

## Supporting Information

# **An AIE-active tetra-aryl imidazole-derived chemodosimeter for turn-on recognition of hydrazine and its bioimaging in living cells**

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## 1. General methods

Aniline, terephthalaldehyde, acetic acid, p-Anisil, ammonium acetate, malonitrile, ethanol, deionized water,  $\text{N}_2\text{H}_4$ , Cys,  $\text{ClO}^-$  and other analytical reagents were all purchased from Innochem Technology Co., Ltd.

$^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were measured with a Bruker AVB-600 spectrometer and Bruker AVB-600 spectrometer, TMS was used as the internal reference, and LC-MS spectra were measured with a Waters2695 spectrometer. Fluorescence spectra were recorded by an F7000 spectrofluorimeter from Hitachi PharmaSpec. Fluorescence imaging of  $\text{N}_2\text{H}_4$  in HeLa cells was recorded on a Nikon A1R<sup>+</sup> (Japan) laser scanning confocal microscope.

**SWJT-31** was dissolved in DMSO to prepare 1.0 mM stock solution,  $\text{N}_2\text{H}_4$  and other other analytical such as: triethylamine, aniline, n-butylamine,  $\text{NH}_4^+$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Zn}^{2+}$ , Cys, Hcy, GSH,  $\text{ClO}^-$ ,  $\text{HCO}_3^-$ ,  $\text{NO}_2^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Br}^-$ ,  $\text{S}^{2-}$ , urea,  $\text{H}_2\text{O}_2$ ,  $\text{ONOO}^-$ , Ser, Trp, Leu, Glu were dissolved in distilled water to prepare 10.0 mM stock solution. Ultraviolet-visible absorption and fluorescence spectra were recorded after 20.0  $\mu\text{L}$  of **SWJT-31** stock solution and 20.0  $\mu\text{L}$  of  $\text{N}_2\text{H}_4$  stock solution were diluted to 2.0 mL with DMSO/HEPES (6: 4, v/v, pH = 7.4) buffer solution and incubated at 37°C for 30 min. The selectivity for the  $\text{N}_2\text{H}_4$  is investigated by adding other analytical reagents solutions instead of  $\text{N}_2\text{H}_4$  in a similar way. For fluorescence spectra of AIE effect test, the excitation was set at 420 nm, and the excitation and emission gaps were 10/5 nm. However, fluorescence spectra for detecting hydrazine hydrate, the excitation was set at 340 nm, and the excitation and emission gaps were 5/2.5 nm.

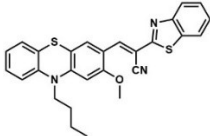
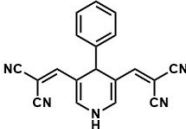
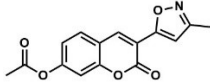
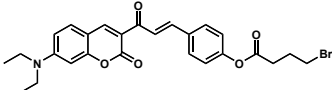
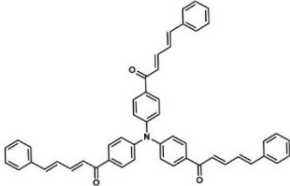
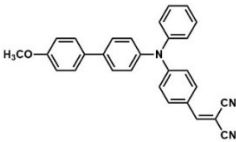
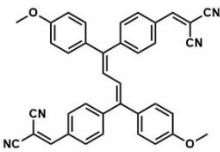
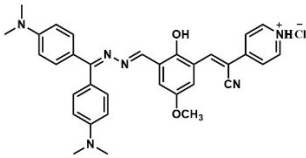
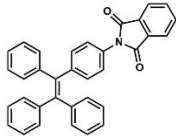
HeLa cells (human cervical cancer cells) were plated in 96-well plates and incubated in a 5% carbon dioxide, 37°C incubators for 24h. After removing the medium, add probes of different concentrations (0, 5.0  $\mu\text{M}$ , 10.0  $\mu\text{M}$ , 15.0  $\mu\text{M}$ , 20.0  $\mu\text{M}$ ) to the fresh medium for 24 hours in an incubator. Then, 10.0  $\mu\text{L}$  of MTT (5 mg/mL in PBS, pH 7.4)

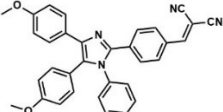
was added to each well, and after the cells were incubated for another 2 hours, the culture supernatant was removed, and the resulting formazan crystals were dissolved in 150  $\mu$ L of dimethyl sulfoxide. Next, the plates were incubated for an additional 10 minutes at 37°C on a shaker at 60-70 rpm. Finally, the absorbance (A) of each well at 570 nm was measured using a microplate reader. (Cell relative viability =  $A_{\text{sample}} / A_{\text{control}} * 100\%$ ).

HeLa cells were plated in glass bottom dishes with fresh medium and incubated in a 37°C incubator for 24 hours. After washing with PBS buffer, fresh medium and 10.0  $\mu$ M probe were added and incubated for 30 min. After washing three times with PBS buffer, the medium and 40.0  $\mu$ M  $N_2H_4$  were added for treatment for 30 min. Finally, after washing three times with PBS buffer, the dishes were placed on a confocal laser scanning microscope for cell imaging.

