

Supplementary Information

For

**Axial-phenyl Constrained Bis(imino)acenaphthene-Nickel Precatalysts Enhancing
Ethylene Polymerization**

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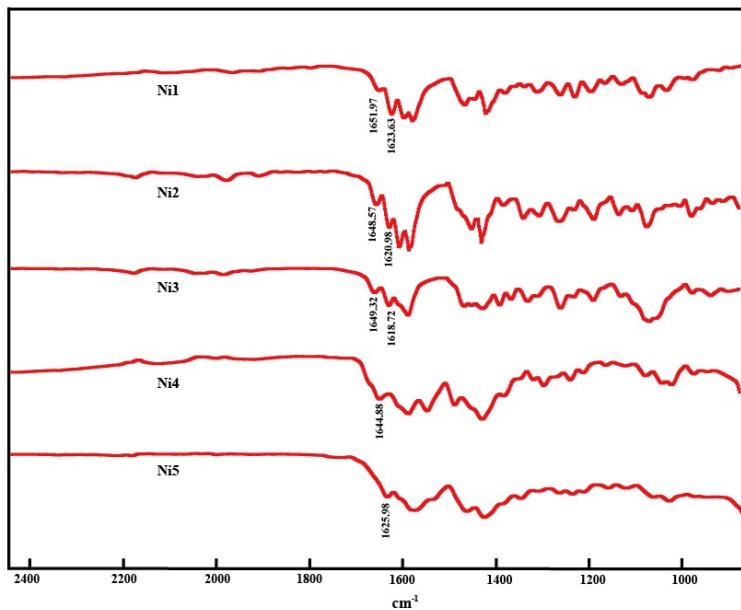


Figure S1. FT-IR spectrum of Ni1-Ni5.

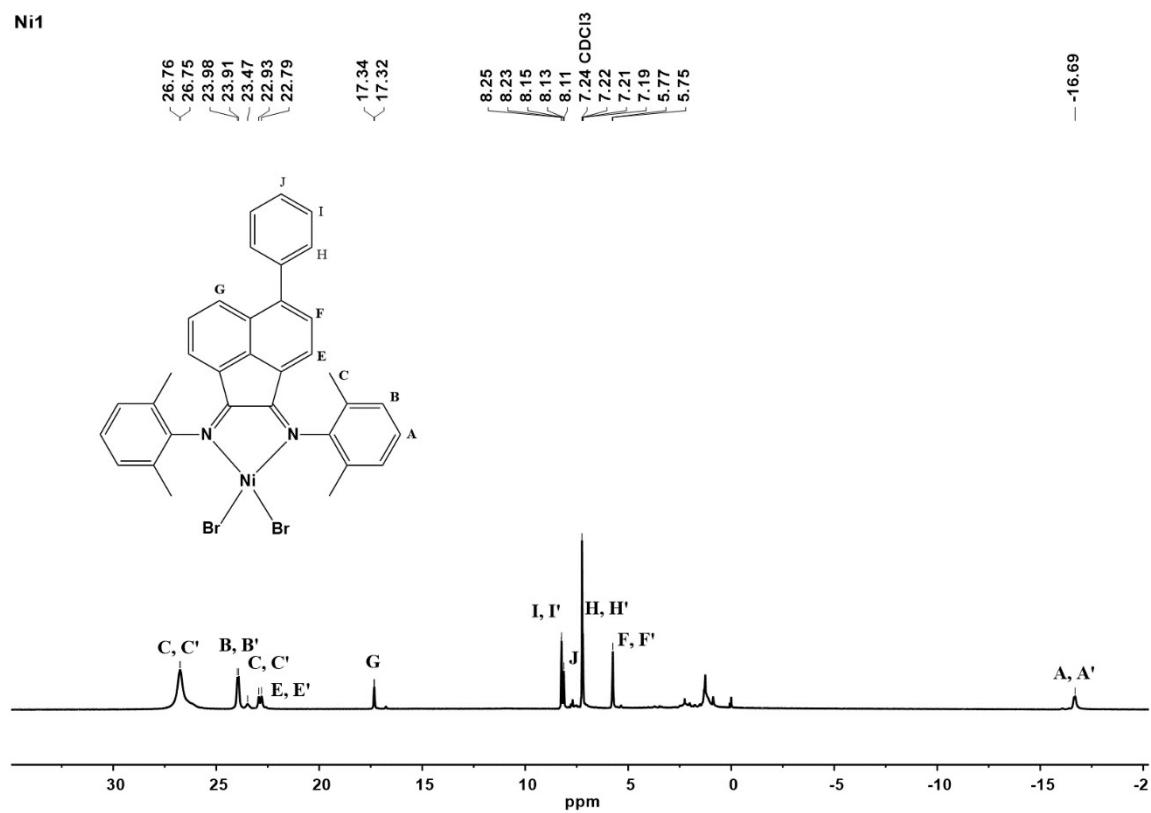


Figure S2. ¹H NMR (400 MPa) spectrum of Ni1 (recorded in CDCl₃ at 25 °C).

Ni2

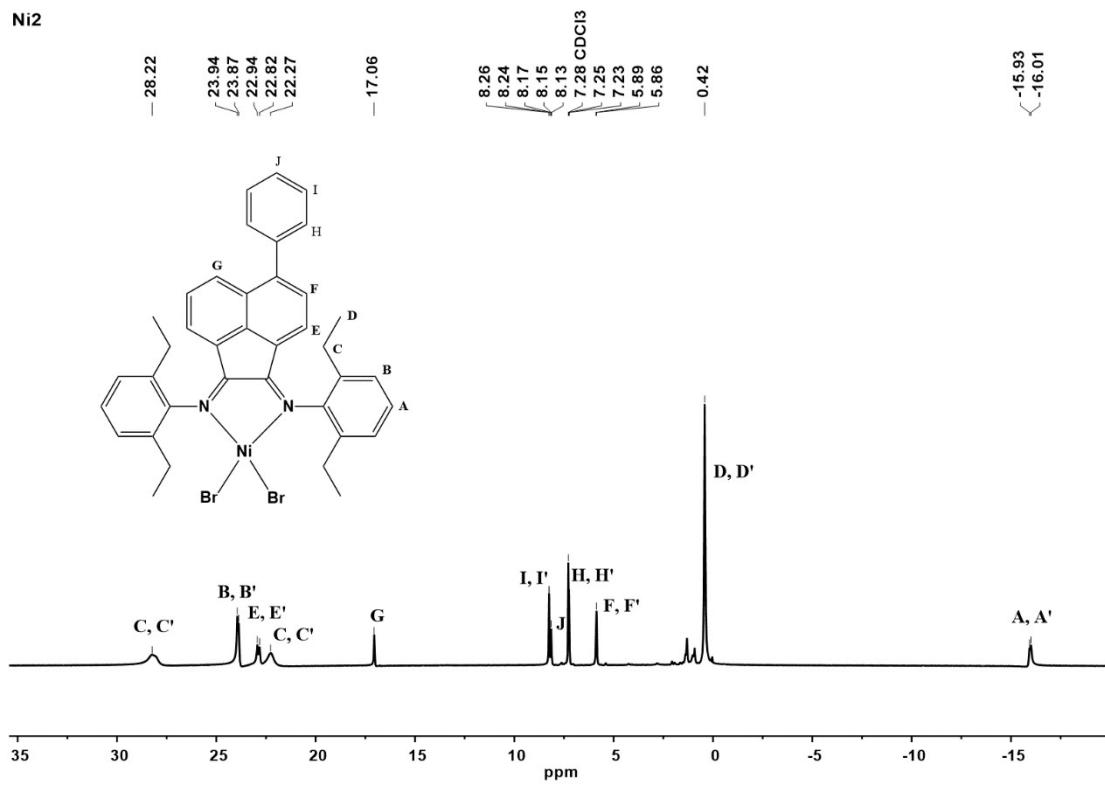


Figure S3. ¹H NMR (400 MPa) spectrum of Ni2 (recorded in CDCl₃ at 25 °C).

Ni3

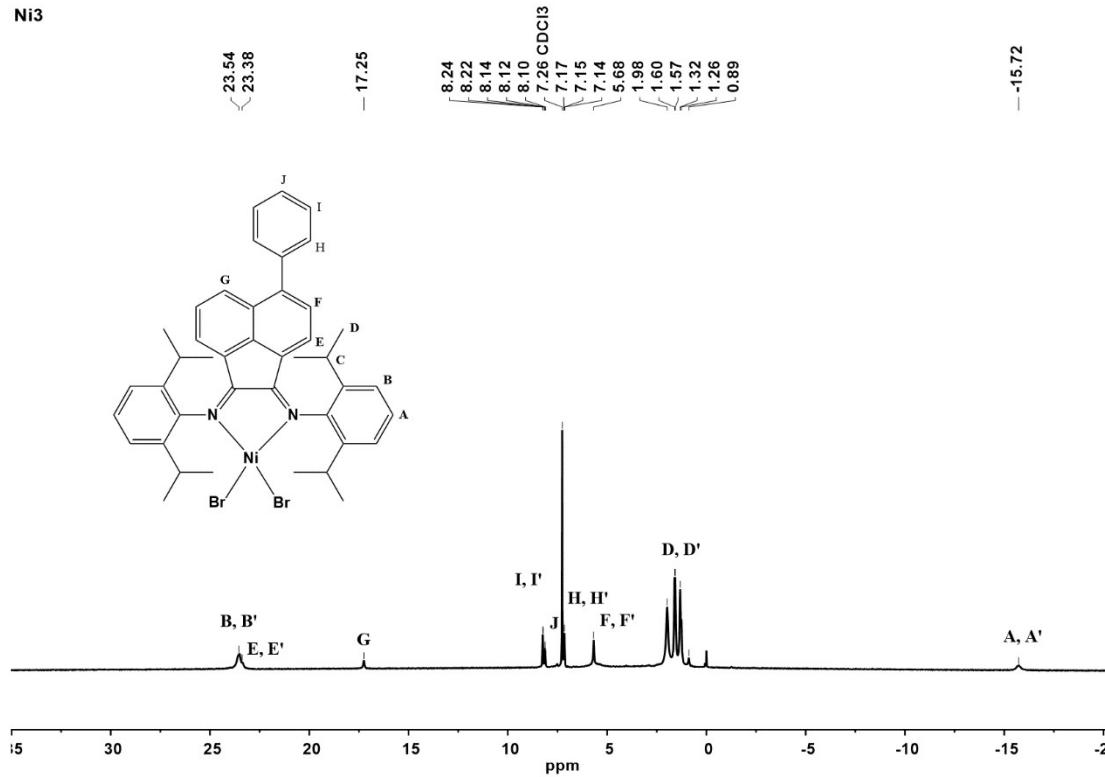


Figure S4. ¹H NMR (400 MPa) spectrum of Ni3 (recorded in CDCl₃ at 25 °C).

