

## **Supporting Information**

### **Effect of Length and Functionality of the Linker ‘between the Main Chain and the Chiral Pendant’ on the Helical Nature of Chiral Poly(Ionic Liquids)**

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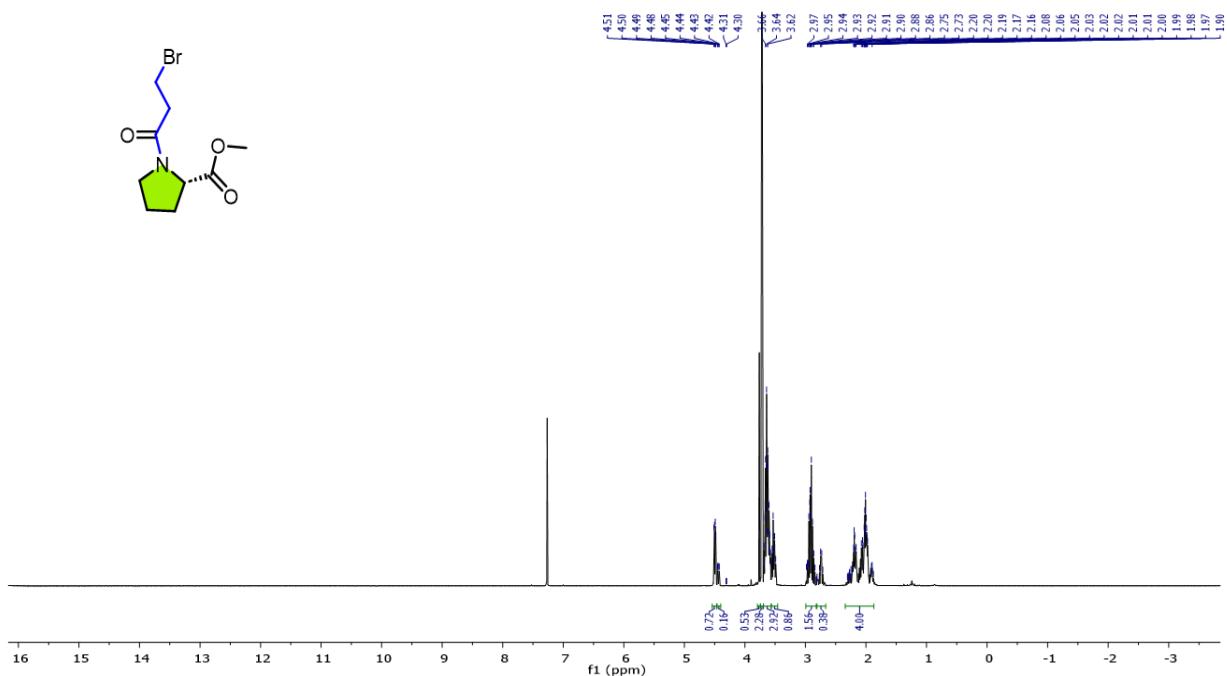
School of Basic Sciences,

Indian Institute of Technology,

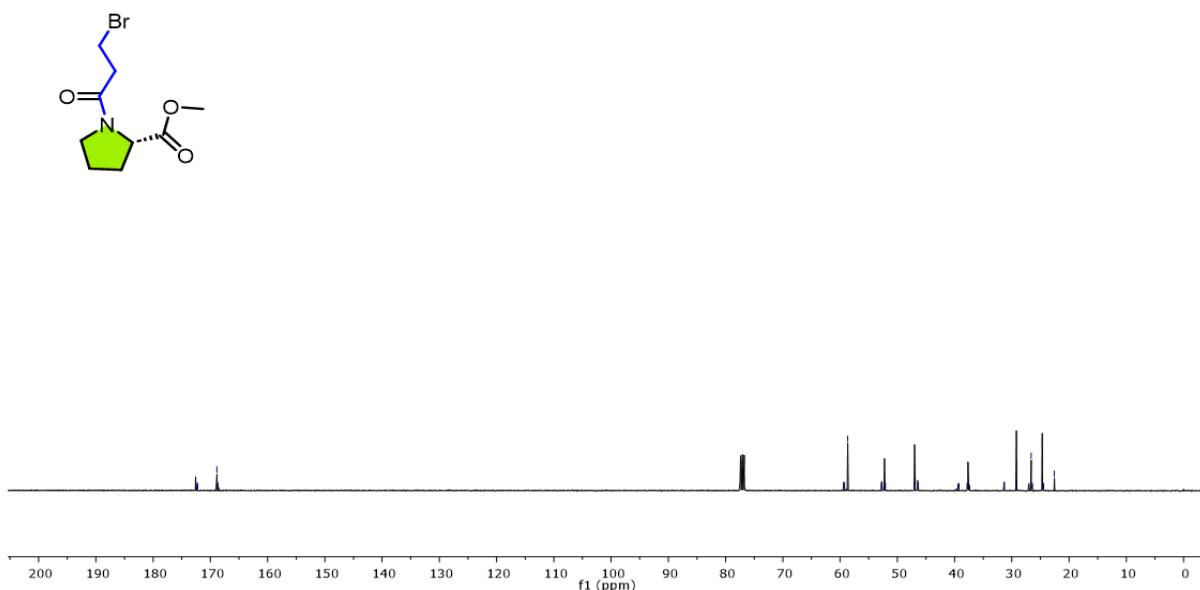
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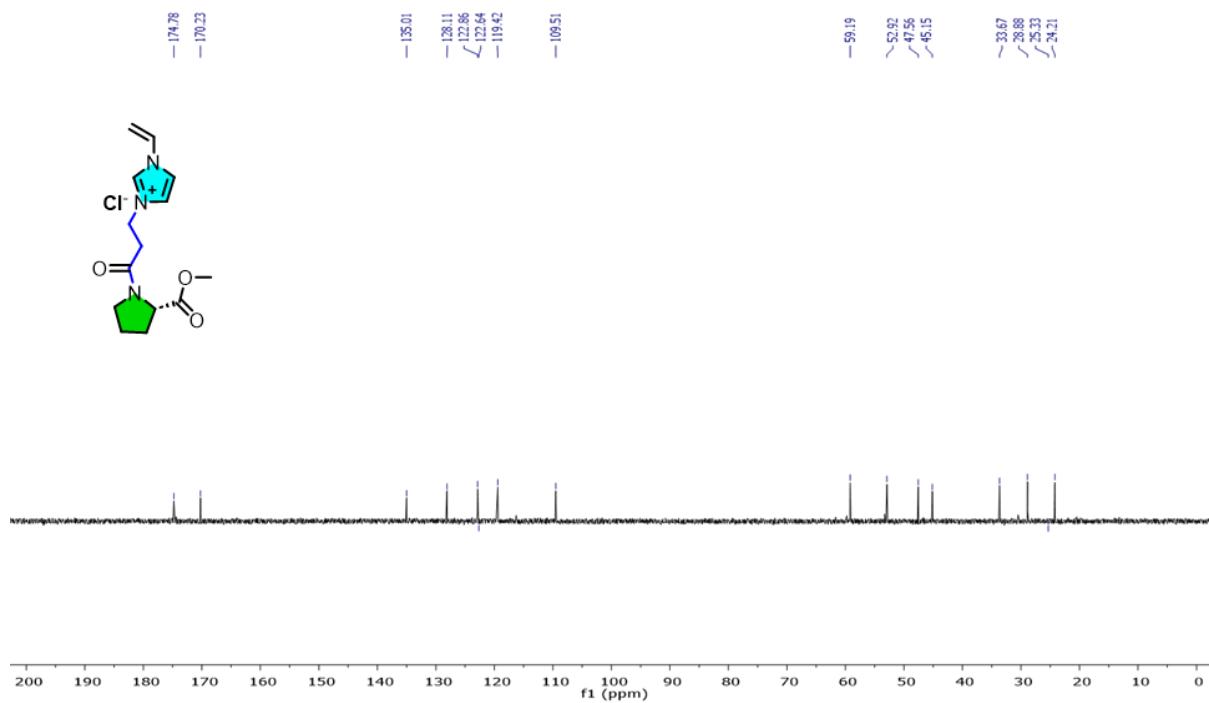
## 1. Supplementary figures



