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Electronic supporting information for:

Polyglyoxylamide hydrogels for the traceless stimulus-mediated release of covalently-immobilized drugs

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Synthesis of TEG-N₃-30-PGAm. This polymer was synthesized by the same procedure as TEG-N₃-40-PGAm. It was prepared from: PEtG (200 mg, 2.0 mmol of pendent ester, 1.0 equiv.), 2-azidoethylamine (95 mg 1.1 mmol, 0.56 equiv.), dry 1,4-dioxane (5 mL) and TEG-amine (1120 mg, 5.9 mmol, 5.0 equiv.). A tacky solid was obtained. Yield: 92%. ¹H NMR (CDCl₃, 400 MHz): δ 8.66 – 7.73 (br s, 0.9H), 5.71 (br s, 1.0H), 3.76 – 3.50 (m, 9.4H), 3.43 (s, 1.4H), 3.37 (s, 2.4H). ¹³C {¹H} NMR (CDCl₃, 400 MHz): δ 167.2, 96.6, 71.9, 70.5, 69.1, 58.9, 39.3. FT-IR: 3516-3148, 2870, 2099, 1674, 1545 cm⁻¹. SEC (DMF, PMMA): $M_n = 56$ kg/mol, $M_w = 90$ kg/mol, D = 1.61.

Synthesis of *o*NB-TEG-N₃-20-PGAm. The synthetic procedure was same as *o*NB-TEG-N₃-30-PGAm. The copolymer was prepared from: TEG-N₃-30-PGAm (100 mg, 0.15 mmol of azide, 1.0 equiv.), alkyne-functionalized *o*-nitrobenzyl alcohol (11.9 mg, 0.051 mmol, 0.33 equiv.), sodium ascorbate (10.1 mg, 0.051 mmol, 0.33 equiv.) and copper sulfate (8.14 mg, 0.051 mmol, 0.33 equiv.) A yellow tacky solid was obtained. Yield: 89%. ¹H NMR (DMSO-*d*₆, 400 MHz): δ 8.60 – 7.79 (m, 1.5H), 5.56 (br s, 1.0H), 4.89 (br s, 0.2H), 4.47 (br s, 0.4H), 3.82-3.38 (m, 8.7H), 3.24 (s, 2.2H). FT-IR: 3533-3115, 2902, 2101, 1624, 1524 cm⁻¹.

Synthesis of Phe-*o*NB-TEG-N₃-20-PGAm. The conjugation of Phe methyl ester to the 20% azide-functionalized copolymer followed the same procedure as for the preparation of Phe-*o*NB-TEG-N₃-30-PGAm except that *o*NB-TEG-N₃-20-PGAm (115 mg, 0.052 mmol of *o*NB linker, 1.0 equiv.), CDI (84 mg, 0.52 mmol, 10 equiv.) and Phe methyl ester (26 mg, 0.14 mmol, 3.0 equiv.) were used. Yield: 49% ¹H NMR (DMSO-*d*₆, 400 MHz): δ 8.84 – 7.87 (m, 1.7H), 7.46 –7.02 (m, 0.5H), 5.56 (br s, 1H), 4.88 (br s, 0.2H), 4.63 – 4.25 (m, 0.4H), 3.87 – 2.87 (m, 15H).

Table S1. The formulations of hydrogels containing 4-arm-PEG-alkyne and phenylalanine conjugated PGAm copolymer in 300 μ L DMF/water (v/v = 4/1).

15% w/v Hydrogels							
	4-arm-PEG-alkyne	Phe- <i>o</i> NB-TEG-N ₃ -PGAm	CuSO ₄	Sodium ascorbate			
Mass (mg)							
20% azide	23	22	2.7	3.3			
30% azide	27	17	3.5	4.3			



Figure S1. ¹H NMR spectrum of 30 kg/mol PEtG (CDCl₃, 400 MHz). A DP_n of \sim 335 was determined based on the relative integrals of the methyl CH₃ peak at 0.9 ppm compared to the backbone methine peak at 5.6 ppm.



Figure S2. Conversion of 40% of the pendent esters of **PEtG** to pendent 2-azidoethylamides in the synthesis of **TEG-N₃-40-PGAm**. The reaction was monitored by ¹H NMR spectroscopy. The percent functionalization was determined by comparing the integral of the -CH (methine) peak from the polymer backbone at 5.66 ppm with that of the -CH₂ peak from the pendent ester groups at 4.25 ppm. The reaction was stopped when the integral of -CH₂ peak at 4.25 ppm was ~1.20, indicating ~40% azide functionalization was achieved.



Figure S3. Conversion of 30% of the pendent esters of PEtG to pendent 2-azidoethylamides in the synthesis of TEG-N₃-30-PGAm. The reaction was monitored by ¹H NMR spectroscopy. The percent functionalization was determined by comparing the integral of the -CH (methine) peak from the polymer backbone at 5.66 ppm with that of the -CH₂ peak from the pendent ester groups at 4.25 ppm. The reaction was stopped when the integral of -CH₂ peak at 4.25 ppm was ~1.40, indicating ~30% azide functionalization was achieved.



Figure S4. ¹H NMR spectrum of TEG-N₃-40-PGAm (CDCl₃, 400 MHz).



Figure S5. ¹H NMR spectrum of TEG-N₃-30-PGAm (CDCl₃, 400 MHz).



Figure S6. ¹H NMR spectrum of *o*NB-TEG-N₃-30-PGAm (DMSO-*d*₆, 400 MHz).



Figure S7. ¹H NMR spectrum of *o*NB-TEG-N₃-20-PGAm (DMSO-*d*₆, 400 MHz).



Figure S8. ¹H NMR spectrum of Phe-*o*NB-TEG-N₃-30-PGAm (DMSO-*d*₆, 400 MHz).



Figure S9. ¹H NMR spectrum of Phe-*o*NB-TEG-N₃-20-PGAm (DMSO-*d*₆, 400 MHz).



Figure S10. ${}^{13}C{}^{1}H$ NMR spectrum of PEtG (CDCl₃, 100 MHz). Note that end-cap peaks are not observed in the spectrum due to the high molar mass.



Figure S11. ${}^{13}C{}^{1}H$ NMR spectrum of TEG-N₃-40-PGAm (CDCl₃, 100 MHz).



Figure S12. ${}^{13}C{}^{1}H$ NMR spectrum of TEG-N₃-30-PGAm (CDCl₃, 100 MHz).



Figure S13. FT-IR spectrum of PEtG.



Figure S14. FT-IR spectrum of TEG-N₃-40-PGAm.



Figure S15. FT-IR spectrum of TEG-N₃-30-PGAm.



Figure S16. FT-IR spectrum of *o*NB-TEG-N₃-30-PGAm.



Figure S17. FT-IR spectrum of *o*NB-TEG-N₃-20-PGAm.



Figure S18. Overlay of the FT-IR spectra of 4-arm-PEG-alkyne, Phe-*o*NB-TEG-N₃-30-PGAm and the corresponding hydrogel.



Figure S19. Photos of the 15% w/v **Phe**-*o***NB-TEG-N₃-30-PGAm** hydrogel a) immediately after gelation and b) after removal of copper and the DMF.



Figure S20. Cyclic loading (compression) and unloading (relaxation) stress-strain curves of a representative **Phe**-*o***NB-TEG-N₃-30-PGAm** 15% w/v hydrogel, measured in PBS.



Figure S21. The calibration curve for the fluorescamine assay for Phe methyl ester showing the intensity of fluorescence at 470 nm versus Phe methyl ester concentration.