Ni4

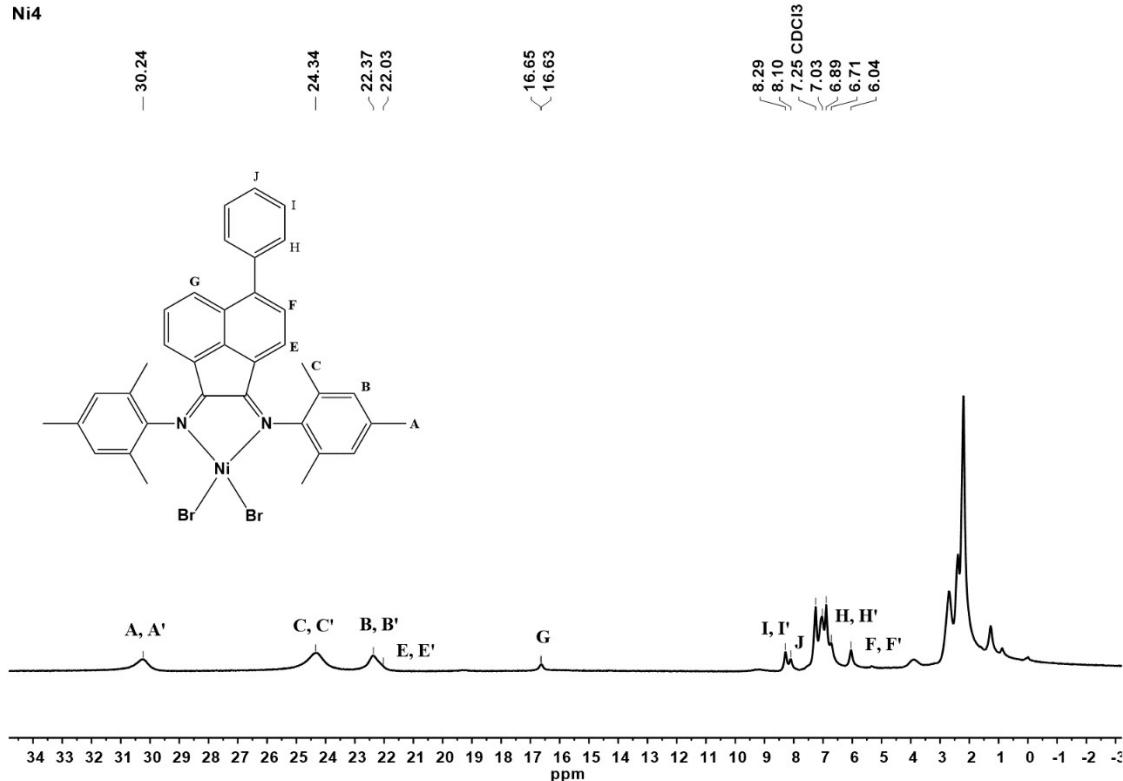


Figure S5. ¹H NMR (400 MPa) spectrum of Ni4 (recorded in CDCl₃ at 25 °C).

Ni5

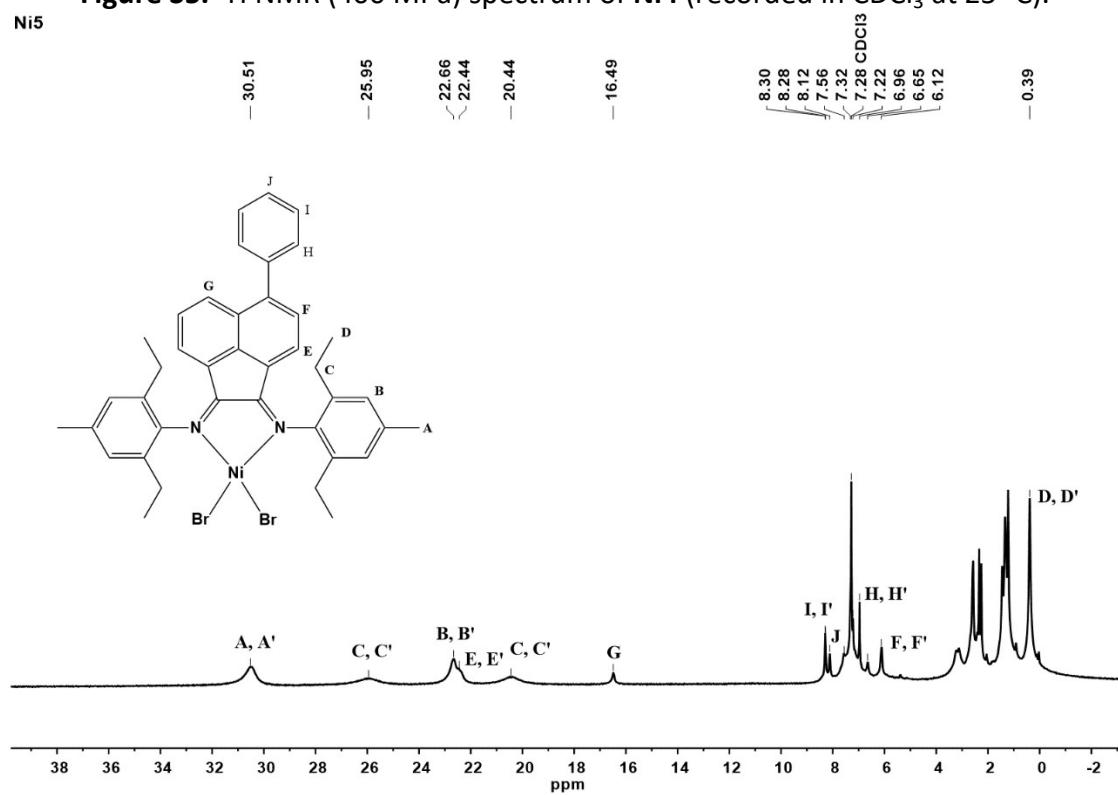


Figure S6. ¹H NMR (400 MPa) spectrum of Ni5 (recorded in CDCl₃ at 25 °C).

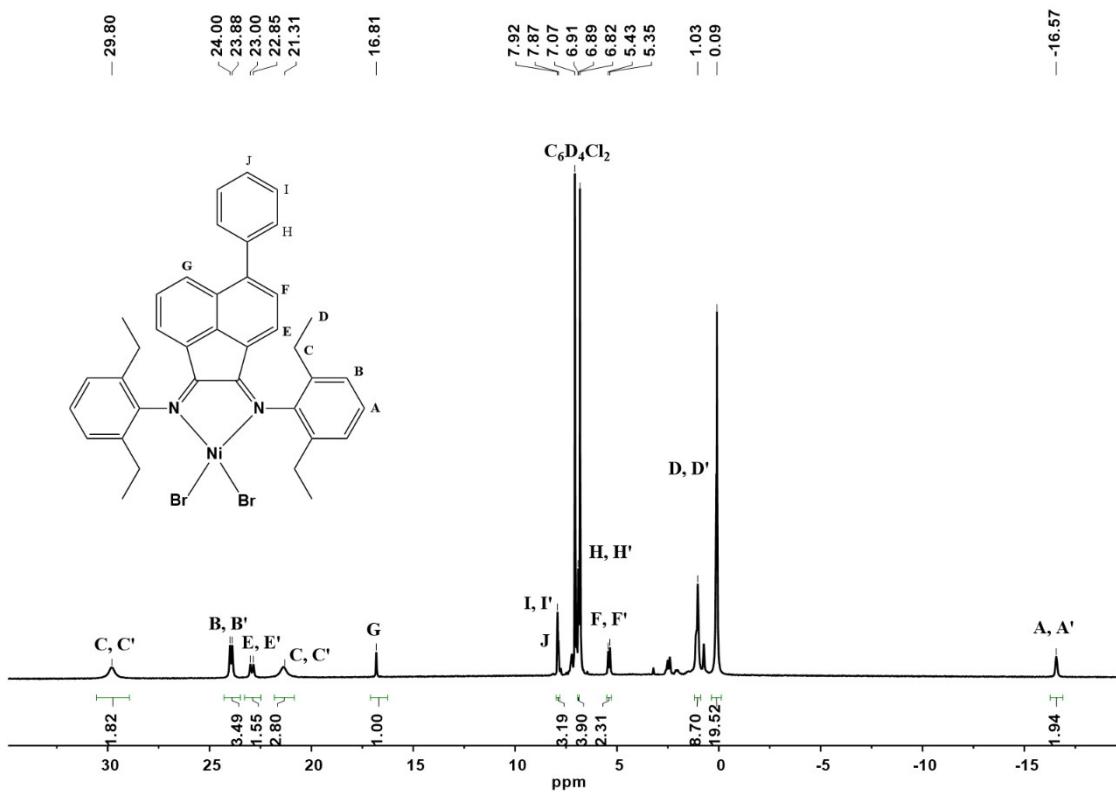


Figure S7. ^1H NMR (500 MPa) spectrum of Ni2 (recorded in $\text{C}_6\text{D}_4\text{Cl}_2$ at 25 °C).