## 2. Summary of AIE-based fluorescent probes for hydrazine.

**Table S1. The comparison of the reported AIE-based hydrazine fluorescent probes.**

probes	Solvent system (v/v)	LOD	Response time	reference
	DMSO/PBS (9/1)	0.11 nM	-	30
	DMSO/PBS (2/3)	11 nM	60 min	31
	CH <sub>3</sub> CN/H <sub>2</sub> O (3/7)	2.90 ppb	45 min	32
	DMSO/PBS (6/4)	0.04 μM	25 min	33
	THF/H <sub>2</sub> O (9:1)	0.119 nM	-	34
	DMSO/tris-HCl (3/1)	0.196 μM	20 min	35
	DMSO/PBS (9/1)	2.88 ppb	3 min	36
	DMSO/H <sub>2</sub> O (1/99)	1.2 ppb	2 min	37
	CH <sub>3</sub> CN/HEPES (3/97)	6.4 ppb	-	38

	DMSO/HEPES (6/4)	33.8 nM	30 min	This work
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### 3. Synthesis of the M-CHO and SWJT-31.

**Q-CHO** was synthesized according to previous work<sup>36</sup>. Aniline (93 mg, 1.0 mmol) and terephthalaldehyde (134 mg, 1.0 mmol) were dissolved in acetic acid (15 mL) and reacted at room temperature. After 1 h, p-Anisil (270 mg, 1.0 mmol) and ammonium acetate (539 mg, 7.0 mmol) were added to the reaction flask to react at 120 °C for 10 h to obtain **Q-CHO**.

The synthetic procedures of **SWJT-31** were shown in Scheme 2. **Q-CHO** (92 mg, 0.2 mmol) and malononitrile (12.6  $\mu$ L, 0.2 mmol) were dissolved in anhydrous ethanol (2.0 mL), followed by piperidine (8.0  $\mu$ L). Next, the mixture was stirred at room temperature overnight, and the orange yellow precipitate was precipitated and then filtered. After precipitation and purification, the orange yellow powder **SWJT-31** (71 mg) was obtained with a yield of 69.8%. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.76 (d, *J* = 8.6 Hz, 2H), 7.66 (s, 1H), 7.59 (d, *J* = 8.6 Hz, 2H), 7.53 (d, *J* = 8.4 Hz, 2H), 7.39–7.33 (m, 3H), 7.10–7.03 (m, 4H), 6.83–6.76 (m, 4H), 3.79(s, 3H), 3.77 (s, 3H) ppm. <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 159.5, 158.8, 158.7, 144.1, 139.2, 136.9, 136.4, 132.3(2C), 131.8, 130.7(2C), 129.9, 129.5(2C), 129.0, 128.9(2C), 128.4(2C), 128.3(2C), 126.7, 122.1, 114.0(2C), 113.9, 113.7(2C), 112.7, 81.9, 55.2, 55.1 ppm. LC-MS: *m/z* 475.2 [M + H]<sup>+</sup>.

#### 4. $^1\text{H}$ , $^{13}\text{C}$ NMR spectra and LC-MS of SWJT-31.

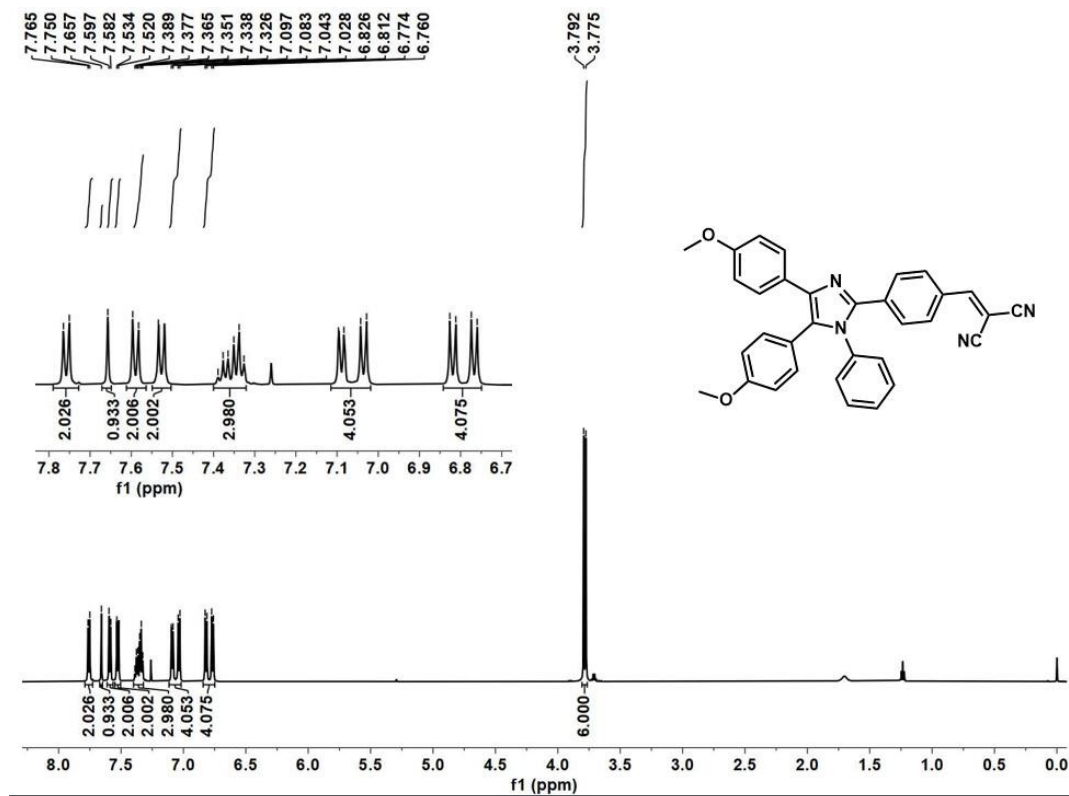


Figure S1.  $^1\text{H}$  NMR spectrum of SWJT-31 in  $\text{CDCl}_3$ .

