

Supporting Information for

**Concise Synthesis of Light/Temperature/pH/CO₂-Quadruple Responsive
Azobenzene Functionalized Homopolymer for Reversible Photopatterning**

Ke Wang^{a,*}, Meiyu Si^a, Xiaofang Liu^a, Zilong Wang^a, Guiyan Liu^a, Yongfei Zeng^{a,*}

^a Tianjin Key Laboratory of Structure and Performance for Functional Molecules, College of Chemistry, Tianjin Normal University, Tianjin, 300387, China. E-mail: kewang@tjnu.edu.cn, yfzeng@nankai.edu.cn.

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Characterization

NMR spectrum were performed on Bruker Avance III 400MHz Nuclear Magnetic Resonance Spectrometer. FTIR spectrum were measured by a Nicolet IR200 FT-IR spectrometer. An Agilent PL-GPC50 instrument was used to carry out molecular weight and polydispersity index (eluent: THF; standard sample: polystyrene; chromatographic column: PLMIXED-C, 5 μm, 7.5 × 50 mm; detector: differential refraction detector; flow rate: 1.0 mL/min). The UCST, LCST and UV/vis absorption spectrum were tested on a UH-4150 UV-vis spectrophotometer. TEM was observed by a FEI TECNAI G2 F20 (accelerating voltage: 200 kV). NanoBrook Omni laser light scattering spectrometer was used to the analysis of dynamic light scattering (DLS). Intensity of 365 nm UV lamp: 90 W/cm², intensity of 430 nm blue light lamp: 30 W/cm².

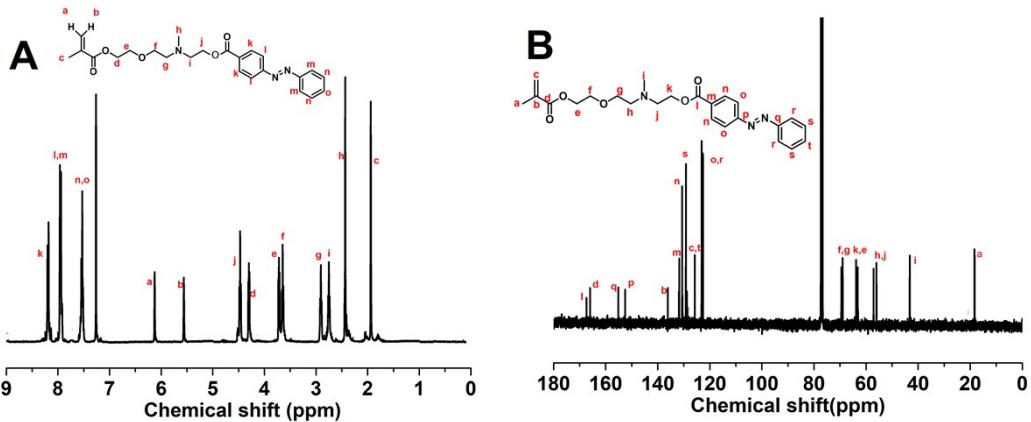


Figure S1. The ^1H NMR spectra (A) and ^{13}C NMR spectra (B) of Azo-N-EOMA in CDCl_3 .

(E)-2-((2-(methacryloyloxy) ethoxy)ethyl) (methyl)aminoethyl 4-(phenyldiazenyl)benzoate (Azo-N-EOMA) (78%). ^1H NMR (400 MHz, CDCl_3) δ 8.22 - 8.15 (d, $J = 8.4$ Hz, 2H), 7.95 (dd, $J = 8.2, 1.7$ Hz, 4H), 7.55 - 7.50 (m, 3H), 6.12(s, 1H), 5.58 - 5.53 (m, 1H), 4.46 (t, $J = 5.9$ Hz, 2H), 4.29 (t, $J = 5.8$ Hz, 2H), 3.74 - 3.69 (m, 2H), 3.64 (t, $J = 5.6$ Hz, 2H), 2.89 (t, $J = 5.7$ Hz, 2H), 2.75 (t, $J = 5.7$ Hz, 2H), 2.42 (s, 3H), 1.94 (s, 3H). ^{13}C NMR (101 MHz, CDCl_3) δ 167.4, 166.0, 155.1, 152.7, 136.1, 131.8, 130.6, 129.2, 125.2, 123.1, 122.8, 69.5, 69.1, 63.8, 63.2, 57.2, 56.0, 43.2, 18.2.