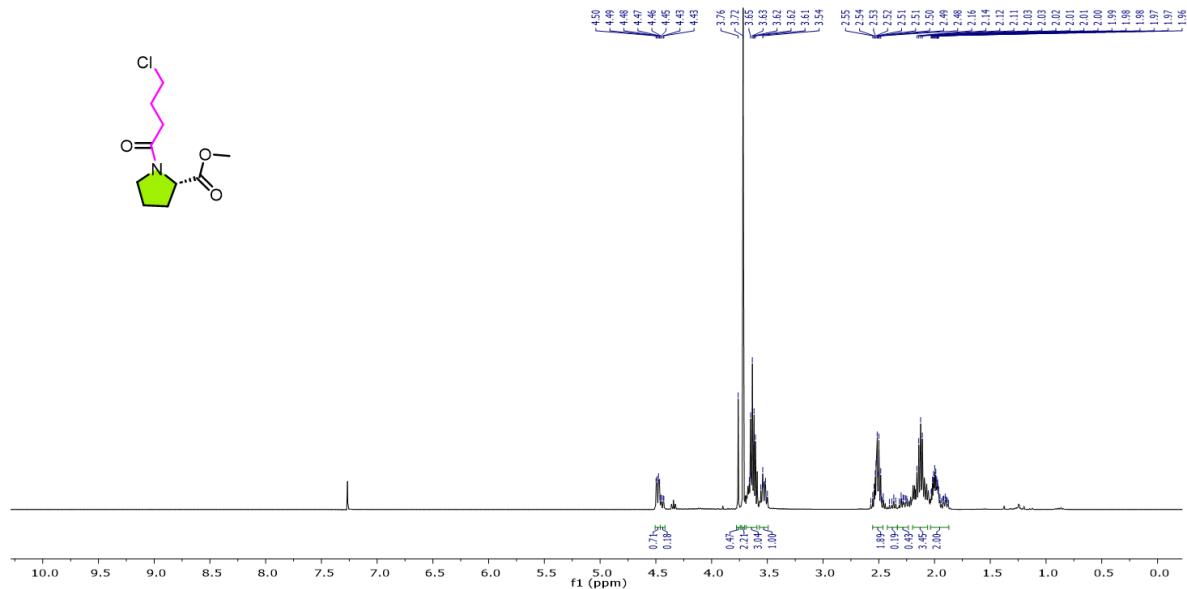
**Fig. S1**  $^1\text{H}$  NMR spectrum of (L-)-3.



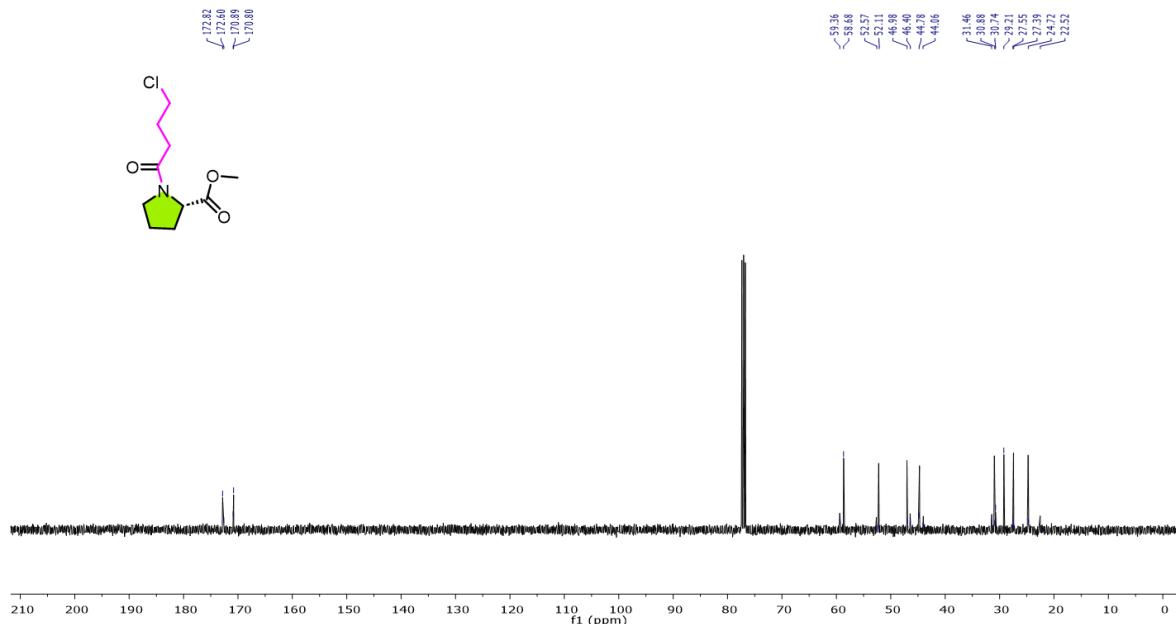
**Fig. S2**  $^{13}\text{C}$  NMR spectrum of (L-)-3.



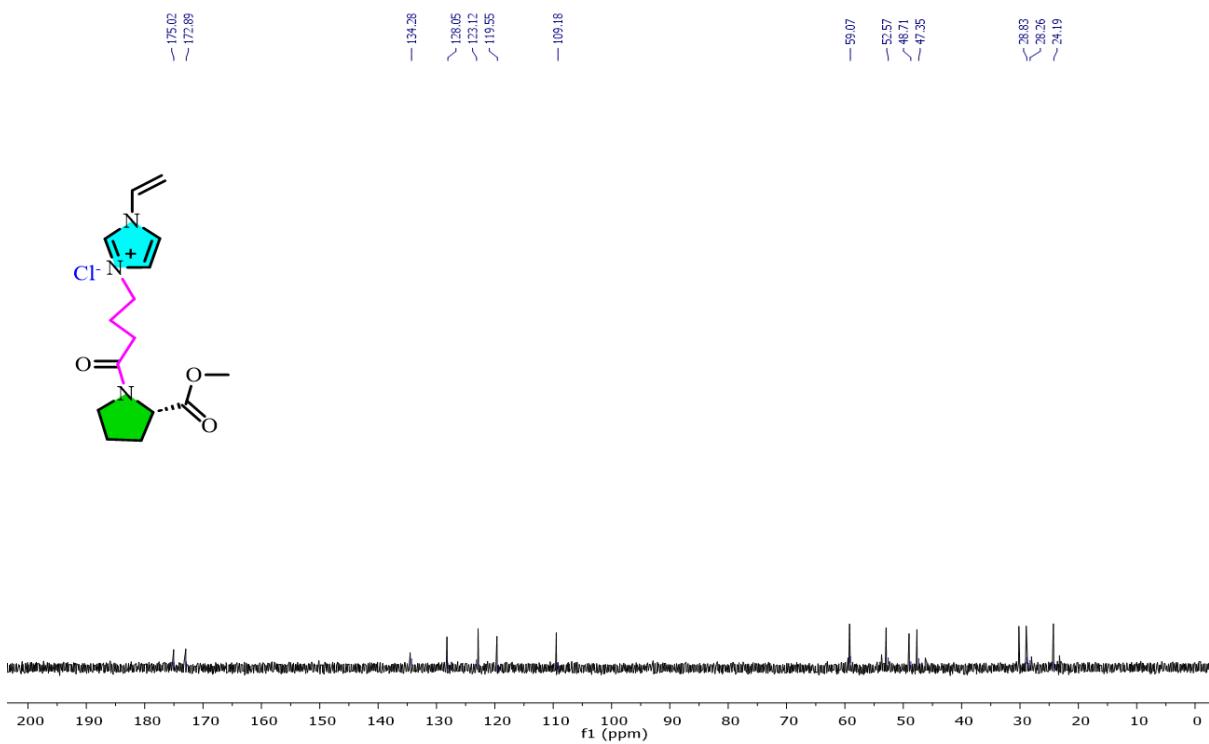
**Fig. S3**  $^{13}\text{C}$  NMR spectrum of mono-(L)-3-Cl.



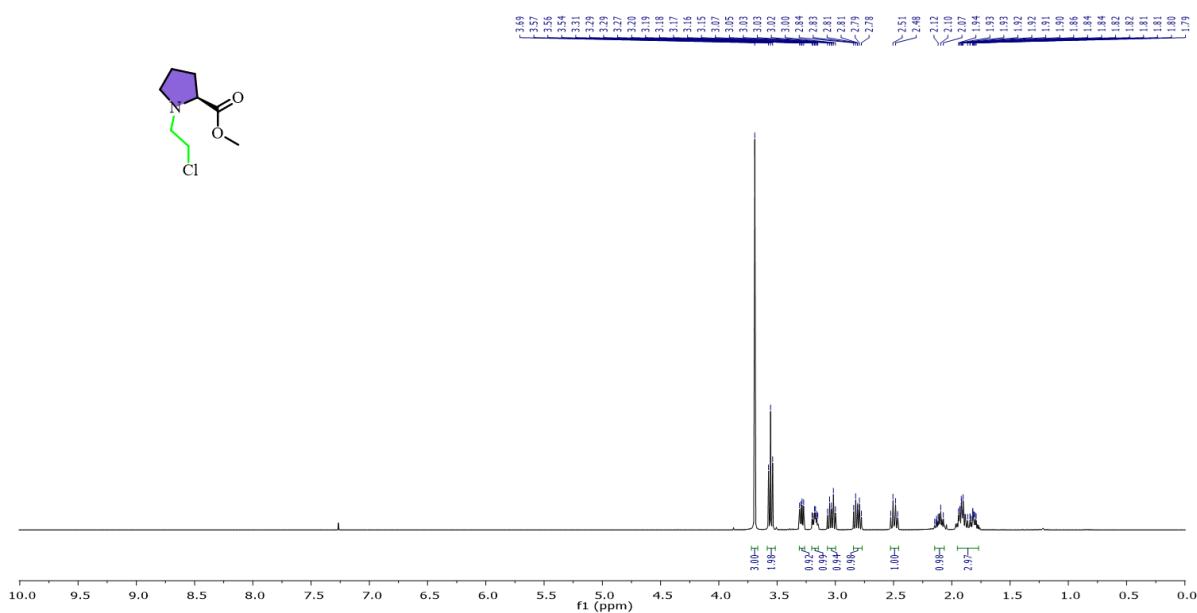
**Fig. S4**  $^1\text{H}$  NMR spectrum of (L)-4.