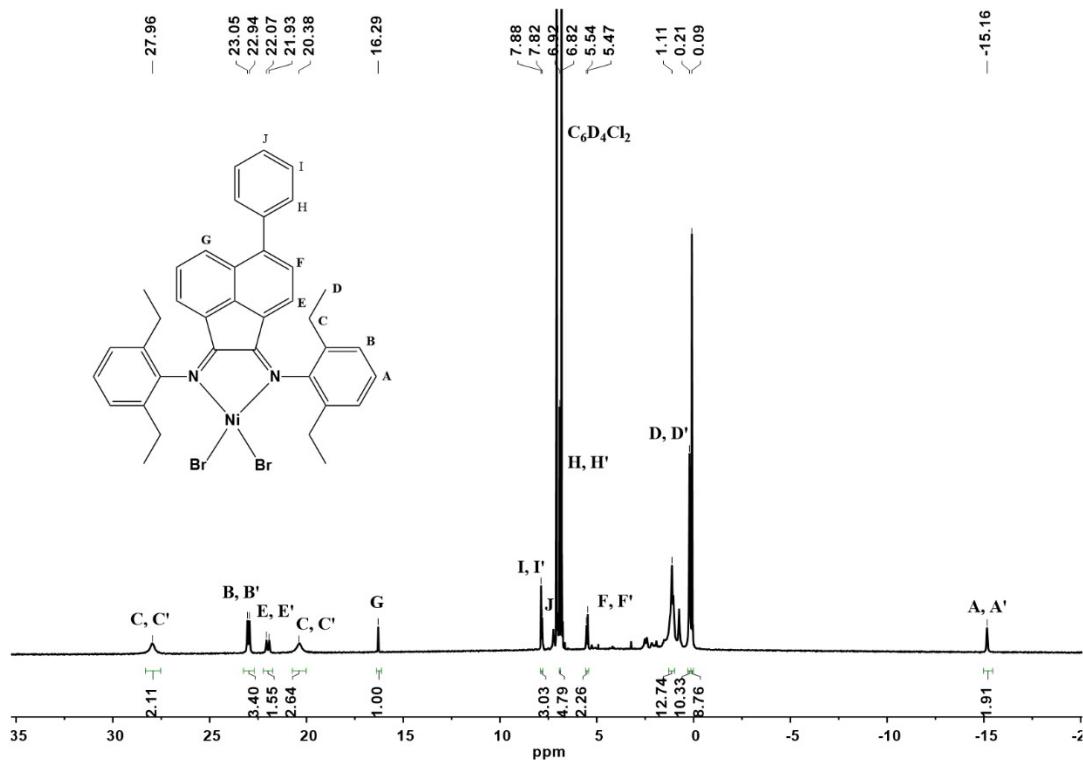


Figure S8. ^1H NMR (500 MPa) spectrum of Ni2 (recorded in $\text{C}_6\text{D}_4\text{Cl}_2$ at 40 °C).

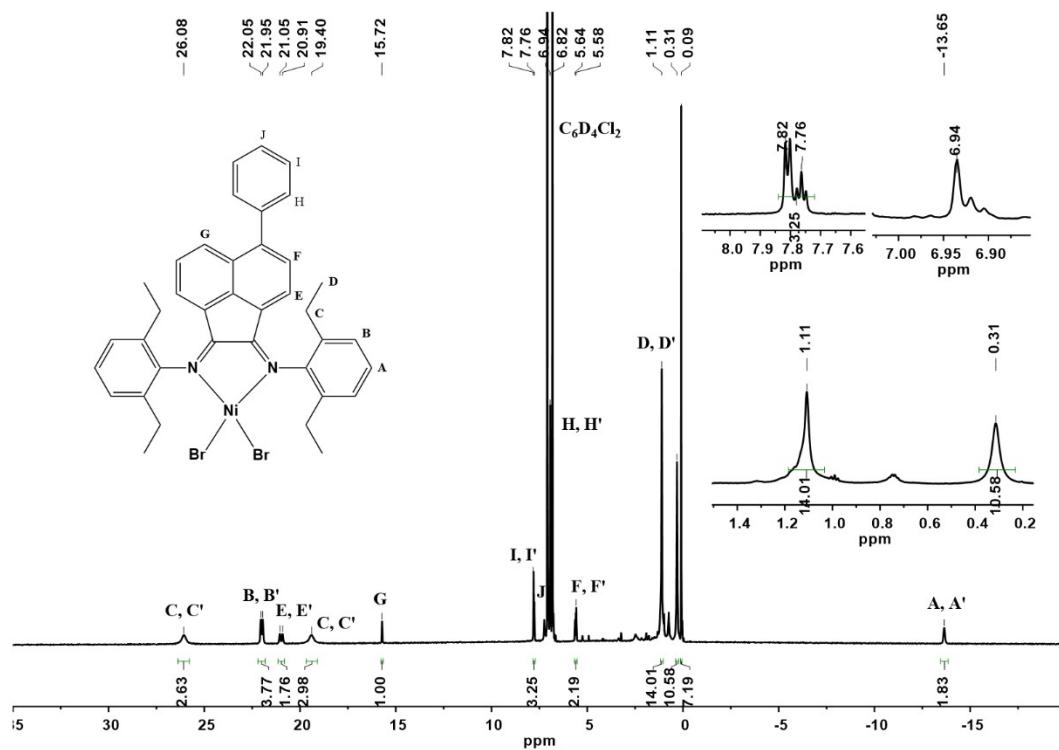


Figure S9. ^1H NMR (500 MPa) spectrum of **Ni2** (recorded in $\text{C}_6\text{D}_4\text{Cl}_2$ at 60 °C).

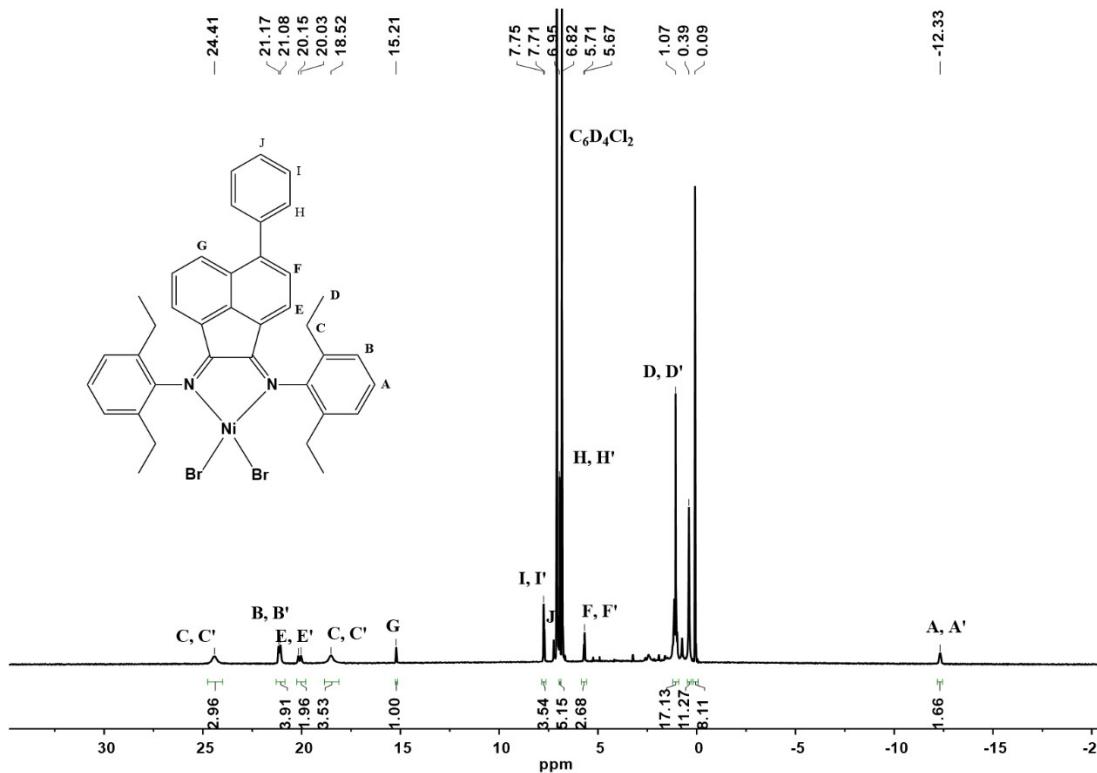


Figure S10. ^1H NMR (500 MPa) spectrum of **Ni2** (recorded in $\text{C}_6\text{D}_4\text{Cl}_2$ at 80 °C).

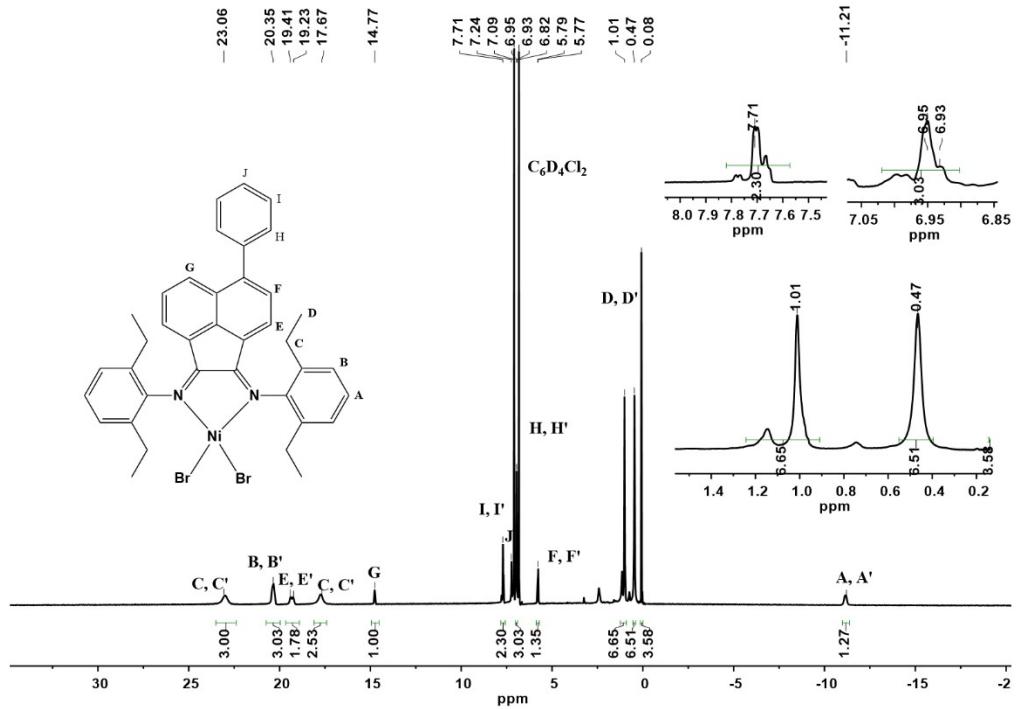


Figure S11. ^1H NMR (500 MPa) spectrum of **Ni2** (recorded in $\text{C}_6\text{D}_4\text{Cl}_2$ at 100 °C).

