150	63	9	7.0
	180	65	10	6.5

Table S4. Site-selectivity for the isobutyration of **6** at different time intervals.^{*a*}

Ph O H	$\frac{O}{O} + O + O + O + O + O + O + O + O + O +$	talyst 1 (1.5 mol%) <mark> PrCO)₂O</mark> (1.1 mol. equiv.) CH₃CN, 0 °C - rt, time	Ph 0 HO i-Pr	Ph 0 i-Proco $j_0 OMe$ 11	нооме
	Time (min	n) 10 (%)	11 (%)	Ratio of 10/11	
	10	46	5.6	8.2	
	30	64	7.9	8.1	
	50	68	8.6	7.9	
	70	71	9.3	7.6	
	90	73	9.8	7.4	
	120	75	10	7.5	

^{*a*}Reactions were conducted in 30 mg (0.106 mmol). Product ratios determined by HPLC analysis.

Table S5.	Site-selectivity	y for the	pivaloylation	of 6 at	different	time intervals. ^a
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Ph O HO HO	HO _{OMe} Ca	talyst 1 (1.5 mol%) CO) ₂ O (2.2 mol. equ ₃ CN, 0 ºC - rt, time	Ph O niv.) Ho t	BuOCO 13 Ph OC t-BuOCO t-BuOCO	HO OMe
	Time (h)	13 (%)	14 (%)	Ratio of 13/14	
	1	17	4	4.2	
	2	33	9	3.6	
	3	40	12	3.3	
	4	47	14	3.3	
	6	56	16	3.5	
	8	64	18	3.5	

10	68	19	3.6
12	72	20	3.6
14	75	22	3.4
16	77	23	3.3

Table S6. Site-selectivity for the benzoylation of 6 at different time intervals.^{*a*}

Ph	HO HO OMe	Catalyst 1 (1.5 mol%) Bz ₂ O (2.2 mol. equiv.) CH ₃ CN, 0 °C - rt, time	Ph O HO HO 15	z_{OMe}^{O} + $\frac{Ph}{BzO}$ + $\frac{O}{BzO}$ + $\frac{Ph}{BzO}$ + $\frac{O}{H}$ + $\frac{Ph}{BzO}$ + Ph	-0 10 0 Me
	Time (h)	15 (%)	16 (%)	Ratio of 15/16	
	1	22	5	4.4	
	2	32	7	4.6	
	3	42	9	4.6	
	4	47	10	4.7	
	6	50	11	4.5	
	8	52	11.5	4.5	
	10	54	12	4.5	

^aReactions were conducted in 30 mg (0.106 mmol). Product ratios determined by HPLC analysis.

Table S7. Site-selectivity for the acetylation of **40** at different time intervals.^{*a*}

$\begin{array}{c} \text{nol\%}) \\ \begin{array}{c} \text{equiv}) \\ \text{t, time} \end{array} \begin{array}{c} \text{TrO} \\ \text{HO} \\ \text{AcO} \\ \text{HO} \\ \text{HO} \\ \text{HO} \\ \text{OMe} \end{array} \begin{array}{c} \text{TrO} \\ \text{AcO} \\ \text{HO} \\ \text{HO} \\ \text{HO} \\ \text{HO} \\ \text{HO} \\ \text{OMe} \end{array}$	atalyst 1 (1.5 n Ac ₂ O (1.1 mol. o CHCl ₃ , 0 °C- n	HO HO HO 40	TrC HO ⁻ HO
42 (%) Ratio of 41/42) 41 (%)	Time (mi	
3 6.6	20	5	
6 5.6	34	10	
7 6.1	43	15	
8 7.1	57	25	
8.5 8.0	68	35	
9 8.2	74	45	
42 (76) Ratio of 41/4 3 6.6 6 5.6 7 6.1 8 7.1 8.5 8.0 9 8.2	20 34 43 57 68 74	5 10 15 25 35 45	

HO HO HO 40	catalyst 1 (* (<i>i</i> -PrCO) ₂ O (CHCI ₃ , 0 DMe	1.5 mol%) 1.1 mol. equiv °C- rt, time	() <i>i-</i> ProCo	HO _{OMe} HO _{OMe}	
	Time (min)	43 (%)	44 (%)	Ratio of 43/44	-
	5	35	2.5	14	
	10	53	3.6	14.5	
	15	69	4.7	14.7	
	20	80	5.5	14.5	
	30	89	6	14.8	

Table S8. Site-selectivity for the isobutyration of 40 at different time intervals.^{*a*}

^aReactions were conducted in 30 mg (0.106 mmol). Product ratios determined by HPLC analysis.