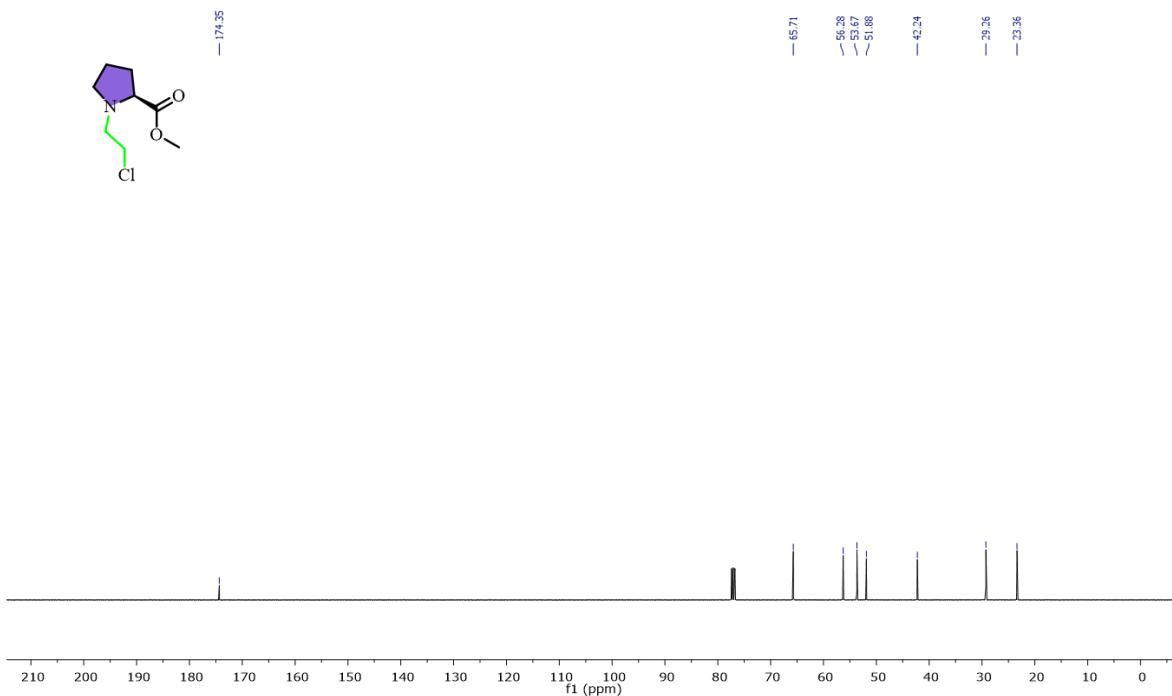
**Fig. S5** <sup>13</sup>C NMR spectrum of (L)-4.



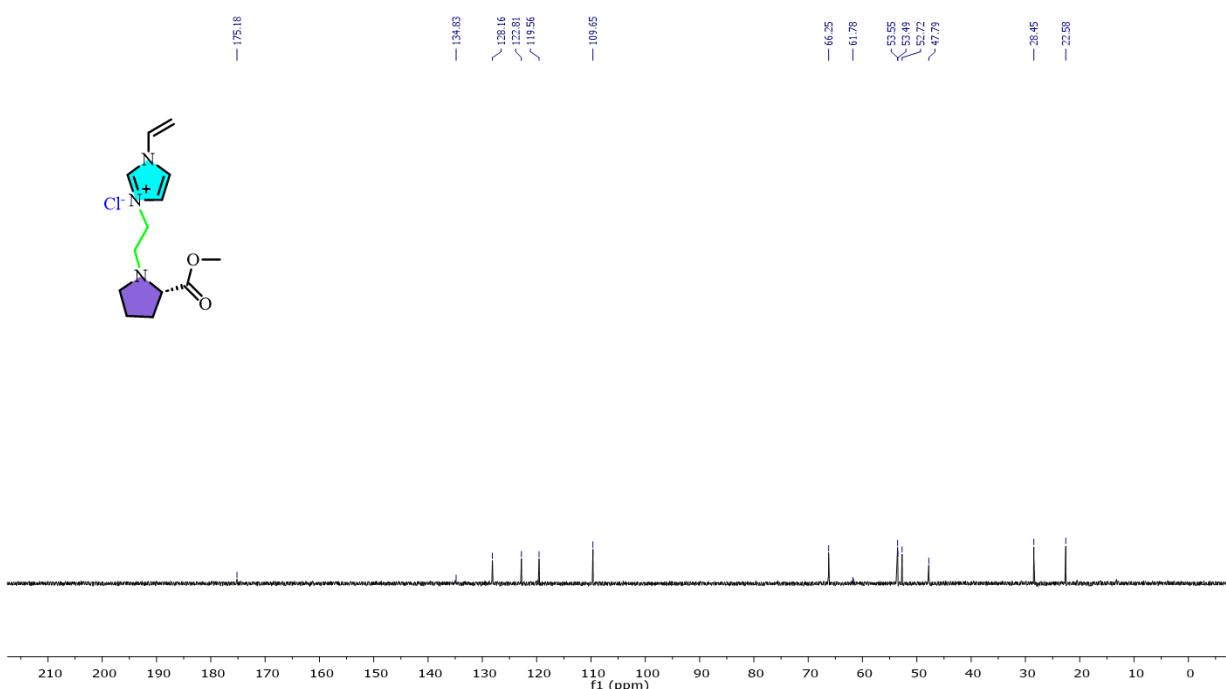
**Fig. S6** <sup>13</sup>C NMR spectrum of mono-(L)-4-Cl.



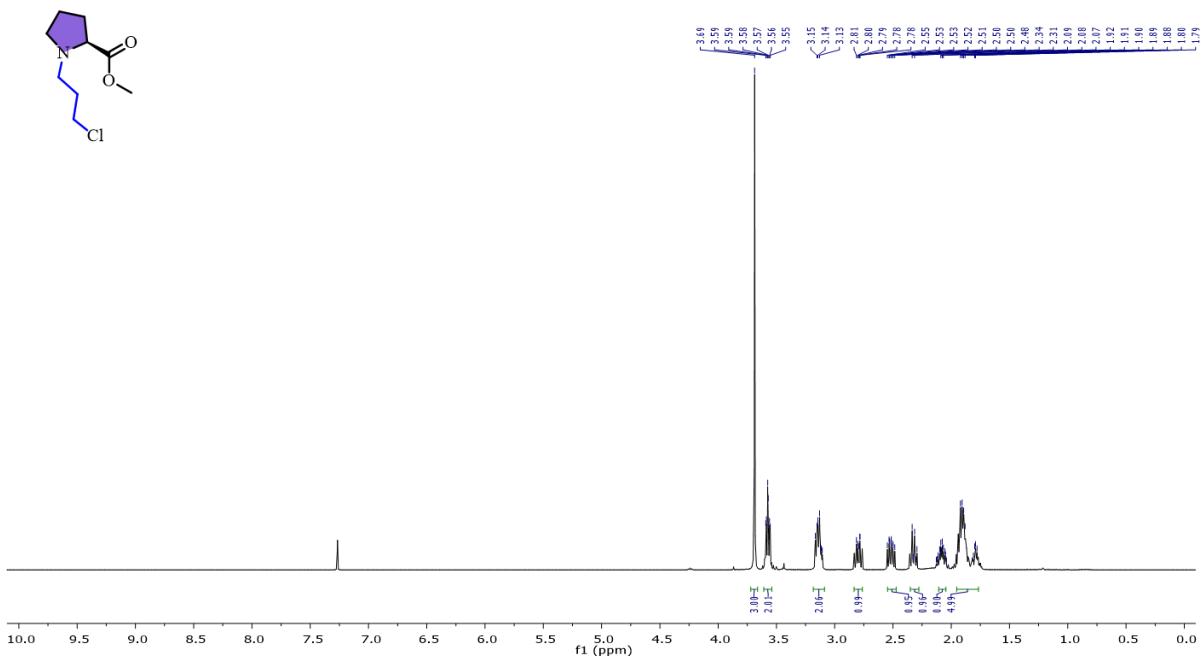
**Fig. S7**  $^1\text{H}$  NMR spectrum of (L)-5.



