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# **Electronic SupplementaryInformation**

# New routes of azomethine ylide generation from prolines to synthesize diverse *n*-heterocycles: a DFT supported *endo*-selective mechanism

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#### 1. Structure elucidation of 6g: regio and stereo-selectivity



Azomethine ylide derived from L-proline undergoes a [3+2] cycloaddition reaction through the abovementioned HOMO-LUMO intereaction, which accounts for the regioselectivity. There is lots of theoretical and experimental evidence in the literature.<sup>1</sup> The reverse selectivity is observed with azomethine ylide derived from L-proline ester without decarboxylation.<sup>2</sup> The endo-selectivity is discussed in the manuscript with the help of literature.

From the NOESY spectra (SI figure 1) of compound **6g**, it is very clear that more contours are present in the aromatic region apart from the interaction of orthoprotonsin aromatic rings. These excess interactions (blue cloured) among the aromatic protons are shown below.



SI Figure 1: NOESY spectra of compound 6g



#### 2. References

K. Alimohammadi, Y. Sarrafi, M. Tajbakhsh, S. Yeganegi, M. Hamzehloueian, *Tetrahedron*, 2011, 67
 1589;(a) Y. Toma, M. Kunigami, K. Watanabe, M. Higashi, S. Arimitsu, *J. Flu. Chem.*, 2016, 189, 22; (c)
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 V. V. Lipson, D. V. Atamanuk, *Mol Divers*, 2016, 20, 299.

2. K. Yamazaki, P. Gabriel, G. D. Carmine, J. Pedroni, M. Farizyan, T. A. Hamlin, D. J. Dixon, *ACS Catal.*, 2021, **11**, 7489.

3. <sup>1</sup>H and <sup>13</sup>C-NMR spectra of all the compounds 6a-j,7a-l, 8a-d, 9a-j, 10a-d and <sup>19</sup>F-NMR spectra of 6g, 7e, 9b

SI Figure 2: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 6a



# SI Figure 3: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 6b



# SI Figure 4: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 6c



## SI Figure 5: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 6d











SI Figure 9: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 6h





SI Figure 10: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 6i





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SI Figure 13: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound7b





SI Figure 14: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 7c





SI Figure 15: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 7d





SI Figure 16: <sup>1</sup>H,<sup>13</sup>Cand <sup>19</sup>F-NMR spectra of compound 7e







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## SI Figure 17: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 7f



## SI Figure 18: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 7g



## SI Figure 19: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 7h



# SI Figure 20: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 7i



#### SI Figure 21: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 7j



## SI Figure 22: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 7k





# SI Figure 24: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 8a



# SI Figure 25: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound8b





SI Figure 26: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 8c





SI Figure 27: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 8d





SI Figure 28: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 9a





SI Figure 29: <sup>1</sup>H,<sup>13</sup>Cand<sup>19</sup>F-NMR spectra of compound 9b





SI Figure 30: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 9c



SI Figure 31: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 9d



SI Figure 32: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 9e



#### SI Figure 33: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 9f



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## SI Figure 34: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 9g



SI Figure 35: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 9h



# SI Figure 36: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 9i





SI Figure 37: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 9j





SI Figure 38: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 10a





SI Figure 39: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 10b





SI Figure 40: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 10c





SI Figure 41: <sup>1</sup>H and <sup>13</sup>C-NMR spectra of compound 10d



![](_page_45_Figure_0.jpeg)

SI Figure 42: <sup>1</sup>H,<sup>13</sup>C and <sup>19</sup>F-NMR spectra of compound 10e

![](_page_45_Figure_2.jpeg)

![](_page_46_Figure_0.jpeg)

20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210 -22 f1 (ppm)

#### 4. (a) Crystal Summary Data of Compound 9a(CCDC 1832647)

![](_page_47_Figure_1.jpeg)

- Chemical formula and formula weight (M): C16 H18 N2 O2 and 270.14
- Crystal system: Triclinic
- ✤ Unit-cell dimensions (angstrom or pm, degrees) and volume, with edges: a=7.827(4)
  b=10.029(4) c=17.446(8)alpha=89.147(6) beta=87.469(6) gamma=82.695(6)
- Temperature: 296(2)
- ✤ Space group symbol: P -1
- ✤ No. of formula units in unit cell (Z): 4
- ♦ Number of reflections measured and/or number of independent reflections, Rint: 2746
- ✤ Final R values (and whether quoted for all or observed data):0.0484

#### (b) Crystal Summary Data of Compound 10e (CCDC 2312186)

![](_page_47_Figure_11.jpeg)

- Chemical formula and formula weight (M): C23 H23 F N2 O2and 378.43
- Crystal system: Monoclinic
- Unit-cell dimensions (angstrom or pm, degrees) and volume, with edges: a=19.48(2)b=5.587(5) c=18.462(17)alpha=90beta=101.83(3)gamma=90
- ✤ Temperature: 297(2)
- ✤ Space group symbol: P2(1)/c
- ✤ No. of formula units in unit cell (Z): 4
- ♦ Number of reflections measured and/or number of independent reflections, Rint: 2521
- ✤ Final R values (and whether quoted for all or observed data):0.1224

#### **5.**Mass spectra data of intermediates

#### Mass spectra were taken in infusion method.

#### **Parameters** :

Source - capillary (kv): 3.06, Sampling cone: 85, source offset: 80 Temp (°C) - Source: 120, Desolvation: 300 Gas flows – Cone gas (L/h) 50, Desolvation gas (L/h) 600, Lock spray capillary (kv) 2.50

In order to ascetain the reaction pathway we carried out mass kinetics, for that we withdraw a little aliquot from the reaction mixture and performed mass analysis to find out any probable intermediate(s). To our delight for both of the reactions we got peaks corresponding to the probable intermediates which are shown below and the corresponding time is also shown.

![](_page_48_Figure_5.jpeg)

![](_page_49_Figure_0.jpeg)

![](_page_50_Figure_0.jpeg)

![](_page_51_Figure_0.jpeg)

![](_page_52_Figure_0.jpeg)

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![](_page_53_Figure_0.jpeg)