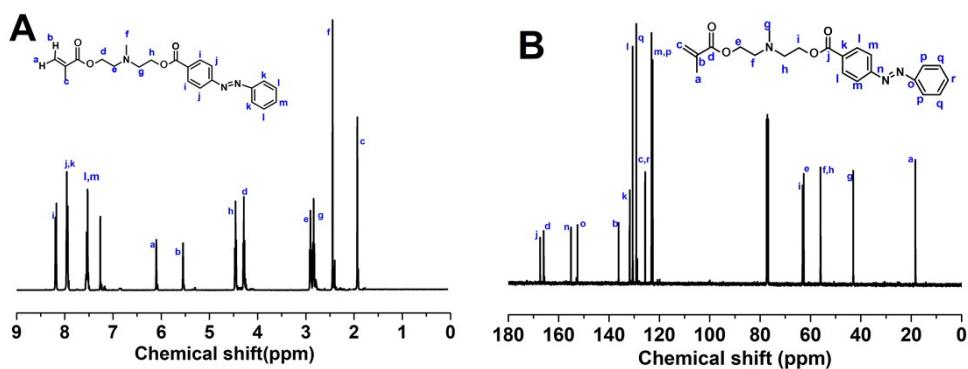


Figure S2. The ^1H NMR spectra (A) and ^{13}C NMR spectra (B) of Azo-N-MA in CDCl_3 .

(E)-2-((2-(methacryloyloxy) ethyl) (methyl)aminoethyl 4-(phenyldiazenyl)benzoate (Azo-N-MA) (83%). ^1H NMR (400 MHz, CDCl_3) δ 8.22 -

8.15 (d, $J = 8.6$ Hz, 2H) , 7.95(dd, $J = 8.2, 1.7$ Hz, 4H), 7.55 - 7.50 (m, 3H), 6.12(s, 1H), 5.58 - 5.53 (m, 1H), 4.46 (t, $J=5.8$ Hz, 2H), 4.29 (t, $J = 5.8$ Hz, 2H), 2.89 (t, $J=5.8$ Hz, 2H), 2.75 (t, $J= 5.8$ Hz, 2H), 2.42 (s, 3H), 1.94 (s, 3H). ^{13}C NMR (101 MHz, CDCl_3) δ 167.2, 165.9, 155.2, 152.7, 136.0, 132.3, 130.9, 129.2, 125.5, 123.0, 122.9, 62.9, 62.2, 56.1, 56.0, 43.1, 18.3.

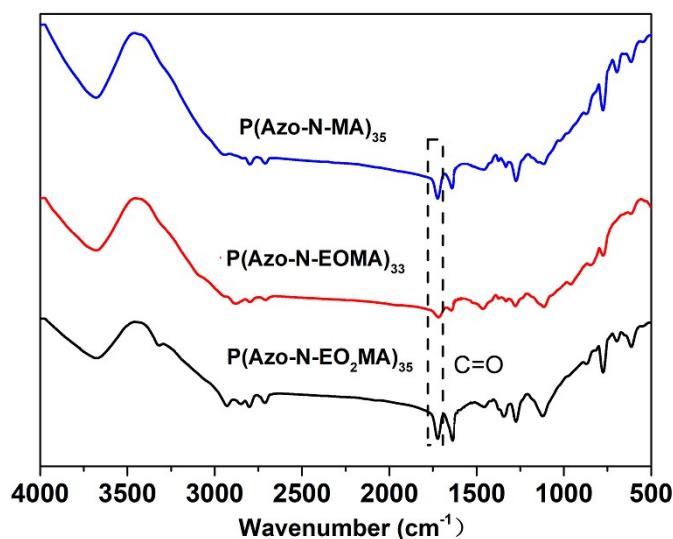
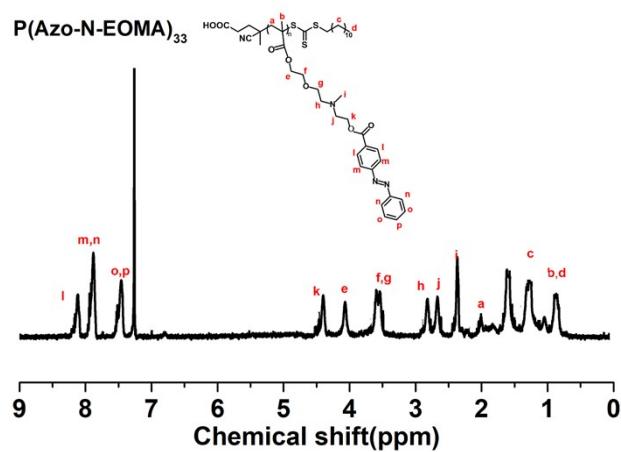


Figure S3. The FT-IR spectra of $\text{P}(\text{Azo-N-EO}_2\text{MA})_{35}$, $\text{P}(\text{Azo-N-EOMA})_{33}$ and $\text{P}(\text{Azo-N-MA})_{35}$.