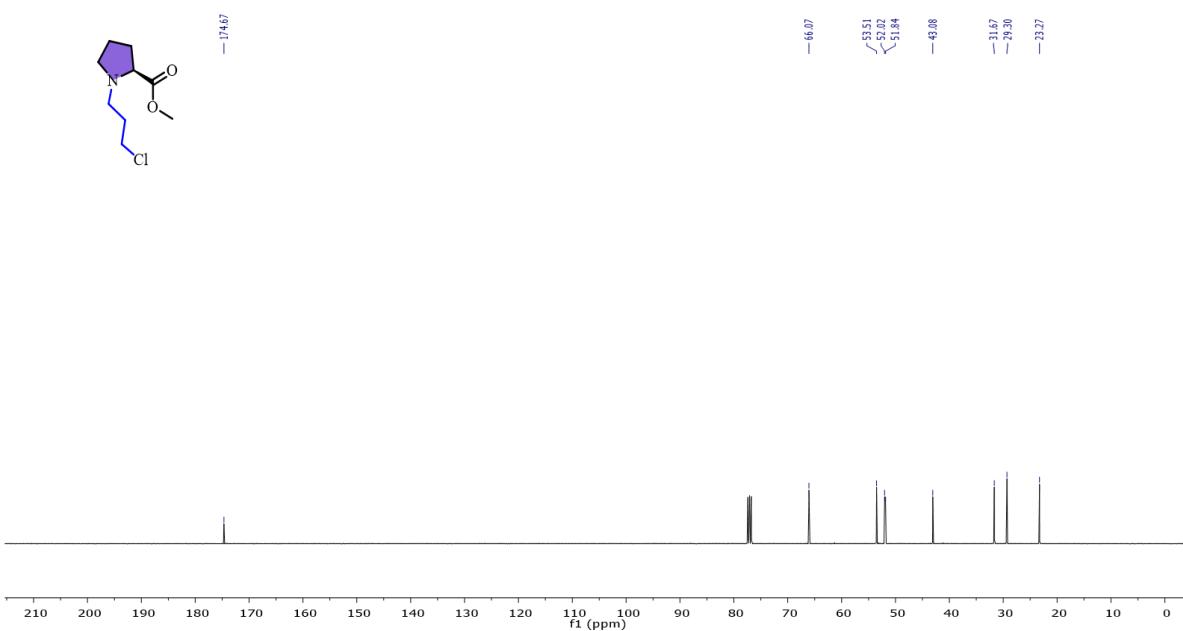
**Fig. S8**  $^1\text{H}$  NMR spectrum of (L)-5.



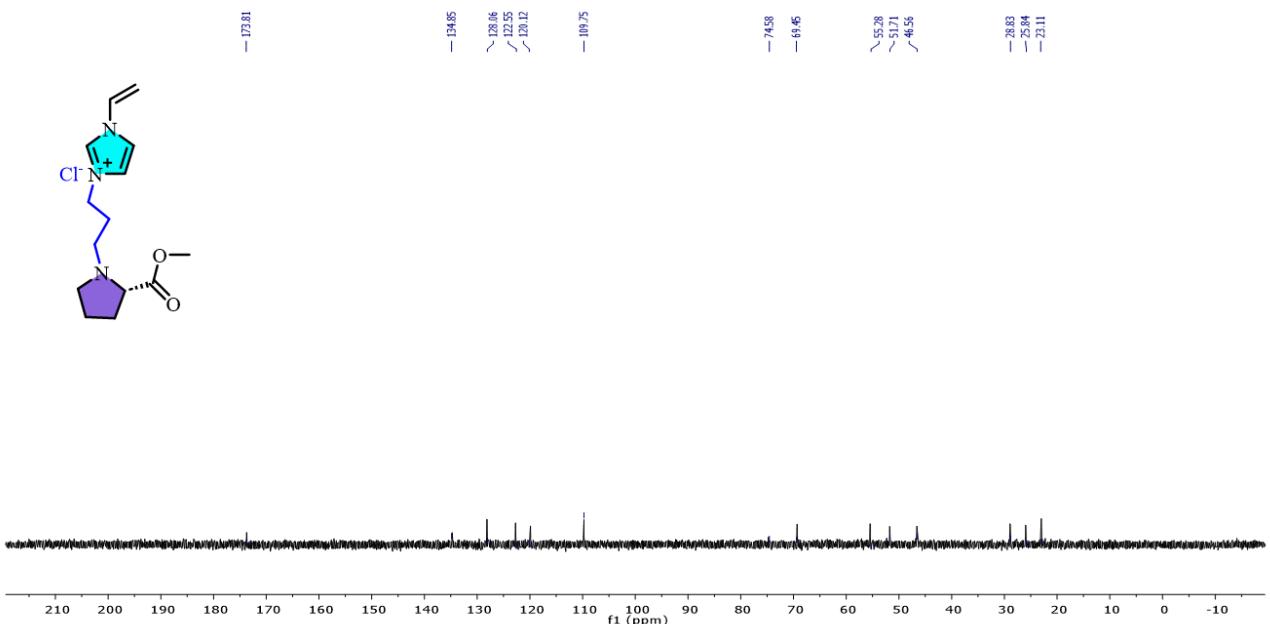
**Fig. S9**  $^{13}\text{C}$  NMR spectrum of mono-(L)-5-Cl.



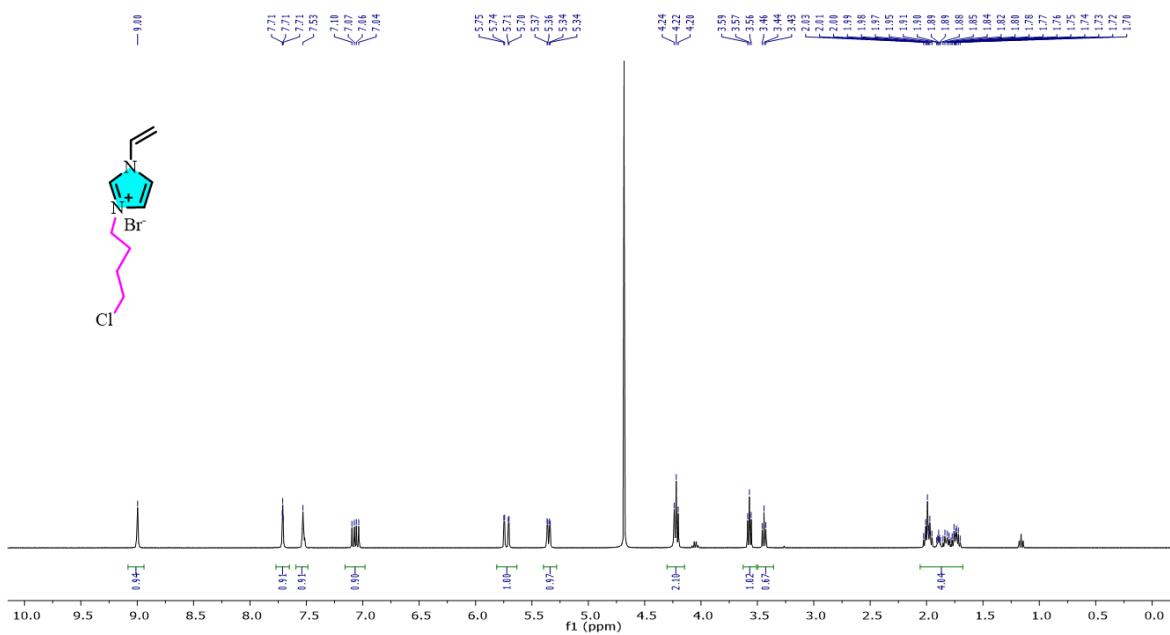
**Fig. S10**  $^1\text{H}$  NMR spectrum of (L)-5.