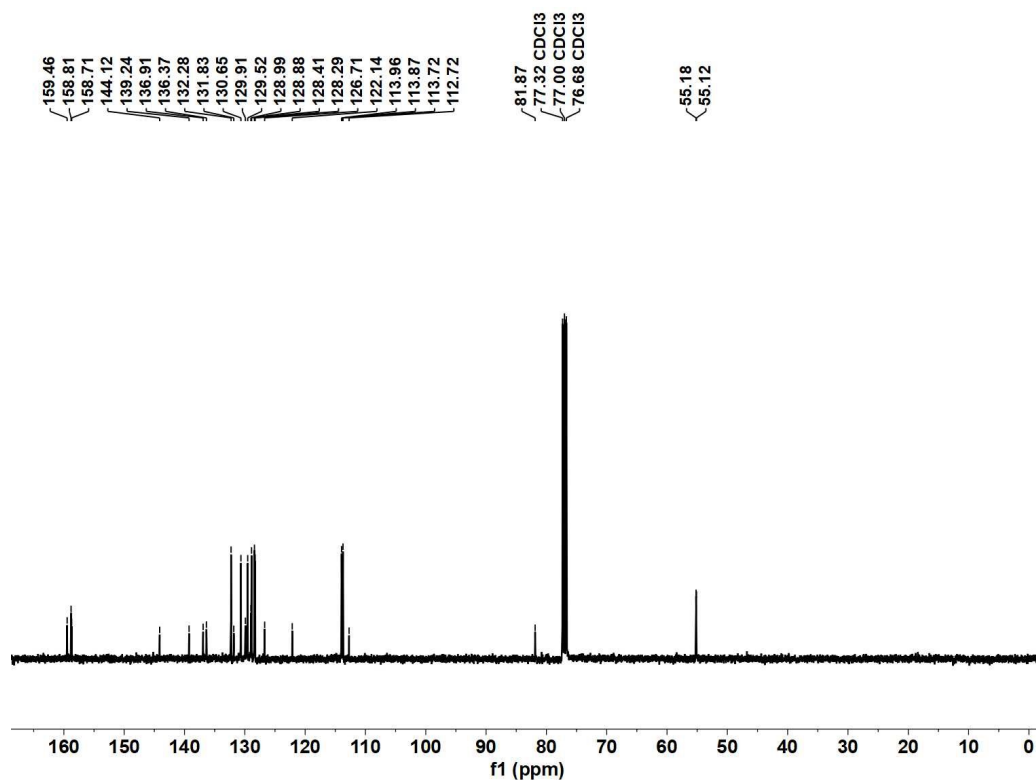
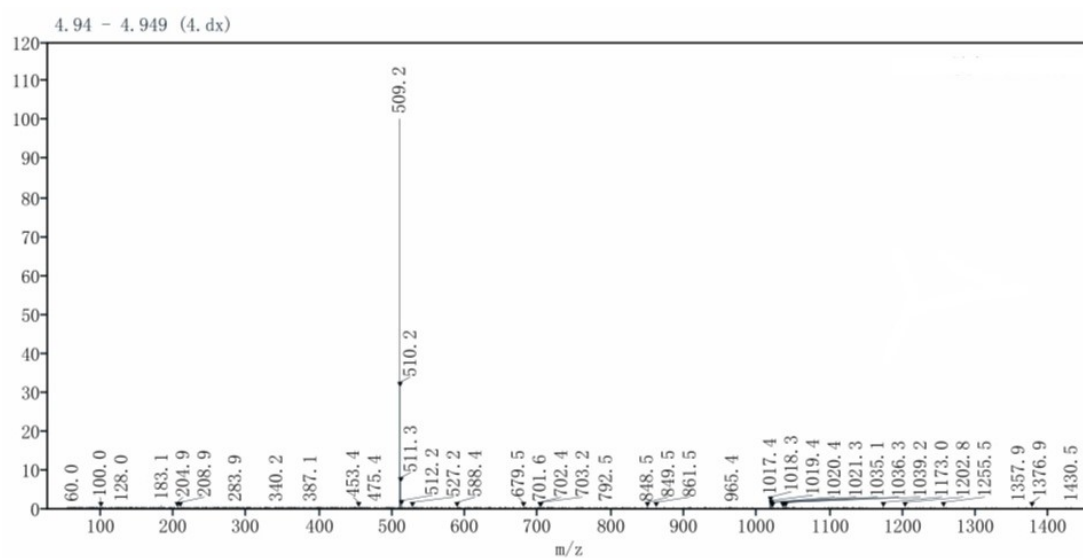