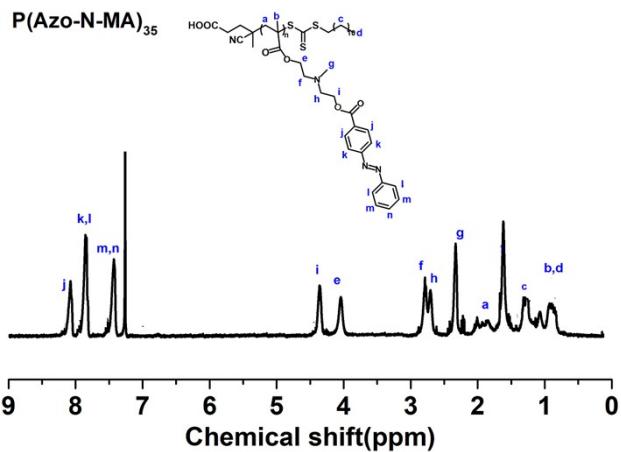


Figure S4. The ^1H NMR spectra of $\text{P}(\text{Azo-N-EO}_2\text{MA})_{33}$ and $\text{P}(\text{Azo-N-MA})_{35}$.

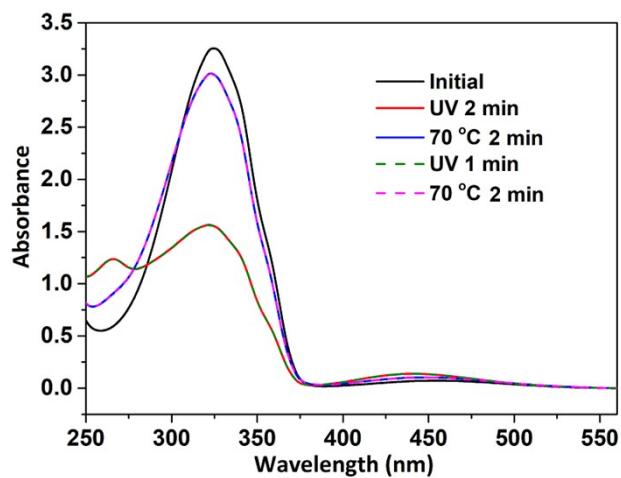


Figure S5 UV/vis absorption spectra of $\text{P}(\text{Azo-N-EO}_2\text{MA})_{35}$ in THF solution (0.1 mg/mL) under UV/heating.

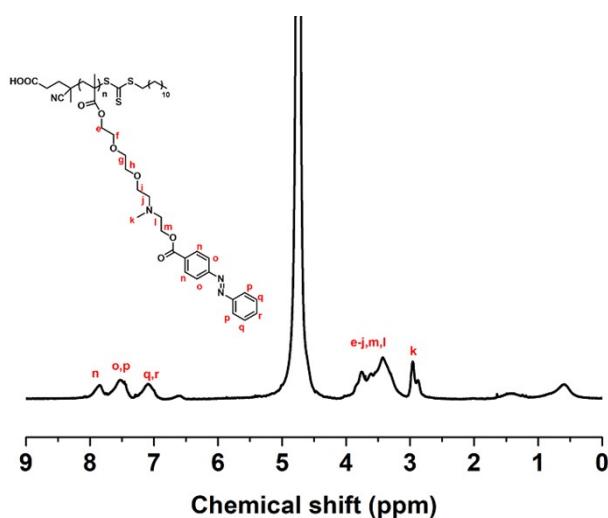


Figure S6 ^1H NMR spectra of $\text{P}(\text{Azo-N-EO}_2\text{MA})_{35}$ in D_2O ($\text{C} = 5 \text{ mg/mL}$, $\text{pH} = 4$).