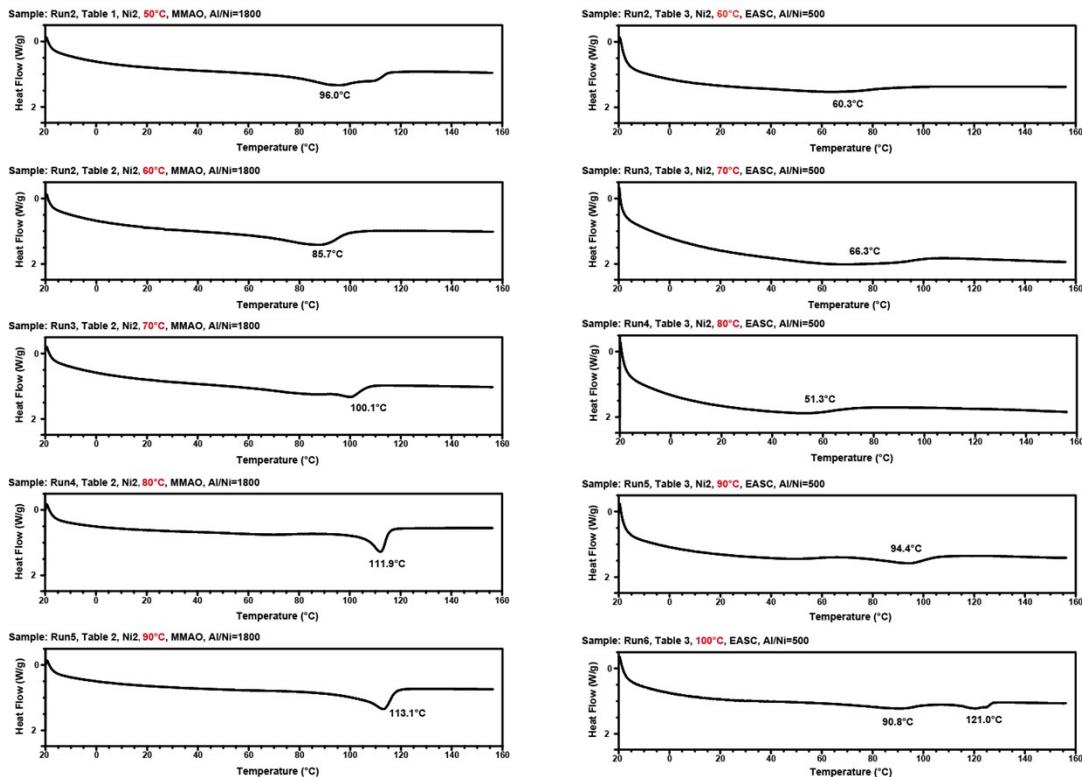


Figure S12. DSC spectrum of the polyethylenes obtained at different polymerization temperatures using MMAO or EASC as cocatalyst.

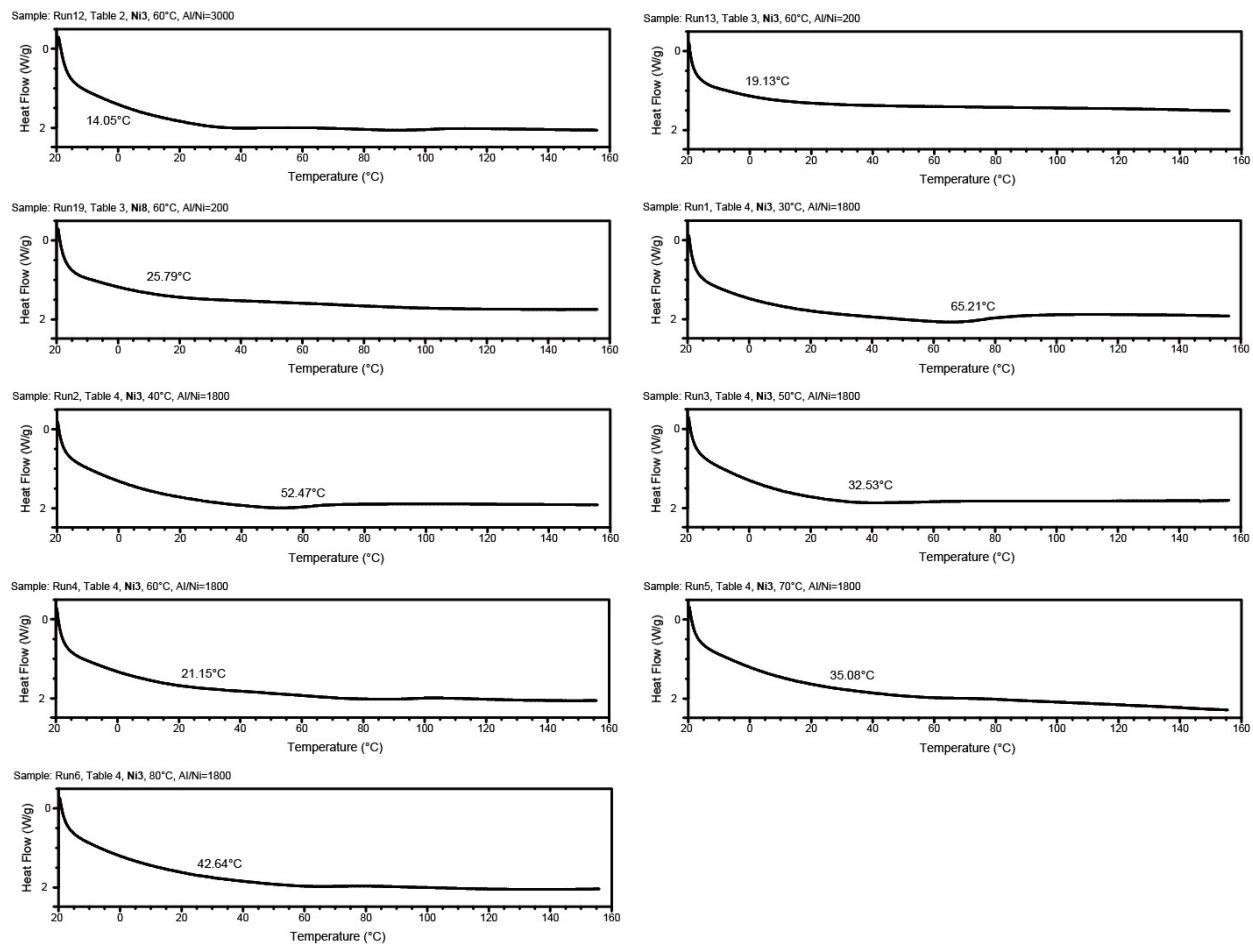


Figure S13. DSC spectrum of amorphous polyethylenes.

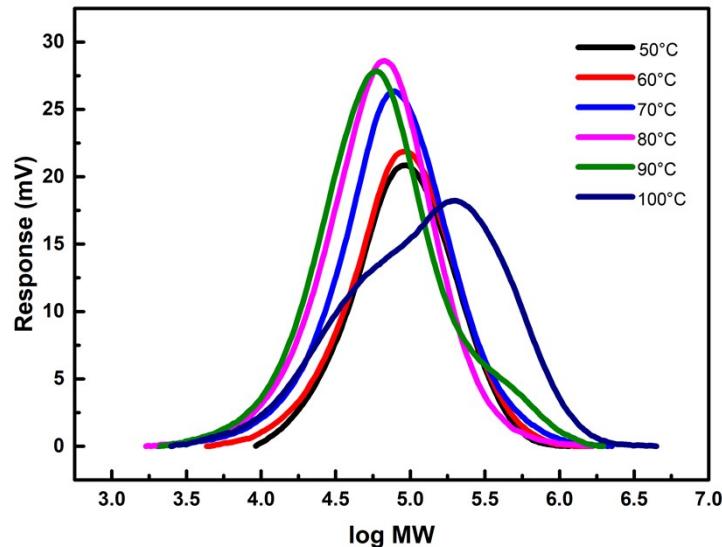
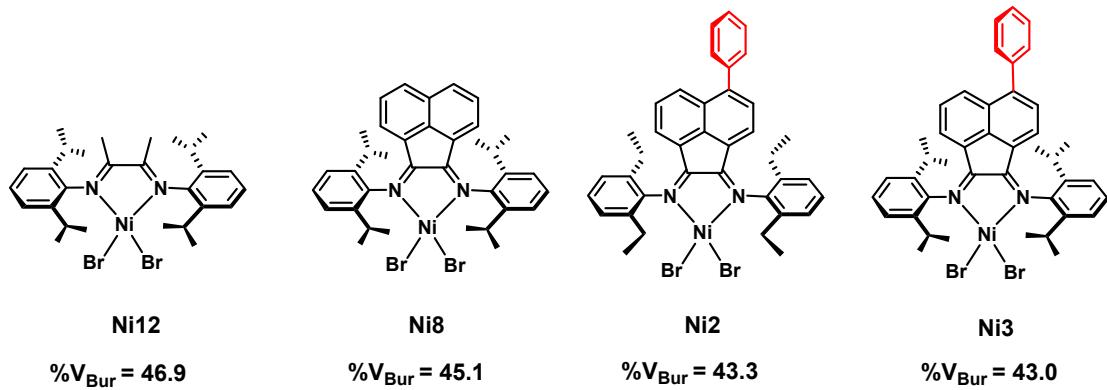


Figure S14. GPC traces of the polyethylenes produced by Ni2/EASC (Run 1-6, Table 3).



The stronger the skeleton rigidity, the lower the buried volume value

Rotation suppression of N-aryl group

Figure S15. The structures of **Ni9**, **Ni8**, **Ni2** and **Ni3** and their value of the buried volume ($\%V_{Bur}$).

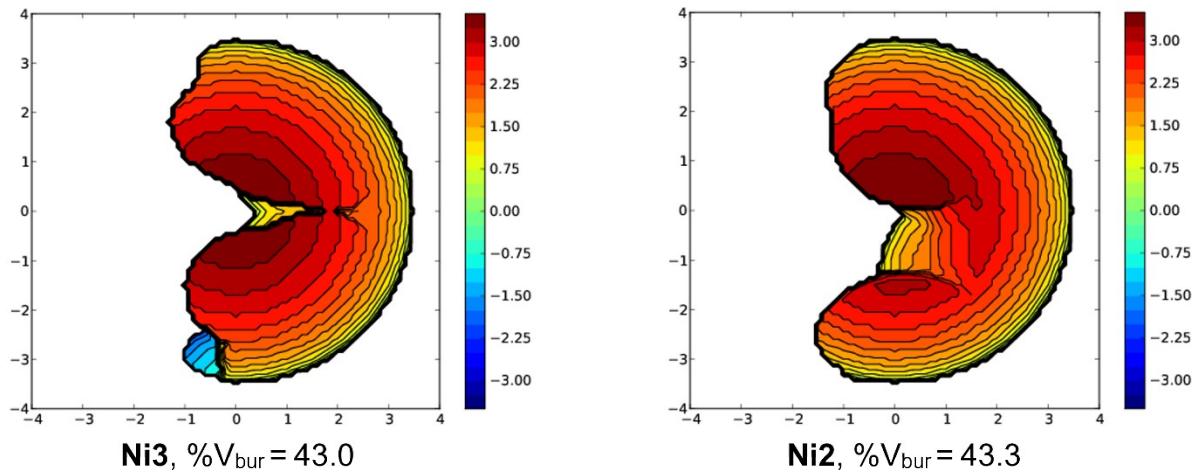


Figure S16. Topographical stereograms of **Ni3** and **Ni2**.