Table S9. Site-selectivity for the pivaloylation of 40 at different time intervals.^{*a*}

HO HO HO 40 ^H	catalys O O OMe CHO	t 1 (1.5 mol%))₂O (2.2 mol. equiv) Cl ₃ , 0 °C- rt, time	TrO− HO∽∖ t-BuOCO−	HOOMe t-BuOCO HOOMe 45 46	нооме
	Time (h)	45 (%)	46 (%)	Ratio of 45/46	
	1	19	3	6.1	
	2	35	5.5	6.3	
	3	45	7	6.4	
	4	53	9	5.8	
	5	59	10	5.9	
	7	66	11	6	
	9	73	12	6.0	
	11	78	12.5	6.2	
	13	82	13.5	6.0	
	16	84	14	6	

^aReactions were conducted in 30 mg (0.106 mmol). Product ratios determined by HPLC analysis.

	catalyst 1 (1.5 mol%) Bz ₂ O (2.2 mol. equiv)	TrO HO		Ż
40 HO Me	CHCl ₃ , 0 °C- rt, time	BzO H 47	о _{ОМе} 48	 OMe
Time (h)	47 (%)	48 (%)	Ratio of 47/48	
0.5	12	3	4	
1	23	5	4.6	
2	35	8	4.4	
4	50	14	3.6	
5	58	18	3.2	
6	64	20	3.2	
8	73	23	3.2	

Table S10. Site-selectivity for the benzoylation of 40 at different time intervals.^a

No significant change in the regioisomeric ratios was observed at different time intervals for the acylations.

4. Turnover Frequency (TOF) for major regioisomeric product:

Catalyst 1 (1.5 mol%) was used in each case. Conversion to major site-selective product at different time intervals are given in Table 3-10. TOF = mole of product formed / (mole of catalyst used \times time)²

Table 11. Comparison of TOF for acylation of **6** with varied acylation agents.

Reaction type	TOF (h ⁻¹)	Reaction Type	TOF (h ⁻¹)
Acetylation of 6	21.8	Acetylation of 40	92.4
Isobutyration of 6	54.6	Isobutyration of 40	184
Pivaloylation of 6	7.8	Pivaloylation of 40	8.8
Benzoylation of 6	9.3	Benzoylation of 40	8.3

5. HPLC traces and determination of the product ratios and conversion.

(Note: Void volume appears between 0 and ~ 3.5 min. in all HPLC traces).



Acylation of methyl-6-O-trityl glucopyranoside (40)

Figure S1. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 90:10) for acylation of 40 by DMAP.

Retention times: 7.6 min. (major isomer 41), 27.6 min. (starting material 40).



Figure S2. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 90:10) for acylation of **40** by **1**. Retention times: 7.7 min. (major isomer **41**), 27.8 min. (starting material **40**) and remaining peaks correspond to the minor isomers.



Isobutyration of methyl-6-*O*-trityl glucopyranoside (40)

Figure S3. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of DMAP-catalyzed isobutyration of **40**. Retention times: 4.7 min. (major isomer **43**), 42.2 min. (starting material **40**) and remaining peaks correspond to the minor isomers.



Figure S4. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed isobutyration of **40**. Retention times: 4.7 min. (major isomer **43**), 42.2 min. (starting material **40**) and remaining peaks correspond to the minor isomers.