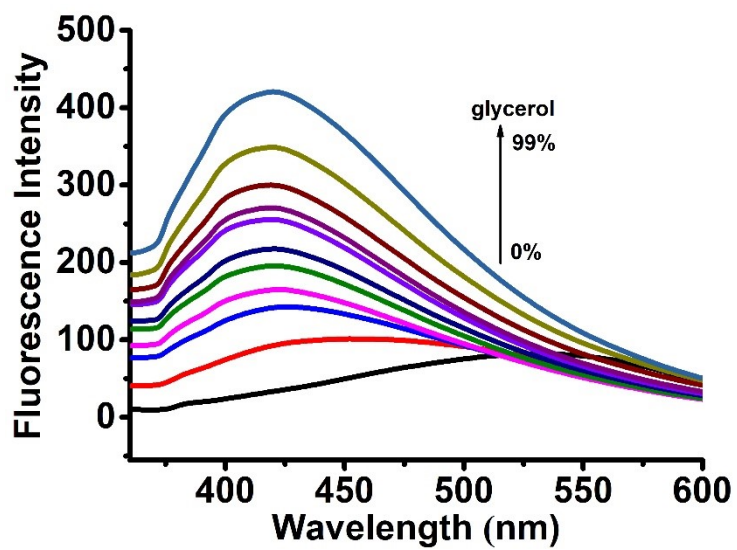
Figure S2.  $^{13}\text{C}$  NMR spectrum of SWJT-31 in  $\text{CDCl}_3$ .



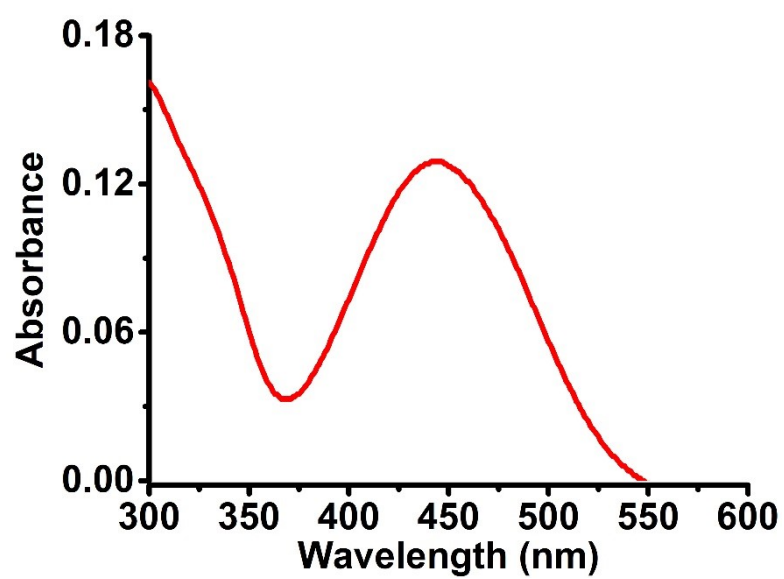
**Figure S3.** LC-MS spectrum of SWJT-31.



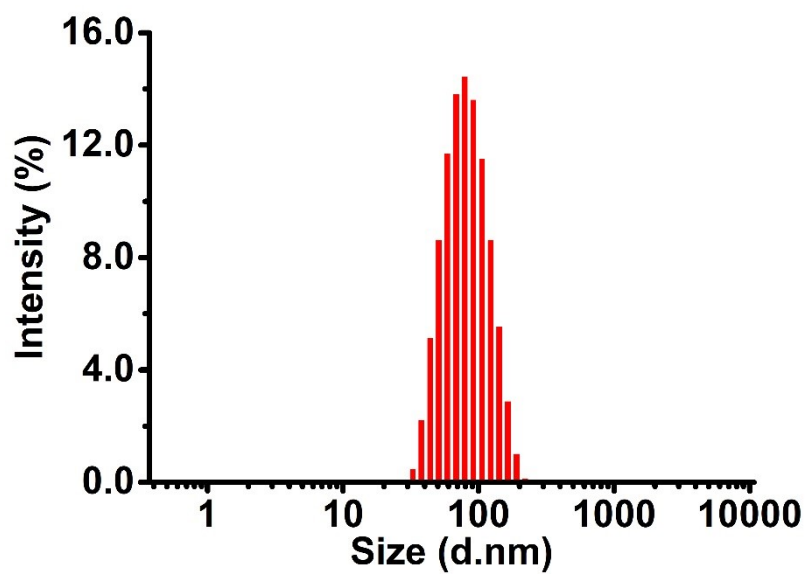
## 5. Basic properties of SWJT-31.



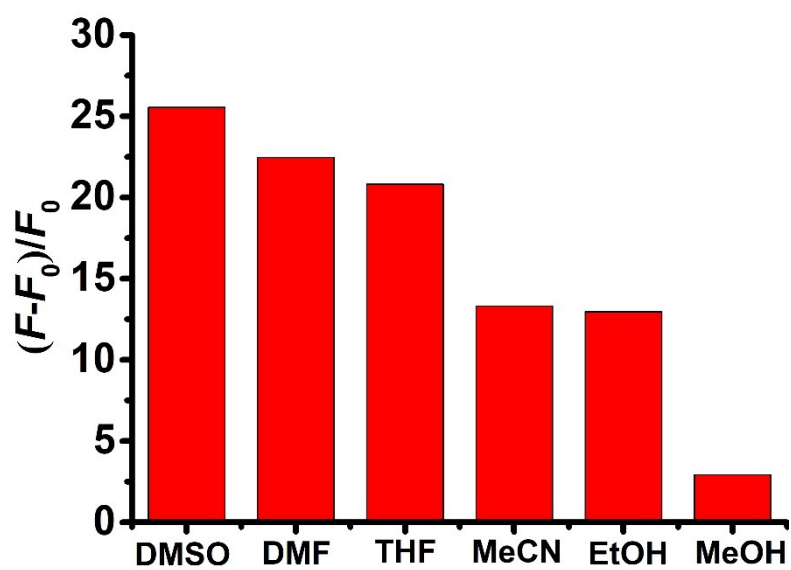
**Figure S4.** Fluorescence spectra of SWJT-31 (10.0 μM) in the presence of different viscosity in glycerol/ DMSO fraction.