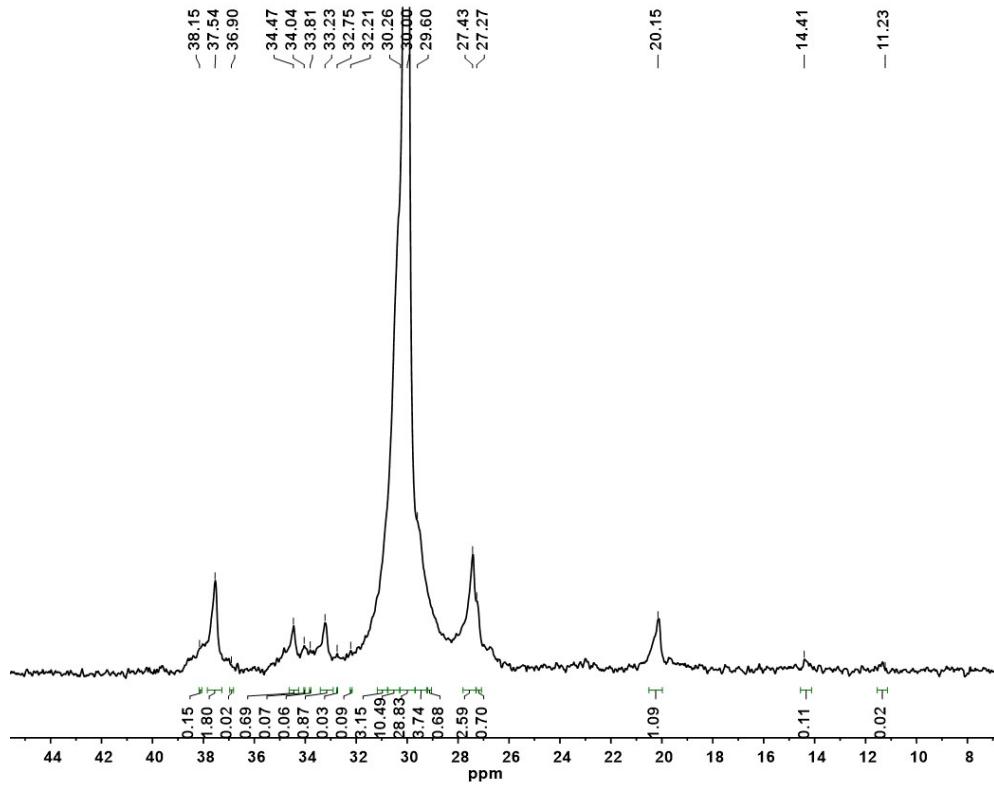


Figure S17. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}2-60-\text{M}1}$ (recorded in 1,1,2,2-tetrachloroethane- d_2 at 100 °C).

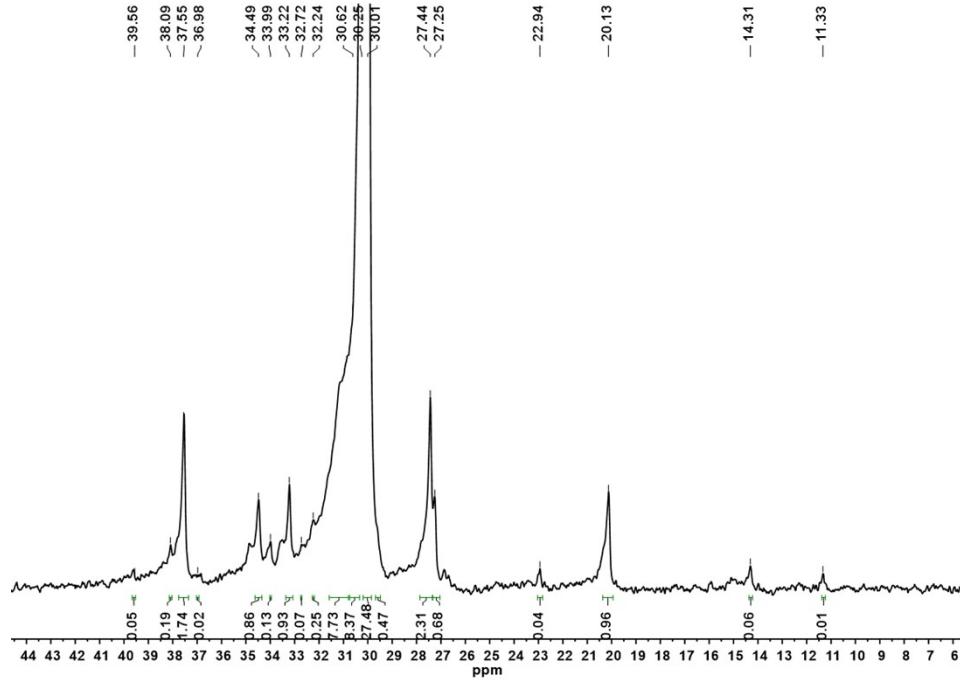


Figure S18. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}2-80-\text{M}1}$ (recorded in 1,1,2,2-tetrachloroethane- d_2 at 100 °C).

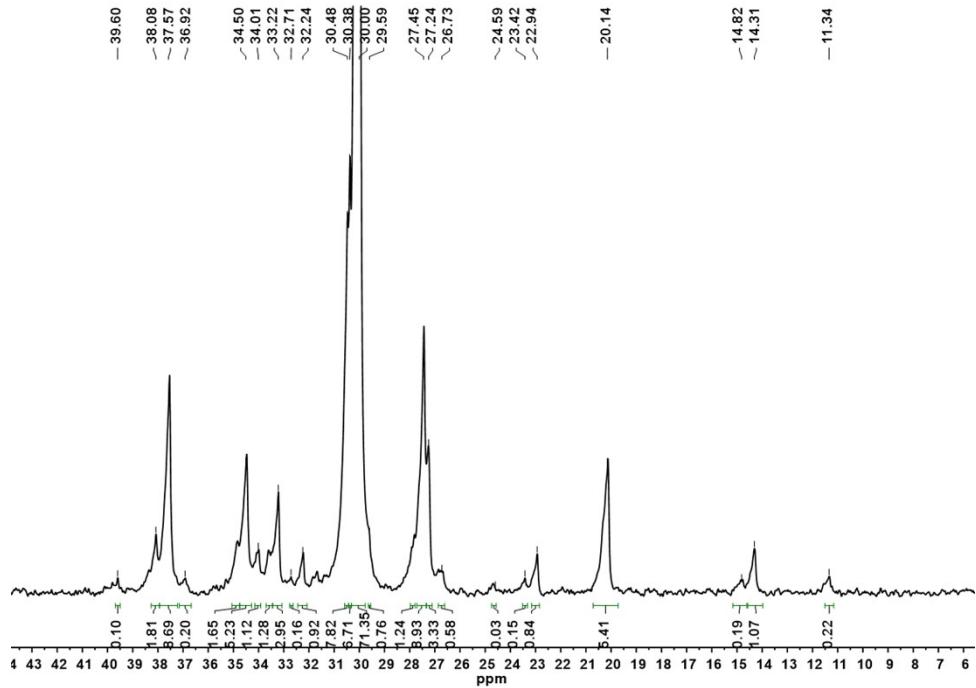


Figure S19. ¹³C NMR (125 MPa) spectrum of PE_{Ni2-60-E5} (recorded in 1,1,2,2-tetrachloroethane-*d*₂ at 100 °C).

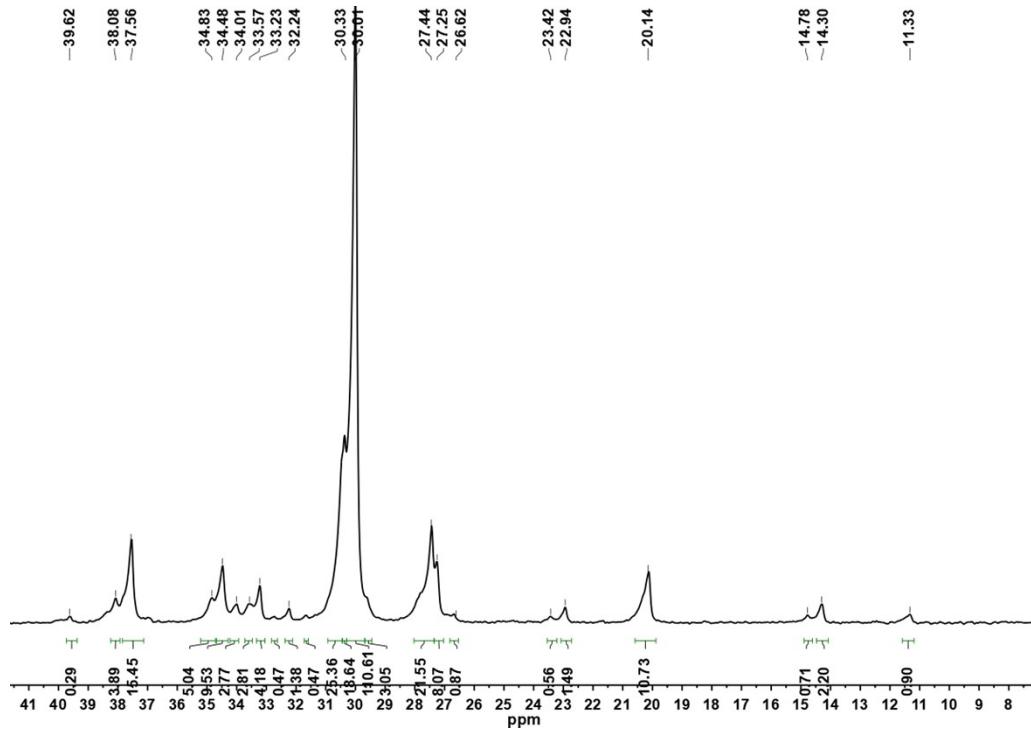


Figure S20. ¹³C NMR (125 MPa) spectrum of PE_{Ni2-80-E5} (recorded in 1,1,2,2-tetrachloroethane-*d*₂ at 100 °C).