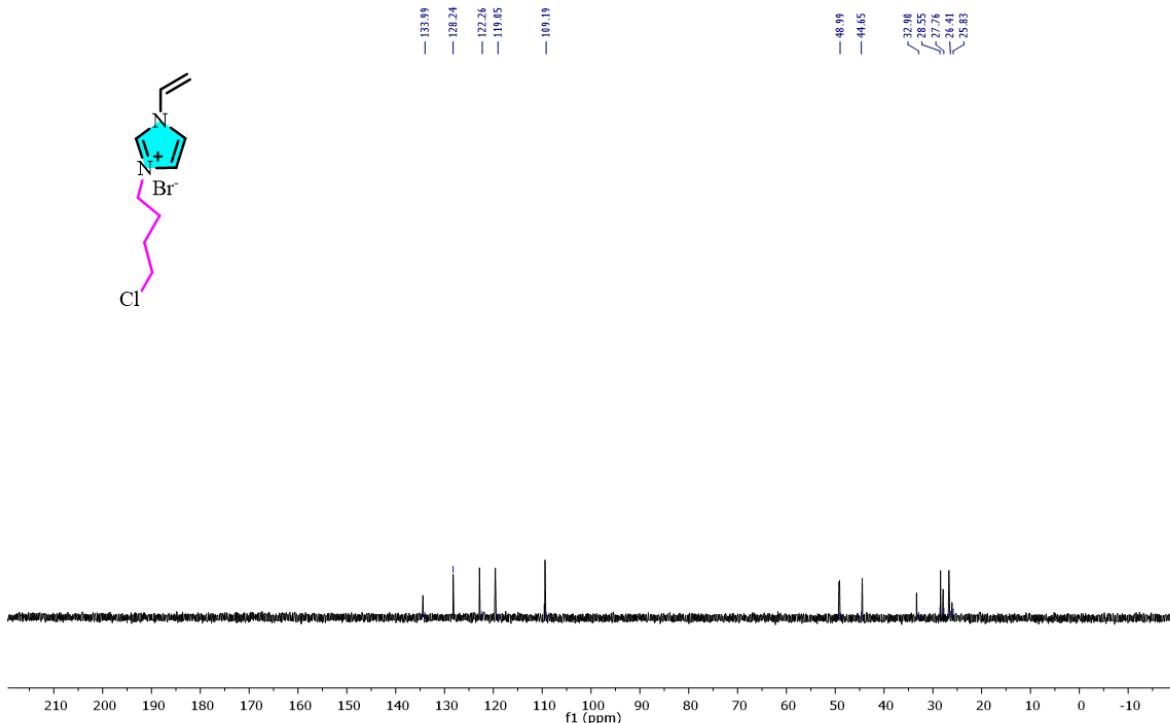
**Fig. S11**  $^{13}\text{C}$  NMR spectrum of (L)-5.



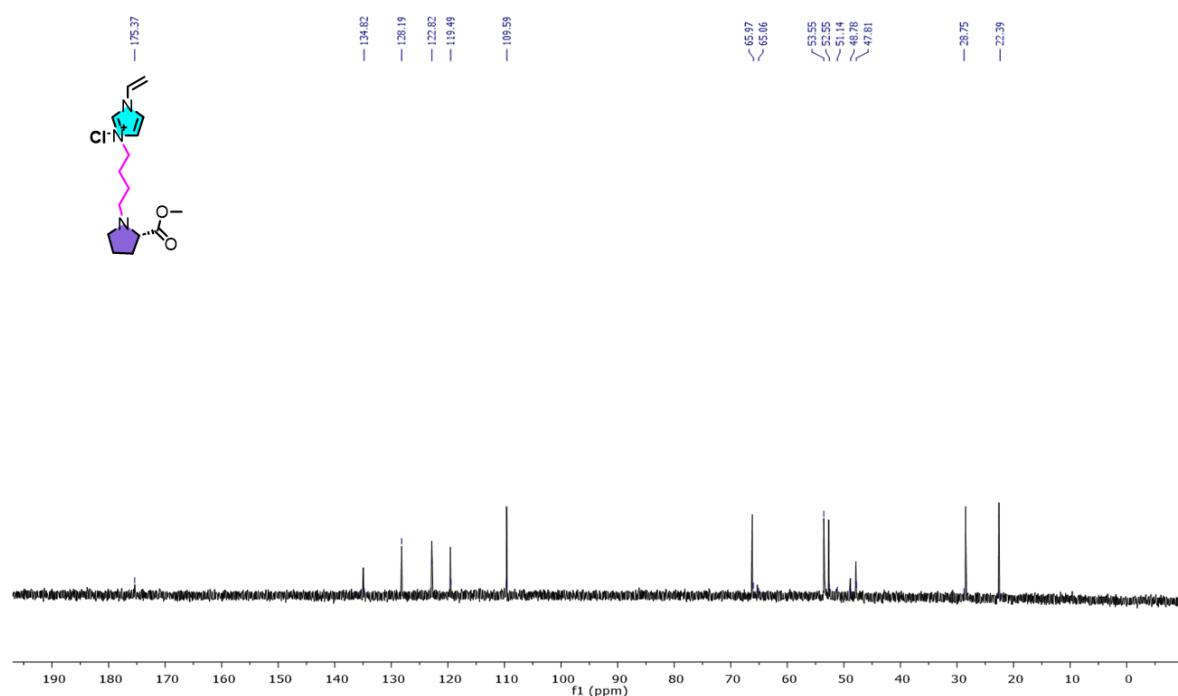
**Fig. S12**  $^{13}\text{C}$  NMR spectrum of mono-(L)-6-Cl.



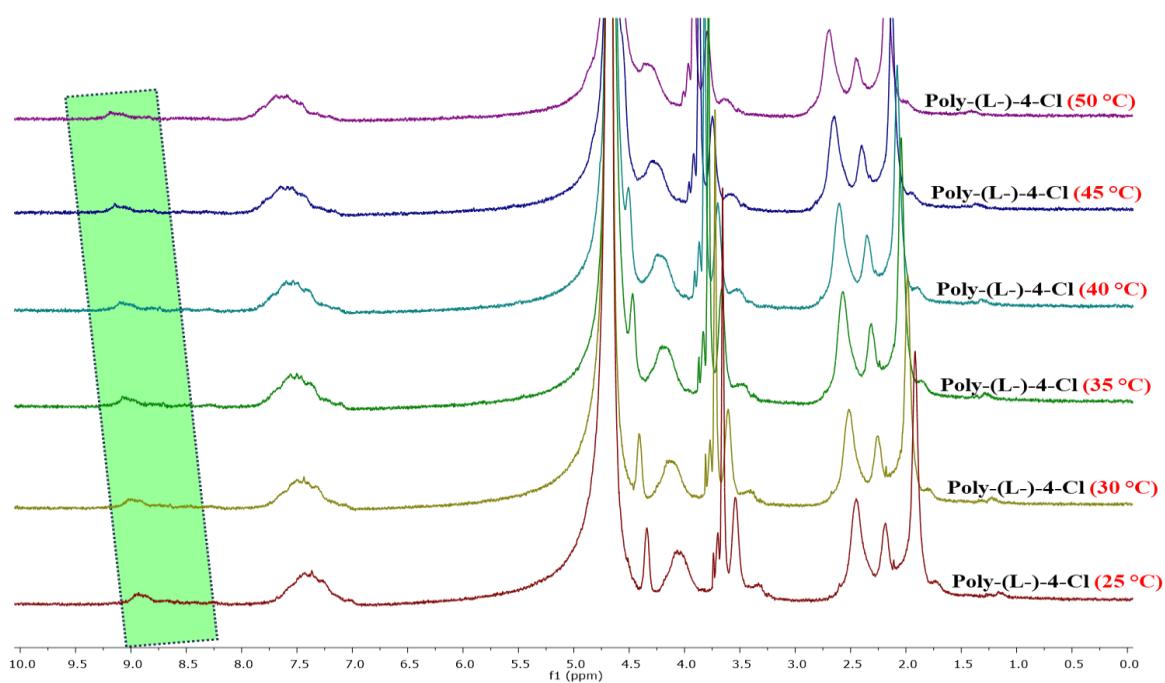
**Fig. S13**  $^1\text{H}$  NMR spectrum of (L)-7.