Pivaloylation of methyl-6-*O***-trityl glucopyranoside (40)**

Figure S5. HPLC chromatogram (eluant: EtOAc-n-hexane = 60:40) of DMAP-catalyzed pivaloylation of **40**. Retention times: 4.1 min. (major isomer **45**), 40.7 min. (starting material **40**) and remaining peaks correspond to the minor isomers.



Figure S6. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed pivaloylation of **40**. Retention times: 4.1 min. (major isomer **45**), 42.5 min. (starting material **40**) and remaining peaks correspond to the minor isomers.



Benzoylation of methyl-6-O-trityl glucopyranoside (40)

Figure S7. HPLC chromatogram (eluant: EtOAc-n-hexane = 60:40) of DMAP-catalyzed benzoylation of **40**. Retention times: 4.3 min. (major isomer **47**), 41.9 min. (starting material **40**) and remaining peaks correspond to the minor isomers.



Figure S8. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed benzoylation of **40**. Retention times: 4.3 min. (major isomer **47**), 42.9 min (starting material **40**) and remaining peaks correspond to the minor isomers.



Acylation of methyl 6-*O*-trityl-α-D-mannopyranoside (49)

Figure S9. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of DMAP-catalyzed acylation of **49**. Retention times: 8.2 min. (major isomer **50**), 18.2 min. (starting material **49**) and remaining peaks correspond to the minor isomers.



Figure S10. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed acylation of **49**. Retention times: 8.2 min. (major isomer **50**), 18.4 min. (starting material **49**) and remaining peaks correspond to the minor isomers.



Isobutyration of methyl 6-*O*-trityl-α-D-mannopyranoside (49)

Figure S11. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of DMAP-catalyzed isobutyration of **49**. Retention times: 4.4 min. (major isomer **52**), 18.4 min. (starting material **49**) and remaining peaks correspond to the minor isomers.



Figure S12. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed isobutyration of **49**. Retention times: 4.4 min. (major isomer **52**), 18.6 min. (starting material **49**) and remaining peaks correspond to the minor isomers.



Pivaloylation of methyl 6-*O***-trityl-α-D-mannopyranoside (49)**

Figure S13. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of DMAP-catalyzed pivaloylation of **49**. Retention times: 3.9 min. (major isomer **54**), 18.1 min. (starting material **49**) and remaining peaks correspond to the minor isomers.



Figure S14. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed pivaloylation of **49**. Retention times: 3.9 min. (major isomer **54**), 18.2 min. (starting material **49**) and remaining peaks correspond to the minor isomers.



Benzoylation of methyl 6-*O*-trityl-α-D-mannopyranoside (49)

Figure S15. HPLC chromatogram (eluant: EtOAc-n-hexane = 60:40) of DMAP-catalyzed benzoylation of **49**. Retention times: 5.4 min. (major isomer **56**), 18.4 min. (starting material **49**) and remaining peaks correspond to the minor isomers.



Figure S16. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed benzoylation of **49**. Retention times: 5.4 min. (major isomer **56**), 18.2 min. (starting material **49**) and remaining peaks correspond to the minor isomers.



Acylation of methyl 6-O-trityl-α-D-galactopyranoside (58)

Figure S17. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of DMAP-catalyzed acylation of **58**. Retention times: 15.3 min. (major isomer **59**), 37.5 min. (starting material **58**) and remaining peaks correspond to the minor isomers.



Figure S18. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed acylation of **58**. Retention times: 14.9 min. (major isomer **59**), 36.7 min. (starting material **58**) and remaining peaks correspond to the minor isomers.



Isobutyration of methyl 6-*O*-trityl-α-D-galactopyranoside (58)

Figure S19. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of DMAP-catalyzed isobutyration of **58**. Retention times: 5.8 min. (major isomer **61**), 37.2 min. (starting material **58**) and remaining peaks correspond to the minor isomers.



Figure S20. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed isobutyration of **58**. Retention times: 5.8 min. (major isomer **61**), 37.3 min. (starting material **58**) and remaining peaks correspond to the minor isomers.