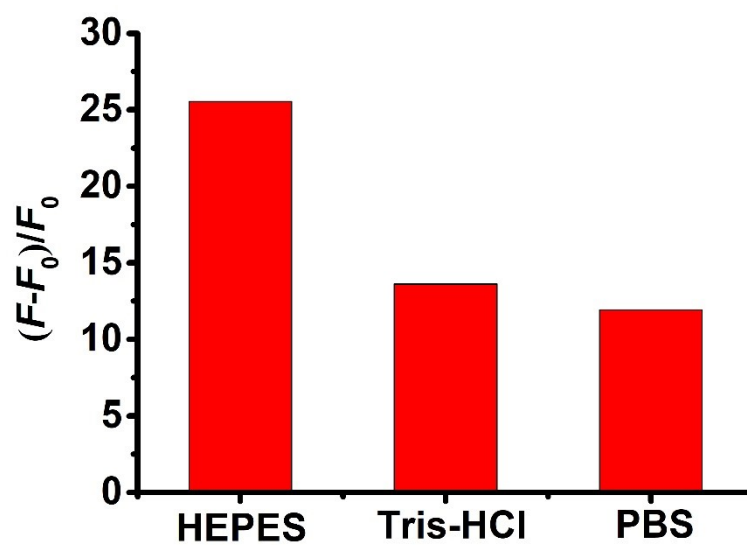
**Figure S5.** In DMSO/HEPES (v/v=1:99, pH = 7.4) buffer solution, absorption spectra of SWJT-31 (10.0  $\mu$ M).



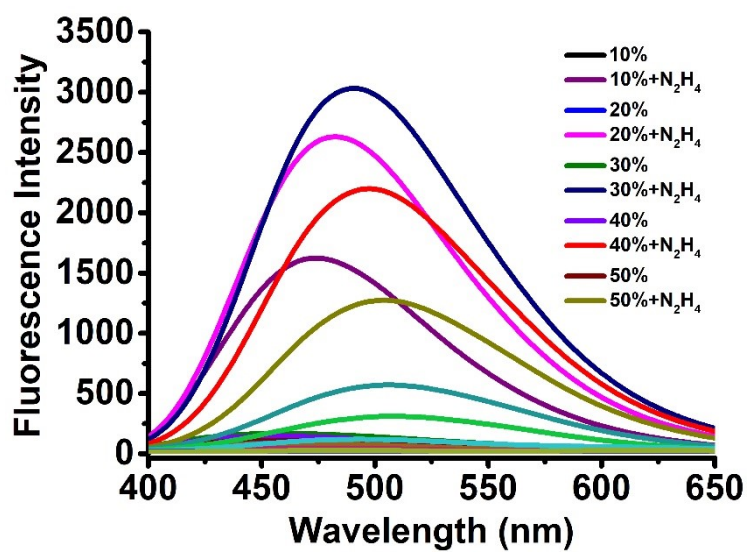
**Figure S6.** Size distribution of SWJT-31 in the aggregated state in 99% HEPES buffer solution (pH = 7.4).



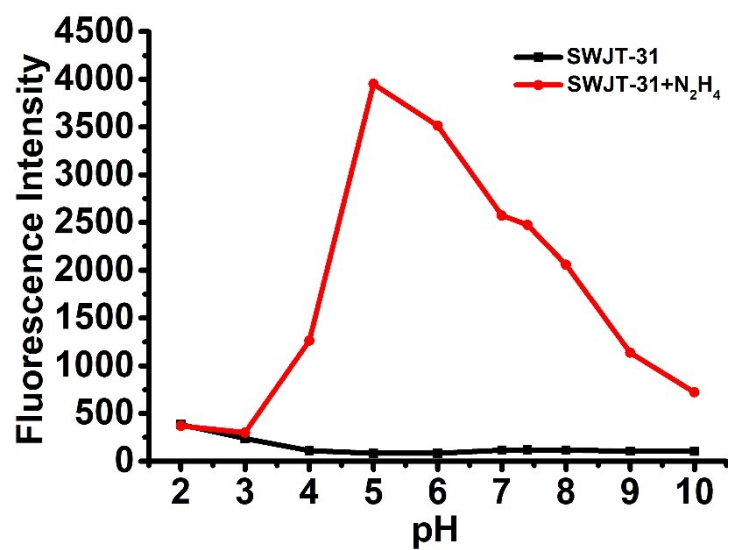
**Figure S7.** The fluorescence enhancement times of SWJT-31 (10.0  $\mu\text{M}$ ) and SWJT-31+  $\text{N}_2\text{H}_4$  (100.0  $\mu\text{M}$ ) in different organic phases at 497nm ( $\lambda_{\text{ex}} = 340$  nm).



**Figure S8.** The fluorescence enhancement times of SWJT-31 (10.0  $\mu\text{M}$ ) and SWJT-31 +  $\text{N}_2\text{H}_4$  (100.0  $\mu\text{M}$ ) in different organic phases at 497nm ( $\lambda_{\text{ex}} = 340 \text{ nm}$ ).



**Figure S9.** Fluorescence spectra versus the content of the DMSO/HEPES mixture of SWJT-31 in the presence of N<sub>2</sub>H<sub>4</sub>(100.0 μM) ( $\lambda_{\text{ex}} = 340 \text{ nm}$ ).