**Fig. S14**  $^{13}\text{C}$  NMR spectrum of (L-)-7.

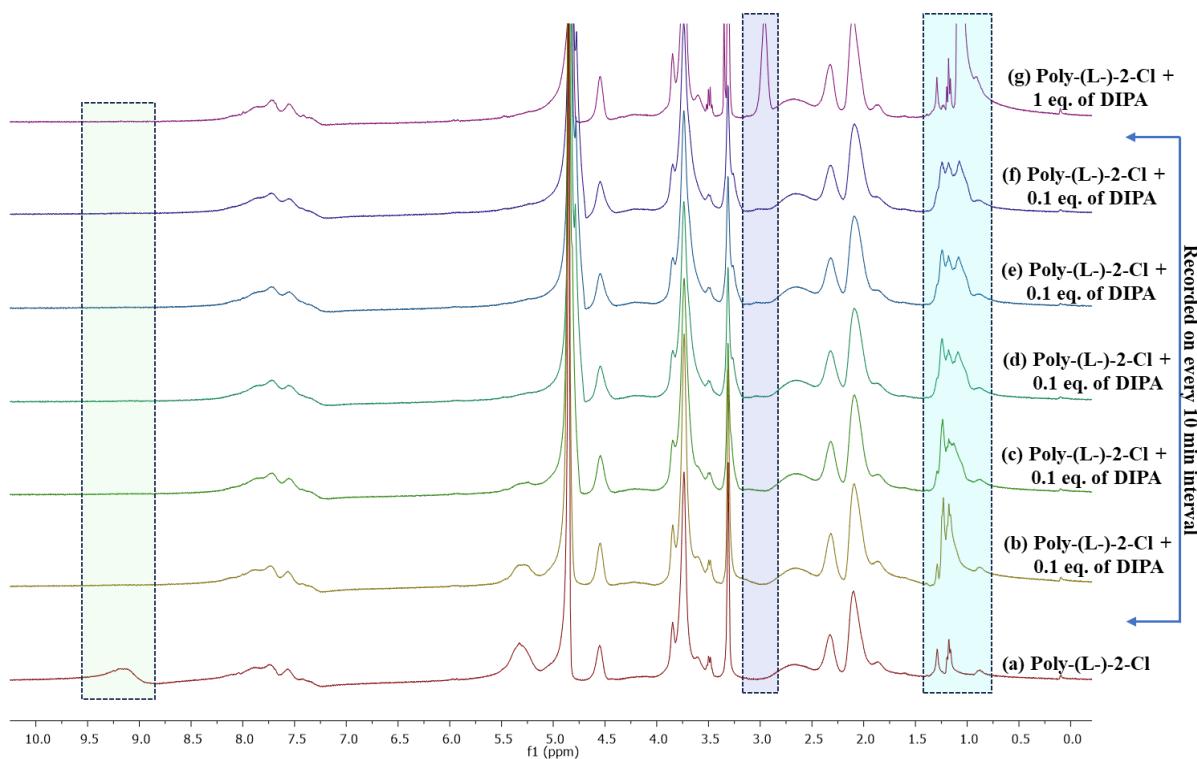


**Fig. S15**  $^{13}\text{C}$  NMR spectrum of mono-(L)-7-Cl.

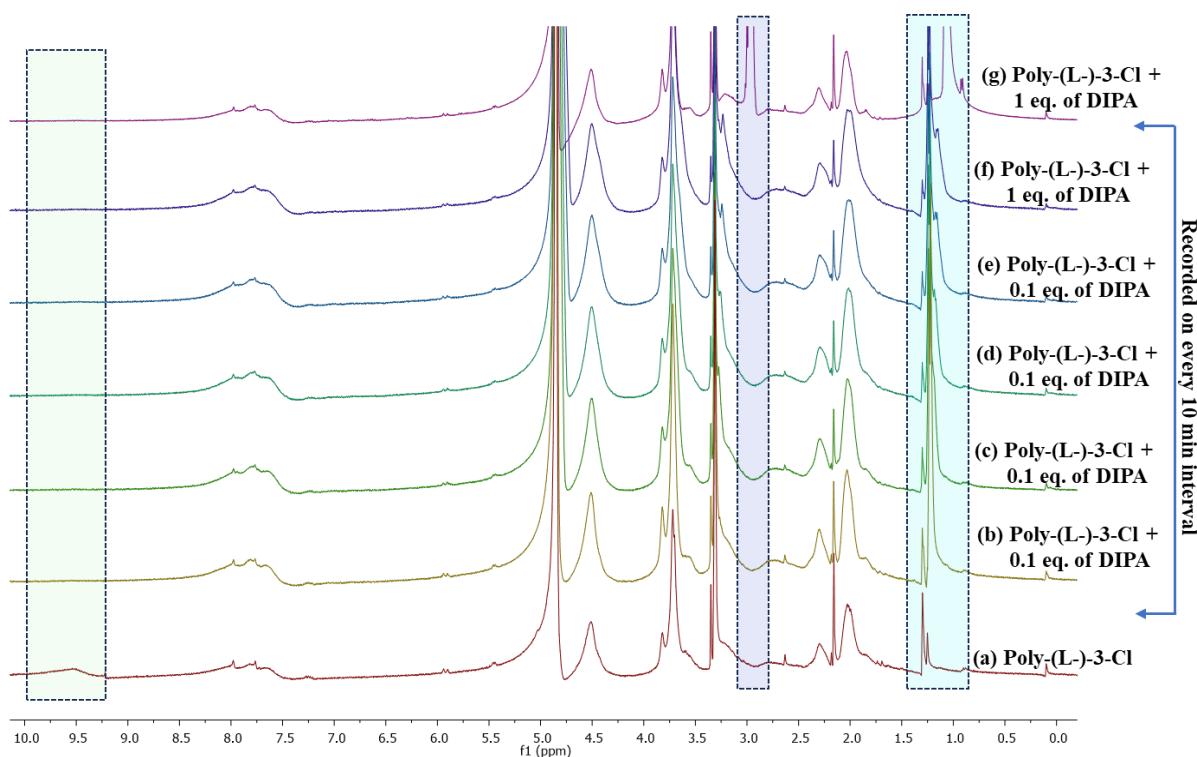


**Fig. S16**  $^1\text{H}$  (in  $\text{D}_2\text{O}$ ) NMR of poly-(L)-4-Cl at different temperatures.

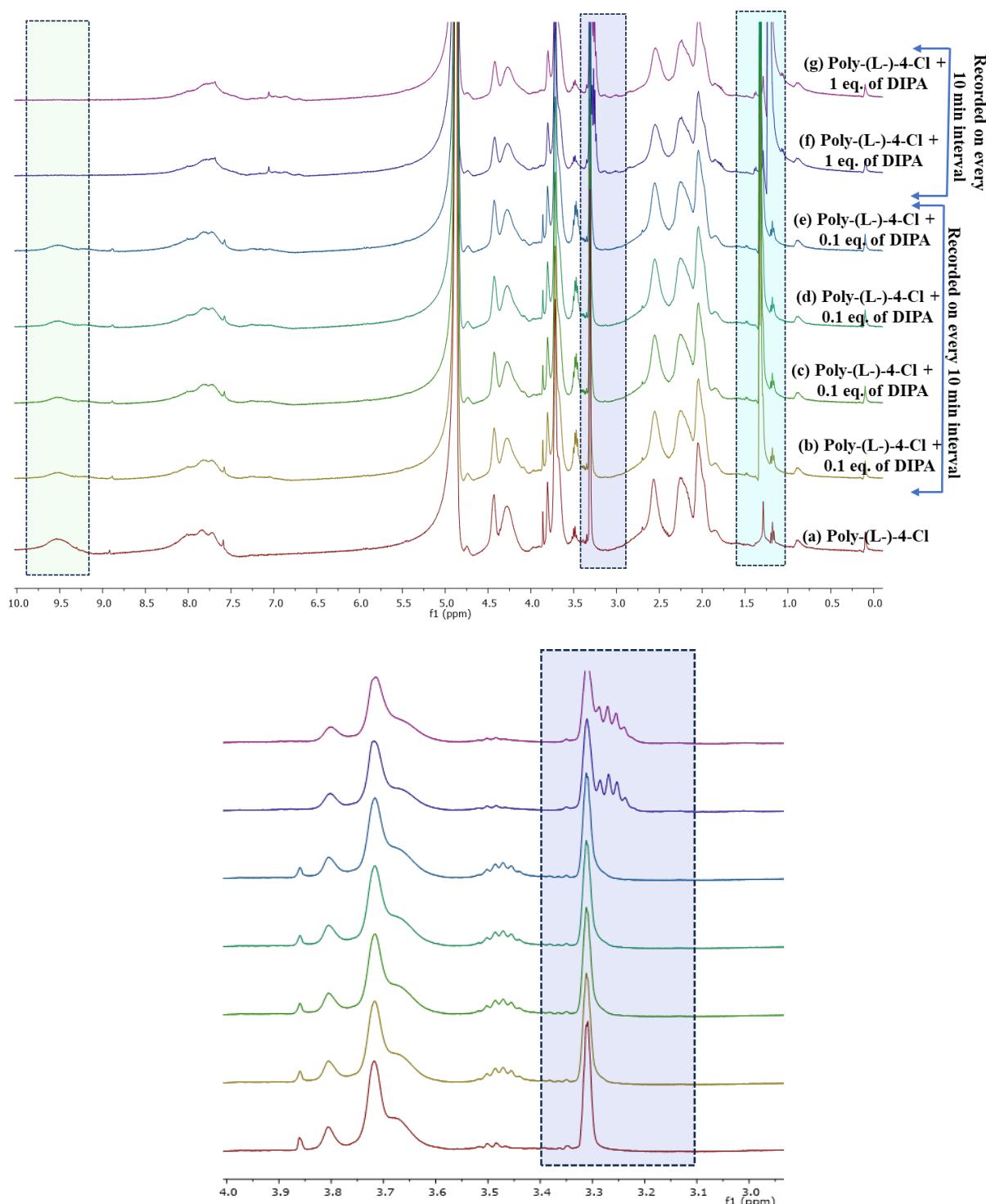
A)



B)



C)