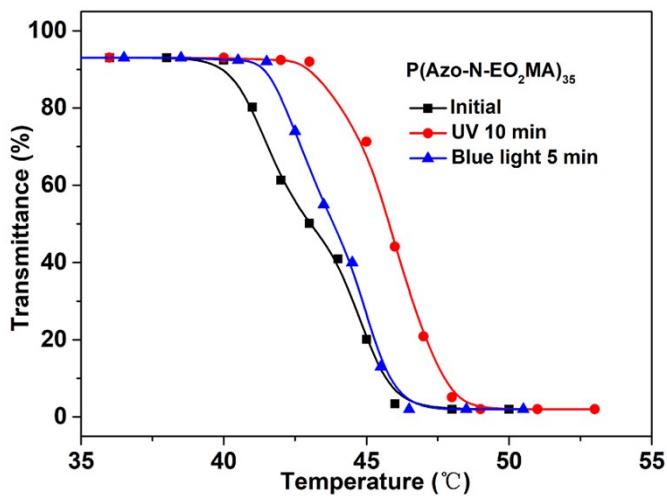


Figure S7 The effect of the irradiation of UV/blue light on the LCST of P(Azo-N-EO₂MA)₃₅.

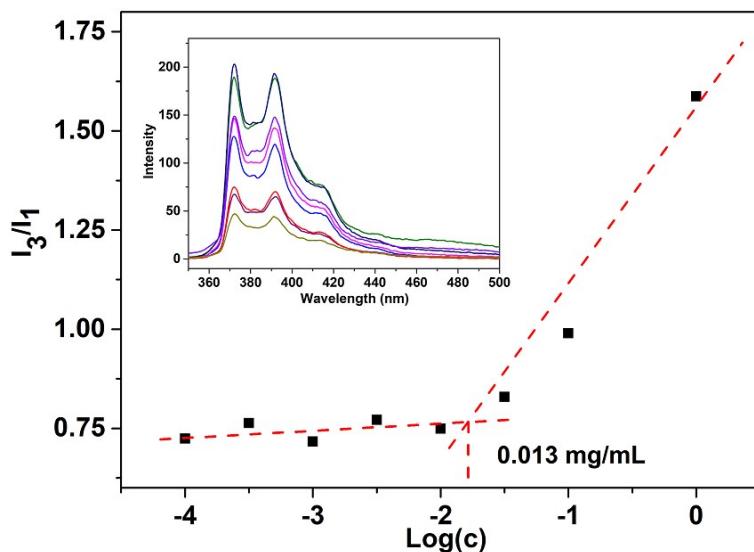


Figure S8 Fluorescence spectra of P(Azo-N-EO₂MA)₃₅ in water with pyrene and plots of I_3/I_1 vs concentration for P(Azo-N-EO₂MA)₃₅ in deionized water with different concentrations (from 0.0001 to 1 mg/mL).

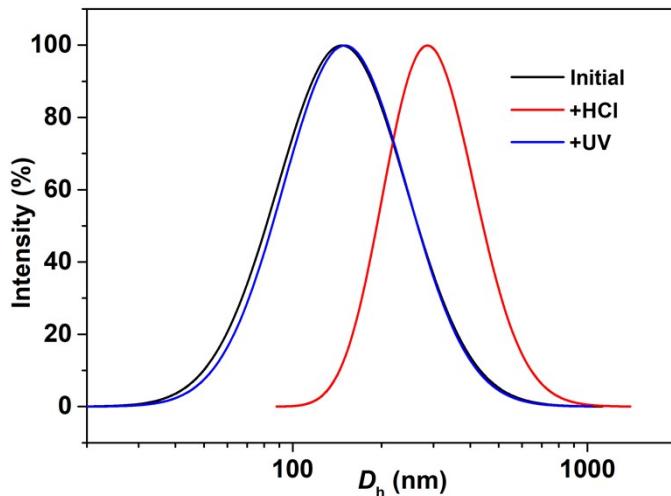


Figure S9 The D_h of micelles of $\text{P}(\text{Azo-N-EO}_2\text{MA})_{35}$, after treated with HCl and after irradiation of UV.

Table S1. Summary of LCST-type CP ($^{\circ}\text{C}$), UCST-type CP ($^{\circ}\text{C}$) and Critical pH of homopolymers.

	pH=3		pH=5 (LCST, $^{\circ}\text{C}$)			pH=7		BuOH/H ₂ O (UCST, $^{\circ}\text{C}$)			Critical pH		
	A ^a	U ^b	A	U	B ^c	A	U	A	U	B	A	U	B
P(Azo-N-EO ₂ MA) ₃₅	S ^d	S	43	46	43.5	I	I	34	25	32	4.65	5.0	4.73
P(Azo-N-EOMA) ₃₃	S	S	38	42	38.5	I	I	66	56	63	3.8	4.2	3.9
P(Azo-N-MA) ₃₅	I ^e	I	I	I	I	I	I	I	I	I	I	I	I

^aA is the initial state of the solution; ^bU is the solution under UV radiation for 10 min; ^cB is the solution under blue light radiation for 5 min; ^dS represents the sample being soluble; ^eI represents the sample being insoluble.