**Figure S10.** Effects of pH on the reaction of SWJT-31 (10.0  $\mu\text{M}$ ) with  $\text{N}_2\text{H}_4$  (100.0  $\mu\text{M}$ ) ( $\lambda_{\text{ex}} = 340 \text{ nm}$ ).

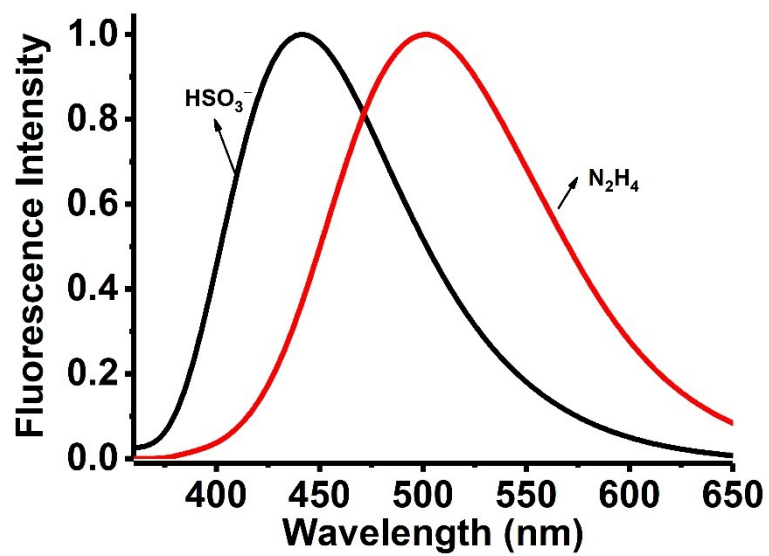
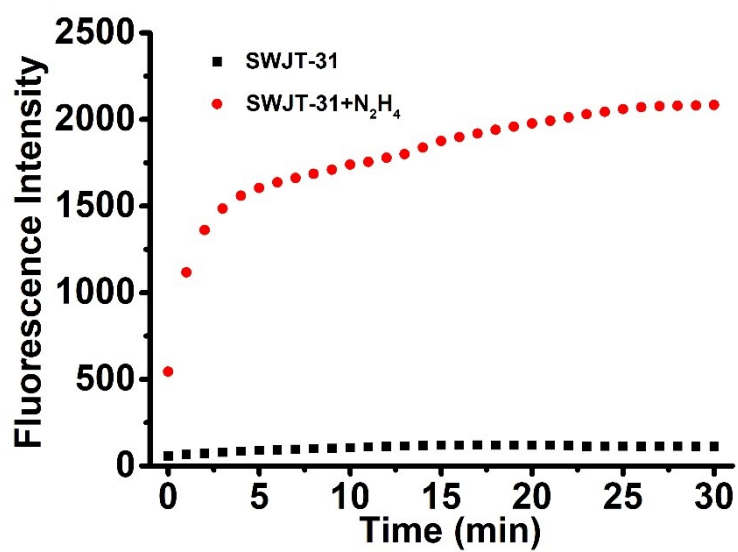
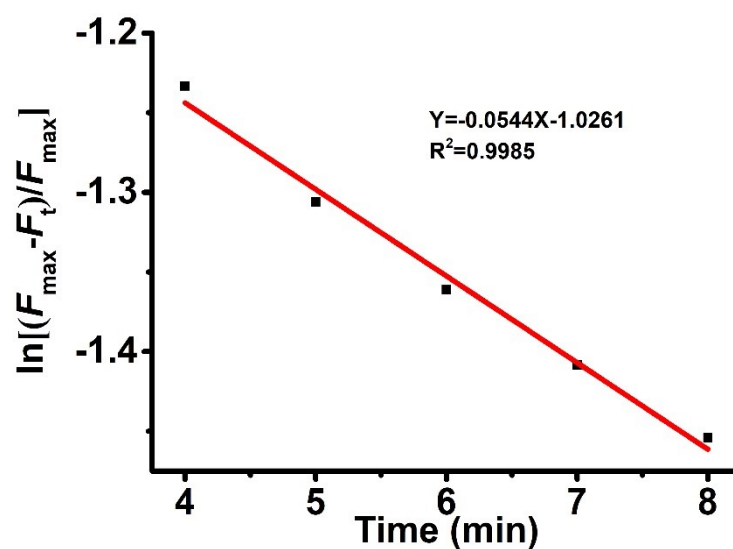


Figure S11. Fluorescence spectra of SWJT-31+  $\text{N}_2\text{H}_4$  and SWJT-31+  $\text{SO}_3^{2-}$ .





**Figure S12.** In the absence or presence of N<sub>2</sub>H<sub>4</sub> (10.0 μM), time-dependent fluorescence intensity of SWJT-31 (10.0 μM) in DMSO: HEPES (6:4, v/v, pH = 7.4) buffer solution at 497 nm ( $\lambda_{\text{ex}} = 340$  nm).



**Figure S13.** Pseudo first-order kinetic plots of SWJT-31 (10.0  $\mu\text{M}$ ) with the addition of  $\text{N}_2\text{H}_4$  in HEPES buffer solution (60% DMSO, pH = 7.4) ( $\lambda_{\text{ex}} = 340 \text{ nm}$ ).