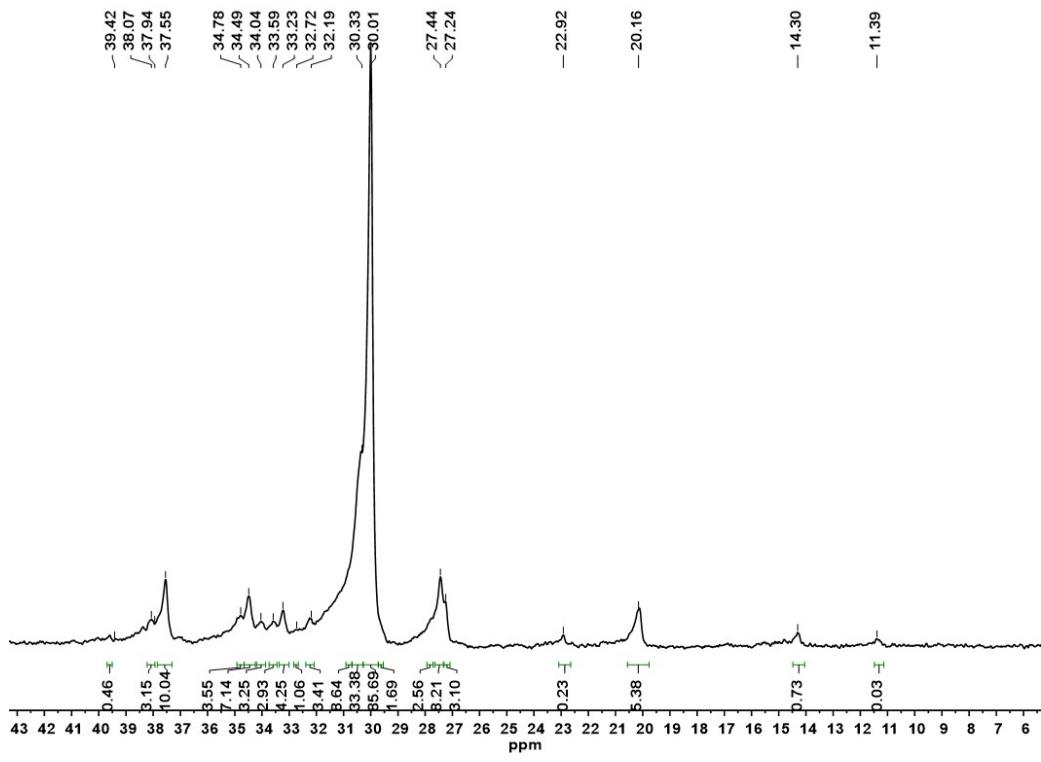


Figure S21. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}2-100-\text{E}5}$ (recorded in 1,1,2,2-tetrachloroethane-

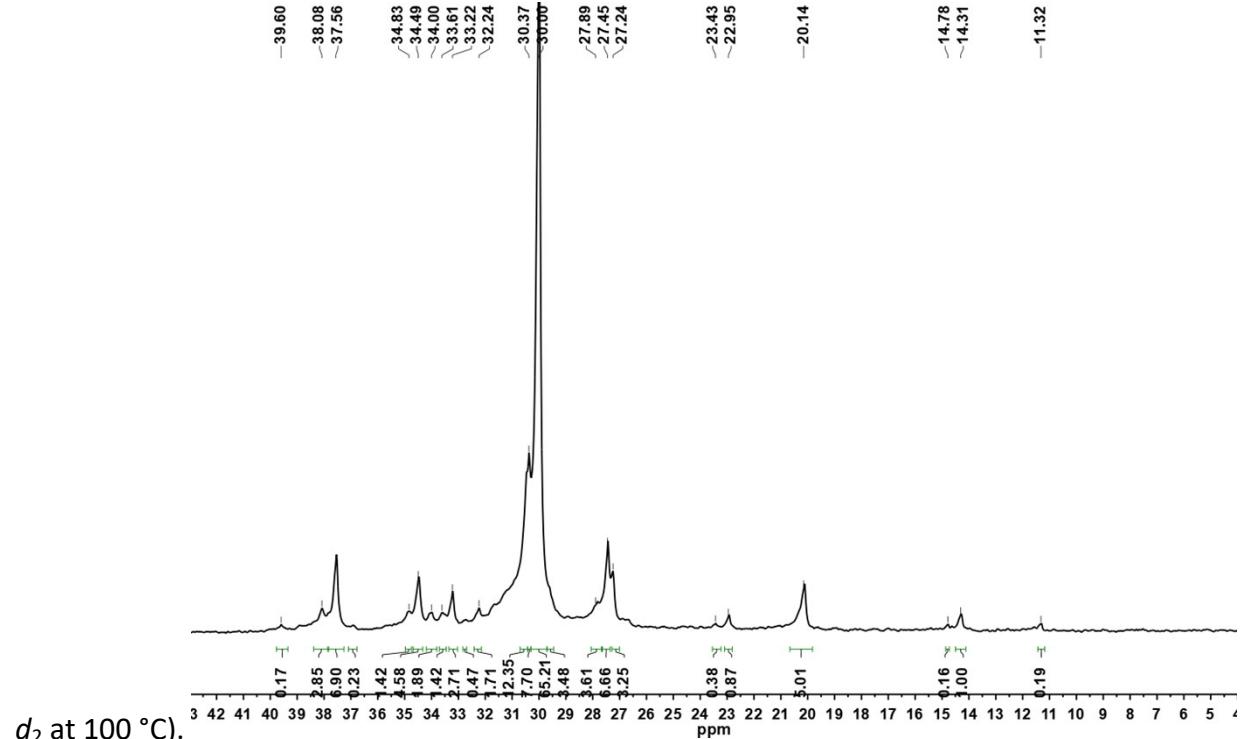


Figure S22. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}2-60-\text{E}2}$ (recorded in 1,1,2,2-tetrachloroethane- d_2 at 100 °C).

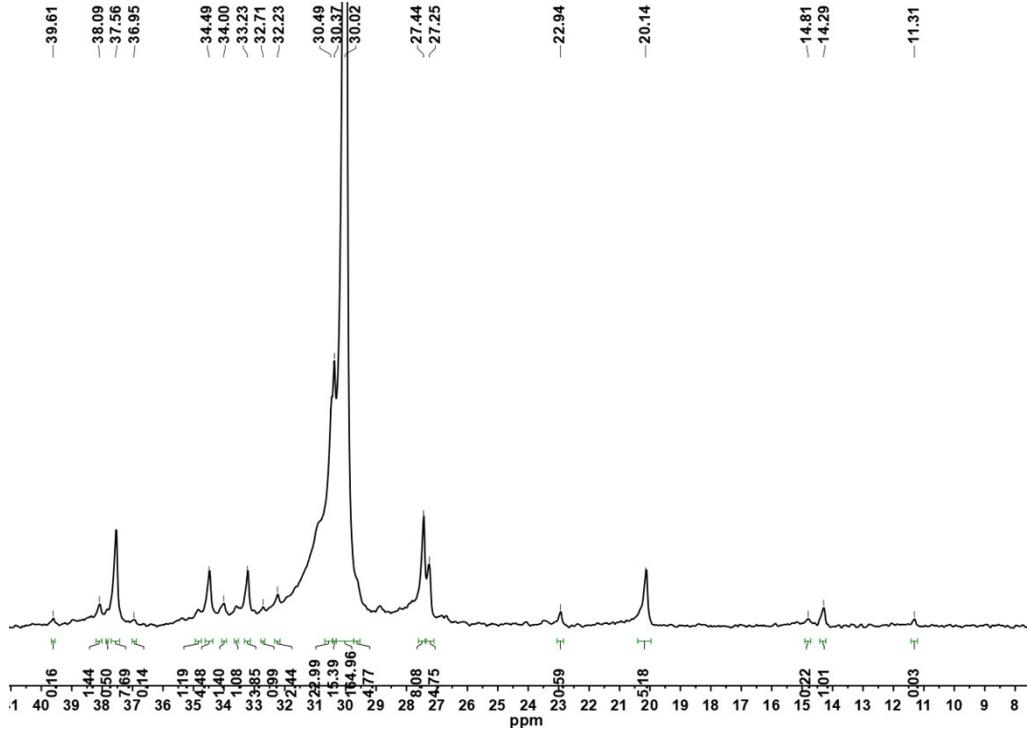


Figure S23. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}1\text{-}60\text{-E}2}$ (recorded in 1,1,2,2-tetrachloroethane- d_2 at 100 °C).

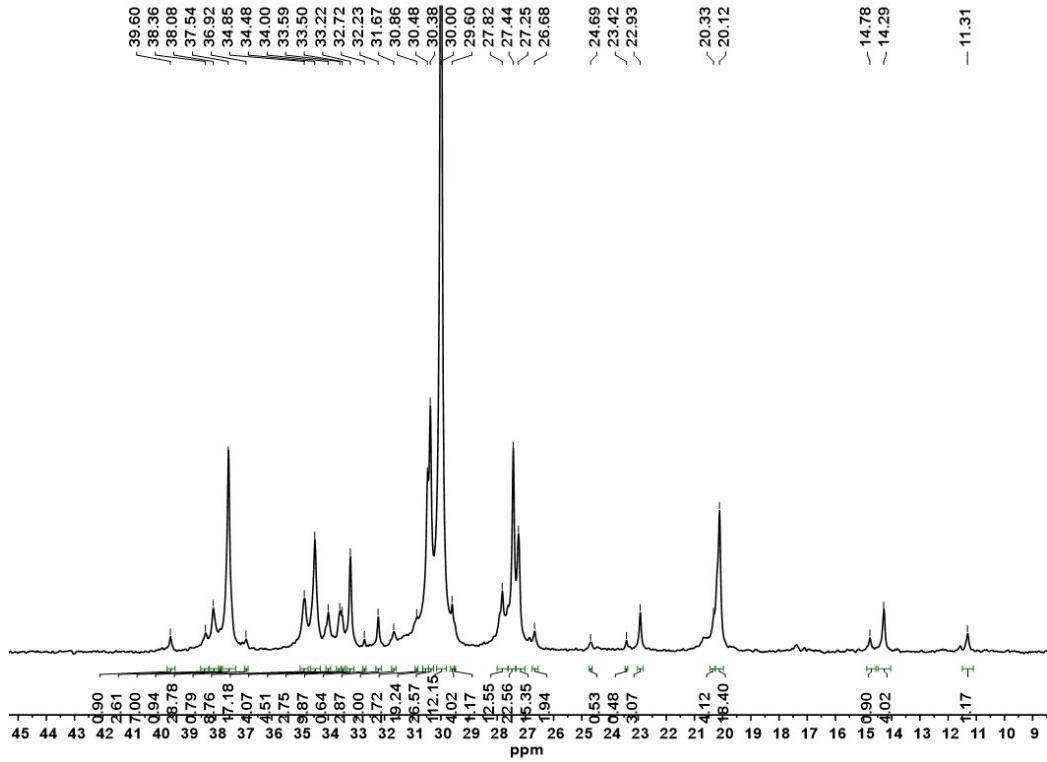


Figure S24. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}3\text{-}60\text{-E}2}$ (recorded in 1,1,2,2-tetrachloroethane- d_2 at 100 °C).