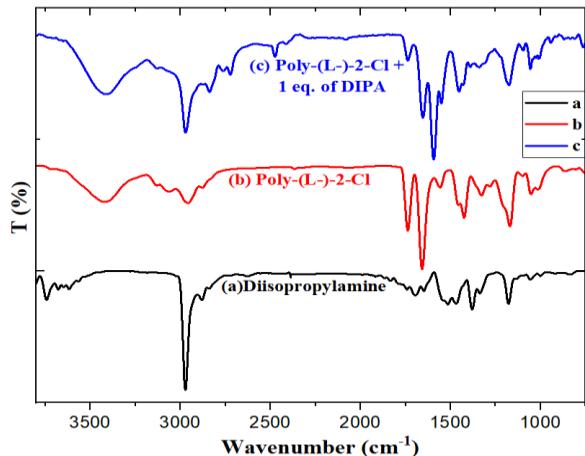


(Expanded region i.e. 3 – 4 ppm for poly-(L)-4-Cl)

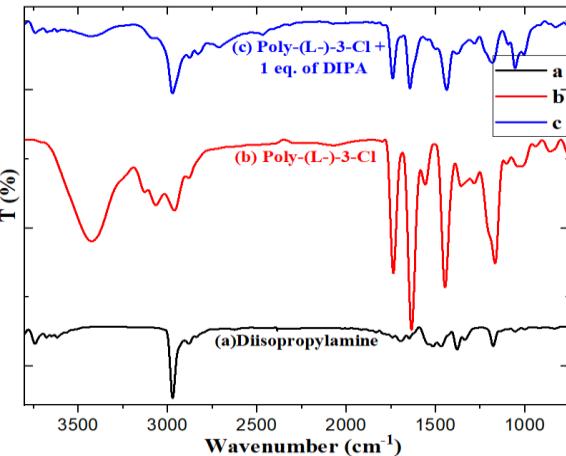
**Fig. S17**  $^1\text{H}$  (in MeOD) NMR of A) poly-(L)-2-Cl B) poly-(L)-3-Cl and C) poly-(L)-4-Cl in presence of 0.1 and 1 equivalent of diisopropylamine at different time interval.

**FT-IR spectroscopy of chiral monomers and their corresponding chiral polymers in presence of diisopropylamine:** \* appearing due to some instrumental error.

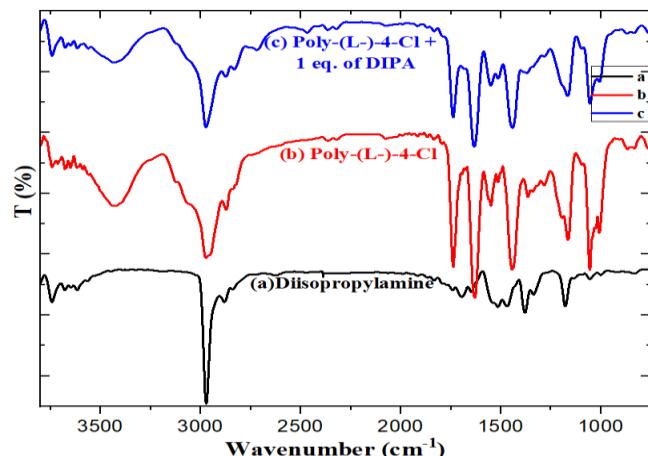
A)



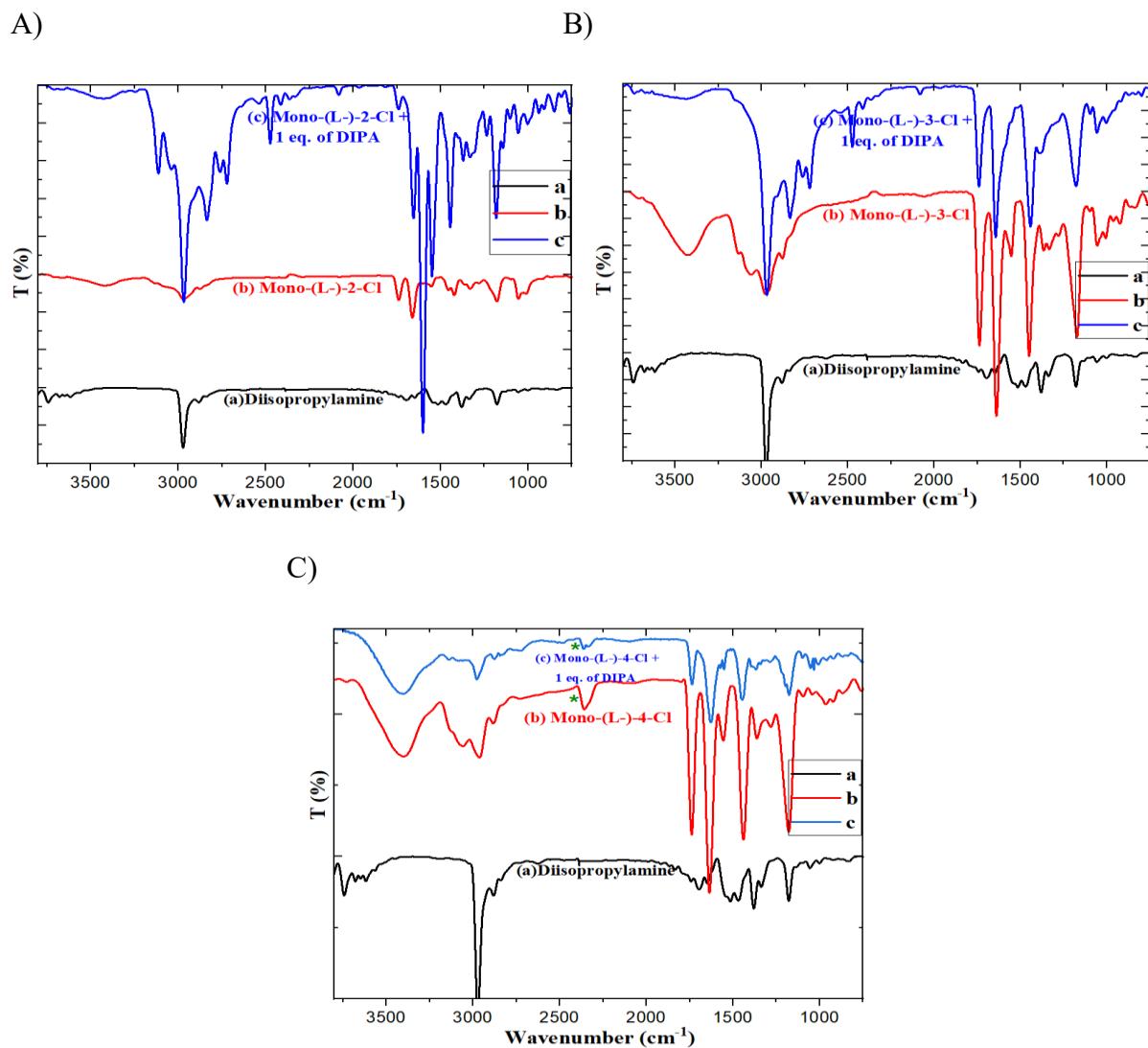
B)



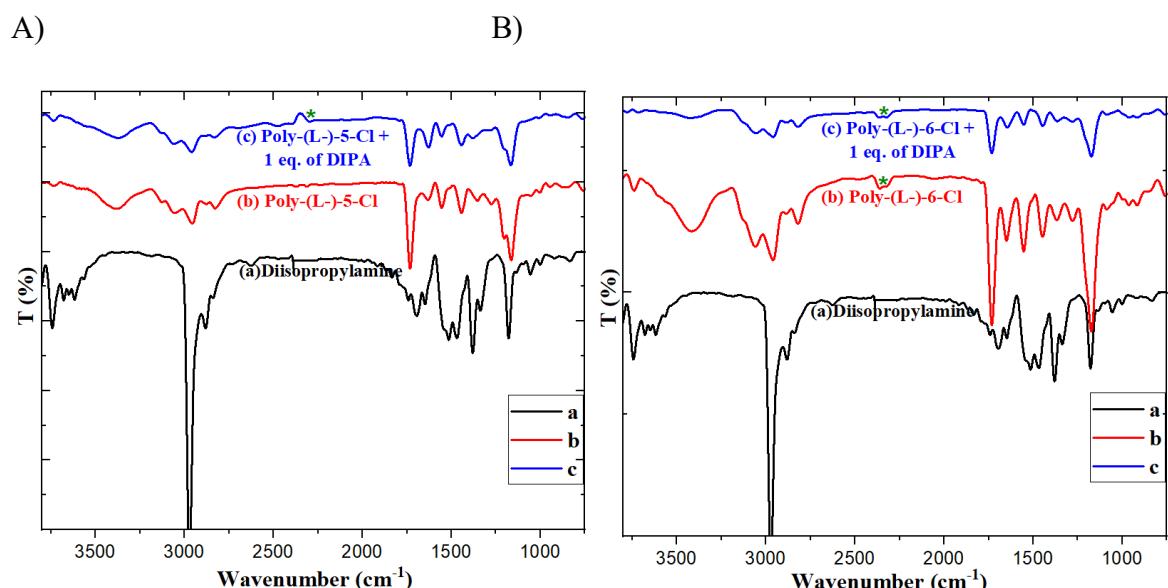
C)



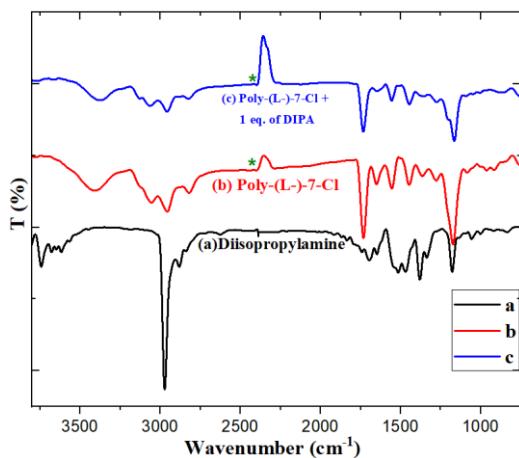
**Fig. S18** FT-IR overlay of A) poly-(L)-2-Cl, B) poly-(L)-3-Cl and C) poly-(L)-4-Cl in presence of diisopropylamine.