Pivaloylation of methyl 6-*O***-trityl-α-D-galactopyranoside (58)**

Figure S21. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 65:35) of DMAP-catalyzed pivaloylation of **58**. Retention times: 4.5 min. (major isomer **63**), 31 min. (starting material **58**) and remaining peaks correspond to the minor isomers.



Figure S22. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 65:35) of **1**-catalyzed pivaloylation of **58**. Retention times: 4.6 min. (major isomer **63**), 31 min. (starting material **58**) and remaining peaks correspond to the minor isomers.



Benzoylation of methyl 6-*O*-trityl-α-D-galactopyranoside (58)

Figure S23. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of DMAP-catalyzed benzoylation of **58**. Retention times: 5.3 min. (major isomer **65**), 36.7 min. (starting material **58**) and remaining peaks correspond to the minor isomers.



Figure S24. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 60:40) of **1**-catalyzed benzoylation of **58**. Retention times: 5.3 min. (major isomer **65**), 36.7 min. (starting material **58**) and remaining peaks correspond to the minor isomers.



Figure S25. HPLC chromatogram (eluant: EtOAc-*n*-hexane = 90:10) for acylation of **40** by DMAP and Et₃N (1.1 mol. equiv.). Retention times: 7.6 min. (major isomer **41**), 27.6 min. (starting material **40**).

6. General characterization information of catalysts 1 and 5.

The DAAP catalysts **1** and **5** were characterized by NMR spectroscopies and mass spectrometry. Characteristics of ¹H NMR spectra were the changes of the chemical shifts of the CH_2 - moieties at the peripheries, depending on the nature of the attached functional group. Thus, resonance of - CH_2Cl at ~ 3.60 ppm (t) is replaced by resonance at ~ 3.91 ppm (t) of - CH_2 -DAAP. ¹³C NMR chemical shifts also ascertained the peripheral – CH_2 - moiety, that of ~ 43 ppm for - CH_2Cl and ~ 56 ppm for - CH_2 -DAAP. Further, differences in the resonance values for -NCH₃- moiety of DAAP prior to (**3**) and after functionalization (**1** and **5**) were noted. Mass spectrometry ascertained the structural homogeneity of derivatives **1** and **5**.

7. References

- P. G. Goekjian and S. Vidal, Regioselective Protection at the Secondary Positions of Carbohydrates with Acyclic Protecting Groups, in *Protecting Groups – Strategies and Applications in Carbohydrate Chemistry*, ed. S. Vidal, Wiley-VCH: Weinheim, 2019; ch. 4, pp. 109-144.
- 2. K. Price and D. T. McQuade, Chem. Commun., 2005, 1714.

NMR Spectra



Figure S26. ¹H NMR spectrum of 1 (CDCl₃, 400 MHz).



Figure S27. ¹³C NMR spectrum 1 (CDCl₃, 100 MHz).

Figure S28. ¹H NMR spectrum of compound 4 (CDCl₃, 400 MHz).

Figure S29. ¹³C NMR spectrum of compound 4 (CDCl₃, 100 MHz).

Figure S30. ¹H NMR spectrum of 5 (D₂O, 400 MHz).

Figure S31. ¹³C NMR spectrum of **5** (D₂O, 100 MHz).

Figure S32. ¹H NMR spectrum of 7 (CDCl₃, 400 MHz).

Figure S33. ¹³C NMR spectrum of 7 (CDCl₃, 100 MHz).

Figure S34. ¹H NMR spectrum of 10 (CDCl₃, 400 MHz).

Figure S35. ¹³C NMR spectrum of 10 (CDCl₃, 100 MHz).

Figure S36. ¹H NMR spectrum of 13 (CDCl₃, 400 MHz).

Figure S37. ¹³C NMR spectrum of 13 (CDCl₃, 100 MHz).

Figure S38. ¹H NMR spectrum of 15 (CDCl₃, 400 MHz).

Figure S39. ¹³C NMR spectrum of 15 (CDCl₃, 100 MHz).

Figure S40. ¹H NMR spectrum of 20 (CDCl₃, 400 MHz).