The result of the analysis as follows:

$$\ln [(F_{\max} - F_t) / (F_{\max})] = -k_{\text{obs}}t$$

$$t_{1/2} = \ln 2 / k_{\text{obs}}$$

Where  $F_{\max}$  and  $F_t$  are the fluorescent intensity at maximum emission wavelength, and time  $t$ .  $k_{\text{obs}}$  is the pseudo-first-order rate constant.

$$k_{\text{obs}} = 9.06 \times 10^{-4} \text{ s}^{-1}$$

$$t_{1/2} = 12.74 \text{ min}$$

**6. Application of SWJT-31 as test strips.**

**Table S2. Determination of N<sub>2</sub>H<sub>4</sub> in water samples.**

<b>Sample</b>	<b>spiked(μM)</b>	<b>Found mean ± SD<sup>a</sup>(μM)</b>	<b>Recovery</b>
<b>Tape water</b>	0	Not Detected	
	2	1.854±0.027	92.69%
	6	5.669±0.115	94.49%
	10	10.736±0.029	107.4%
<b>Spring water</b>	0	Not Detected	
	2	1.916±0.012	95.81%
	6	6.259±0.027	104.3%
	10	10.17±0.239	101.7%
<b>Jing Lake</b>	0	Not Detected	
	2	1.887±0.043	94.37%
	6	5.214±0.106	86.90%
	10	9.724±0.106	97.24%

7.  $^1\text{H}$ ,  $^{13}\text{C}$  NMR spectra and LC-MS of the product SWJT-N2.

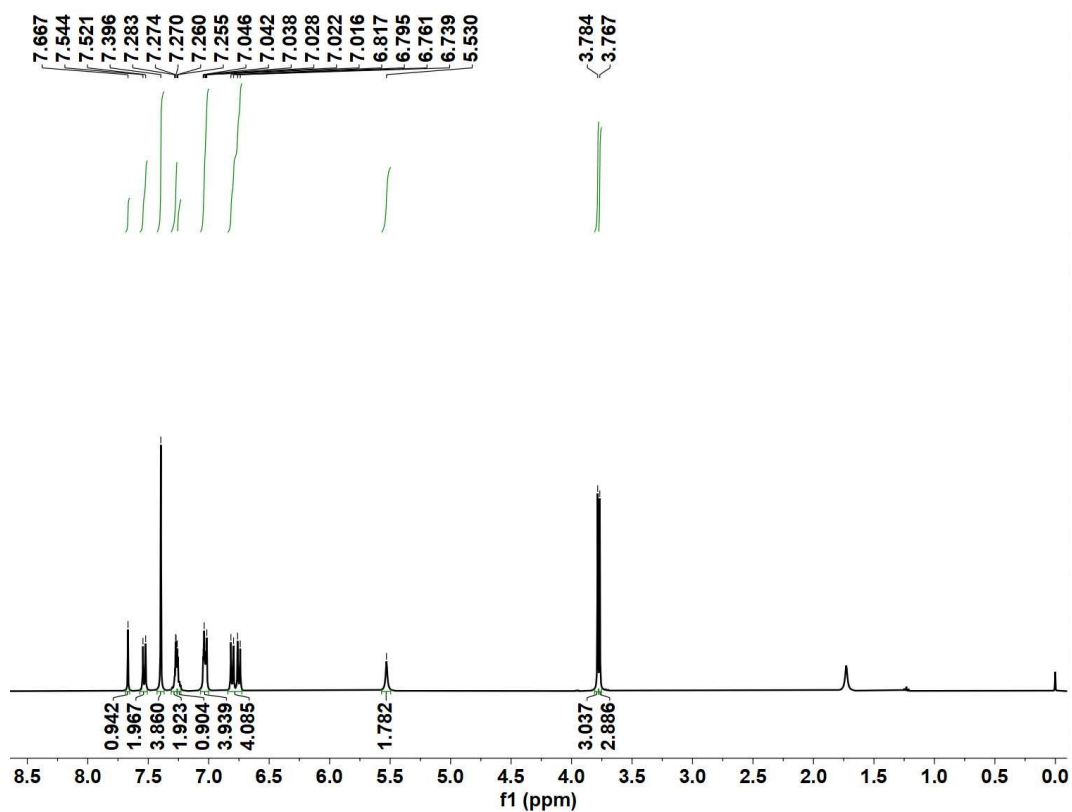


Figure S14.  $^1\text{H}$  NMR spectrum of the product SWJT-N2 in  $\text{CDCl}_3$ .

