Supporting Information

Anthracene carboxyimide-based selenide as fluorescent probe for

ultrasensitive detection of hypochlorous acid

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1. Preparation of reactive oxygen species (ROS) and reactive nitrogen species (RNS)

All the stocking solutions of ROS/RNS were prepared based on the reported literature [1]. The stock hydrogen peroxide (H₂O₂), sodium hypochloride (NaOCl) and tert-Butyl hydroperoxide (TBHP) slolutions were purchased from Sigma-Aldrich. Superoxide anion (O₂⁻⁾ solution was prepared by fully dispersing the potassium dioxide in anhydrous DMSO via ultrasonic treatment. Hydroxyl radicals (\cdot OH) and tert-butoxy radical (tBuO \cdot) were prepared by Fenton reaction, the molar ratio of FeSO₄:H₂O₂ and FeSO₄:TBHP was 1:10. Peroxyl radicals (ROO \cdot) was generated from 2,2'-azobis(2-amidinopropane)dihydrochloride. Peroxynitrite (ONOO⁻) solution was prepared by 3-morpholinosydnonimine hydrochloride (SIN-1). NO \cdot were diluted from the commercially available 2,2'-azobis (2-amindinopropane) dihydrochloride and sodium nitroferricyanide(III) dihydrate (SNP) to ultrapure water.

2. Determination of the detection limit

The detection limit was calculated based on the method reported in the previous literature by the equation as follows:

Detection limit = $3\sigma/k$ [2]

Where σ is the standard deviation of blank measurement, k is the slope of the equation between fluorescence intensity and the concentrations of NaOCl. We measured the fluorescence intensity of the probe **AC-Se** without NaOCl for six times to obtain the standard deviation, and the slope k was obtained according to the linear equation of the fluorescence intensity of F₅₄₂ with the increasing concentration of NaOCl.



3. HR-MS data of AC-Se and the reaction mixture of AC-Se with ClO-

Fig. S1 HR-MS spectra of AC-Se and the reaction mixture of AC-Se with ClO-.

4. Photostability of probe AC-Se and the oxidation product toward ClO-



Fig. S2 Time-dependent fluorescence intensity changes of AC-Se (10 μ M) under the irradiation by a 450W lamp, $\lambda_{ex} = 470$ nm, slits = 2/2 nm.

$\begin{array}{c} 1.0 \\ 0.8 \\ 0.8 \\ 0.6 \\ 0.4 \\ 0.2 \\ 0.0 \\ \hline 0.0 \\ \hline \hline Control \end{array}$ $\begin{array}{c} 1 \\ 5 \\ 10 \\ \hline 1 \\ 5 \\ 10 \\ \hline 15 \\ 20 \\ 30 \\ \hline Ac-Se (\mu M) \end{array}$

5. Cytotoxicity assays

Fig. S3 Cell viability of HeLa cells treated with different concentrations of AC-Se (0, 5, 10, 15, 20, 30μ M) for 24 h.

6. Real-time imaging



Fig. S4 (a) Time course of fluorescence intensity (0-30 minutes) of HeLa cells after addition of AC-Se (5 μ M). (b) Time course of fluorescence intensity (0-30 minutes) of local HeLa cells from Fig. S4a. (c)Average fluorescence intensity from image S4b respectively. $\lambda_{exc} = 488$ nm, $\lambda_{em} = 540-560$ nm.

7. NMR spectra



Fig. S5 ¹H NMR (400 MHz) spectrum of AC-Br in CDCl_{3.}



Fig. S7 ¹H NMR (600 MHz) spectrum of AC-Se in DMSO-*d*₆.



Fig. S8 ¹³C NMR (151 MHz) spectrum of AC-Se in DMSO-*d*₆.

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