Figure S41. ¹³C NMR spectrum of 20 (CDCl₃, 100 MHz).

Figure S42. ¹H NMR spectrum of a mixture of 22 (30%) and 23 (70%) (CDCl₃, 400 MHz).

Figure S43. ¹³C NMR spectrum of a mixture of 22 and 23 (CDCl₃, 100 MHz).

Figure S44. ¹H NMR spectrum of a mixture of 24 (21%) and 25 (79%) (CDCl₃, 400 MHz).

Figure S45. ¹³C NMR spectrum of a mixture of 24 and 25 (CDCl₃, 100 MHz).

Figure S46. ¹H NMR spectrum of a mixture of 26 (23%) and 27 (77%) (CDCl₃, 400 MHz).

Figure S47. ¹³C NMR spectrum of a mixture of 26 and 27 (CDCl₃, 100 MHz).

Figure S48. ¹H NMR spectrum of 31 (CDCl₃, 400 MHz).

Figure S49. ¹³C NMR spectrum of **31** (CDCl₃, 100 MHz).

Figure S50. ¹H NMR spectrum of 34 (CDCl₃, 400 MHz).

Figure S51. ¹³C NMR spectrum of 34 (CDCl₃, 100 MHz).

Figure S52. ¹H NMR spectrum of 36 (CDCl₃, 400 MHz).

Figure S53. ¹³C NMR spectrum of 36 (CDCl₃, 100 MHz).

Figure S54. ¹H NMR spectrum of 38 (CDCl₃, 400 MHz).

Figure S55. ¹³C NMR spectrum of **38** (CDCl₃, 100 MHz).

Figure S56. ¹H NMR spectrum of 41 (CDCl₃, 400 MHz).

Figure S57. ¹³C NMR spectrum of 41 (CDCl₃, 100 MHz).

Figure S58. ¹H NMR spectrum of 43 (CDCl₃, 400 MHz).

Figure S59. ¹³C NMR spectrum of 43 (CDCl₃, 100 MHz).

Figure S60. ¹H NMR spectrum of 45 (CDCl₃, 400 MHz).

Figure S61. ¹³C NMR spectrum of 45 (CDCl₃, 100 MHz).

Figure S62. ¹H NMR spectrum of 47 (CDCl₃, 400 MHz).

Figure S63. ¹³C NMR spectrum of 47 (CDCl₃, 100 MHz).

Figure S64. ¹H NMR spectrum of a mixture of 50 (70%) and 51 (30%) (CDCl₃, 400 MHz).

Figure S65. ¹³C NMR spectrum of a mixture of 50 and 51 (CDCl₃, 100 MHz).

Figure S66. ¹H NMR spectrum of 52 (CDCl₃, 400 MHz).

Figure S67. ¹³C NMR spectrum of 52 (CDCl₃, 100 MHz).

Figure S68. ¹H NMR spectrum of 54 (CDCl₃, 400 MHz).

Figure S69. ¹³C NMR spectrum of 54 (CDCl₃, 100 MHz).

Figure S70. ¹H NMR spectrum of 56 (CDCl₃, 400 MHz).

Figure S71. ¹³C NMR spectrum of 56 (CDCl₃, 100 MHz).

Figure S72. ¹H NMR spectrum of 59 (CDCl₃, 400 MHz).

Figure S73. ¹³C NMR spectrum of 59 (CDCl₃, 100 MHz).

Figure S74. ¹H NMR spectrum of 61 (CDCl₃, 400 MHz).

Figure S75. ¹³C NMR spectrum of 61 (CDCl₃, 100 MHz).

Figure S76. ¹H NMR spectrum of a mixture of 63 (77%) and 64 (23%) (CDCl₃, 400 MHz).

Figure S77. ¹³C NMR spectrum of a mixture of 63 and 64 (CDCl₃, 100 MHz).

Figure S78. ¹H NMR spectrum of 65 (CDCl₃, 400 MHz).

Figure S79. ¹³C NMR spectrum of 65 (CDCl₃, 100 MHz).