**Fig. S19** FT-IR overlay of A) **mono-(L)-2-Cl**, B) **mono-(L)-3-Cl** and C) **mono-(L)-4-Cl** in presence of diisopropylamine.

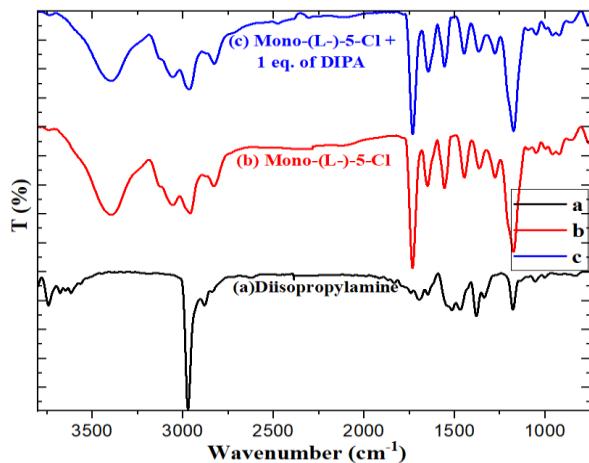


C)

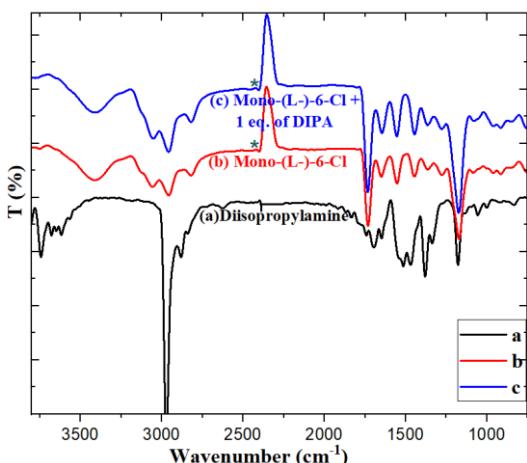


**Fig. S20** FT-IR overlay of A) **poly-(L)-5-Cl**, B) **poly-(L)-6-Cl** and C) **poly-(L)-7-Cl** in presence of diisopropylamine.

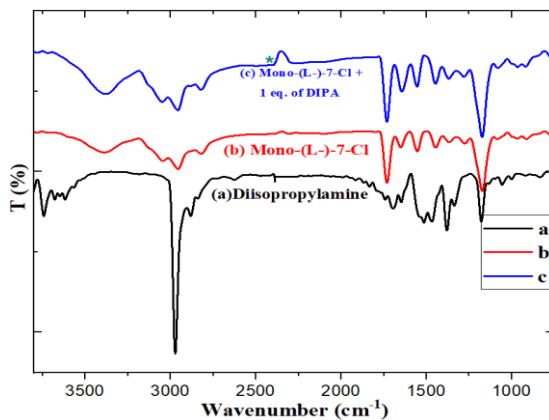
A)



B)

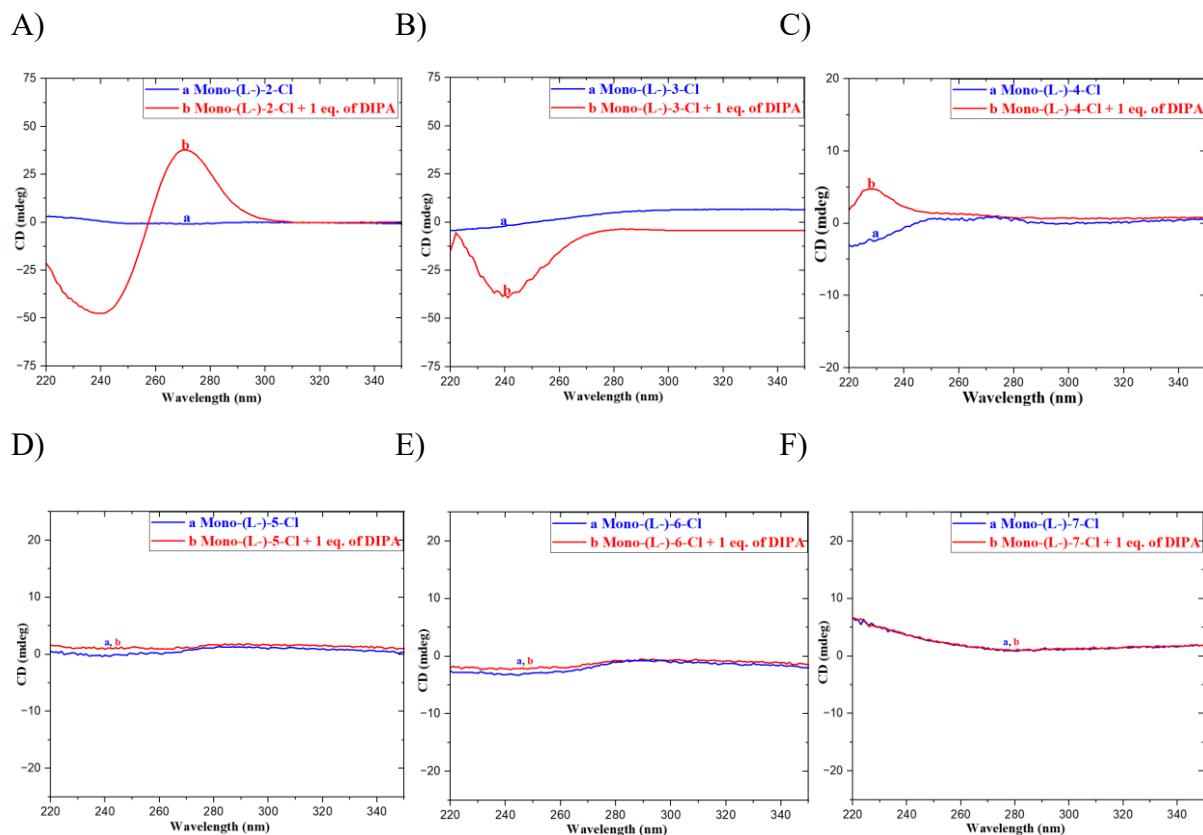


C)

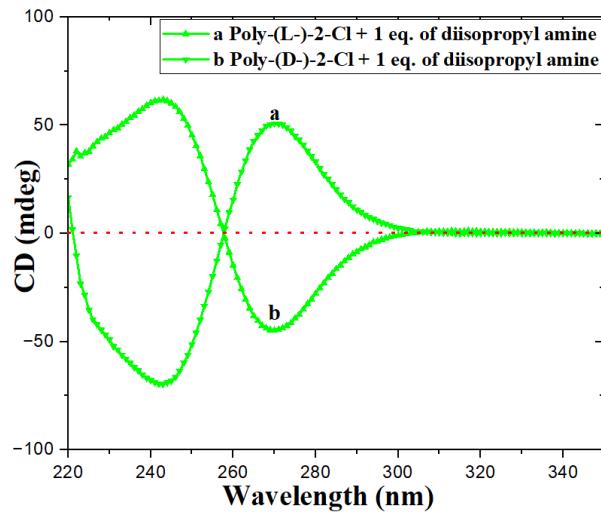


**Fig. S21** FT-IR overlay of A) **mono-(L)-5-Cl**, B) **mono-(L)-6-Cl** and C) **poly-(L)-7-Cl** in presence of diisopropylamine.

**CD absorption spectra of chiral monomes:**



**Fig. S22** CD absorption spectra of chiral monomers before and after addition of diisopropylamine.



**Fig. S23** CD absorption spectra of **Poly-(L)-2-Cl** and **Poly-(D)-2-Cl** in presence of diisopropylamine (solvent = methanol, Temp = 25 °C).