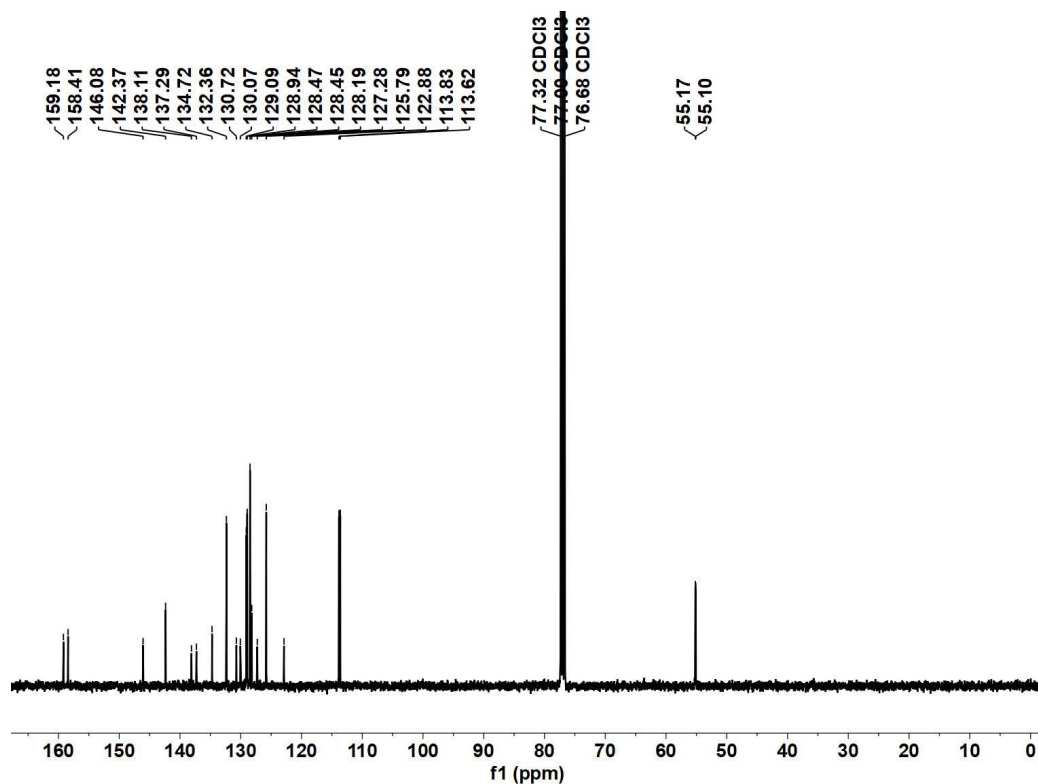
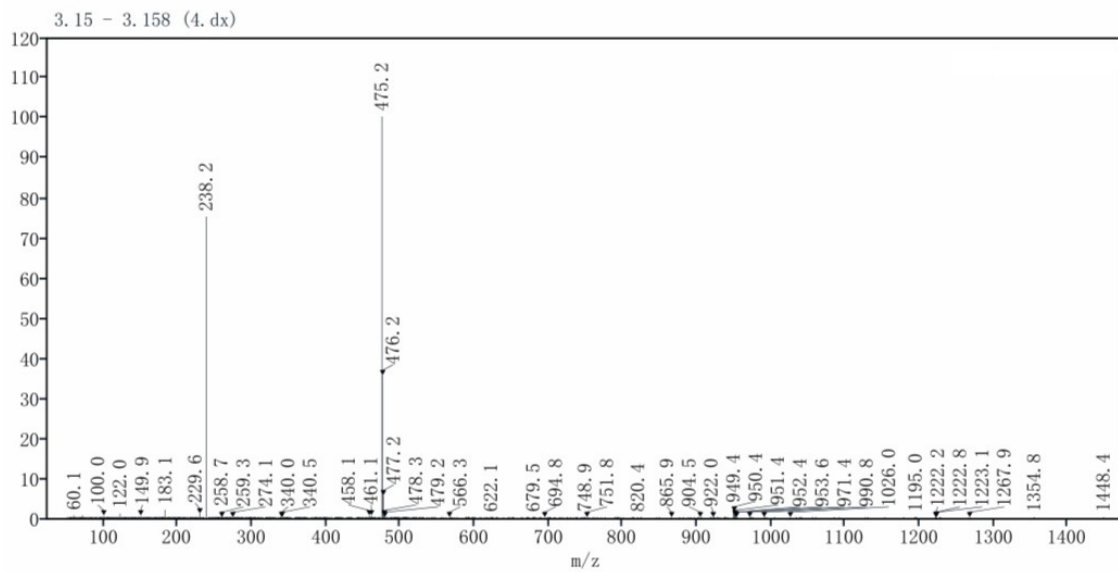
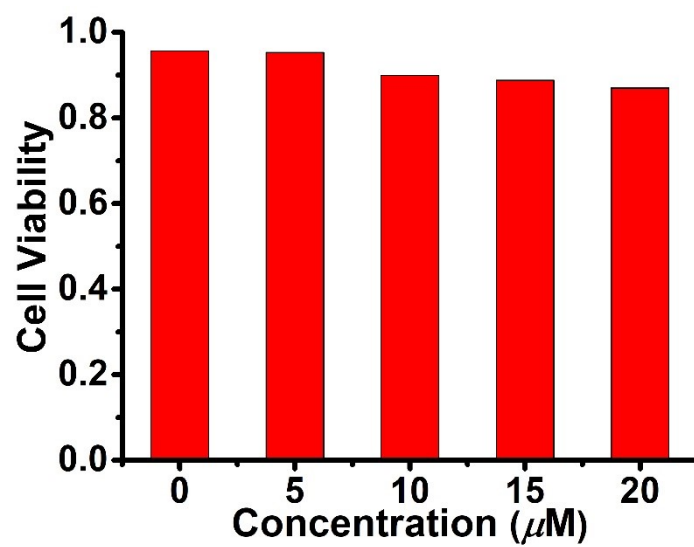


Figure S15.  $^{13}\text{C}$  NMR spectrum of the product SWJT-N2 in  $\text{CDCl}_3$ .



**Figure S16.** LC-MS spectrum of the product SWJT-N2.

## 8. Cytotoxicity of SWJT-31 in living HeLa cells.



**Figure S17.** The viability of HeLa cells was determined by MTT assay after incubation with different concentrations of SWJT-31 for 24 h.