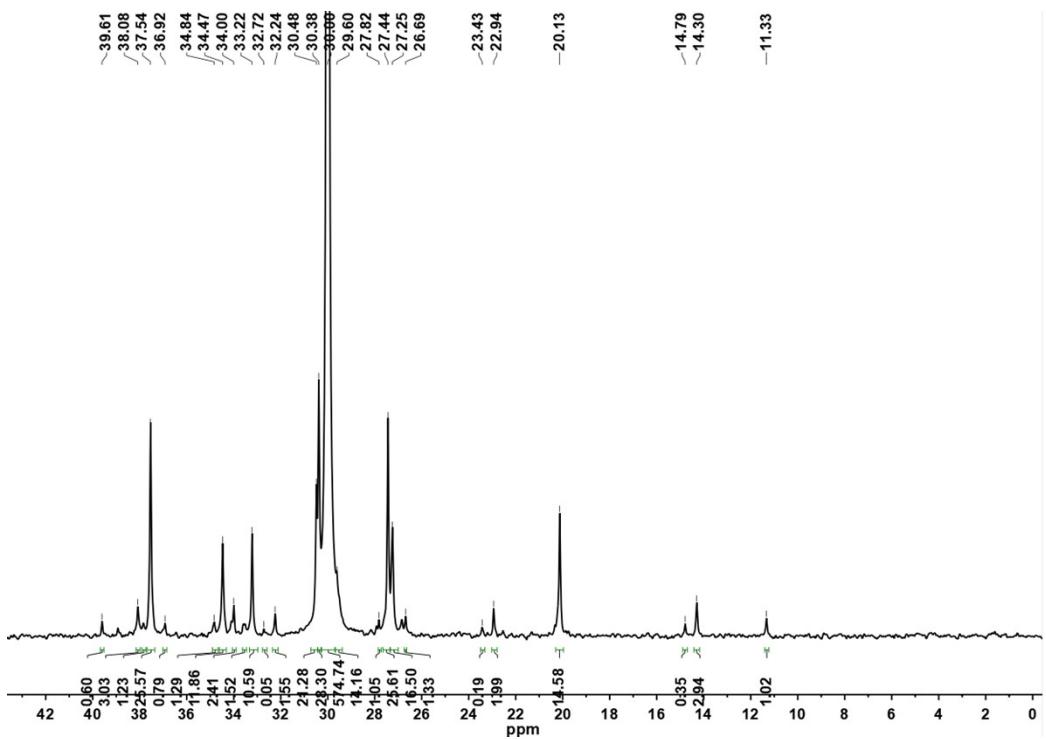


Figure S25. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}4-60-\text{E}2}$ (recorded in 1,1,2,2-tetrachloroethane- d_2 at 100 °C).

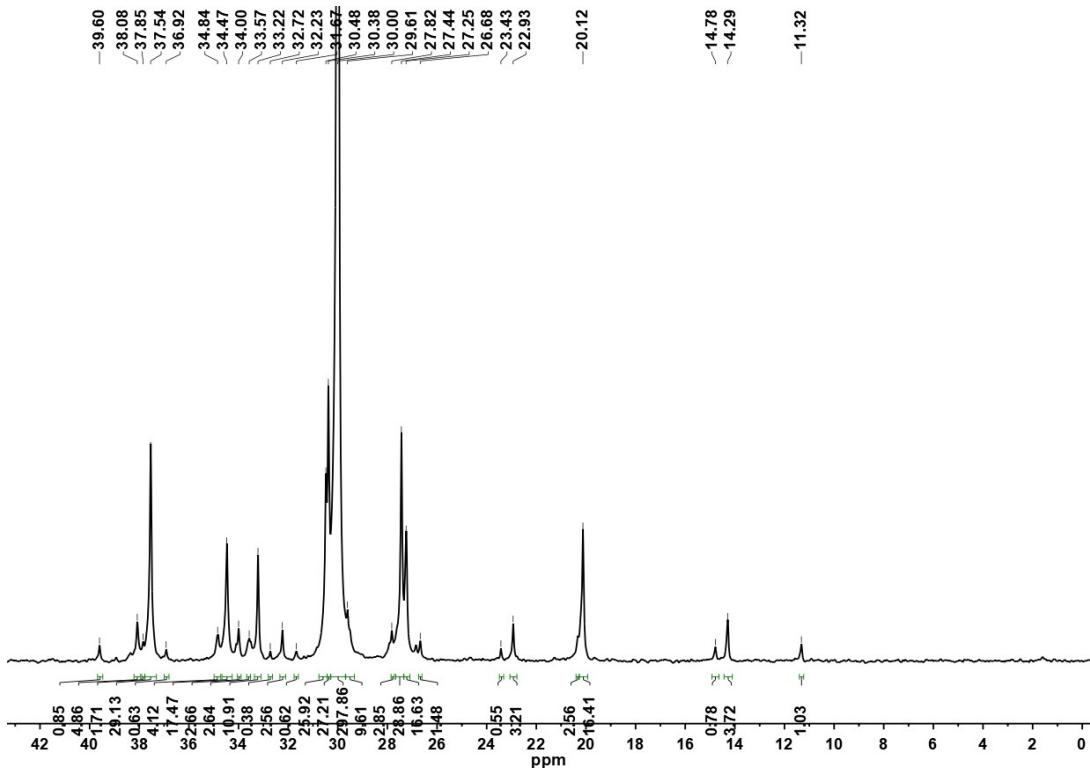


Figure S26. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}5-60-\text{E}2}$ (recorded in 1,1,2,2-tetrachloroethane- d_2 at 100 °C).

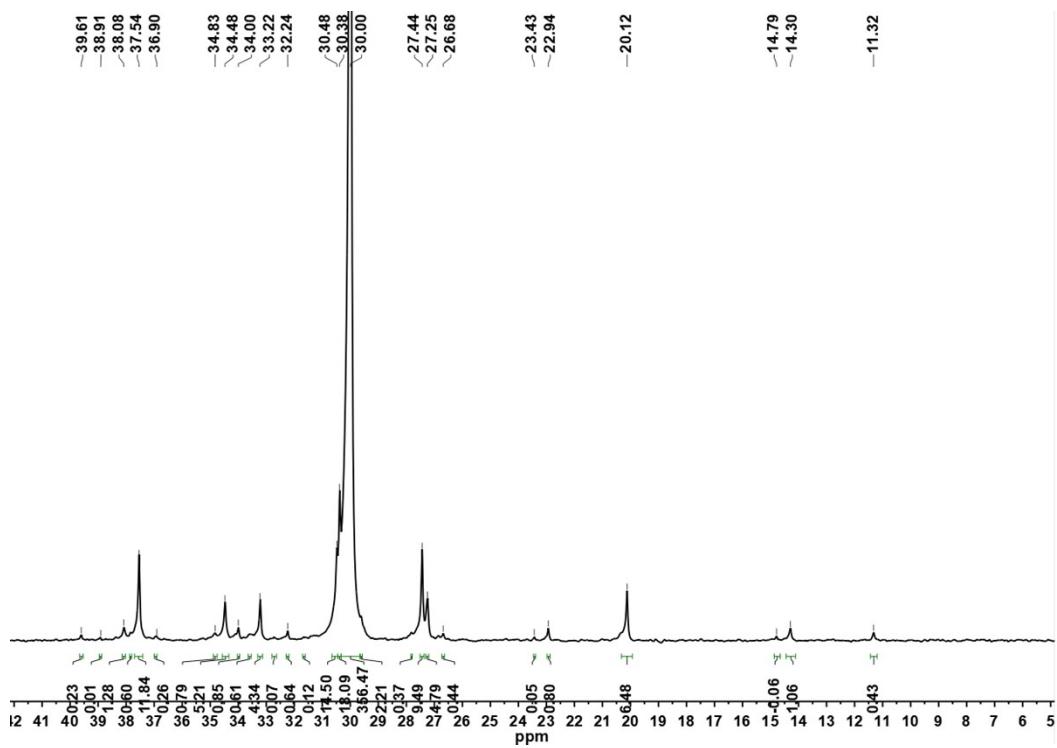


Figure S27. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}6-60-\text{E}2}$ (recorded in 1,1,2,2-tetrachloroethane- d_2 at 100 °C).

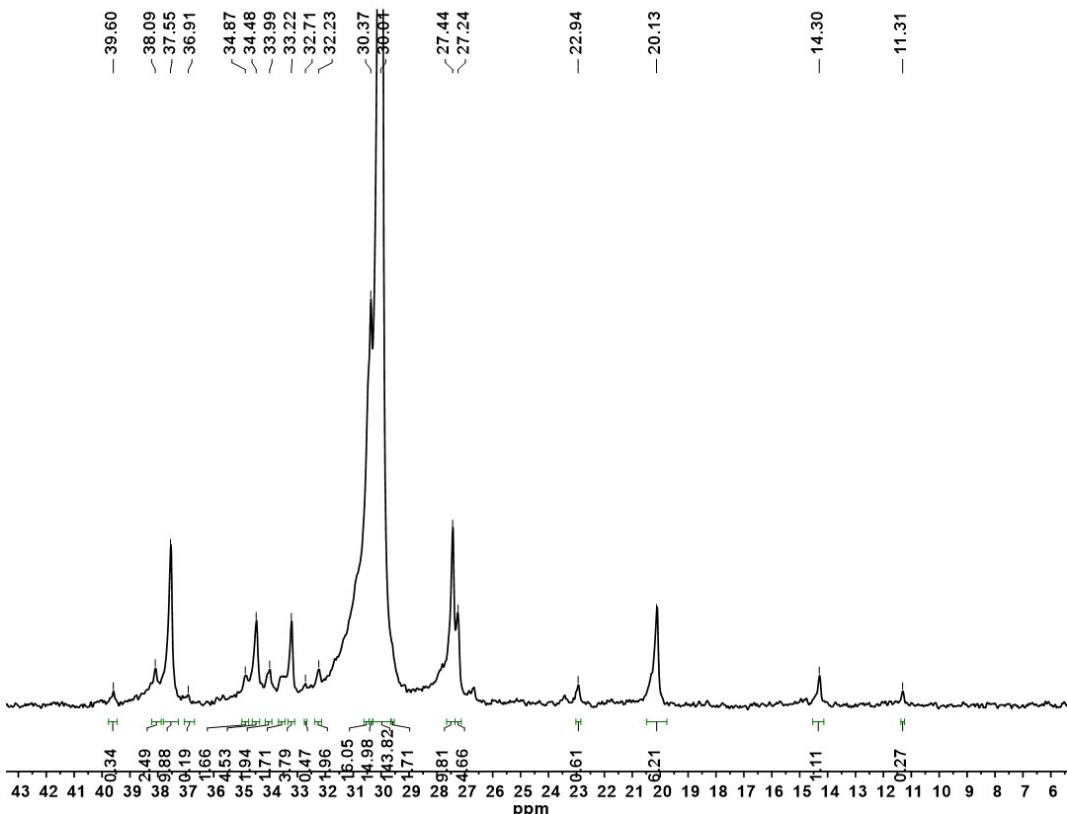


Figure S28. ^{13}C NMR (125 MPa) spectrum of $\text{PE}_{\text{Ni}7-60-\text{E}2}$ (recorded in 1,1,2,2-tetrachloroethane- d_2 at 100 °C).

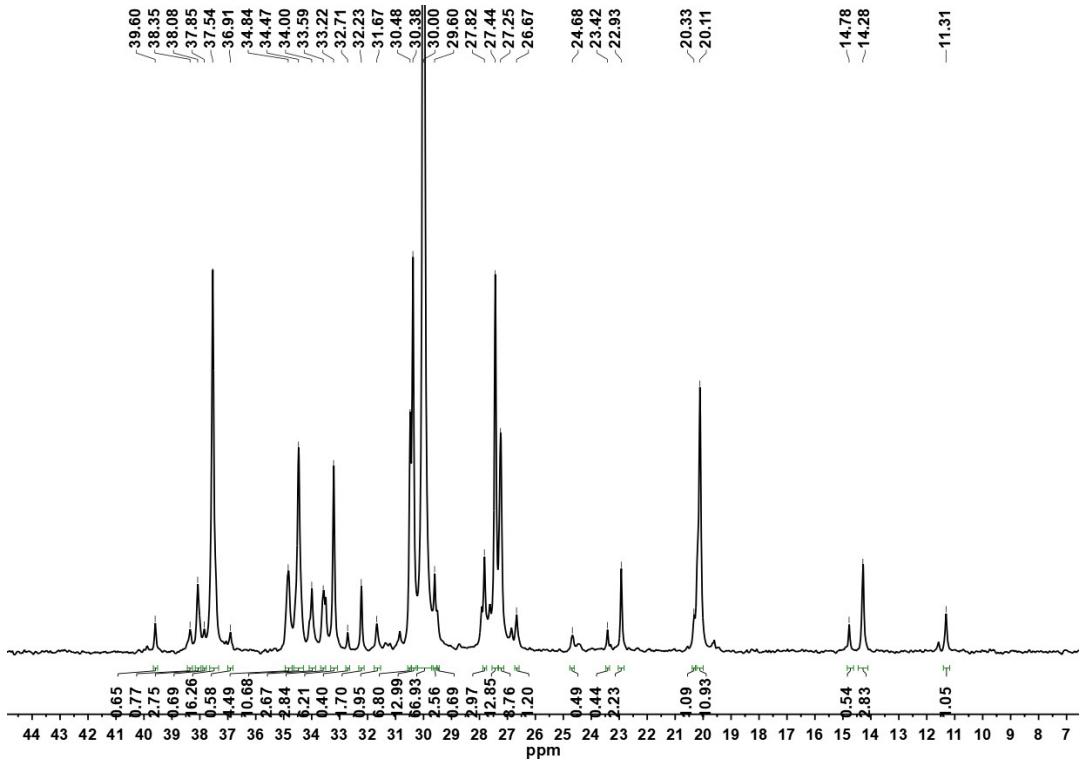


Figure S29. ¹³C NMR (125 MPa) spectrum of PE_{Ni8-60-E2} (recorded in 1,1,2,2-tetrachloroethane-*d*₂ at 100 °C).

Table S1. Crystal data and structure refinement of **Ni2** and **Ni3**.

	Ni2	Ni3
CCDC number	2331173	2331174
Empirical formula	C ₃₈ H ₃₆ Br ₂ N ₂ Ni	C ₈₄ H ₁₈₈ Br ₄ N ₄ Ni ₂
Formula weight	739.22	1590.64
Temperature/K	169.96(13)	169.98(11)
Crystal system	monoclinic	triclinic
Space group	P2 ₁ /n	P-1
a/Å	12.9593(8)	10.1206(4)
b/Å	15.7707(6)	11.7415(6)
c/Å	17.4731(10)	18.2647(5)
α/°	90	100.897(3)
β/°	110.829(7)	90.722(3)
γ/°	90	104.200(4)
Volume/Å ³	3337.7(3)	2062.19(15)
Z	4	1
ρ _{calc} g/cm ³	1.471	1.281
μ/mm ⁻¹	3.852	3.153
F(000)	1504.0	816.0
Crystal size/mm ³	0.1 × 0.06 × 0.05	0.15 × 0.08 × 0.05
Radiation	Cu Kα (λ = 1.54184)	Cu Kα (λ = 1.54184)
2θ range for data collection/°	7.794 to 152.49	7.924 to 151.494
Index ranges	-16 ≤ h ≤ 16, -18 ≤ k ≤ 19, -20 ≤ l ≤ 20	-12 ≤ h ≤ 12, -14 ≤ k ≤ 13, -22 ≤ l ≤ 22
Reflections collected	23084	24318
Independent reflections	6489 [R _{int} = 0.0513, R _{sigma} = 0.0368]	8177 [R _{int} = 0.0618, R _{sigma} = 0.0563]
Data/restraints/parameters	6489/751/578	8177/0/432
Goodness-of-fit on F ²	1.094	1.033
Final R indexes [I>=2σ (I)]	R ₁ = 0.1258, wR ₂ = 0.2328	R ₁ = 0.0723, wR ₂ = 0.1817
Final R indexes [all data]	R ₁ = 0.1349, wR ₂ = 0.2363	R ₁ = 0.0866, wR ₂ = 0.1975
Largest diff. peak/hole / e Å ⁻³	1.29/-0.90	1.89/-0.96

Table S2. Peak assignments of ^{13}C NMR spectrum of the polyethylenes ^a

Peak	Attribution	Peak	Attribution	Peak	Attribution
1	1B_2	12	βB_1 1,4- βB_1 $1,5\text{-}\beta\text{B}_1$ 1,6- βB_1 4 B_5	23	1,4-br B_1
2	1B_4 1B_5 1B_n	13	1,6- $\beta'\text{B}_1$	24	αB_2
3	1B_3	14	3 B_4	25	4 B_4
4	1B_1 1,5- B_1 1,6- B_1	15	4 B_n	26	αB_{3-5} 5 B_5 n B_n αB_n 1,4- αB_n 1,4-n B_n
5	1,4-1 B_1	16	δB_{1-n}	27	1,4- $\alpha'\text{B}_1$
6	2 B_3	17	γB_1 1,4- γB_1 $1,5\text{-}\gamma\text{B}_1$ 1,6- γB_1	28	3 B_3
7	2 B_5 2 B_n 1,4-2 B_n	18	γB_{2-5} 1,4- γB_n 1,4-(n-2) B_n γB_n	29	αB_1 1,4- αB_1 1,5- αB_1 1,6- αB_1 1,6-a' B_1
8	2 B_{4-8}	19	1,4- $\alpha'\text{B}_n$	30	br B_3 1,5- $\alpha'\text{B}_1$
9	1,5- $\beta'\text{B}_1$	20	3 B_n 1,4-3 B_n	31	br $\text{B}_{4,5,n}$
10	2 B_2	21	3 B_5	32	1,4-br B_n
11	βB_{2-5} βB_n (n-1) B_n 1,4- βB_n 1,4-(n-1) B_n	22	br B_1 1,5-br B_1 1,6-br B_1	33	br B_2

^a ^{13}C NMR spectrum of the polyethylenes (**PE_{Ni3-60-E2}**) was shown in Figure 12 in manuscript.

Table S3. Branching analysis of selected samples of polyethylenes.^a

Samples	Branches /1000 C's	Methyl branches/%	Ethyl branches/ %	Propyl branches/%	Butyl branches/ %	Amyl branches/ %	Long branches/ %	Details
PE _{Ni2-60-M1}	81	93.30	0.36	0.72	1.27	1.09	3.26	Run 2, Table 2
PE _{Ni2-80-M1}	83	84.36	0.62	0.74	2.41	2.60	9.27	Run 4, Table 2
PE _{Ni2-60-E5}	122	81.22	1.99	1.75	5.67	1.40	8.07	Run 2, Table 3
PE _{Ni2-80-E5}	167	83.12	1.41	3.10	6.06	1.71	4.60	Run 4, Table 3
PE _{Ni2-100-E5}	119	69.76	0.09	3.05	9.22	3.60	14.27	Run 6, Table 3
PE _{Ni1-60-E2}	65	64.58	0.44	1.15	5.73	8.11	20.00	Run 8, Table 3
PE _{Ni2-60-E2}	123	65.31	0.55	2.22	10.93	4.53	16.47	Run 12, Table 3
PE _{Ni3-60-E2}	185	76.22	4.83	2.46	7.10	2.00	7.39	Run 13, Table 3
PE _{Ni4-60-E2}	48	85.29	2.55	2.60	4.30	0.16	5.10	Run 14, Table 3
PE _{Ni5-60-E2}	97	83.67	2.19	1.77	4.52	1.07	6.78	Run 15, Table 3
PE _{Ni6-60-E2}	30	85.68	1.98	2.31	3.99	0.62	5.41	Run 16, Table 3
PE _{Ni7-60-E2}	74	82.47	0.32	0.82	5.81	1.72	8.85	Run 17, Table 3
PE _{Ni8-60-E2}	180	74.89	3.37	3.15	8.46	2.18	7.95	Run 19, Table 3

^a Data determined from the ¹³C NMR spectrum.

Table S4. Stress-strain properties of the selected polyethylene samples.

Samples	Maximum load /N	Modulus of elasticity /%	Rupture Stress /MPa	Rupture Strain /%
PE_{Ni2-60-M1}	99.68566	31.16010	20.86127	989.17561
PE_{Ni2-80-M1}	121.83333	25.94073	26.47687	1407.09152
PE_{Ni1-60-M3}	35.26493	231.10084	5.11201	441.05134
PE_{Ni3-60-M3}	34.90348	3.92895	7.15527	945.84150
PE_{Ni4-60-M3}	63.21949	327.03846	10.63216	1357.50637
PE_{Ni5-60-M3}	98.77724	25.40926	22.83357	1008.34370
PE_{Ni2-60-E5}	55.29499	14.95573	12.10514	1189.17704
PE_{Ni2-80-E5}	43.55141	9.05498	10.36686	2019.59190